

A STUDY OF THE RETENTION OF DYESTUFFS  
ON PAPER MAKING FIBERS UNDER  
VARIOUS CONDITIONS

A thesis submitted by

Ward Duncan Harrison

B.S. in Chem. E. 1932, Iowa State College,

M.S. 1934, Institute of Paper Chemistry

in partial fulfillment of the requirements of  
the Institute of Paper Chemistry for the degree  
of Doctor of Philosophy from Lawrence College,  
Appleton, Wisconsin.

*Ward D. Harrison*

June 1936

## TABLE OF CONTENTS

Chapter	Page
I INTRODUCTION	8
II METHOD OF ATTACK	11
III MATERIALS, EQUIPMENT, AND PROCEDURES	
A. Materials.....	15
B. Equipment.....	17
C. Experimental Procedures	
1. Method of Dyeing.....	21
2. Method of Determining Dye Retention by the Use of Beer's Law.....	22
3. Method of Determining Dye Retention by the Use of the Kubelka and Monk Equations.....	24
4. Methods of Determining Chemical Constants for Pulps.....	26
IV DISCUSSION OF EXPERIMENTAL PROCEDURE	
A. Beer's Law.....	28
B. Correction for Turbidity of White Waters.....	30
C. Method of Determining Dye Retention by the Use of Beer's Law.....	36
D. Limits of Error in the Retention Method Using Beer's Law.....	38
E. Results	47
1. Retention on Sulphite Pulp Bleached to Various Degrees.....	48

Chapter

Page

2.	Retention on Kraft Pulp Bleached to Various Degrees.....	55
3.	Retention on Bleached and Unbleached Sulphite and Kraft, Bleached Soda, Groundwood and Rag Pulps.....	63
4.	Retention on Unbleached and Bleached Sulphite and Kraft at Various pH's.....	71
5.	Retention on Unbleached and Bleached Sulphite at Various Consistencies.....	92
6.	Retention on Unbleached and Bleached Sulphite at Various Temperatures.....	102
7.	Retention on Unbleached and Bleached Sulphite at Various Reaction Times.....	112
8.	Retention on Unbleached Sulphite at Various Freenesses.....	121
9.	Retention of Various Dyes on Unbleached Sulphite.....	127
10.	Retention on Unbleached Sulphite When Various Acids were Used with and without Size.....	131
F.	Applicability of the Kubelka and Monk Equations for Determining Dye Retentions	133
1.	Kubelka and Monk Equations.....	134
2.	Method of Determining Dye Retention by the Use of the Kubelka and Monk Equations.....	137

Chapter	Page
3. Limits of Error in the Retention Method Using the Kubelka and Monk Equations.....	144
G. Applicability of the Freundlich Adsorption Equation to the Dye Retention Results.....	153
V CONCLUSIONS	161
VI SUMMARY	165
VII BIBLIOGRAPHY	166
VIII APPENDIX	
A. Calculations.....	167
B. Dye-Standard Graphs.....	227
C. Dye Formulas.....	252
D. Standard White Water Data.....	259

LIST OF ILLUSTRATIONS

	Page
Figure 1. Sheet Making Apparatus.....	18
Figure 2. Per Cent Retention of Du Pont Victoria Green SC as Per Cent Transmittance is Varied for Various Weight Dyeings.....	41
Figure 3. Per Cent Retention of Du Pont Brilliant Crocein FL as Per Cent Transmittance is Varied for Various Weight Dyeings.....	42
Figure 4. Per Cent Retention of Pontamine Fast Red 8BL as Per Cent Transmittance is Varied for Various Weight Dyeings.....	43
Figure 5. Retention of Du Pont Victoria Green SC on Sulphite Pulp at Various Degrees of Bleach.....	50
Figure 6. Retention of Du Pont Brilliant Crocein FL on Sulphite Pulp at Various Degrees of Bleach.....	51
Figure 7. Retention of Pontamine Fast Red 8BL on Sulphite Pulp at Various Degrees of Bleach.....	52
Figure 8. Retention of Du Pont Victoria Green SC on Kraft Pulp at Various Degrees of Bleaching	57
Figure 9. Retention of Du Pont Brilliant Crocein FL on Kraft Pulp at Various Degrees of Bleaching.....	58
Figure 10. Retention of Pontamine Fast Red 8BL on Kraft Pulp at Various Degrees of Bleaching.....	59
Figure 11. Retention of Du Pont Victoria Green SC on Various Pulps.....	66
Figure 12. Retention of Du Pont Brilliant Crocein FL on Various Pulps.....	67
Figure 13. Retention of Pontamine Fast Red 8BL on Various Pulps.....	68

	Page
Figure 14. Retention of Du Pont Victoria Green SC at Various pH's on Unbleached Sulphite Pulp.....	75
Figure 15. Retention of Du Pont Brilliant Crocein FL at Various pH's on Unbleached Sulphite Pulp.....	76
Figure 16. Retention of Pontamine Fast Red 8BL at Various pH's on Unbleached Sulphite Pulp..	77
Figure 17. Retention of Du Pont Victoria Green SC at Various pH's on Bleached Sulphite Pulp.	78
Figure 18. Retention of Du Pont Brilliant Crocein FL at Various pH's on Bleached Sulphite Pulp.....	79
Figure 19. Retention of Pontamine Fast Red 8BL at Various pH's on Bleached Sulphite Pulp....	80
Figure 20. Retention of Du Pont Victoria Green SC at Various pH's on Unbleached Kraft Pulp.....	81
Figure 21. Retention of Du Pont Brilliant Crocein FL at Various pH's on Unbleached Kraft Pulp.....	82
Figure 22. Retention of Pontamine Fast Red 8BL at Various pH's on Unbleached Kraft Pulp.....	83
Figure 23. Retention of Du Pont Victoria Green SC at Various pH's on Bleached Kraft Pulp....	84
Figure 24. Retention of Du Pont Brilliant Crocein FL at Various pH's on Bleached Kraft Pulp.	85
Figure 25. Retention of Pontamine Fast Red 8BL at Various pH's on Bleached Kraft Pulp.....	86
Figure 26. Retention of Du Pont Victoria Green SC at Various Consistencies on Unbleached Sulphite Pulp.....	94
Figure 27. Retention of Du Pont Brilliant Crocein FL at Various Consistencies on Unbleached Sulphite Pulp.....	95

	Page
Figure 28. Retention of Pontamine Fast Red 8BL at Various Consistencies on Un bleached Sulphite Pulp.....	96
Figure 29. Retention of Du Pont Victoria Green SC at Various Consistencies on Bleached Sulphite Pulp.....	97
Figure 30. Retention of Du Pont Brilliant Crocein FL at Various Consistencies on Bleached Sulphite Pulp.....	98
Figure 31. Retention of Pontamine Fast Red 8BL at Various Consistencies on Bleached Sulphite Pulp.....	99
Figure 32. Retention of Du Pont Victoria Green SC at Various Temperatures on Unbleached Sulphite Pulp.....	105
Figure 33. Retention of Du Pont Brilliant Crocein FL at Various Temperatures on Unbleached Sulphite Pulp.....	106
Figure 34. Retention of Pontamine Fast Red 8BL at Various Temperatures on Unbleached Sulphite Pulp.....	107
Figure 35. Retention of Du Pont Victoria Green SC at Various Temperatures on Bleached Sulphite Pulp.....	108
Figure 36. Retention of Du Pont Brilliant Crocein FL at Various Temperatures on Bleached Sulphite Pulp.....	109
Figure 37. Retention of Pontamine Fast Red 8BL at Various Temperatures on Bleached Sulphite Pulp.....	110
Figure 38. Retention of Du Pont Victoria Green SC at Various Times on Unbleached Sulphite Pulp.....	114
Figure 39. Retention of Du Pont Brilliant Crocein FL at Various Times on Unbleached Sulphite Pulp.....	115
Figure 40. Retention of Pontamine Fast Red 8BL at Various Times on Unbleached Sulphite Pulp.....	116

	Page
Figure 41. Retention of Du Pont Victoria Green SC at Various Times on Bleached Sulphite Pulp.....	117
Figure 42. Retention of Du Pont Brilliant Crocein FL at Various Times on Bleached Sulphite Pulp.....	118
Figure 43. Retention of Pontamine Fast Red 8BL at Various Times on Bleached Sulphite Pulp..	119
Figure 44. Retention of Du Pont Victoria Green SC at Various Freenesses on Unbleached Sulphite Pulp.....	124
Figure 45. Retention of Du Pont Brilliant Crocein FL at Various Freenesses on Unbleached Sulphite Pulp.....	125
Figure 46. Retention of Pontamine Fast Red 8BL at Various Freenesses on Unbleached Sulphite Pulp.....	126
Figure 47. Relation of Per Cent Reflectivity to Calculated Grams of Du Pont Victoria Green SC in a Sulphite Sheet.....	147
Figure 48. Relation of Per Cent Reflectivity to Calculated Grams of Du Pont Brilliant Crocein FL in a Sulphite Sheet.....	148



	Page
Figure 49. Relation of Per Cent Reflectivity to Calculated Grams of Pontamine Fast Red 8BL in a Sulphite Sheet.....	149
Figure 50. Equilibrium Adsorption Curves for Du Pont Victoria Green SC on Various P Pulps.....	158
Figure 51. Equilibrium Adsorption Curves for Du Pont Brilliant Crocein FL on Various Pulps.....	159
Figure 52. Equilibrium Adsorption Curves for Pontamine Fast Red 8BL on Various Pulps...	160

## CHAPTER I

### INTRODUCTION

Paper dyeing has increased considerably in the past decade, primarily because of the large growth in the speciality business. This growth, coupled with increased technical knowledge, has induced the dye manufacturers to continually increase their facilities for serving the paper maker. Due to the fact that it has long been recognized that this can be accomplished only when a technical viewpoint is taken, the use of dyes has been put on a greater scientific basis. In the relentless search for new and better colors, the dye companies have built up a wealth of technical knowledge which can not be overestimated. Out of this have come many new refinements of which the determinations of optimum physical and chemical conditions for the dyeing reaction are good examples.

Besides the dye manufacturers, other institutions have studied the dyeing reaction. The most notable probably are the textile research laboratories, and it should be pointed out that a great deal of knowledge in paper dyeing has been taken from these sources.

However, even though a great deal of work has been carried out in the paper dyeing field, there is an apparent scarcity of quantitative data concerning the retention of dyes. Work, for the most part, has been aimed in a direct way at the economics of the problem. This has been done by actually matching whatever was to be dyed using all possible dye formulas, within the limits of the shade, and then calculating the most economical one by taking into consideration the cost of each individual dye. If, however, the quantitative retentions of the various dyes available were known for the particular conditions of the problem, these data, along with their costs, would facilitate the elimination of some dyestuffs immediately in working out a color formula from an economic standpoint. In addition, and even more important in so far as this work is concerned, quantitative retention data obtained under well selected conditions should throw some light upon the trends to be expected in commercial practice. Also it is hoped that this work will be a stepping stone in the right direction for more complete understanding of the dyeing mechanism.

The study of the retention of dyestuffs on paper

making fibers under various conditions has been undertaken with the following objectives:

1. To develop a method for determining the retention of dye on paper making fibers.
2. To study the effect of variables on the retention of dye, such as type of dye and its concentration; temperature, pH, and consistency of the pulp; time of reaction; type of fiber, its chemical constants, and method of preparation; and type of acid used.
3. To point out the applicability of the retention data to theoretical equations.

## CHAPTER II

### METHOD OF ATTACK

Paper is ordinarily dyed by the addition of dye to the beater during the processing of the pulp. The pulp dyed at this stage is later formed into a sheet on the paper machine, at which time a considerable amount of water, commonly called white water, carrying with it fiber and dye, is removed from the pulp. The retention of the dye is the amount of dye left in the sheet. The problem may be attacked by two different methods. The first method consists in determining the amount of dye in the sheet after its manufacture, while the second method is to determine the amount of dye left in the white water. In the latter method, the amount of dye in the sheet may be calculated by subtracting the amount found in the white water from the amount added to the beater.

At first, no suitable way could be found for directly determining the amount of dye in the sheet. A colorimetric method for determining dye concentration in white waters using Nessler tubes was considered, but, due to its subjective nature, was discarded. An accurate objective method which would

show small differences in retention was necessary if reliable results were to be obtained. A spectrophotometric method, using "Beer's Law" as its fundamental principle for determining the amount of dye in the white water was then developed and found to be satisfactory.

When paper making fibers are dyed, the retention results obtained are dependent upon the conditions used during the dyeing operation. The most important of these conditions are type of fiber, its chemical constants, and method of preparation; type of dye and its concentration; temperature, pH, and consistency of the pulp; time of reaction; and type of acid used. With these conditions in mind, a systematic study was made in which one variable was considered at a time, within the limits that are ordinarily found in commercial practice. In order to have a base line for the results, standard conditions were set up and maintained, except for the variable under consideration, throughout the entire investigation. These standard conditions were 10 minutes reaction time, 20 degrees Centigrade temperature, 0.6 per cent consistency, unprocessed unbleached sulphite pulp, and 4.9 pH obtained by the use of sulphuric acid.

In connection with the retention of dyes under the conditions outlined above, it was thought desirable to study the applicability of the Kubelka and Monk equations, which involve an absorption coefficient, and the Freundlich adsorption equation to these data.

By the above method of attack it was hoped that retention results would be obtained which would facilitate the working out of scientific color formulas as well as throw some light upon the retention trends to be expected in commercial practice. It was also hoped that the results would be applicable to theoretical absorption and adsorption equations which, no doubt, in time to come, will help to solve more completely the dyeing mechanism.

Terms and phrases used in this investigation which might have several meanings are defined at this point.

#### DEFINITIONS

Natural White Water: White water obtained when no dye has been used in the sheet making process.

Natural Sheet: Sheet obtained when no dye has been used in the sheet making operation.

Transmission: The process of transmitting rays of light, considered as a function of wavelength (in the range 400 to 700 millimicrons). Example: transmission curve of distilled water.

Transmittance: The relative capacity of a dye solution to pass rays of light of a specified wavelength. The wavelength used for any given dye is at the point where the transmittance is at its maximum deviation from the transmission curve of distilled water.

Reflection: The process of throwing back rays of incident light, considered as a function of wavelength (in the range 400 to 700 millimicrons). Example: the reflection curve of a natural sheet.

Reflectivity: The relative capacity of a sheet of paper of infinite thickness to throw back rays of incident light of a specified wavelength. The wavelength used for any given dye is at the point where the reflectivity is at its maximum deviation from the reflection curve of a natural sheet.

Scattering: The process of deflecting light energy from its original direction.

Absorption: The process of disappearance of light upon passing through a medium.



## CHAPTER III

### MATERIALS, EQUIPMENT, AND PROCEDURES

In carrying out the work outlined in the method of attack the following materials, equipment, and procedures were used.

#### A. Materials.

The unbleached sulphite pulp used was a hard commercial pulp made from hemlock and as a result was on the red side in color. Bleached sulphite in five different grades was produced from this unbleached pulp by using 3, 6, 10, 20, and 30 per cent bleach (based on 35 per cent available chlorine) in a single stage, high density (12 per cent consistency, calcium hypochlorite treatment. In each case the pulp was well washed after complete exhaustion of the bleach liquors.

The unbleached kraft pulp used was an extremely hard grade of commercial jack pine kraft. This pulp was bleached to four different degrees using 5, 15, 30, and 40 per cent bleach (based on 35 per cent available chlorine) with calcium hypochlorite. The bleach using 5 per cent was carried out in a single stage at 12 per cent consistency; the 15 and 30 per cent bleaches were carried out in two stages, both at 12

per cent consistency, by adding two-thirds of the bleach in the first stage and one-third in the second; the 40 per cent bleach was carried out exactly the same as the 30 per cent bleach except that an additional 10 per cent of bleach was added in a third stage at 12 per cent consistency. In the case of multiple stage bleachings, the second or third stage was never started until after the chlorine from the previous stage was completely exhausted and the bleach residues were washed out.

The dyes used represented Du Pont Company products and are sold under the following trade names: Du Pont Victoria Green Small Crystals; Du Pont Rhodamine B; Du Pont Auramine Concentrated; Du Pont Methylene Blue ZX; Du Pont Basic Brown BR; Du Pont Safranine T Extra; Du Pont Methyl Violet NE; Du Pont Brilliant Crocein FL; Du Pont Anthraquinone Blue B; Du Pont Nigrosine WSB Powder; Du Pont Quinoline Yellow Concentrated; Du Pont Orange II Concentrated; Pontacyl Violet S4B; Pontamine Fast Red 8BL; Du Pont Purpurine 4B Concentrated; Pontamine Yellow SXP; Pontamine Fast Scarlet 4BS; Pontamine Black E; Du Pont Brilliant Paper Yellow Concentrated; and Pontamine Fast Yellow>NNL. These dyes were used in 0.3 gram per liter solutions

which were freshly prepared each day.

Two types of acid, both of reagent grade, were used, namely, sulphuric acid and alum (aluminum sulphate).

The size used was neutral in a three per cent solution made from "G" gum rosin.

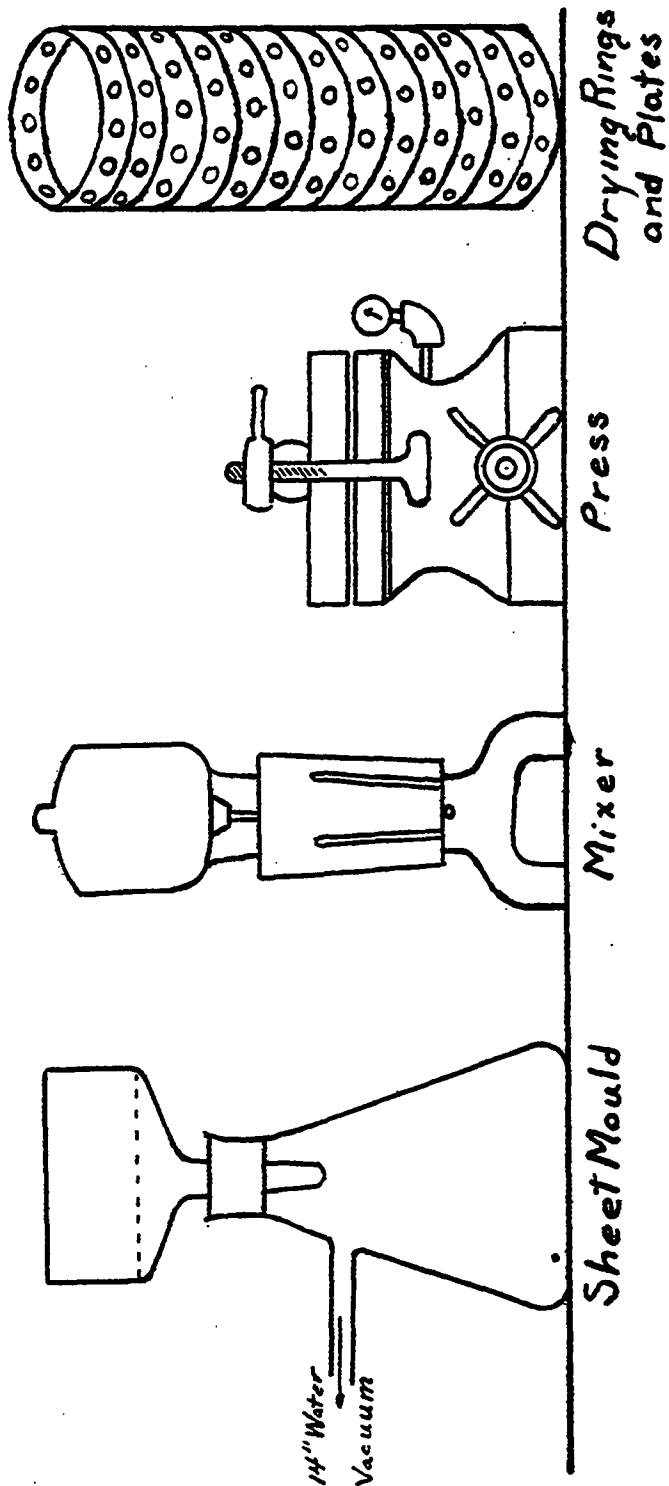
#### B. Equipment.

