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Sponsor technical cont	act	Sponsor iss	uing office				
BRIAN ANDREEN GRANTS	PROGRAM COORD	DR JOHN I	SCHAEFER	PRESIDENT			
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School of Chemistry (404) 894-4002 (Tel.) (404) 894-7452 (Fax) **Georgia Institute of Technology** Atlanta, Georgia 30332 A Unit of the University System of Georgia

July 3, 1990

Dr. John W. Robson Grants Program Associate Research Corporation 6840 East Broadway Blvd. Tuscon, Arizona 85710-2815

Dear Dr. Robson:

Enclosed are two copies of the Terminal Research Report for my Research Corporation Grant numbered R-21. This report has been signed both by me and by a repesentative of the Georgia Institute of Technology.

Please let me know if you need any other information at this time.

Sincerely yours,

Herbert O. House Professor of Chemistry

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REPORT OF RESEARCH CORPORATION GRANT

(Submit original and one legible copy)

Please check one)

Interim Report

I Terminal Report

Reply to: 6840 E. Broadway

Tucson, Arizona 85710

NSTITUTION AND ADDRESS Georgia Institute of Technology Atlanta, GA 30332

RINCIPAL INVESTIGATOR Herbert 0. House

PHONE (404)894-4044

CADEMIC RANK AND DEPARTMENT Professor, School of Chemistry

HORT TITLE OF RESEARCH SUPPORTED BY GRANT

Derivatives of 1,8-Diphenylanthracene as Potential Catalysts or Reagents for Asymmetric Synthesis

STARTING DATE June 15, 1988

UMMARY OR PRINCIPAL FINDINGS AND THEIR SIGNIFICANCE (State succinctly in language understandable to one not necessarily xpert in this field. Include extent to which original goals have been realized and any changes to original plan made or contemplated.)

The initial goal of this program, the synthesis of various 1,8-diaryl-2,7-dimethylanthracenes, requires reasonable quantities of the precursor, 1,8-dichloro-2,7-dimethylanthracene. Much of the time during the past two years has been devoted to developing a good workable synthesis for this precursor. Of the various methods examined (detailed in my Interim Progress Report), the method we are now using consists of the following steps: (1) reaction of isoprene with benzoquinone followed by oxidation to form 6-methyl-1,4-naphthoquinone; (2) reaction of 6-methyl-1,4-naphthoquinone with isoprene followed by oxidation to form a mixture of comparable amounts of the isomeric 2,6- and 2,7-dimethylanthraquinones; (3) separation of the isomeric dimethylyanthraquinones by acetone extraction follow by fracional crystallization; (4) chlorination of the 2,7-isomer by reaction of the quinone with chlorine at about 100° C in sulfuric acid solution; (5) separation of the desired 2,7-dimethyl-1,8-dichloroanthraquinone by a combination of column chromatography and recrystallization; (6) reduction of the dichlorodimethyl quinone with zinc in aqueous ammonia followed by treatment with acid to dehydrate the crude alcohol product forming 1,8-dichloro-2,7-dimethylanthracene.

In the course of this study, we found that earlier literature reports describing the preparations of 1,8-dichloro-2,7-dimethylanthraquinone and of 1,5-dichloro-2,6-dimethylanthraquinones had not given pure compounds. Thus, we have obtained pure samples of these quinones for the first time. By use of the reduction procedure noted above, each of these quinones was reduced to the corresponding dichloro-dimethyl-anthracene. The next phase of this work will involve the nickel-catalyzed coupling of the above 1,8-dichloro-2,7-dimethylanthracene precursor with phenylmagnesium bromide and with xylylmagnesium bromide. The latter coupled product will be used to prepare the chiral molecule that is the objective of this research.

While the above methodology was being developed, we also devoted some time to the development of a synthesis for 4-bromo-2-keto-bicyclo-[2.2.2]-octane, a possible precursor for the unknown and presumably very highly strained enone, 2-keto-delta-3,4-bicyclo-[2.2.2]-octene, that we expect to have unusual physical and chemical properties. The synthesis involved the following steps: (1) a Diels-Alder reaction of 2-methoxybutadiene with 3-buten-2-one to form 1-methoxy-4-acetylcyclohexene; (2) cyclization of the enol ether with anydrous p-toluenesulfonic acid to form 4-methoxy-2-keto-bicyclo-[2.2.2]-octane; (3) reaction of the bridgehead methyl ether with boron tribromide to form the corresponding ketone with a bromine at the bridgehead. With this bromo ketone in hand we are beginning to explore methods that might be used to generate (and then trap) the very strained enone mentioned above.

REPORT OF RESEARCH CORPORATION GRANT

Page 2

STUDENT PARTICIPATION (Give names of students working on the project, their roles in the research, their achievements and their career plans.)

Jay Holt - graduate student now completing his third year of graduate study toward a PhD in chemistry. All of the experimental work on this project has been performed by Mr. Holt.

PAPERS AND SCIENTIFIC TALKS (Give titles and references to papers and talks resulting from the work. Attach two copies of any reprints acknowledging Research Corporation support, if not previously forwarded.)

None yet.

OTHER SUPPORT (List amounts and sources - including institutional-of other contributions received or expected.)

None other than limited support by the Georgia Tech School of Chemistry for services and supplies.

EXPENDITURE OF RESEARCH CORPORATION GRANT FUNDS (List cumulative expenditures.)

a. Equipment supplies (Itemize major expenditures) \$1796.35 - purchase of chemicals and expendable supplies during the period June 15, 1988 to June 14, 1990

b. Stipends (Academic status, rates, periods of appointment) \$11,199.96 - Research Assistantship (6-15-88 to 6-14-89) \$11,599.96 - Research Assistantship (6-15-89 to 6-14-90)

c. Other expenditures (Itemize and give purpose) None

Signature of principal investigator

Signeture of institution (required for terminal report only) ASST. TO VP/GEN. MGR.

Name and position of authorized officer of institution

ly 10, 1990

REPORT OF RESEARCH CORPORATION GRANT

(Submit original and one legible copy)

Reply to: 6840 E. Broadway

Tucson, Arizona 85710

ISTITUTION AND ADDRESS Georgia Institute of Technology Atlanta, GA 30332

RINCIPAL INVESTIGATOR Herbert 0. House

Please check one)

Terminal Report

PHONE (404)894-4044

CADEMIC RANK AND DEPARTMENT Professor, School of Chemistry

HORT TITLE OF RESEARCH SUPPORTED BY GRANT

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c. Other expenditures (Itemize and give purpose) None

ignature of principal investigator

gnature of authorized officer of institution (required for terminal report only) E. FAITH GLEASON

ASST. TO VP/GEN MCP ame and position of authorized officer of institution

The terminal report must be approved by an authorized officer of the institution.