

GEORGIA INSTITUTE OF TECHNOLOGY
OFFICE OF CONTRACT ADMINISTRATION
SPONSORED PROJECT INITIATION

Date: 6/29/79

Project Title: *Effect of Localized Corrosion on Dental Amalgams*

Green Card - NIH

Project No: *E-19-B04 (Continuation of E-19-B03)*

Project Director: *Dr. M. Marek*

Sponsor: *DHEW/PHS/National Institute of Dental Research*

Agreement Period: From 6/1/79 Until 5/31/81

Type Agreement: *Grant No. 2R01 DEO 3601-07*

Amount: \$41,844 (PHS, E-19-B04)

4,993 (GIT, E-19-344)

\$46,837 TOTAL

Reports Required: *Interim Progress Report; Terminal Progress Report*

Sponsor Contact Person (s):

Technical Matters

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Contractual Matters

(thru OCA)

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Dept. of Health, Education & Welfare
Bethesda, MD 20014*

Defense Priority Rating: *n/a*

Assigned to: Chemical Engineering (School/Laboratory)

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Other _____

GEORGIA INSTITUTE OF TECHNOLOGY
OFFICE OF CONTRACT ADMINISTRATION
SPONSORED PROJECT TERMINATION

Date: June 10, 1980

Project Title: Effect of Localized Corrosion on Dental Amalgams

Project No: E-19-B04

Project Director: Dr. M. Marek

Sponsor: DHEW/PHS/National Institute of Dental Research

Effective Termination Date: 6/30/80

Clearance of Accounting Charges: 6/30/80

Grant/Contract Closeout Actions Remaining:

- ☐ Final Invoice and Closing Documents
- ☒ Final Fiscal Report (Interim)
- ☒ Final Report of Inventions (Interim)
- ☐ Govt. Property Inventory & Related Certificate
- ☐ Classified Material Certificate
- ☐ Other _____

CONTINUED BY E-19-B05

Assigned to: Chemical Engineering (School/Laboratory)

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Other OCA Research Property Coord.

SECTION IV

| | | | |
|---|--|-------------------------------|----------|
| APPLICANT: REPEAT GRANT NUMBER SHOWN ON PAGE 1 → | | GRANT NUMBER | |
| SECTION IV—SUMMARY PROGRESS REPORT | | DE 03601-08 | |
| PRINCIPAL INVESTIGATOR OR PROGRAM DIRECTOR (Last, First, Initial) | | PERIOD COVERED BY THIS REPORT | |
| Marek, Miroslav | | FROM | THROUGH |
| NAME OF ORGANIZATION | | 07/01/79 | 06/30/80 |
| Georgia Institute of Technology | | | |
| TITLE (Repeat title shown in Item 1 on first page) | | | |
| EFFECT OF CORROSION ON DENTAL AMALGAMS | | | |

1. List publications: (a) published and not previously reported; (b) in press. Provide five reprints if not previously submitted.
2. List all additions and deletions in professional personnel and any changes in effort.
3. Progress Report. (See Instructions)

Publications

M. Marek: "Corrosion Test for Dental Amalgam,"
Journal of Dental Research, 59, 63-69 (1980)

Changes

Effort: No Change

Personnel: Dr. Steven Reese, postdoctoral trainee, participated in one part of the program (no salary compensation).

Progress ReportOBJECTIVESA. Overall Objectives

To improve the understanding of the role of corrosion in the deterioration of dental amalgam restorations by investigating the mechanism of corrosion, the effects of corrosion on the properties, and by developing suitable testing methods.

B. Goals For The Current Year

- a. To investigate the mechanism of corrosion of high copper dental amalgam.
- b. To investigate the relationship between creep and corrosion of dental amalgam.
- c. To initiate a study of the relationship between porosity, mercury content and corrosion of dental amalgam.
- d. To evaluate the corrosion test based on controlled potential coulometry.

2. STUDIES AND RESULTSa. A Study of the Mechanism of Corrosion of High Copper Dental Amalgam

Earlier studies have shown that in the corrosion affected zone copper is depleted, the Cu_6Sn_5 deteriorates, and some of the γ_1 phase changes into β_1 . In this study the reactions and the mass transport have been examined. In one part of the study the electrolyte (1% NaCl) in which specimens were actively corroding was analyzed periodically by atomic absorption spectrophotometry for metallic ions, and solid corrosion products were analyzed by electron microprobe. The results show that initially tin dissolves but solubility is quickly reached and the main product of the reaction are insoluble tin compounds, mainly tin oxychloride. Copper enters the solution later but the release rate soon exceeds that of copper. Some solid copper corrosion products (chloride) were also found. Since the amalgams did not contain the tin-rich γ_2 phase it is thought that the tin initially originates in the γ_1 phase; this destabilizes the phase and facilitates the transformation into β_1 . Copper is released when the Cu_6Sn_5 phase breaks down and is trans-

ported by diffusion to the surface where it dissolves. Further tests are planned to test this hypothesis.