The sheet making apparatus, figure 1, consisted of an ordinary 6 inch Buchner funnel connected to a 2 liter suction flask carrying a vacuum of 14 inches of water. Auxiliary equipment used in connection with the sheet making apparatus consisted of a Hamilton Beach malted milk mixer, model 25, a 65 mesh wire to fit the Buchner funnel, drying plates and rings such as are used with the British sheet making apparatus<sup>7</sup>, a Jena glass funnel of 1G1 porosity, a good grade of white blotters, and a press capable of holding 7 by 7 inch sheets and exerting a pressure of 100 pounds per square inch.

The pH standards used were the ordinary colorimetric standards put out by the LaMotte Chemical Products Company.

An experimental Valley beater, of one pound capacity, was used for processing the pulp.

Figure 1.  
Sheet Making Apparatus



A General Electric Recording Color Analyzer, developed by Professor A. C. Hardy of the Massachusetts Institute of Technology, was used for the spectrophotometric measurement of transmittance and reflectivity. The operation of this instrument may briefly be described as follows: the light source, a ribbon filament projection lamp, is focused upon the entrance slit of a monochromator which is of the Abbé autocollimating type. The light beam leaving this system, comprising a wavelength band of ten millimicrons width, is split up into two beams. One beam passes to a flicker disk and is reflected one half of the time by the disk to a photoelectric cell, while the other beam passes through a small glass cell containing the liquid to be analyzed, is reflected off the surface of a block of magnesium carbonate, and is caught by the photoelectric cell when it is not viewing the previously mentioned beam. When the instrument is to be used for analyzing color in papers, the glass cell is removed and the paper is inserted in place of the magnesium carbonate. The current from the photoelectric cell is amplified by a three stage resistance coupled amplifier, the output of which feeds the grids of two thyratrons; the plates of these tubes

are in series with the field coils of a small motor, the armature of which is supplied from an independent source of the same frequency. With this arrangement the armature rotates in one direction when the beam from the specimen is the more intense and in the opposite direction when the beam that is reflected from the flicker disk is the more intense. When the intensities are equal, the field currents are equal and opposite thus stopping the motor. By causing the motor to control a sector diaphragm which varies the intensity incident upon the flicker disk, the two beams are automatically balanced. A pen operating on a rotating drum covered with a sheet of graph paper records the position of this diaphragm on the ordinate of the graph while the rotation of the drum, which is synchronized with the optical system, changes the wavelength of the spectral band admitted to the photo-electric cell and is recorded on the abscissa of the graph. When the instrument is to be used for analyzing the color of liquids for their transmittance, the result obtained is relative to the transmittance of the glass cell when filled with distilled water. On the other hand, when analyzing papers for their reflectance, the result obtained at any given wavelength with

a freshly prepared magnesium carbonate block is arbitrarily chosen as 100 per cent reflectivity.

### C. Experimental Procedures.

The following experimental procedures were used in carrying out the work of this investigation.

#### 1. Method of Dyeing.

The dyeing procedure is carried out under well controlled conditions as follows: Exactly 3.0000 grams of pulp, calculated on the bone dry basis, are weighed out on an analytical balance. Sufficient distilled water of the correct temperature is placed upon the pulp so that when the dye and acid are added the mixture will be of the correct consistency. The pulp mixture is then placed in a malted milk mixer, the correct amount of dye run in from a burette, a stop watch started, and the acid used for setting the dye added. At the moment the reaction time is up, the dyed pulp is removed from the stirrer and immediately made into a sheet on the sheet making apparatus. The sheet formed in this way is removed by couching off on a smooth surfaced metal plate (British sheet mould drying plate), pressed between clean blotters at a pressure of 100 pounds per square inch for exactly one minute, air dried under tension (by the use of

British sheet mould drying rings), and stored in the dark for later spectrophotometric examination. The white water caught in the flask of the sheet making apparatus is filtered through a Jena filter to remove any fiber draining through from the sheet and set aside for later spectrophotometric examination.

## 2. Method of Determining Dye Retention by the Use of Beer's Law<sup>2</sup>.

The determination of dye retention for any given dye is carried out in nine steps as follows:

a. Dye standards of known concentration are prepared by adding a known amount of dye to a known volume of distilled water. These standards are used for comparing the unknown dye solutions (white waters) and are adjusted to a corresponding pH. Solutions, in a range from no dye to one containing more dye than is expected in the white waters being examined, are prepared for each dye so that the amount in the unknown white water will fall within the range of the standards.

b. Transmission curves of the standard dye



solutions for the dye in question along with the transmission curve of distilled water are obtained by the use of the color analyzer.

c. A point on the wavelength scale of the transmission curves is picked where maximum deviation occurs for the dye solution relative to distilled water. When the transmittance of a solution is stated, it is hereafter understood to be at this point.

d. The logarithms of the transmittances of the standard dye solutions are plotted against their concentrations (Beer's law<sup>2</sup> is satisfied if the curves are straight lines).

e. Transmission curves of the white water in question and a natural white water, i.e. no dye, are obtained by the use of the color analyzer.

f. The transmittances of the white water and the natural white water are determined at the wavelength of maximum deviation. This point remains the same for any given dye.

g. The transmittance of the white water, corrected for turbidity, is obtained by dividing the product of the transmittance of the white

water and the transmittance of the distilled water by the transmittance of the natural white water.

h. The logarithm of the corrected white water transmittance is determined and the concentration of the dye solution corresponding to this value is read from the standard graph as developed in (d) above.

i. From the concentration of the dye in the white water the per cent retention of the dye is determined.

### 3. Method of Determining Dye Retention by the Use of Kubelka and Monk Equations<sup>6</sup>.

The determination of dye retention for any given dye is carried out in eight steps as follows:

a. The reflection curves of the series of papers under consideration using the same furnish and dyed with the same dye in varying concentrations, as well as a natural sheet (i.e. no dye), are obtained with the help of the color analyzer.

b. The retention of dye for one of the papers is determined by the method using Beer's law.

From this, the weight of dye in the sheet is determined.

c. A point on the wavelength scale of the reflection curve is picked where maximum deviation is obtained for the dyed paper relative to the natural sheet. When the reflectance of the paper is stated it is always considered to be at this point.

d. The K/S values for all the papers are determined by the use of the equation,  $K/S = (R_{\infty} - 1)^2 / 2R_{\infty}$ , where  $R_{\infty}$  is the reflectivity of the paper (Kubelka and Monk equation).

e. The K/S value for the natural sheet is subtracted from the K/S value for each of the other papers in order to get the K/S value due to the dye in each paper. It is assumed that the S values are not affected by the dyeing.

f. The weight of dye left in the sheet as determined by the method using Beer's law, see (b.) above, is set equal to the product of the corresponding K/S value and a constant. From this equation the value of the constant is determined.

g. The product of the constant, as determined in (f.) above, and the K/S value for a given sheet is equal to the weight of dye in that sheet.

h. From the weight of dye in the sheet the per cent retention is calculated.

#### 4. Methods of Determining Chemical Constants for Pulps.

The methods used in determining the chemical constants of the pulps are as follows:

a. Lignin is determined by the Modified Forest Products Laboratory Method<sup>4</sup> (Institute Standard Method 13).

b. Alpha cellulose is determined by the T. A. P. P. I. Method T 203m<sup>10</sup> (Institute Standard Method 421).

c. The permanganate number is determined by a method published by R.H. Wiles<sup>9</sup> (Institute Standard Method 409).

d. The methoxyl content is determined according to the Viebock and Schwappach Method as published by E.P. Clark<sup>1</sup>.

e. Total sulphur is determined by a micro chemical method as follows: The sample (3

grams) is placed in a large test tube, covered with concentrated nitric acid and heated gently until brown nitric acid fumes are evolved freely. Until the pulp is hydrolyzed, the reaction generates enough heat to maintain the rate of reaction. When the violent reaction subsides, heat is applied to complete the oxidation and boil off the excess acid. Small amounts of potassium chlorate are added occasionally to aid in the oxidation and furnish an alkali cation to combine with the sulphates formed by the oxidation. The test tube is boiled just to dryness to drive off all the nitric acid; the residue is then taken up in hydrochloric acid and analyzed for sulphates in the ordinary way using a micro chemical technique. This method is outlined by Holzer<sup>3</sup> and differs only in the weight of sample taken.

f. Inorganic sulphur is determined micro-chemically by analyzing the ash for sulphates in the usual way.

g. Organic sulfur is calculated by subtracting the inorganic from the total sulphur.

## CHAPTER IV

### DISCUSSION OF EXPERIMENTAL PROCEDURE

As has already been pointed out, Beer's law<sup>2</sup> has been used as the fundamental principle in the determination of the retention of dyes. The use of this method is dependent upon dye standards, solutions of known concentration for each dye, and a method of accurately comparing dyed white waters with the standard solutions. Beer's law requires that the logarithm of the transmittance vary linearly with the concentration. With graphs of this type it has been possible to determine the dye concentration in white waters from their corrected transmittance data, hence the retention may be calculated.

#### A. Beer's Law.

In materials such as dye solutions which are homogeneous, the absorption of light by them depends upon the thickness in accordance with Bouger's law. Bouger set forth this law in 1729. It was further developed by Lambert and consequently is frequently called Lambert's law of absorption. Suppose that a layer of unit thickness transmits a fraction  $t$  of the light incident upon it. This layer will absorb a

fraction  $(1 - t)$ . Consequently a thickness  $x$  of the material will transmit the fraction  $t^x$ , and the intensity of the light transmitted is

$$I = I_0 t^x$$

where  $I_0$  is the intensity of the incident light.

This expression may be written

$$I = I_0 e^{-ax}$$

where  $a$ , the absorption coefficient, equals  $-\log t$ .

The absorption coefficient of a solution is in general proportional to the concentration of the solute. The absorption coefficient can therefore be written as

$$a = bc,$$

where  $c$  is the concentration and  $b$  is the absorption coefficient for unit concentration. With this substitution, the equation becomes

$$I = I_0 e^{-bcx},$$

which is known as Beer's law. In the practical application of this law a cell of constant thickness was used to measure the transmittance,  $I / I_0$ , of a dyed solution. Beer's law can therefore be written as

$$I / I_0 = e^{-kc}$$

where  $k$  equals  $bx$ . This expression may further be written as

$$\log I / I_0 = \log e^{-kc} = -kc,$$

where  $I / I_0$  is the transmittance by definition, or, in other words, the logarithm of the transmittance is a linear function of the concentration.

Dye standards for the dyes investigated were prepared and the logarithms of their transmittances were plotted against their concentrations for each dye. These results will be found in section B of the appendix. The resulting straight lines show that this law is applicable under the conditions of this investigation.

#### B. Correction for the Turbidity of White Waters.

The white water obtained from a natural sheet, that is, one to which no dye has been added, has a slightly cloudy appearance in comparison with distilled water even though it has been filtered. This fact shows up conclusively when the transmittance data of the two solutions are examined. For instance, the transmittance of unbleached sulphite white waters at 620 millimicrons is 83.0 per cent while that of distilled water is 97.4 per cent. (See section D of the appendix for natural white water data). A correction has to be made in order to put the white water data on a basis corresponding to that of the distilled water, and thus make the transmittance data of the white waters applicable to the graphical data set up by the



known dye standards. In determining the nature of this correction, dye in known amounts was added to a natural white water and its transmittance determined. It was found that the product of the transmittance of a dyed white water and a constant gave a corrected transmittance reading which, when applied to the graphical results obtained with the standard dye solutions, produced a result within one per cent of the amount of dye added. The constant in these calculations is obtained for any given pulp by dividing the transmittance of distilled water by the transmittance of the natural white water at the same wavelength used in determining the transmittance of the dyed white water. Experimental results in this connection are shown in table 1.

TABLE 1.				
ACTUAL AND CALCULATED DYE CONCENTRATION IN WHITE WATERS				
Dye Added Grams per Liter	Transmittance	Corrected Transmittance	Logarithm Corrected Transmittance	Dye Calculate Grams per Liter
0	83.0	98.3*	1.993	0
0.00375	64.0	75.9	1.880	0.00375
0.00750	49.7	58.9	1.770	0.00745

Note \*. The corrected transmittance for 0 grams dye added is the transmittance of distilled water.

The turbidity of white waters for any given pulp

is very nearly constant due to the fact that the white waters are put through a Jena filter of 1G1 porosity immediately after their separation from the pulp. Any deviation from the natural white water would throw the determination off because of an increase or a decrease in the scattering of light.

It has been noticed from transmittance data on dye standards that every dye solution has a transmittance equal to the transmittance of distilled water at some point in the visible wavelength band. For instance, it was noticed that a solution of Du Pont Victoria Green SC has exactly the same transmittance at 480 millimicrons as distilled water. Both Du Pont Brilliant Crocein FL and Pontamine Fast Red 8BL have transmittances equal to distilled water at 620 millimicrons. From this it may be postulated that a white water containing Du Pont Victoria Green SC should have the same transmittance at 480 millimicrons as its natural white water, that a white water containing Du Pont Brilliant Crocein FL should have the same transmittance at 620 millimicrons as its natural white water, etc. This has been found to be true with few exceptions. Exceptions, however, can be attributed only to a variation in turbidity because a white water is made up of only three things, distilled water, dye,

and the so-called turbidity or light scattering material. If the deviations were in the dye or water, deviations would be seen in the dye standards at this point. These deviations may be corrected for without much loss in accuracy if the results do not show a greater variation than a few tenths of one per cent. For example, if the transmittance of a white water solution containing Du Pont Victoria Green SC is 0.2 per cent low at 480 millimicrons where it should coincide with the transmittance of the natural white water, then 0.2 per cent is added to the transmittance of this solution at 620 millimicrons which is the point used in calculating dye retentions for that particular dye. If, however, the transmittance is 0.2 per cent too high at the coinciding point, then 0.2 per cent is subtracted from the transmittance of the dye at 620 millimicrons, etc. Proof that small corrections of this nature do not impair the accuracy of the results was obtained by studying white waters with varying amounts of turbidity. The white water from a natural sheet of unbleached sulphite was divided into three equal parts. The first part was undiluted, the second part was diluted by one half of its volume with distilled water, while the third part was diluted by twice its volume. These solutions then contained amounts of turbidity per unit volume

in the ratios of one, two thirds, and one third. The transmittances of these solutions, along with distilled water, were then determined at 500 and 600 millimicrons. The results are tabulated in table 2. It is

TABLE 2. TRANSMITTANCE OF WHITE WATER WITH VARYING AMOUNTS OF TURBIDITY				
Turbidity Ratio	0	1/3	2/3	1
Transmittance at 500 millimicrons	94.0	86.6	80.5	75.1
Transmittance at 600 millimicrons	97.4	91.8	87.2	83.0

apparent from table 2 that the difference in transmittance between the ratios of one and two thirds turbidity per unit volume is not the same at 500 and 600 millimicrons being 5.4 and 4.2 per cent respectively. If these differences had been equal, corrections of any magnitude could be made. However, in this case, a maximum correction of only 0.3 per cent can be made without reducing the accuracy of the method. A correction of 0.3 per cent amounts roughly to a 2 per cent change in the amount of turbidity present in the standard, and the assumption can be made for the purpose of a rough calculation that the turbidity

changes as a linear function of the transmittance. Then  $(0.3)(0.33)(100)/5.4$  or 1.85 per cent change in turbidity is equal to a 0.3 per cent change in the transmittance. Now if a 0.3 per cent error was noticed at the coincidence point, 500 millimicrons for example, and corrected for at 600 millimicrons by adding to or subtracting from the transmittance at that point, an error would be introduced. Instead of a 0.3 per cent correction only  $(0.3)(4.2)/5.4$  or 0.233 per cent correction should have been made. However, if a correction no larger than 0.3 per cent is used, the accuracy of the method will not be impaired. The points, 500 and 600 millimicrons, were arbitrarily chosen and the results obtained therefrom are typical of the results that would be obtained with any dye.

Due to the fact that the turbidity in white water settles out with time, it was found necessary to vigorously agitate the white water solution and to make the transmittance determination at a definite time after this agitation. Actual settling rates for the turbidity produced in an unbleached sulphite white water are shown in table 3 at wavelengths of 500 and 600 millimicrons.

<p>TABLE 3</p> <p>SETTLING RATES FOR TURBIDITY IN AN UNBLEACHED</p> <p>SULPHITE WHITE WATER</p>					
Time of settling Minutes	0	3	6	9	12
Transmittance at 500 millimicrons	75.1	76.1	76.7	77.1	77.4
Transmittance at 600 millimicrons	83.0	84.0	84.5	84.9	85.2

### C. Method of Determining Dye Retention by the Use of Beer's Law.

The determination of dye retention by the use of Beer's law has already been given. In addition, a discussion of the method will be given at this point in order to furnish a clear understanding of the way in which the retention calculations were handled. For the most part, dyeings were made at 0.6 per cent consistency by adding 3.0000 grams of pulp to 500 cubic centimeters of water. At other consistencies, the weight of pulp was varied, but the volume of water was always kept constant. The volume of white water obtained when these conditions were used amounted to 470 cubic centimeters.

The transmittance of the dyed white water was determined and corrected in the manner which has

already been discussed. Dye concentration corresponding to the corrected transmittance was then obtained from the plotted standard dye solution data. The concentration obtained in this way was converted into weight of dye present by multiplying by 470 (the volume of the white water in cubic centimeters) and dividing by 1000. The per cent retention of dye was determined by subtracting the amount in the white water from the amount originally used, dividing the result by the amount originally used, and multiplying by 100. Sample calculations, starting with the corrected transmission, are given in table 4 for Du Pont Victoria Green SC on unbleached sulphite. Complete calculations for all the work carried out in this investigation are tabulated in section A of the appendix.

TABLE 4						
SAMPLE CALCULATIONS FOR THE DETERMINATION OF PER CENT						
DYE RETENTION BY THE METHOD USING BEER'S LAW						
Cor- rected Trans- mittance	Log Cor- rected Trans- mittance	Conc. of White Water	Gms. of Dye in White Water	Gms. of Dye Used	Gms. of Dye in Sheet	Per Cent Reten- tion
98.3	1.993	0	0	0	0	100.0
92.5	1.966	.000895	.000420	.0090	.00858	95.3

#### D. Limits of Error in the Retention Method Using Beer's Law.

In examining the errors inherent in the retention method using Beer's law for determining dye retentions, it was found that dyed white water solutions could be produced under identical conditions which would not vary more than one tenth per cent in transmittance. It was also found that the color analyzer would reproduce results within one tenth per cent. With this in mind, the error in retention was determined when the transmittance was purposely changed by one tenth per cent for the  $\frac{1}{2}$ , 1, and 3 pound dyeings of Du Pont Victoria Green SC, Du Pont Brilliant Crocein FL, and Pontamine Fast Red 8BL on unbleached sulphite. These three dyes are representative of the three common classes of dyestuffs and give a general idea of the errors to be expected for any dye.

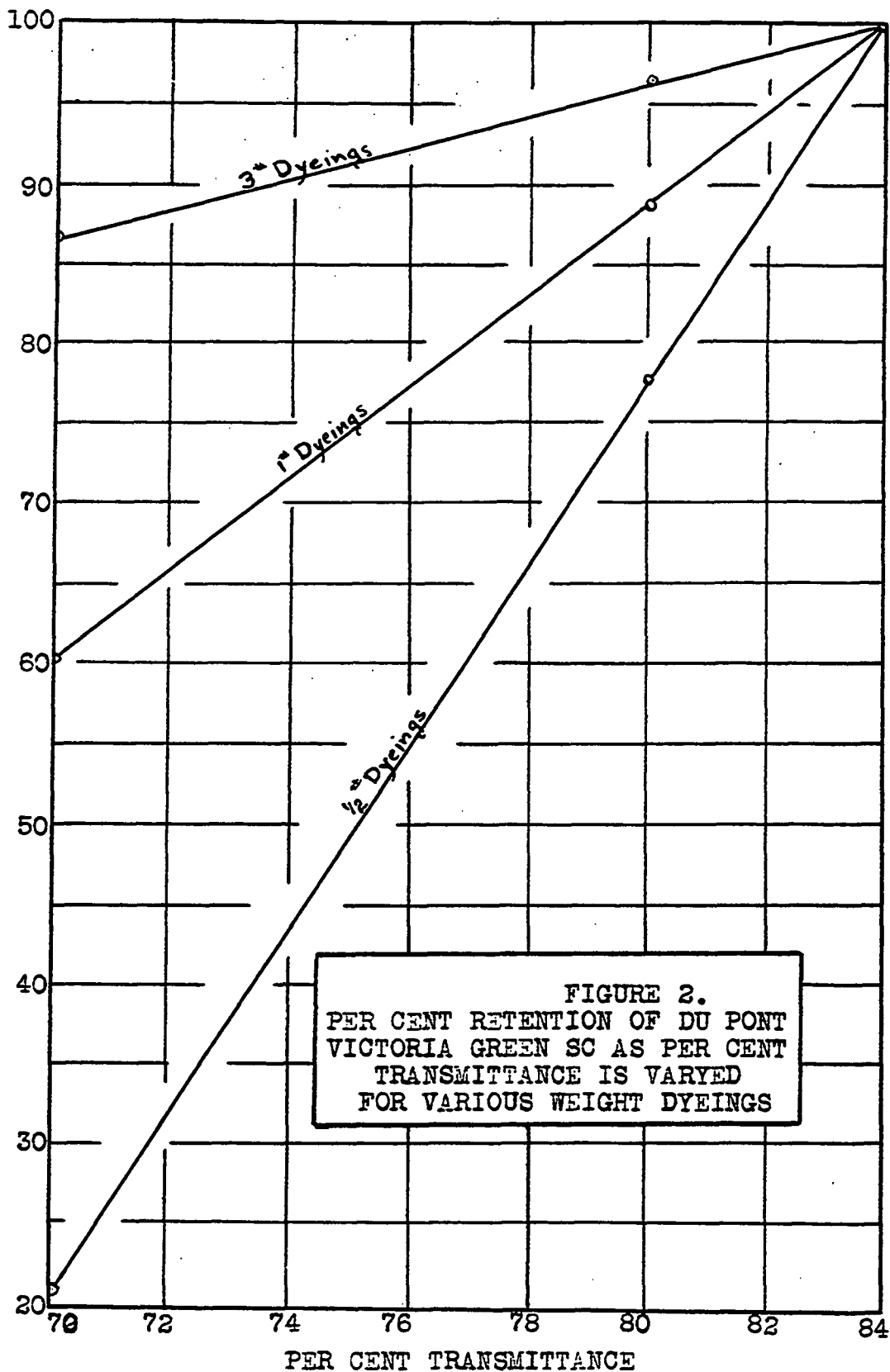
It was found in carrying out this calculation that as the transmittance of a given dye solution for a given weight dyeing was varied, the per cent retention varied as a linear function of the transmittance. In the light of this fact a change in the transmittance could be arbitrarily made to correspond to a possible error, the per cent retention determined, and then the change due to 0.1 per cent change in

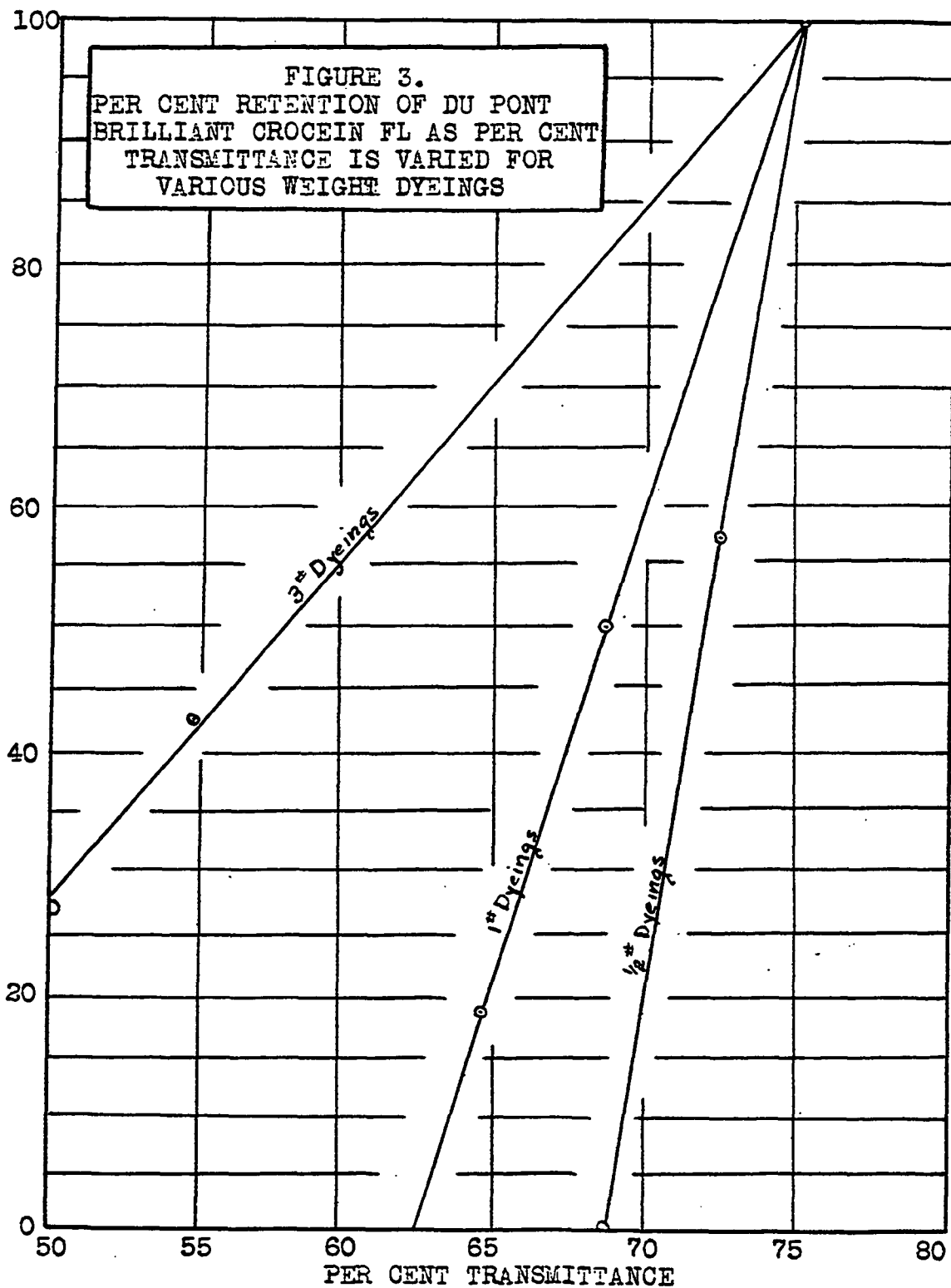


transmittance calculated. Accuracy was conserved in this way. The calculations for Du Pont Victoria Green SC are shown in table 5. A sample calculation will be shown for a one half pound dyeing of this dye. The transmittance changes 14.0 per cent as the retention changes 79.0 per cent. Due to the fact that per cent retention is a linear function of transmittance as is shown graphically in figures 2, 3, and 4, the per cent change in retention due to one tenth per cent change in transmittance can be calculated by dividing 79.0 by 140. The result is 0.564 per cent. Calculations for Du Pont Brilliant Crocein FL and Pontamine Fast Red 8BL are shown in tables 6 and 7 while a summary of all the data concerning errors to be expected with the dyes studied is shown in table 8.

TABLE 5

RETENTION OF DU PONT VICTORIA GREEN 80 AS TRANSMITTANCE IS PURPOSELY VARIED								
POUNDS DYE PER 1000 POUNDS PULP	TRANS- MITTANCE	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GMS. DYE IN WHITE WATER	GMS. DYE USED	GMS. DYE IN SHEET	PER CENT RETENTION
1/2	84.0	98.3	1.993	0	0	0	0	100.0
	80.0	93.6	1.971	.000710	.000335	.0015	.001166	77.8
	70.0	82.0	1.914	.002620	.001185	.0015	.000313	21.0
1	84.0	98.3	1.993	0	0	0	0	100.0
	80.0	93.6	1.971	.000710	.000335	.0030	.002665	88.7
	70.0	82.0	1.914	.002620	.001185	.0030	.001815	60.4
3	84.0	98.3	1.993	0	0	0	0	100.0
	80.0	93.6	1.971	.000710	.000335	.0090	.008665	96.3
	70.0	82.0	1.914	.002620	.001185	.0090	.007815	86.8





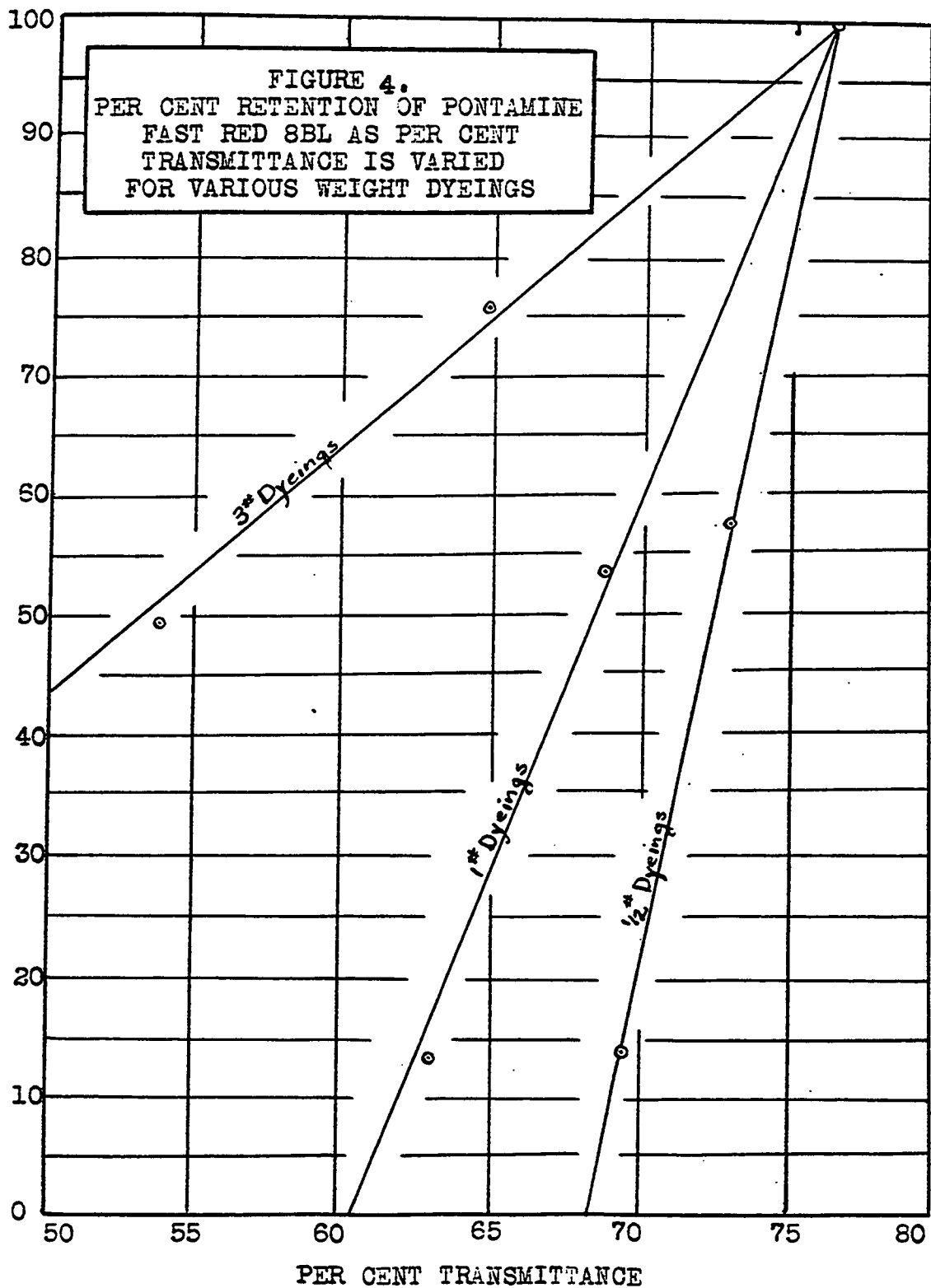


TABLE 6  
RETENTION OF DU PONT BRILLIANT CROCEIN FL AS TRANSMITTANCE IS PURPOSELY VARIED

POUNDS DYE PER 1000 POUNDS PULP	TRANS- MITTANCE	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GMS. DYE IN WHITE WATER	GMS. DYE USED	GMS. DYE IN SHEET	PER CENT RETENTION
1	75.1	94.0	1.973	0	0	0	0	100.0
	72.2	90.4	1.956	.001350	.000635	.0015	.000865	57.7
	68.5	85.7	1.933	.003190	.001500	.0015	0	0
1	75.1	94.0	1.973	0	0	0	0	100.0
	68.5	85.7	1.933	.003190	.001500	.0030	.001500	50.0
	64.5	80.7	1.907	.005200	.002440	.0030	.000560	18.7
3	75.1	94.0	1.973	0	0	0	0	100.0
	54.4	68.1	1.833	.011100	.005240	.0090	.003760	41.8
	50.0	62.6	1.797	.014000	.006580	.0090	.002420	26.9

TABLE 7  
RETENTION OF PONTAMINE FAST RED 8BL AS TRANSMITTANCE IS PURPOSELY VARIED

POUNDS DYE PER 1000 POUNDS PULP	TRANS- MITTANCE	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GMS. DYE IN WHITE WATER	GMS. DYE USED	GMS. DYE IN SHEET	PER CENT RETENTION
1/2	76.7	94.5	1.975	0	0	0	0	100.0
	72.9	89.9	1.954	.001340	.000630	.0015	.000870	58.0
	69.5	58.6	1.932	.002750	.001290	.0015	.000210	14.0
1	76.7	94.5	1.975	0	0	0	0	100.0
	68.8	84.9	1.929	.002940	.001380	.0030	.001620	54.0
	62.8	77.4	1.889	.005520	.002595	.0030	.000405	13.5
3	76.7	94.5	1.975	0	0	0	0	100.0
	64.8	80.0	1.903	.004570	.002150	.0090	.006850	76.1
	53.8	66.4	1.822	.009750	.004580	.0090	.004420	49.1

TABLE 8.			
SUMMARY OF ERRORS TO BE EXPECTED WHEN TRANSMITTANCE VARIES 0.1 PER CENT			
Pounds Dye per 1000 Pounds Pulp	$\frac{1}{2}$	1	3
Du Pont Victoria Green SC			
Difference in Transmittance (%)	14.0	14.0	14.0
Difference in Retention (%)	79.0	39.6	13.2
Error in Retention Due to 0.1% Change in Transmittance (%)	0.564	0.282	0.094
Du Pont Brilliant Crocein FL			
Difference in Transmittance (%)	6.6	10.6	25.1
Difference in Retention (%)	100.0	81.3	73.1
Error in Retention Due to 0.1% Change in Transmittance (%)	1.516	0.767	0.291
Pontamine Fast Red 8BL			
Difference in Transmittance (%)	7.2	13.9	22.9
Difference in Retention (%)	86.0	86.5	50.9
Error in Retention Due to 0.1% Change in Transmittance	1.190	0.622	0.222



## E. Results.

The results of this investigation may be divided up into ten sections, each section of which takes into consideration one of the variables that is important in the dyeing reaction while all the rest of the variables are held constant. In this way it has been possible to study the retention of the various dyes on various stocks under various conditions. The standard conditions used in this work were 0.6 per cent consistency, 4.9 pH obtained by the use of sulphuric acid, a temperature of 20°C., 10 minutes reaction time, and, unless otherwise stated, these conditions were held constant in each determination. Dyes, Du Pont Victoria Green SC, Du Pont Brilliant Crocein FL, and Pontamine Fast Red SBL, which are basic, acid, and direct, respectively, were used as typical examples throughout the investigation. Sulphuric acid was used in most of the work because alum produced a flock in the white water which was undesirable. Results in part 10 of this section justify this use of sulphuric acid.