The transport of copper through the matrix phases was investigated. Samples of γ_1 and β_1 phases were prepared, electroplated with copper, maintained at constant temperature, sectioned, and analyzed by electron microprobe. The results, reported to the 1980 Annual Session of the AADR (1) show that copper diffused about two times faster in the β_1 than in the γ_1 phase. The diffusion parameters were determined. Thus the $\gamma_1 - \beta_1$ transformation facilitates the transport and the corrosion process.

b. A Study of the Relationship Between Creep and Corrosion

Mahler and other investigators have shown that creep correlates well with the incidence of marginal breakdown of amalgam restorations. A recent joint study (2) has shown that both creep and corrosion depend similarly on some variables. In the first part of this study the effect of tin and zinc contents on the creep of the γ_1 and β_1 phases was examined. Specimens of both phases with systematically varied zinc and tin contents were prepared and tested for ADA creep. The results, presented to the 1980 Annual Session of the AADR (3), show that β_1 creeps much more slowly than γ_1 ; tin increases sharply creep of both phases; zinc decreases creep of the γ_1 phase but has little effect on β_1 ; some combinations of tin and zinc results in embrittlement of the matrix. The same specimens used in the creep study are now tested for the corrosion resistance.

The Principal Investigator also participated in a study of the creep of corroded dental amalgam conducted at the Medical College of Georgia. The corroded amalgams showed lower creep indicating that corrosion results in the loss of plasticity (4).

c. A Study of the Effect of Porosity on Corrosion of Dental Amalgam

The study was initiated in the current year. Specimens of a conventional amalgam were prepared and the nominal surface was determined from the dimensions. The actual total surface area was then determined by measuring the adsorption and desorption isotherms at low temperature using Krypton gas. The results show that the actual total surface area, which includes the surface area of the open pores, was about twenty times higher than the nominal area. This demonstrated the very important role of porosity in the corrosion processes. The surface area determination technique will be used in the continuation of this study.

d. Evaluation of the Corrosion Test

The corrosion test based on constant potential coulometry (5) has been used in several phases of this program and the data have been collected for statistical analysis. A study was made of the basic processes involved in the test. Specimens were tested using different exposure times and the relationship between the anodic charge and the corrosion depth has been determined. Examination of the micro-structural changes is in progress.

e. A Study of the Galvanic Interactions Between Dental Amalgam and Other Restorative Materials

This study was conducted in addition to the planned program in view of the important galvanic effects observed in the previous part of the project. In one part of the study the interaction between metals which are not in contact was analyzed theoretically and examined experimentally. In the second part the interactions between metals in contact were measured using two dental amalgams (conventional and high copper) and dental casting alloys which included high-gold, low-gold, silver-palladium, and base-metal alloys. The results, presented to the 1980 Annual Session of the AADR (6), can be summarized as follows: 1. There is no significant interaction between dissimilar metals in the mouth which are not in contact; 2. There is a substantial concentration macrocell due to the difference

between saliva and the tissue fluids, which may have an important effect on the corrosion of the part of the restoration within the tooth cavity; 3. The measurement of open circuit potential differences between various dental materials is of little value in predicting the intensity of the galvanic interactions.

References:

1. S. Reese, R. F. Hochman, and M. Marek: "Low Temperature Diffusion of Copper in the γ_1 and β_1 Phase of Dental Amalgam". Annual Session of the AADR, March 20-23, 1980, Los Angeles, Paper No. 106.
2. M. Marek and D. B. Mahler: "The Corrosion Susceptibility of a High Copper Amalgam as a Function of the Mercury Content". Annual Session of the AADR and the 57th General Session of the IADR, March 29 - April 1, 1979, New Orleans, Paper No. 968.
3. D. F. Averette and M. Marek: "Creep of the Matrix Phases of Dental Amalgam as a Function of Composition". Annual Session of the AADR, March 20-23, 1980, Los Angeles, Paper No. 105.
4. M. B. Butts, T. Okabe, F. J. Mitchell, M. Marek, and C. W. Fairhurst: "Creep of a Corroded High Copper Amalgam", *ibid.*, Paper No. 282.
5. M. Marek: "Corrosion Test for Dental Amalgam". J.D.R., 59, 63-69 (1980).
6. M. Marek: "Galvanic Interactions Between Dental Amalgam and Other Restorative Materials". Annual Session of the AADR, March 20-23, 1980, Los Angeles, Paper No. 1033.

3. SIGNIFICANCE

The proposed research will contribute to the understanding of the basic processes of deterioration of dental amalgam restorations and thus help in the development of improved materials.

4. RESEARCH GOALS FOR THE COMING YEAR

- a. A Study of the Mechanism of Corrosion of High Copper Amalgam: To investigate the breakdown of the Cu_6Sn_5 phase by metallographic examination and microanalysis of specially prepared samples simulating the amalgam microstructure.
- b. A Study of the Relationship Between Mechanical Properties and Corrosion: To complete the determination of the effects of composition on the corrosion of the matrix phases (to complement the creep tests) and to determine the effect of simultaneous creep and corrosion.
- c. A Study of the Relationship Between Porosity and Corrosion: To determine the above relationship for one dental amalgam using the techniques of surface area determinations developed in the current year.
- d. Evaluation of the Electrolytes Used in the In vitro Corrosion Tests: To determine the data unavailable in the literature for the test media used to date (pH, buffering capacity, redox potential).

The undersigned agrees to accept responsibility for the scientific and technical conduct of the project and for provision of required progress reports if a grant is awarded as the result of this application.

4/21/80

Date

Principal Investigator