Before discussing these results in detail, one general conclusion, which is true in every part of this investigation, can be made, namely, that for any given

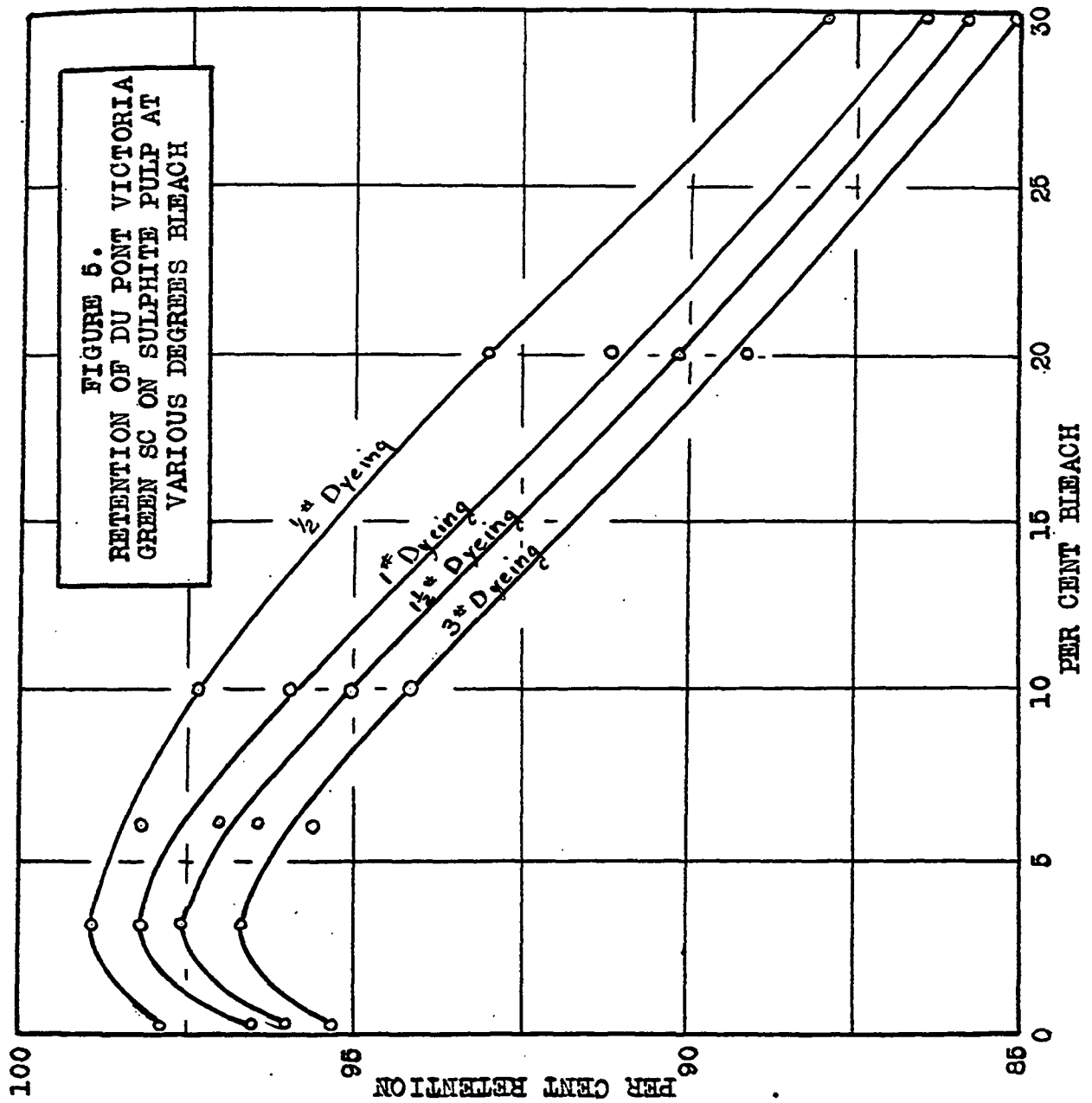
dye under identical conditions of time, temperature, consistency, etc., the one half pound dyeings are retained by a given pulp to a greater extent than those using more dye.

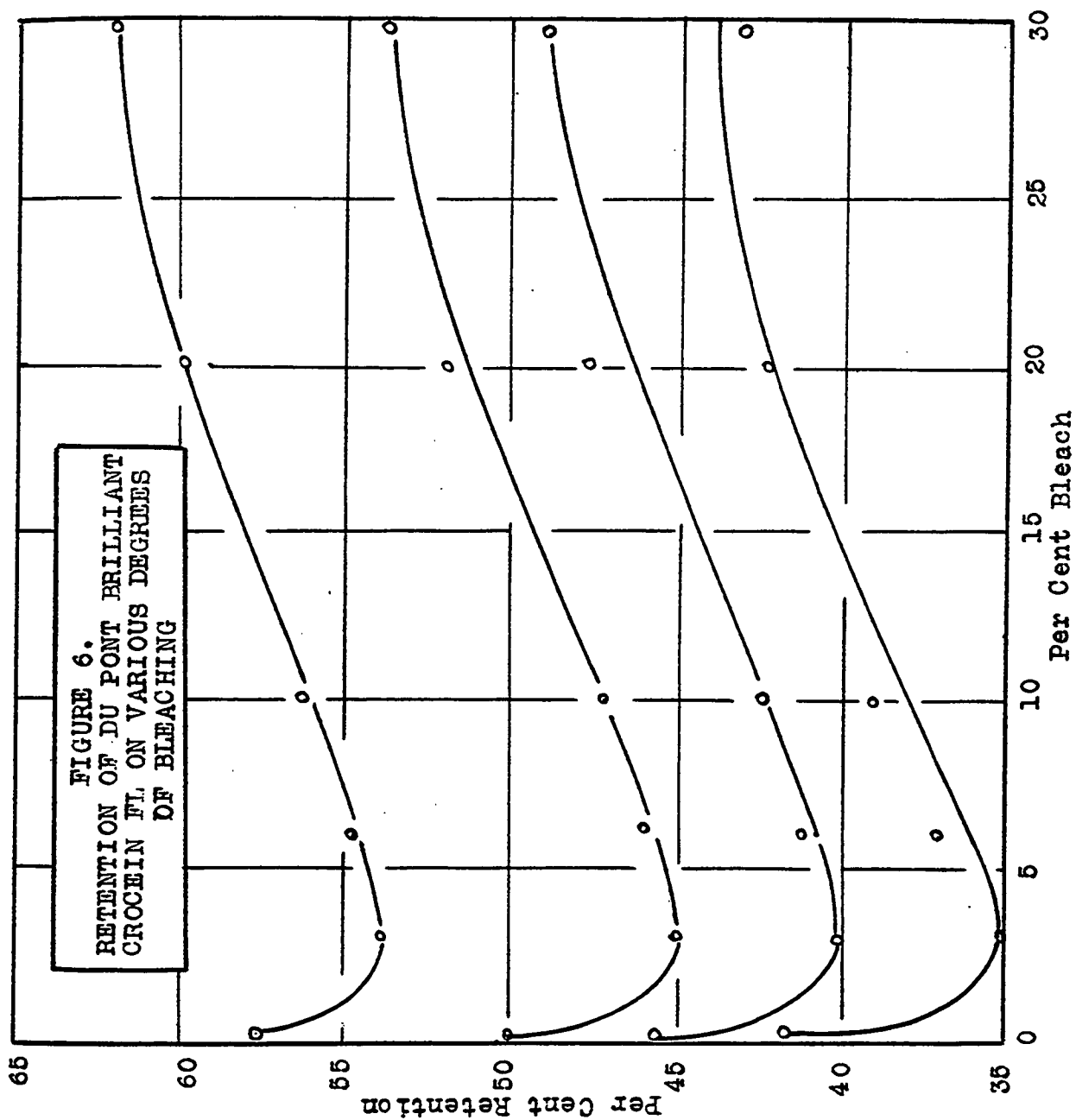
1. Retention on Sulphite Pulp Bleached to Various Degrees.

Unbleached sulphite pulp was bleached to five different degrees and dyed, in addition to the unbleached pulp, with the three typical dyestuffs in four different strengths ( $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings). The retention data are tabulated in table 9 and are shown graphically in figures 5, 6, and 7. The chemical constants of these pulps are tabulated in table 10. Du Pont Victoria Green SC was retained better than either Du Pont Brilliant Crocein FL or Pontamine Fast Red 8BL with the acid dye being retained the least. The basic dyestuff is retained to the extent of 97.9 per cent for a one half pound dyeing on unbleached sulphite, while the acid and direct dyes were retained 57.7 and 58.0 per cent respectively.

It has been believed that basic dyestuffs attach themselves to lignified fibers much better than dyes of other classes. In addition, it has

TABLE 9.				
PER CENT RETENTION ON SULPHITE PULP BLEACHED				
TO VARIOUS DEGREES.				
Per Cent Bleach	Pounds Dye per 1000 Pounds Pulp	Du Pont Victoria Green SC Per Cent	Du Pont Brilliant Crocein FL Per Cent	Pontamine Fast Red 8BL Per Cent
0	$\frac{1}{2}$	97.9	57.7	58.0
	1	96.5	50.0	54.0
	$1\frac{1}{2}$	96.0	45.7	52.2
	3	95.3	41.8	49.1
3	$\frac{1}{2}$	98.9	54.0	48.4
	1	98.2	44.9	44.3
	$1\frac{1}{2}$	97.6	40.1	42.3
	3	96.7	35.0	39.1
6	$\frac{1}{2}$	98.2	54.9	51.8
	1	97.0	46.0	48.5
	$1\frac{1}{2}$	96.4	41.2	46.5
	3	95.7	37.1	43.2
10	$\frac{1}{2}$	97.3	56.4	56.8
	1	96.0	47.1	52.6
	$1\frac{1}{2}$	95.1	42.8	50.9
	3	94.2	39.1	48.4
20	$\frac{1}{2}$	93.0	60.0	59.9
	1	91.2	52.0	56.7
	$1\frac{1}{2}$	90.2	47.6	54.9
	3	89.1	42.4	51.8
30	$\frac{1}{2}$	87.9	62.0	64.0
	1	86.4	53.8	61.0
	$1\frac{1}{2}$	85.8	49.0	59.1
	3	85.0	43.3	56.2





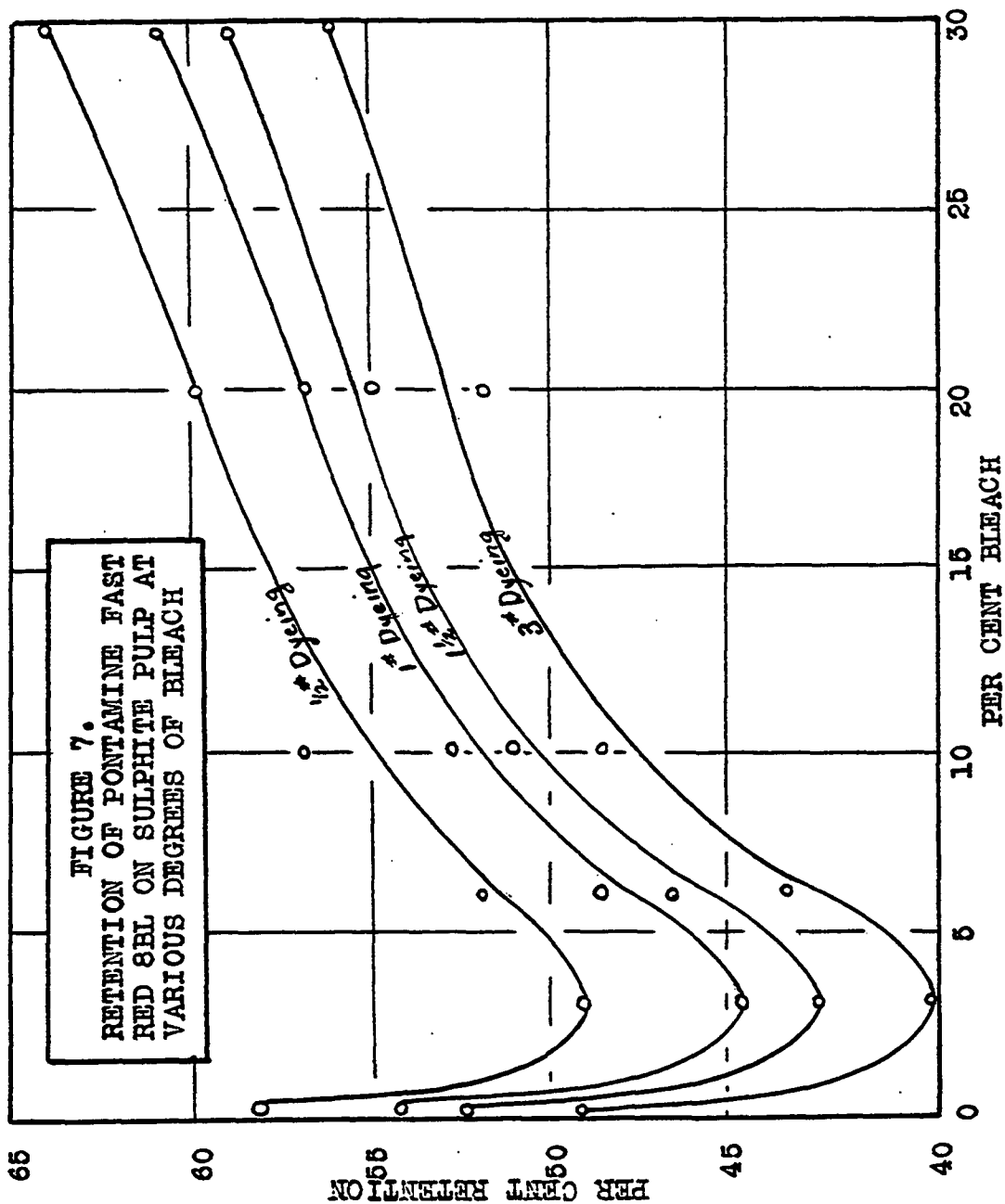


TABLE 10.						
CHEMICAL CONSTANTS OF SULPHITE PULPS						
BLEACHED TO VARIOUS DEGREES						
Per Cent Bleach	0	3	6	10	20	30
Per Cent Lignin	2.45	1.83	0.88	0.42	0.35	0.26
Per Cent Methoxyl	1.34	1.15	0.86	0.62	0.54	0.45
Permanganate Number	17.4	15.7	13.2	11.8	6.3	4.7
Per Cent Alpha Cellulose	82.9	82.7	82.4	81.7	81.0	80.7
Per Cent Total Sulphur	1.96	1.66	1.41	1.10	0.54	0.21
Per Cent Inorganic Sulphur	0.20	0.17	0.12	0.04	0.03	0.02
Per Cent Organic Sulphur	1.76	1.49	1.29	1.06	0.51	0.19

further been accepted that the more lignin present, the better the dye retention. This, however, is not always the case as can be seen from figure 6. (5)

Du Pont Victoria Green SC is retained better on a lignified fiber which has been slightly bleached. It will be noticed that the retention increases from 97.9 per cent for the unbleached sulphite to 98.9 per cent for the same fiber bleached with three per cent bleach. Also it is noticed from table 2 that the lignin content of the slightly bleached

fiber was only 1.83 per cent while the unbleached fiber had a 2.45 per cent lignin content. The most plausible explanation for this result is that the lignin present in the slightly bleached pulp has been chlorinated and, in this condition, has a greater affinity for the basic dye. It must also be taken into account that the unbleached pulp was extremely hard, having a lignin content of 2.45 per cent, a permanganate number of 17.4, and a bleach consumption of 32 per cent. Rys<sup>5</sup> has pointed out that more chlorination takes place during hypochlorite bleaching in hard pulps than in soft pulps. Another point which is evidence that the lignin in the slightly bleached pulp has been chlorinated is its brightness. The slightly bleached pulp has a lower brightness than the unbleached pulp even though its lignin content is lower. The brightness of the unbleached pulp as measured by the General Electric Reflection Meter, filter number 1, is 54.2 per cent while that of the slightly bleached pulp is only 50.8 per cent. As the brightness of the pulp is increased by the use of more bleach, a decided drop in retention is observed. In the light of these facts then, it can be said that basic dyes are attracted by lignin



but in different amounts depending upon the physical and chemical properties of the pulp.

In the case of Du Pont Brilliant Crocein FL and Pontamine Fast Red 8BL, exactly the opposite results are obtained. From figures 7 and 8 it can be seen that, when unbleached sulphite is bleached slightly, a decrease in the retentions are obtained with these dyestuffs. However, with further bleaching, the dyes are retained ever increasingly.

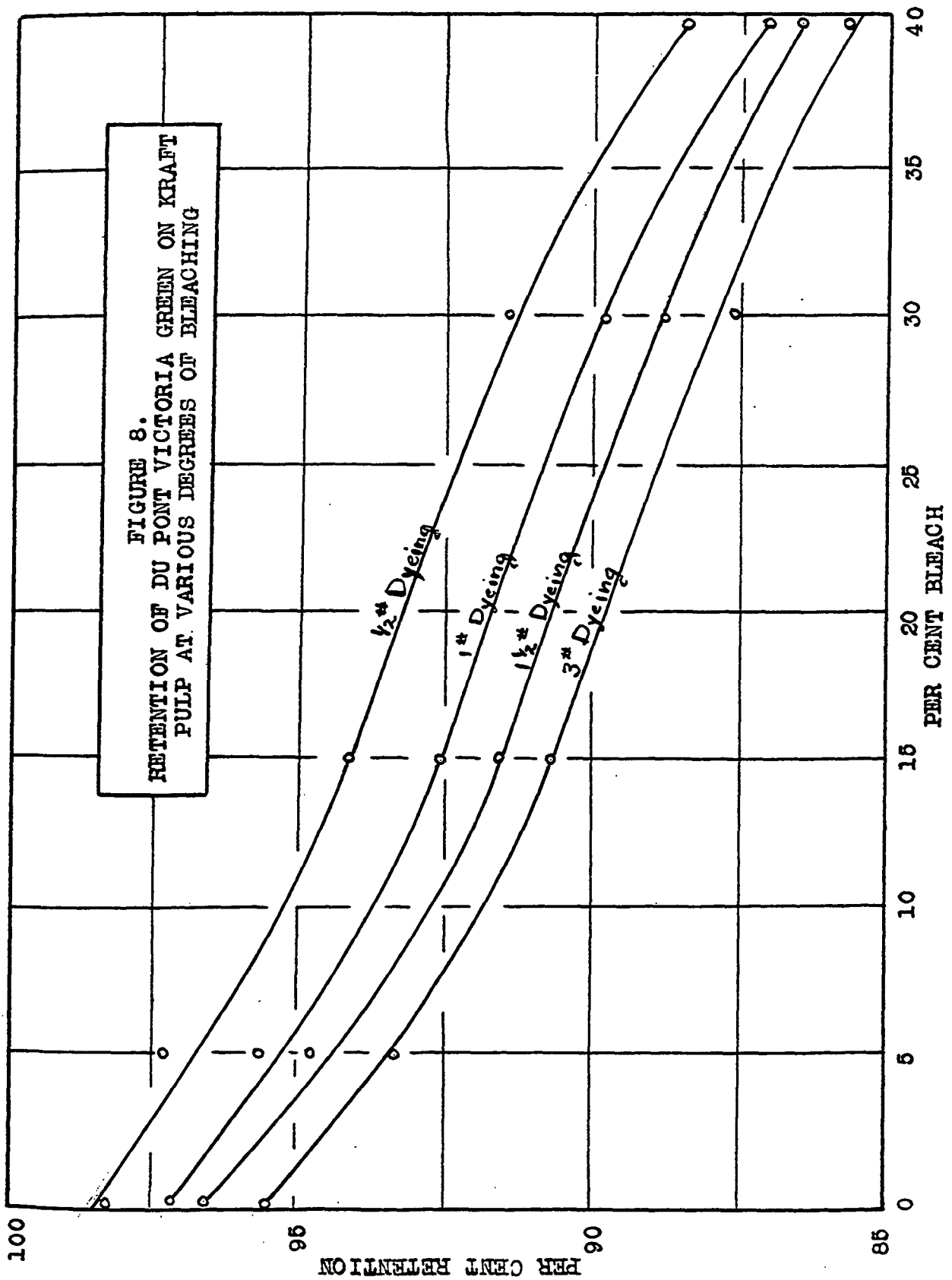
From these results then, it might be concluded that the amount of the lignin, as well as it's character, plays an important role in so far as dye retention on sulphite pulp is concerned.

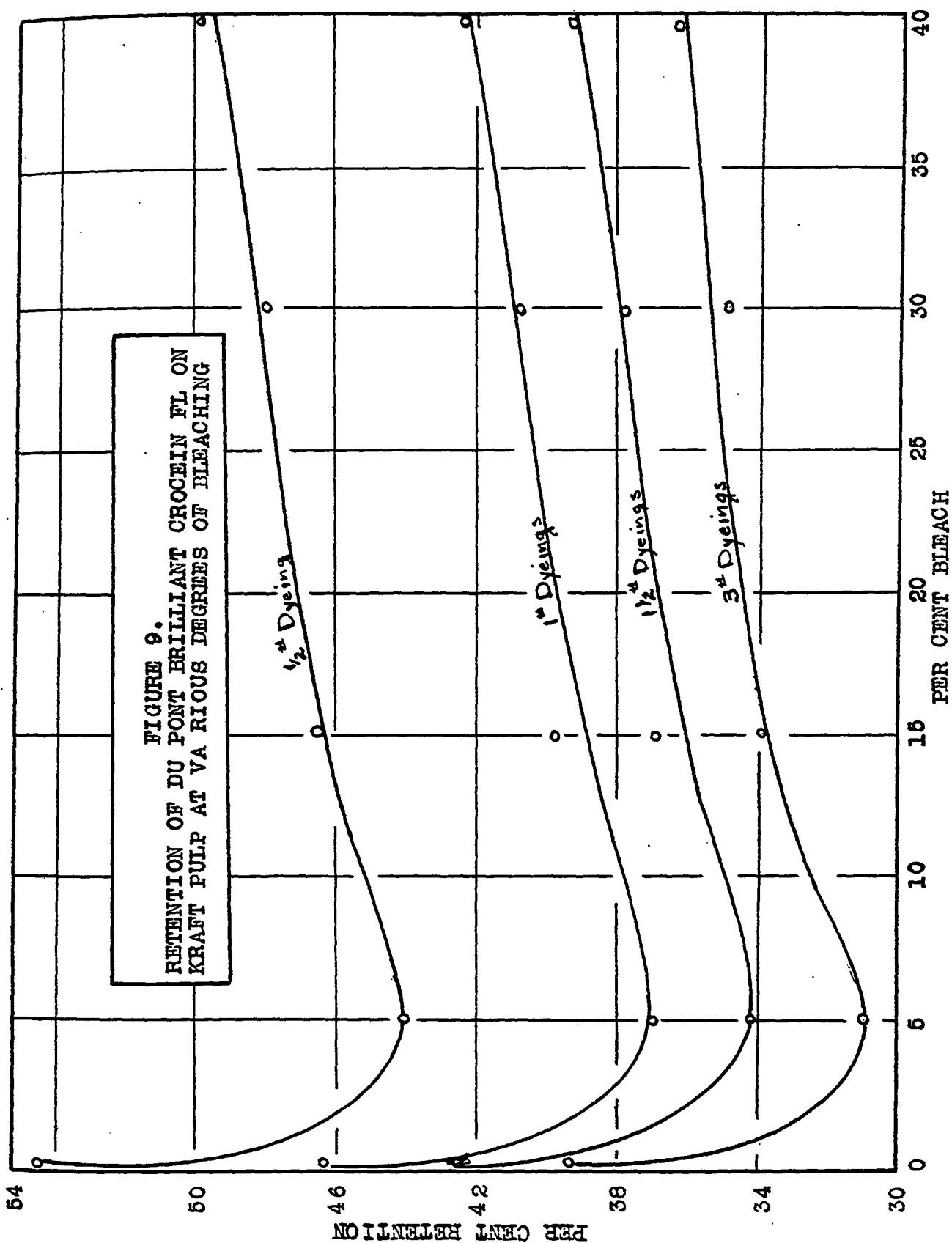
## 2. Retention on Kraft Pulp Bleached to Various Degrees.

In studying the retention of dyes on kraft pulp bleached to various degrees, unbleached kraft pulp was bleached to four different degrees and dyed, in addition to the unbleached pulp, with the three typical dyestuffs in four different strengths ( $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings). The retention data are tabulated in table 11 and are shown graphically in figures 8, 9, and 10. The chemical constants of these pulps are tabulated in table 12.

TABLE 11.				
RETENTION ON KRAFT PULP BLEACHED TO VARIOUS DEGREES				
Per Cent Bleach	Pounds Dye per 1000 Pounds Pulp	Du Pont Victoria Green SC Per Cent	Du Pont Brilliant Crocein FL Per Cent	Pontamine Fast Red 8BL Per Cent
0	$\frac{1}{2}$	98.3	54.5	50.0
	1	97.2	46.3	47.6
	$1\frac{1}{2}$	96.6	42.6	46.6
	3	95.5	39.4	45.0
5 Single Stage	$\frac{1}{2}$	97.3	44.0	54.5
	1	95.7	37.0	51.9
	$1\frac{1}{2}$	94.8	34.2	50.4
	3	93.4	31.1	48.4
15 Double Stage	$\frac{1}{2}$	94.2	46.4	56.8
	1	92.6	39.8	54.2
	$1\frac{1}{2}$	91.6	37.0	52.8
	3	90.7	34.0	50.7
30 Double Stage	$\frac{1}{2}$	91.4	48.0	58.8
	1	89.8	40.8	56.3
	$1\frac{1}{2}$	88.8	37.9	54.9
	3	87.6	34.9	53.0
40 Triple Stage	$\frac{1}{2}$	88.5	49.9	60.0
	1	87.1	42.6	57.4
	$1\frac{1}{2}$	86.5	39.5	56.2
	3	85.7	36.5	54.1

On kraft pulp the basic dye was retained better than any other class of dyes. For one half pound dyeings, the retention of Du Pont Victoria Green SC varied between 88.5 per cent for highly bleached pulp and 98.3 per cent for the unbleached, figure 8. As the lignin decreases in the pulp,





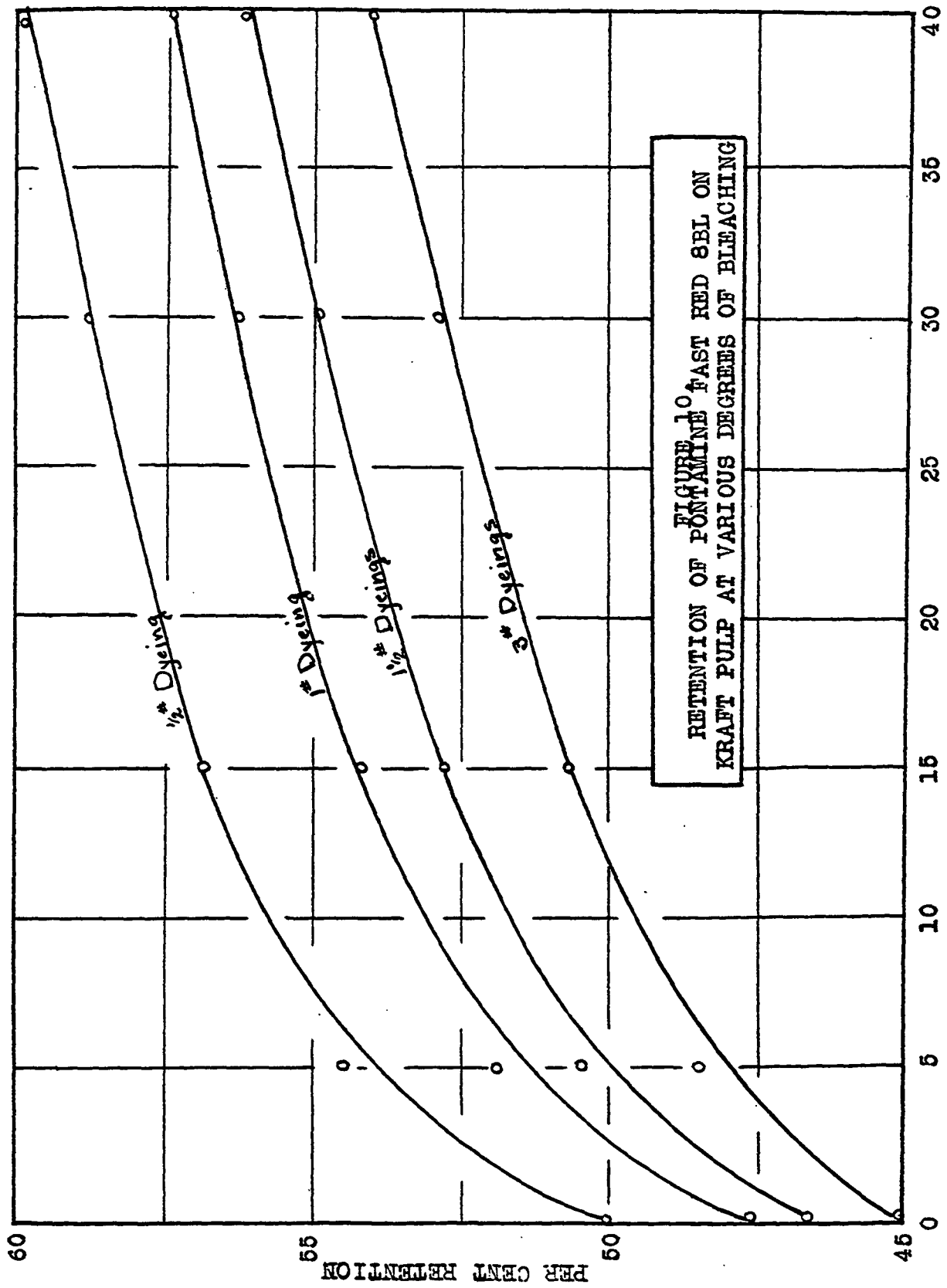


TABLE 12.  
CHEMICAL CONSTANTS OF KRAFT PULPS BLEACHED  
TO VARIOUS DEGREES

Per Cent Bleach	0	5	15	30	40
Per Cent Lignin	8.7	5.7	4.2	2.7	2.0
Per Cent Methoxyl	1.885	1.450	1.120	0.750	0.416
Permanganate Number	26.0	22.4	17.9	8.1	5.0
Per Cent Alpha Cellulose	88.3	87.3	85.1	80.1	78.5
Per Cent Total Sulphur	0.0708	0.0670	0.0595	0.0414	0.0367
Per Cent Inorganic Sulphur	0.0316	0.0300	0.0262	0.0173	0.0149
Per Cent Organic Sulphur	0.0398	0.0370	0.0333	0.0241	0.0218

the per cent retention of basic dyes decreases and in a uniform manner. If the character of the lignin had been changed by bleaching (which was apparently the case with sulphite pulp), slight bleaching of the kraft pulp did not produce a sufficient change in the lignin to cause better retention of the basic dye.

Du Pont Brilliant Crocein FL on kraft pulp bleached to various degrees, however, acted much differently than was expected in view of the

previous results. This dye is retained best on unbleached or highly bleached pulp and is retained the least on slightly bleached pulp as can be seen in figure 9. Sulphite pulp when treated in this same manner acted similarly except that its retention rose more rapidly upon bleaching. In order to explain why unbleached kraft pulp retained more dye than highly bleached pulps several details will have to be taken into consideration. From the experiments carried out on the various sulphite pulps it was concluded that both the character and the amount of lignin present were governing factors in dye retention. In the preceding part of this section it was shown that the character of the lignin in kraft pulp could not have been changed much, if any, because the basic dye used was retained to a greater extent by the lignified fibers. This would tend to show that the character of the lignin was about the same at various degrees of bleaching. In the light of this, then, it would be expected that acid and direct dyes would be retained less on the more lignified fibers. In actual results this was not the case for the acid dye. From the chemical constants tabulated in table 12 no clue could be

found which would lead to an explanation of the results. In addition to the considerations already made, the well known fact must be recalled that basic dyes have the greatest affinity, while acid dyes have the least affinity, for paper making fibers. One plausible explanation of these results is that the character of the lignin in kraft pulp is changed slightly by bleaching. Due to the fact that Du Pont Victoria Green SC is strongly attracted by the pulp, it is retained increasingly with increased lignin content without regard to the slightly changed character of the lignin, and produced results in agreement with those expected. However, in the case of the acid dye, Du Pont Brilliant Crocein FL, which has only a slight affinity for pulps, the lignin was able to attract and hold the dye producing the results that have been shown in figure 9. It should be pointed out, however, that the lignin in kraft pulp was not changed in character nearly so much as it was in the sulphite pulp during bleaching. In the case of sulphite pulp all three of the dyes were affected in an ununiform manner when dyed on slightly bleached pulp.



Pontamine Fast Red 8BL is apparently a strong enough dye to increase the retention upon bleaching. Bleached kraft pulps retain more of this dye than unbleached pulps as is shown in figure 10. For one half pound dyeings, the unbleached pulp retains 50.0 per cent of the direct dye while the bleached pulp retains 60.0 per cent.

From these results then, it may be concluded that the character of the lignin present in the kraft pulp, when only slightly changed by bleaching, has very little effect on the retention of dyes other than that which would be expected by its presence with the exception of acid dyes such as Du Pont Brilliant Crocein FL. The retention of Du Pont Brilliant Crocein FL is decreased in this case because it has only a slight affinity for paper making fibers. Du Pont Victoria Green SC is retained better on kraft pulp than Pontamine Fast Red 8BL, while Du Pont Brilliant Crocein FL is retained the least.

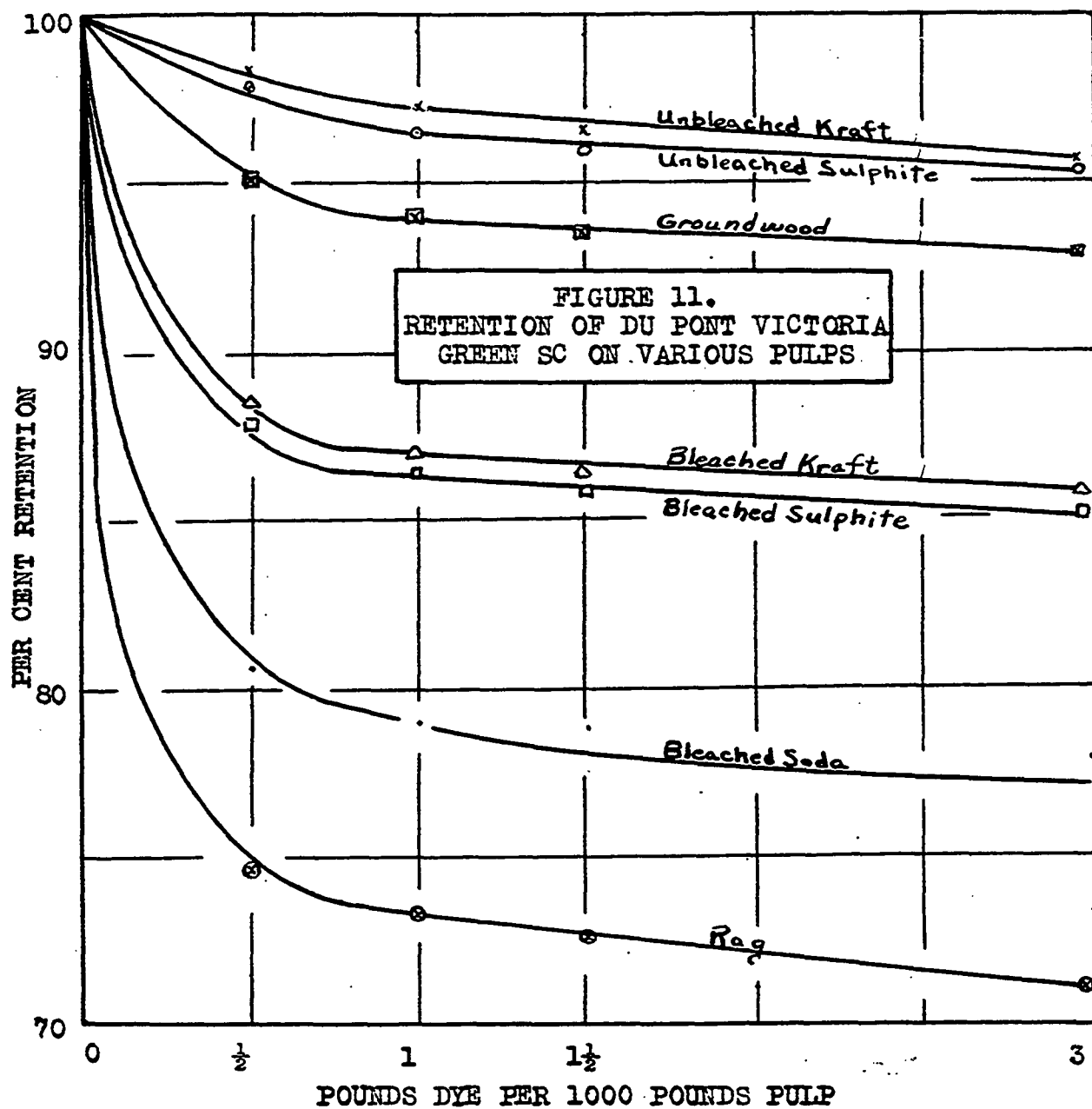
### 3. Retention on Bleached and Unbleached Sulphite and Kraft, Bleached Soda, Groundwood, and Rag Pulps.

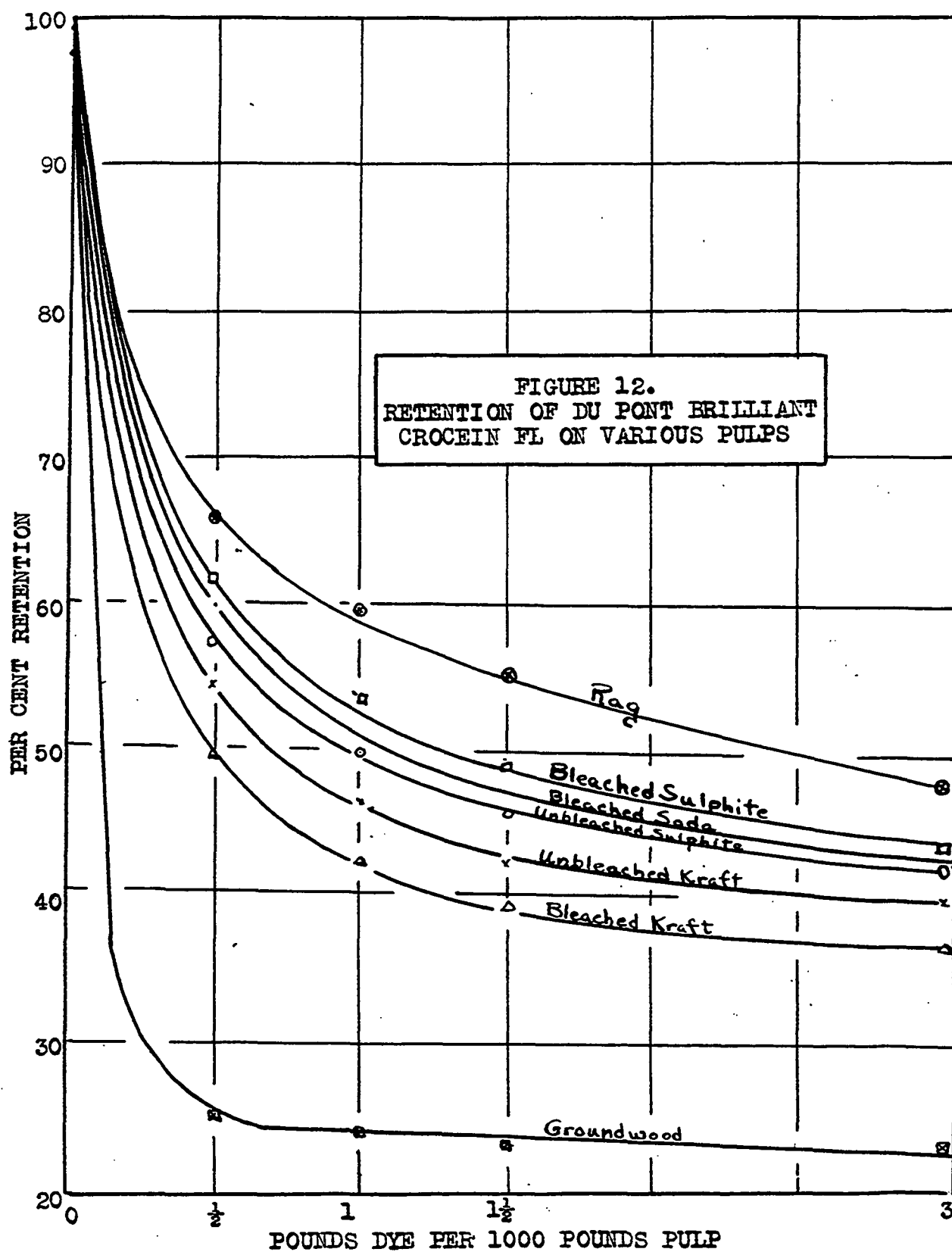
The pulps most generally used in paper making operations, bleached and unbleached sulphite and

kraft, bleached soda, groundwood, and rag, were chosen for the pulp variable in this investigation. The various pulps were dyed with the three typical dyestuffs in four different strengths ( $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings). The retention results are tabulated in table 13 and shown graphically in figures 11, 12, and 13. The chemical constants of the pulps are tabulated in table 14 except for sulphite and kraft which have already been given above in tables 2 and 4 respectively. The retentions of the sulphite and kraft pulps have been given in sections 1 and 2 above, but will be given here again for comparison with the other pulps. The bleached sulphite is the same pulp used in section 1 which was obtained by bleaching unbleached sulphite with thirty per cent bleach while the bleached kraft is the same pulp used in section 2 which was obtained by bleaching unbleached kraft with forty per cent bleach.

Du Pont Victoria Green SC, figure 11, shows that for one half pound dyeings, unbleached kraft retains the dye slightly better than unbleached sulphite, their retentions being 98.3 and 97.9 per cent respectively. Groundwood, unexpectedly, is next with a retention of 95.2 per cent. The

TABLE 13.				
RETENTION ON VARIOUS PULPS				
Pulp	Pounds Dye per 1000 Pounds Pulp	Du Pont Victoria Green SC Per Cent	Du Pont Brilliant Crocein FL Per Cent	Pontamine Fast Red 8BL Per Cent
Unbleached Sulphite	$\frac{1}{2}$	97.9	57.7	58.0
	1	96.5	50.0	54.0
	$1\frac{1}{2}$	96.0	45.7	52.3
	3	95.3	41.8	49.1
Bleached Sulphite	$\frac{1}{2}$	87.9	62.0	64.0
	1	86.4	53.8	61.0
	$1\frac{1}{2}$	85.8	49.0	59.1
	3	85.0	43.3	56.2
Unbleached Kraft	$\frac{1}{2}$	98.3	54.5	50.0
	1	97.2	46.3	47.6
	$1\frac{1}{2}$	96.6	42.6	46.6
	3	95.5	39.4	45.0
Bleached Kraft	$\frac{1}{2}$	88.5	49.9	60.0
	1	87.1	42.6	57.4
	$1\frac{1}{2}$	86.5	39.5	56.2
	3	85.7	36.5	54.1
Bleached Soda	$\frac{1}{2}$	80.7	60.0	55.0
	1	79.0	54.0	52.0
	$1\frac{1}{2}$	78.9	49.5	50.1
	3	78.0	42.0	45.5
Rag	$\frac{1}{2}$	74.7	66.0	63.8
	1	73.3	59.5	60.0
	$1\frac{1}{2}$	72.6	55.0	58.0
	3	71.0	47.2	55.3
Ground- wood	$\frac{1}{2}$	95.2	25.0	33.0
	1	94.0	23.8	30.0
	$1\frac{1}{2}$	93.6	23.0	28.6
	3	92.8	22.9	25.6





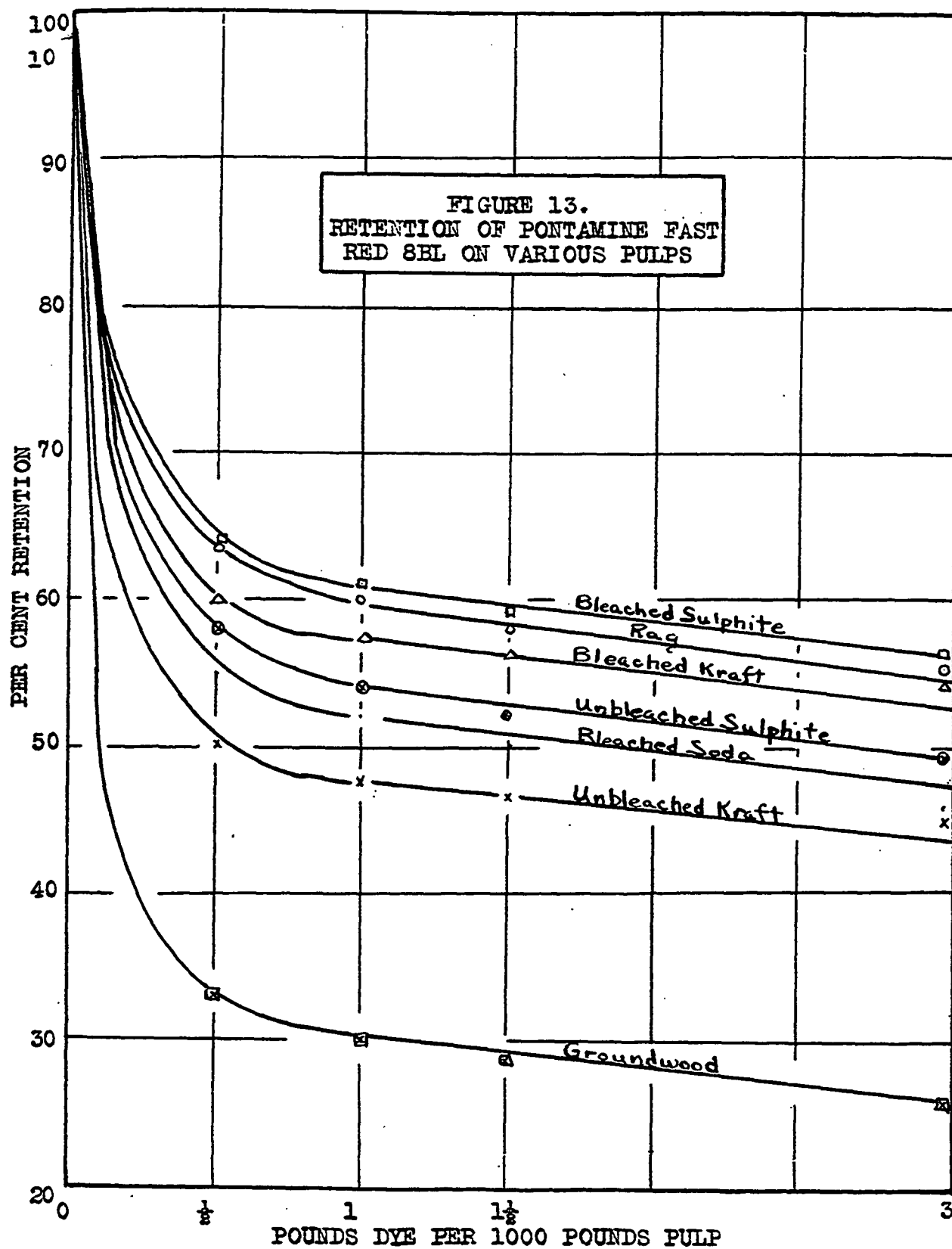


TABLE 14. CHEMICAL CONSTANTS OF BLEACHED SODA, RAG, AND GROUNDWOOD			
Pulp	Bleached Soda	Rag	Groundwood
Per Cent Lignin	0.4	0	27.1
Per Cent Methoxyl	0.085	0.012	5.420
Per Cent Alpha Cellulose	63.05	97.03	-

lignin content of groundwood is 27.1 per cent but can not be considered in a relative way with the other lignified pulps because of its entirely different character. Bleached kraft and sulphite are next having retentions respectively of 88.5 and 87.9 per cent. Bleached soda and rag are last with retentions of 80.7 and 74.7 per cent respectively.

From these results it is evident that the various types of pulps have widely varying affinities for Du Pont Victoria Green SC, but, generally speaking, the amount and character of lignin present has a considerable influence in holding this dye to the fiber. In the case of groundwood, however, which is a mechanical pulp and not comparable with chemical pulps, it should

be noticed that it is well up in the list in its affinity for basic dyes.

Du Pont Brilliant Crocein FL, figure 12, shows that for one half pound dyeings, for example, rag pulp has the best retention which is 66.0 per cent. Bleached sulphite and soda are next best with retentions of 57.7 and 54.5 per cent respectively. Then comes bleached kraft with a retention of 49.9 per cent. Here again we have this same acid dye on bleached kraft pulp which shows an unexpected result. As has already been explained in section 2, the lignin in this particular case has been changed in some way, even though it is present only to the extent of 2.04 per cent, so that it repels the acid dye. Then the highly lignified mechanical pulp was last with the extremely low retention of 25.0 per cent.

From these results it can be concluded that acid dyes show only a small attraction for these pulps, the best of which are those containing the highest amounts of alpha cellulose and the least lignin.

In the case of Pontamine Fast Red 8BL, figure 13, the one half pound dyeings were also retained better than those using greater quantities of dye.



Bleached sulphite, rag, and bleached kraft pulps retain this dye better than the other pulps.

Bleached sulphite has a retention of 64.0, rag 63.6, and bleached kraft 60.0 per cent. Unbleached sulphite, bleached soda, and unbleached kraft are next in line with retentions of 58.0, 55.0, and 50.0 per cent respectively. The bleached soda was expected to be higher in the list but when its alpha cellulose content of only 63.0 per cent is taken into consideration it can be justified. Groundwood with this dye is also last with only 33.0 per cent retention.

These results indicate that the retention of this dye, even though better than that of Du Pont Brilliant Crocein FL, is much less than the retentions obtained with Du Pont Victoria Green SC. In addition, highly purified pulps show dye retentions which are considerably better than those of the more lignified pulps.

#### 4. Retention on Unbleached and Bleached Sulphite and Kraft at Various pH's.

Sulphite and kraft pulps are the most important types of furnishes from a tonnage standpoint that are dyed so they alone were considered regarding the effect of pH on dye retention.

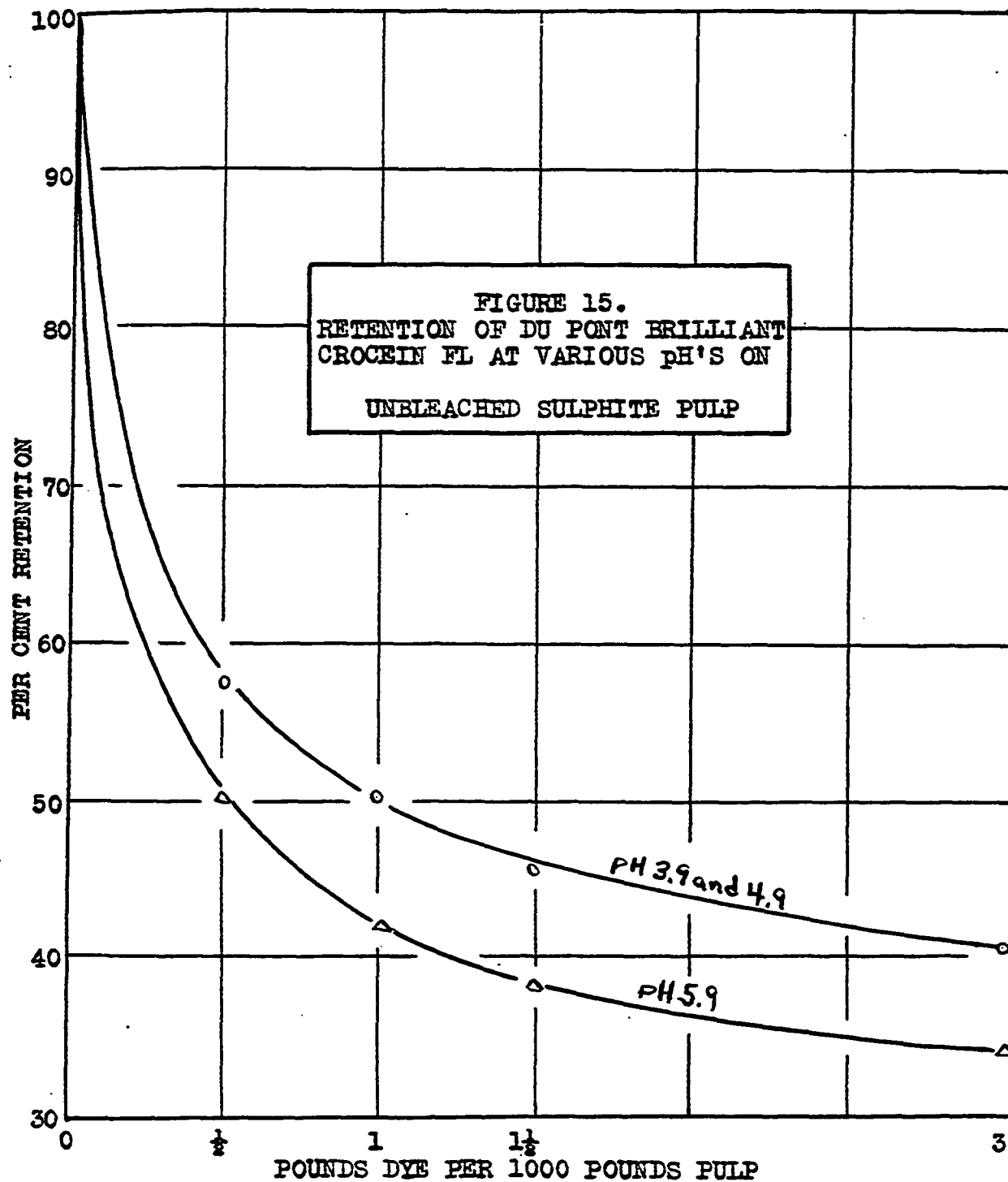
These pulps, both bleached and unbleached, were dyed at three pH's, 3.9, 4.9, and 5.9, with the three typical dyes in four different strengths ( $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings). The results are tabulated in table 15 for the sulphite and in table 16 for the kraft pulp. These same results are plotted graphically in figures 14 to 19 inclusive for the sulphite and 20 to 25 inclusive for the kraft pulp.

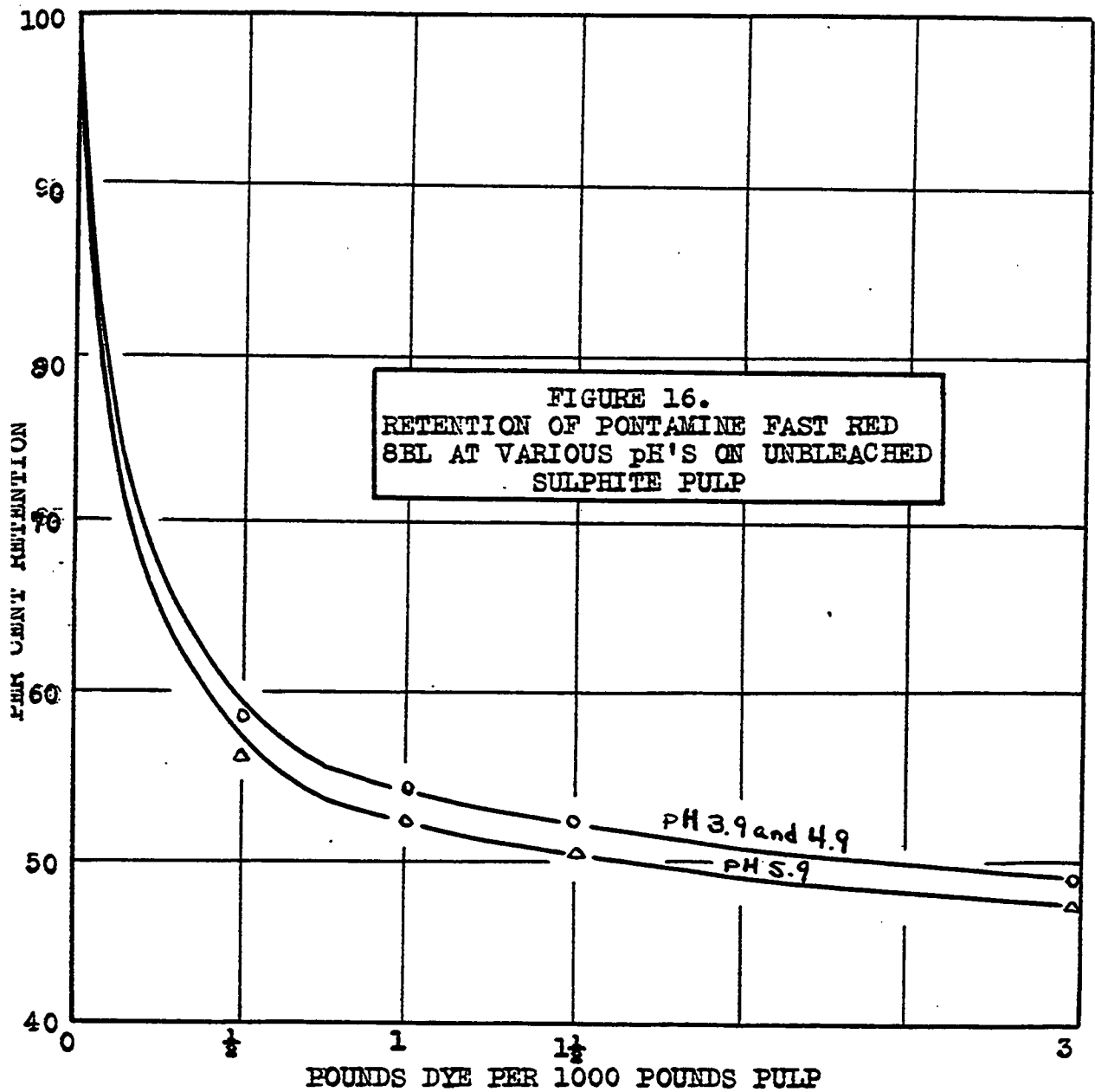
Du Pont Victoria Green SC on unbleached sulphite, figure 14, showed retentions, for example, for one half pound dyeings of 96.0, 97.9, and 97.7 per cent for 3.9, 4.9, and 5.9 pH's respectively. From these data it is apparent that the optimum pH to use with Du Pont Victoria Green SC in obtaining a maximum dye retention on unbleached sulphite pulp is somewhere near 4.9. The effect of higher pH's lowers the retention but not to any where near the extent that lower pH's do.

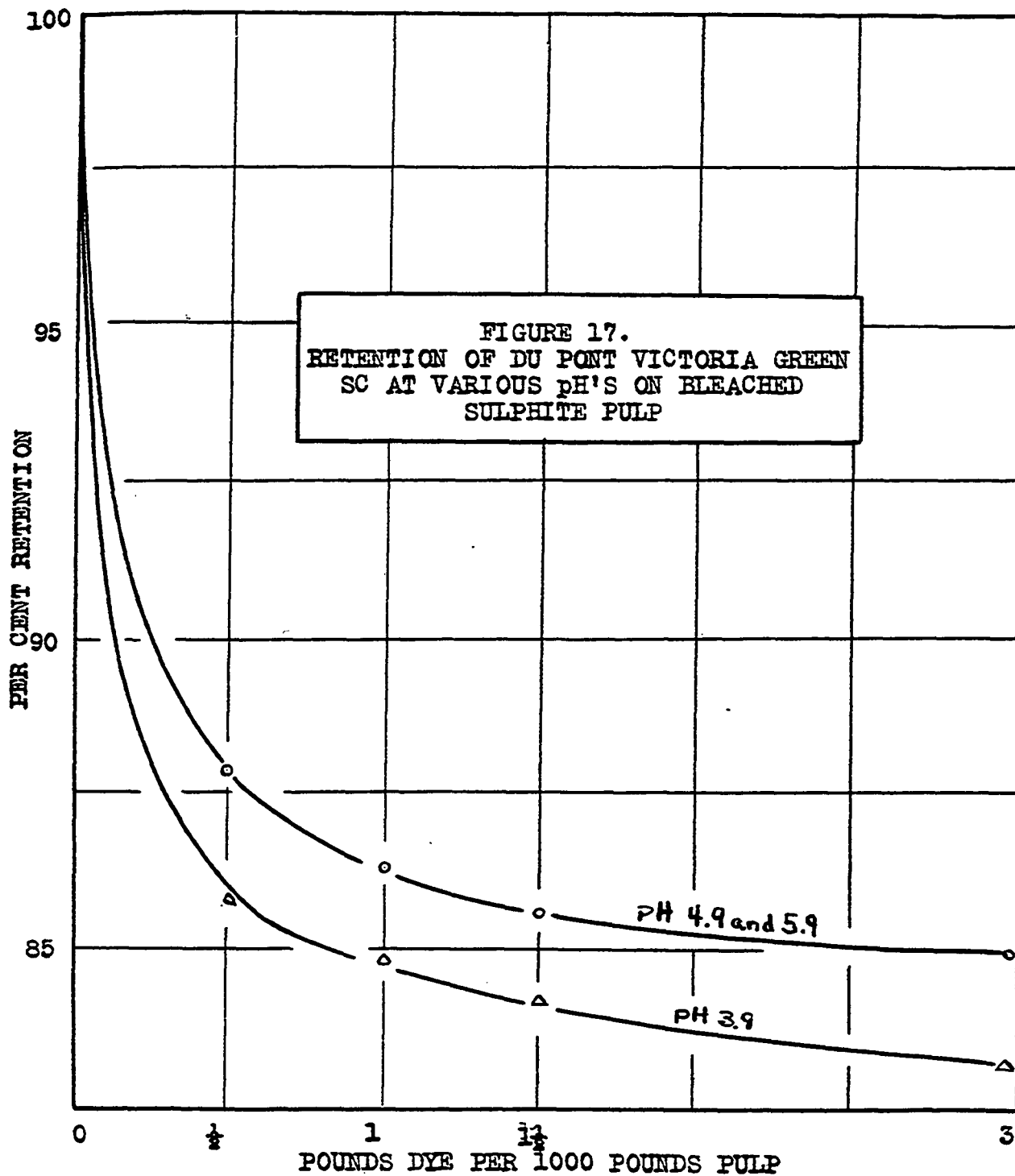
Du Pont Brilliant Crocein FL on unbleached sulphite, figure 15, showed no differences in retention for pH's of 3.9 and 4.9. Above that point, 5.9 for instance, the retention showed a large drop. For one half pound dyeings the retentions 57.7 per cent for pH's of 3.9 and 4.9 and 50.1

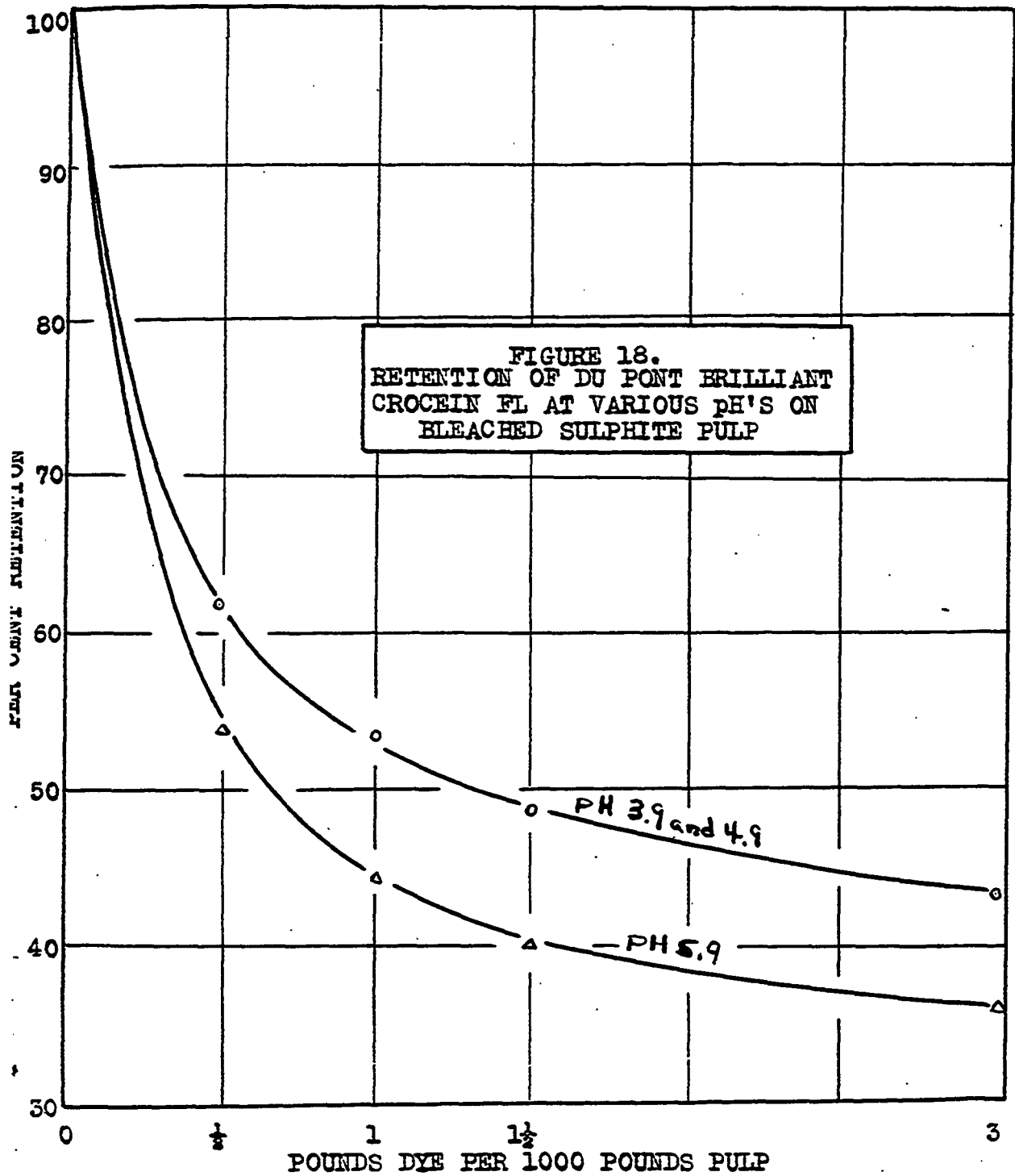
TABLE 15.				
RETENTION OF DYES ON UNBLEACHED AND BLEACHED				
SULPHITE PULP AT VARIOUS pH'S				
Hydrogen Ion Concentration and Pulp	Pounds Dye per 1000 Pounds Pulp	Du Pont Victoria Green SC Per Cent	Du Pont Brilliant Crocein FL Per Cent	Pontamine Fast Red 8BL Per Cent
Unbleached Sulphite 3.9	$\frac{1}{2}$	96.0	57.7	58.2
	1	94.5	50.1	54.1
	$1\frac{1}{2}$	94.1	45.6	52.2
	3	93.8	41.9	49.1
Unbleached Sulphite 4.9	$\frac{1}{2}$	97.9	57.7	58.0
	1	96.5	50.0	54.0
	$1\frac{1}{2}$	96.0	45.7	52.2
	3	95.3	41.8	49.1
Unbleached Sulphite 5.9	$\frac{1}{2}$	97.7	50.1	56.1
	1	96.2	42.2	52.4
	$1\frac{1}{2}$	95.9	38.1	50.4
	3	94.9	34.1	47.3
Bleached Sulphite 3.9	$\frac{1}{2}$	85.8	62.0	64.4
	1	84.8	53.5	61.5
	$1\frac{1}{2}$	84.2	48.9	59.4
	3	83.2	43.1	56.8
Bleached Sulphite 4.9	$\frac{1}{2}$	87.9	62.0	64.0
	1	86.4	53.8	61.0
	$1\frac{1}{2}$	85.8	49.0	59.1
	3	85.0	43.3	56.2
Bleached Sulphite 5.9	$\frac{1}{2}$	87.8	54.0	62.1
	1	86.3	44.7	58.9
	$1\frac{1}{2}$	85.6	40.5	57.3
	3	85.1	36.1	54.3

TABLE 16.				
RETENTION OF DYES ON UNBLEACHED AND BLEACHED				
KRAFT PULP AT VARIOUS pH'S				
Hydrogen Ion Concentration and Pulp	Pounds Dye per 1000 Pounds Pulp	Du Pont Victoria Green Per Cent	Du Pont Brilliant Crocein FL Per Cent	Pontamine Fast Red 8BL Per Cent
Unbleached	$\frac{1}{2}$	97.5	54.8	50.1
Kraft	1	96.5	46.6	47.6
3.9	$1\frac{1}{2}$	96.0	42.9	46.8
	3	94.9	40.0	45.0
Unbleached	$\frac{1}{2}$	98.3	54.5	50.0
Kraft	1	97.2	46.3	47.6
4.9	$1\frac{1}{2}$	96.6	42.6	46.6
	3	95.5	39.4	45.0
Unbleached	$\frac{1}{2}$	98.2	50.0	48.3
Kraft	1	97.1	42.6	45.9
5.9	$1\frac{1}{2}$	96.4	38.1	44.3
	3	95.4	35.1	43.1
Bleached	$\frac{1}{2}$	88.2	50.2	60.1
Kraft	1	87.1	43.0	57.8
3.9	$1\frac{1}{2}$	86.4	40.0	56.8
	3	85.4	36.9	54.7
Bleached	$\frac{1}{2}$	88.5	49.9	60.0
Kraft	1	87.1	42.6	57.4
4.9	$1\frac{1}{2}$	86.5	39.5	56.2
	3	85.7	36.5	54.1
Bleached	$\frac{1}{2}$	88.5	44.6	57.8
Kraft	1	87.0	37.8	55.0
5.9	$1\frac{1}{2}$	86.5	34.4	53.9
	3	85.5	32.1	51.9

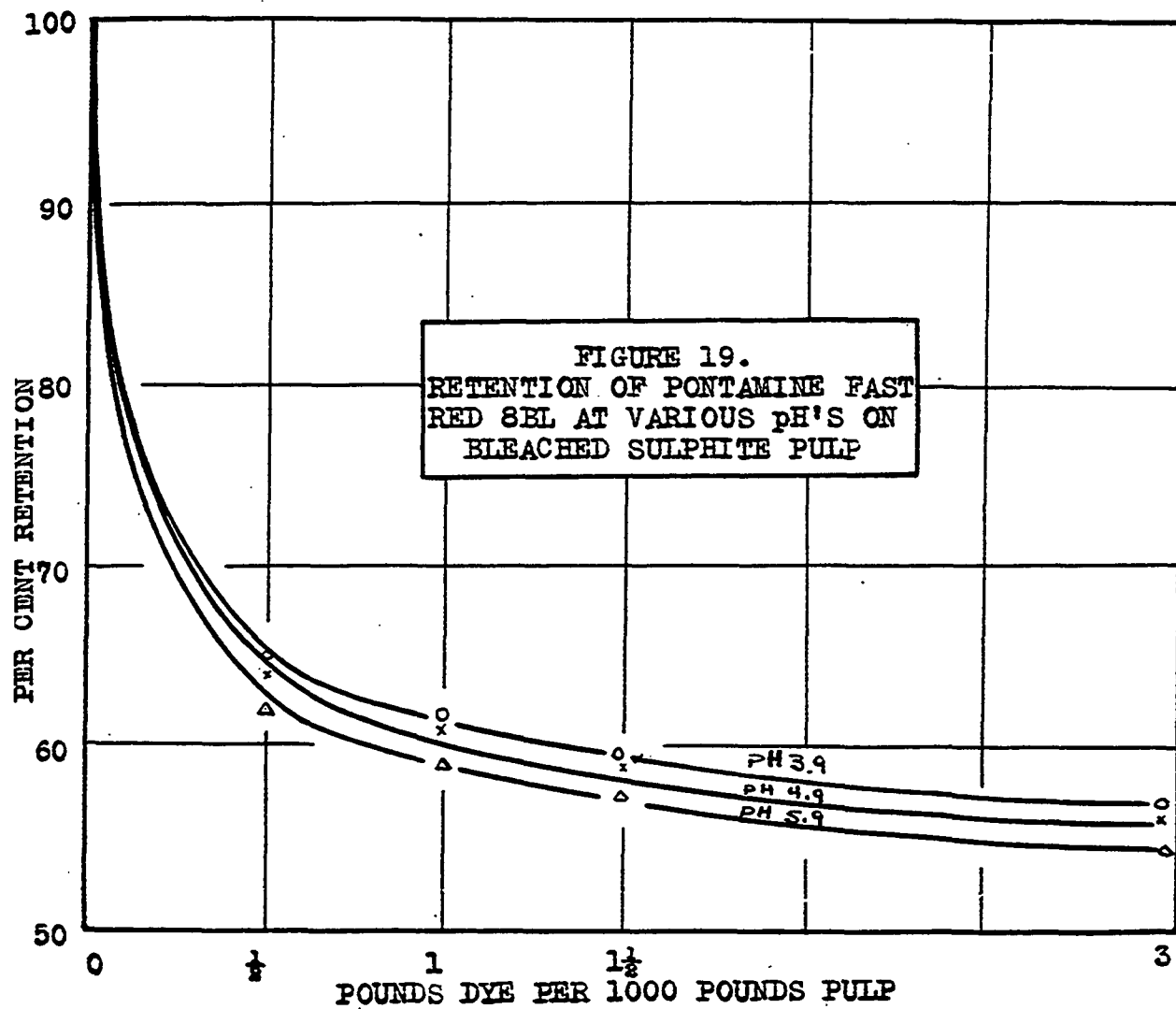


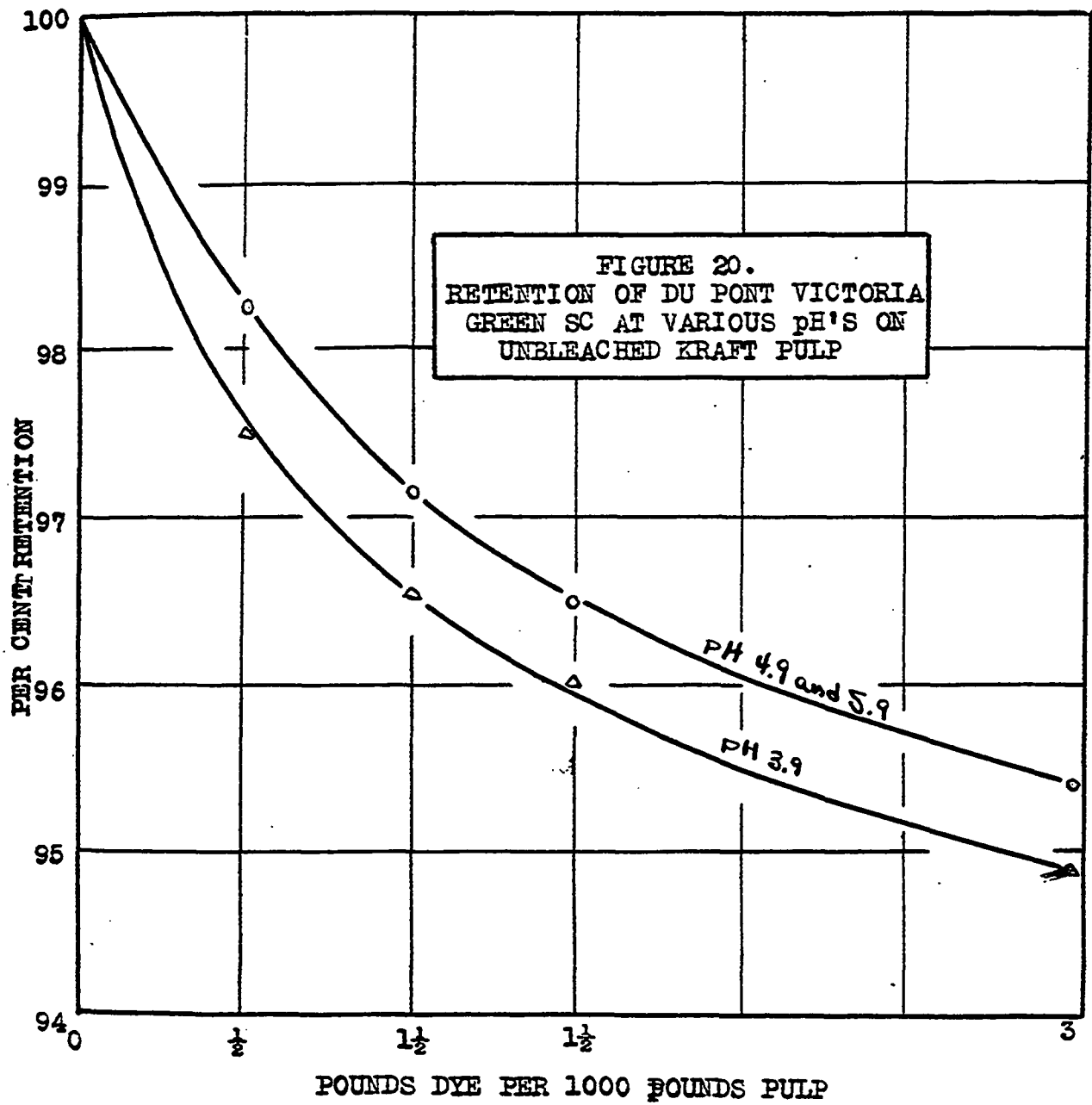


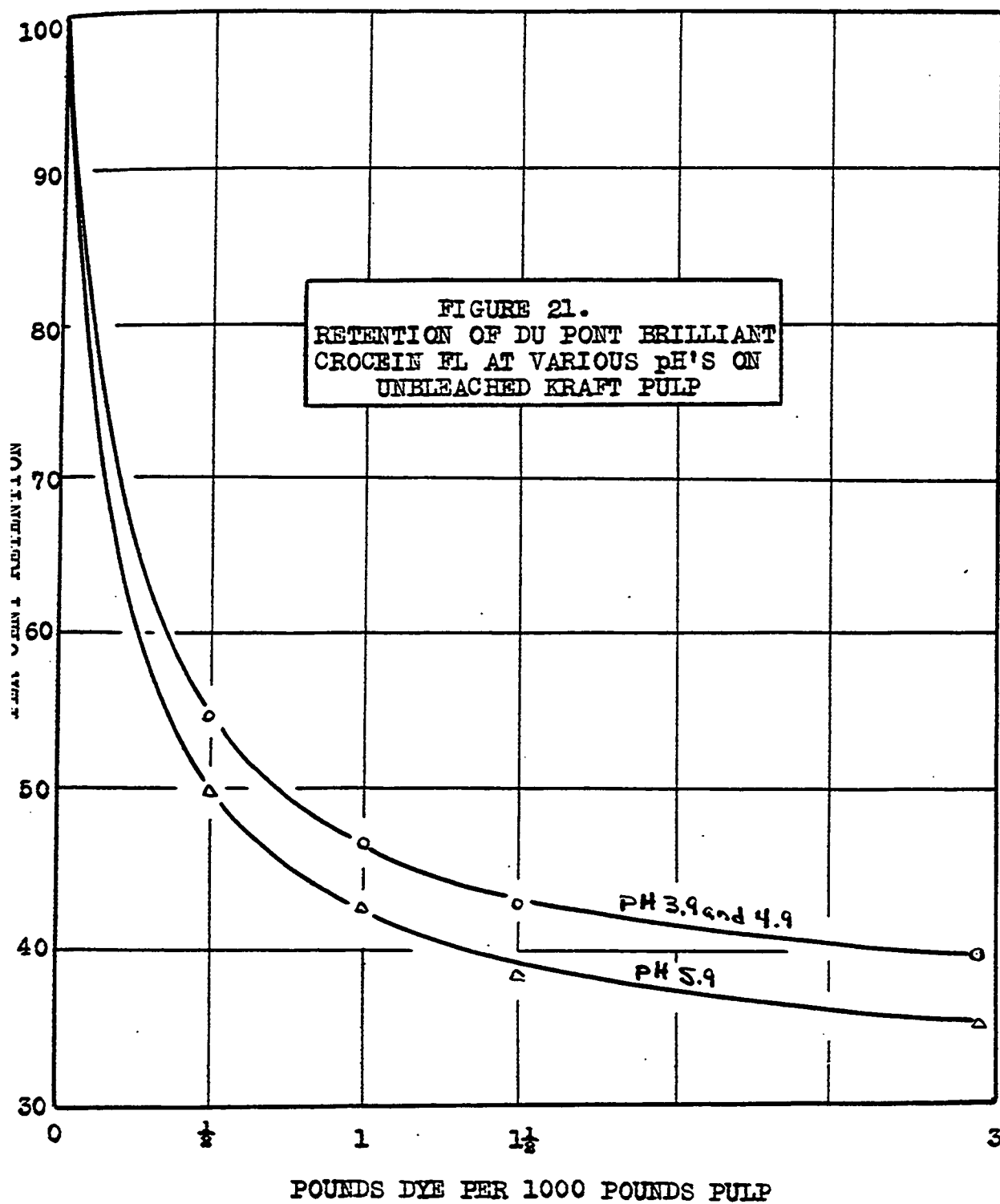


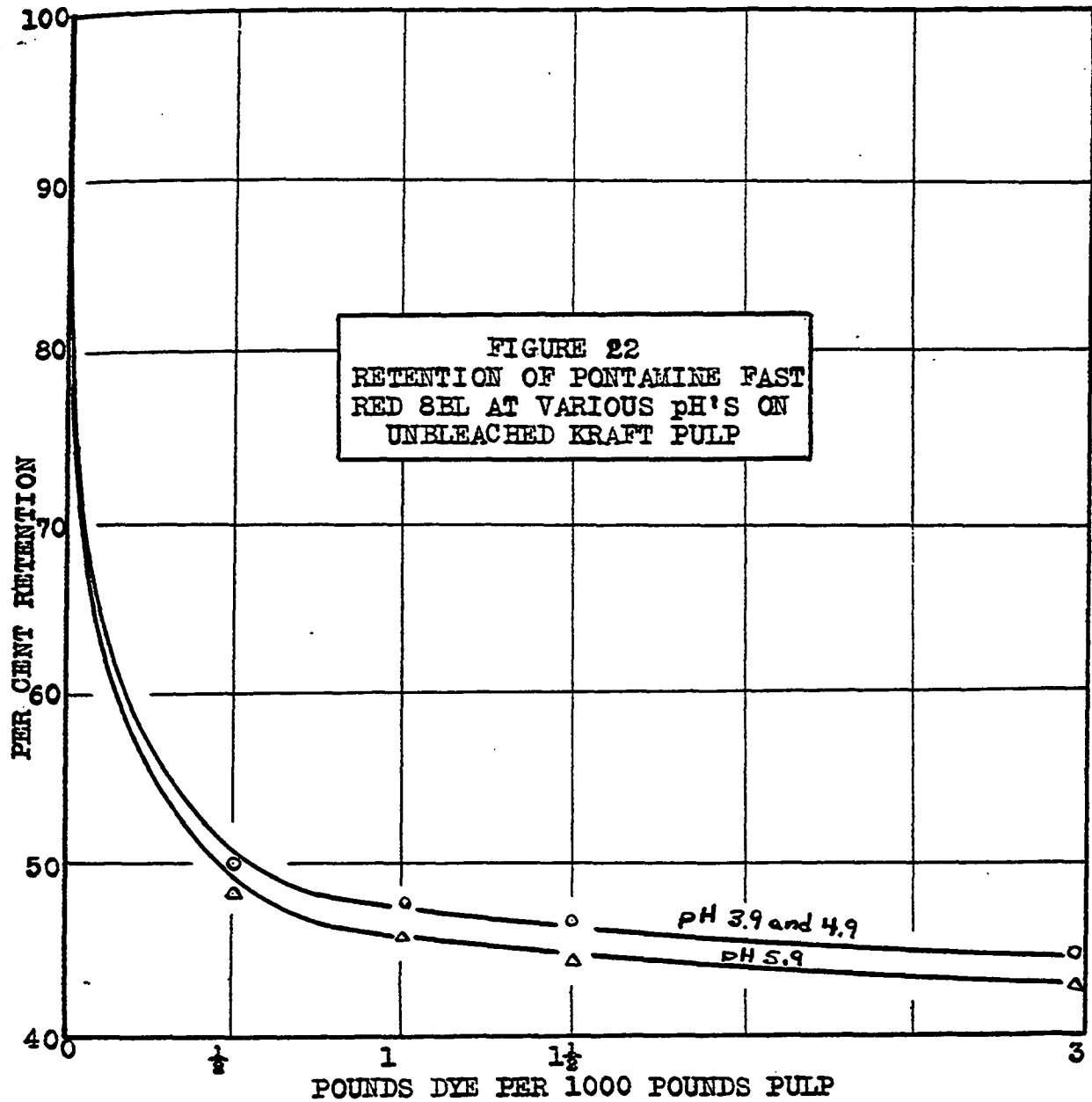


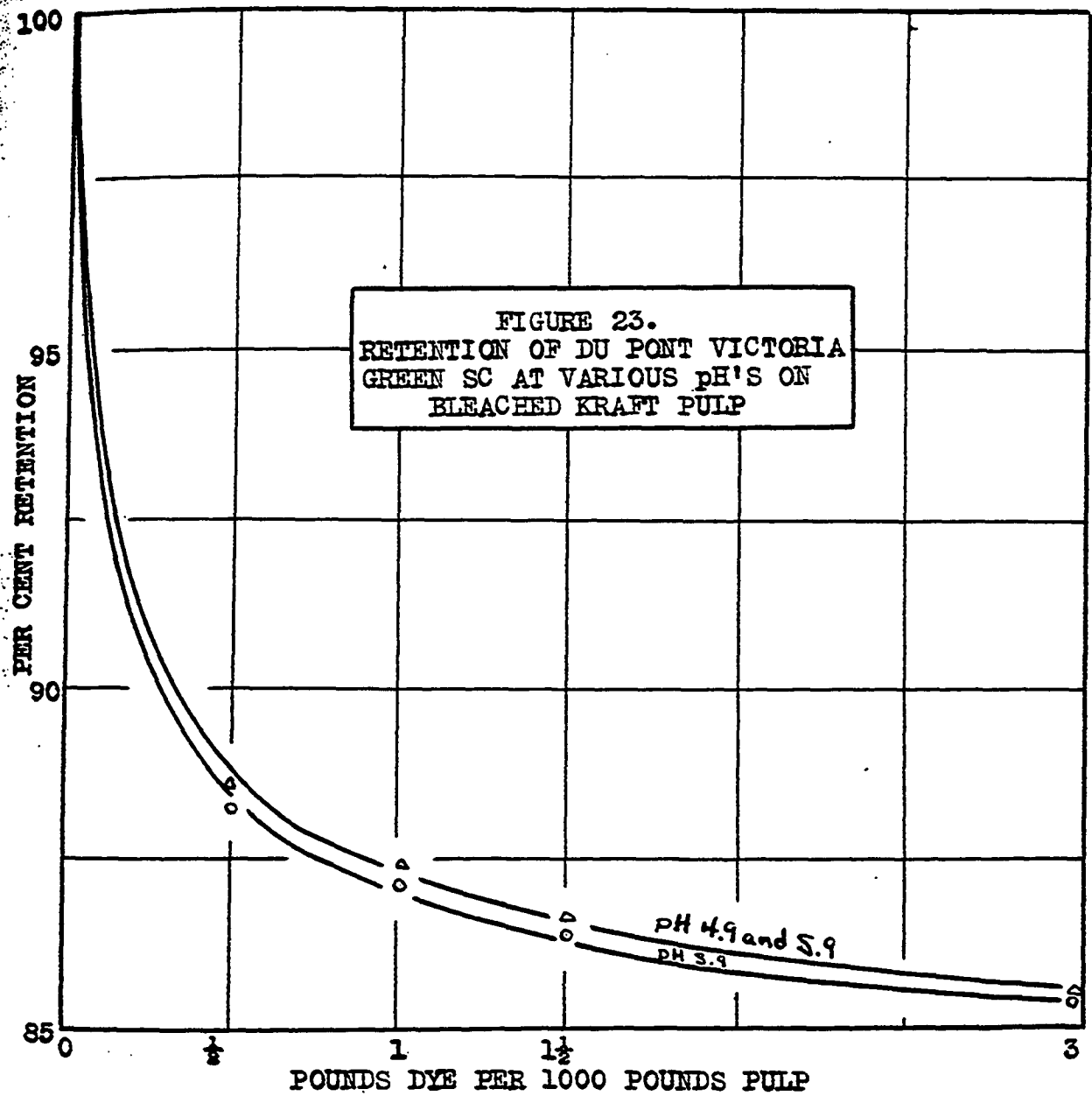


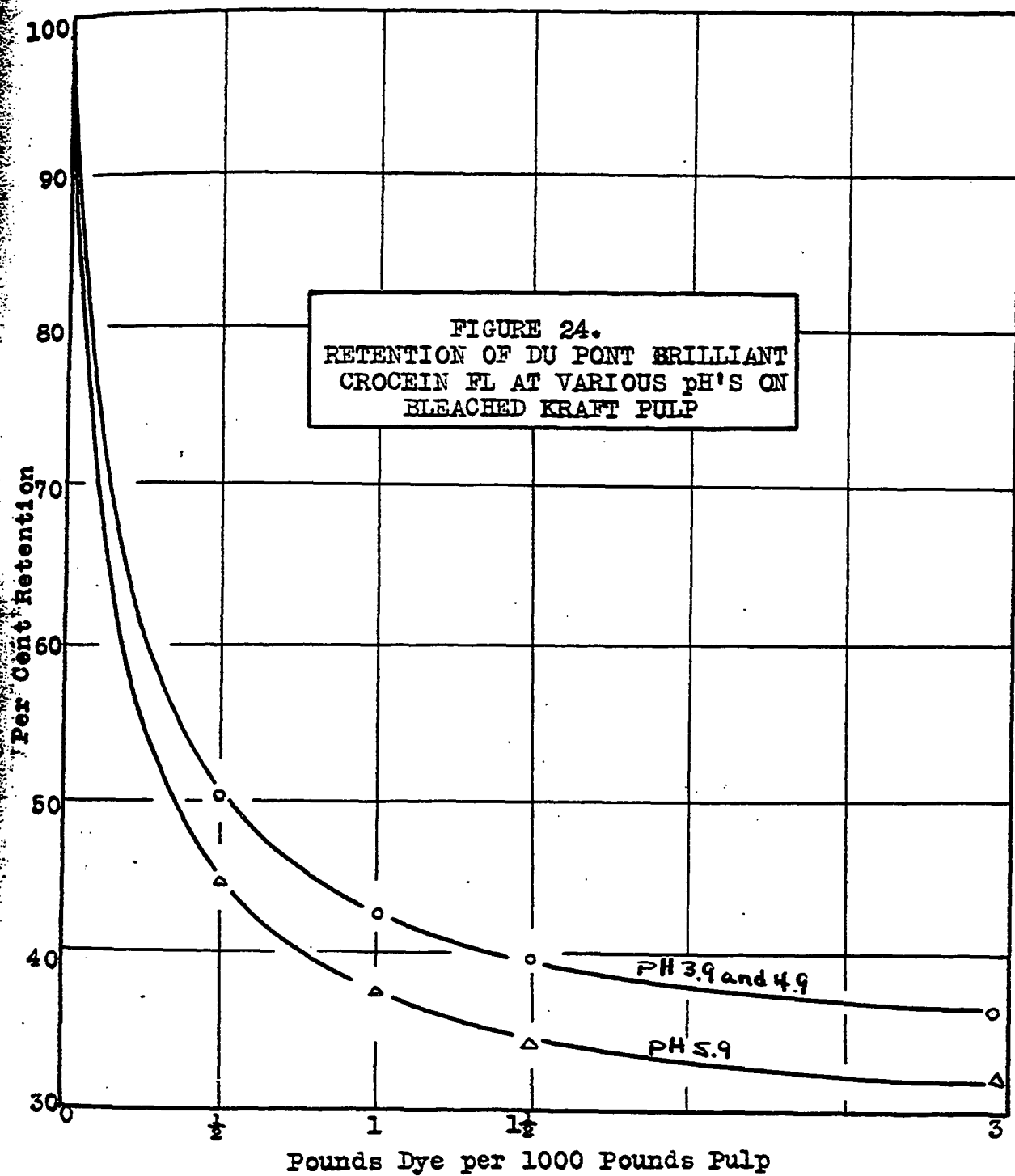


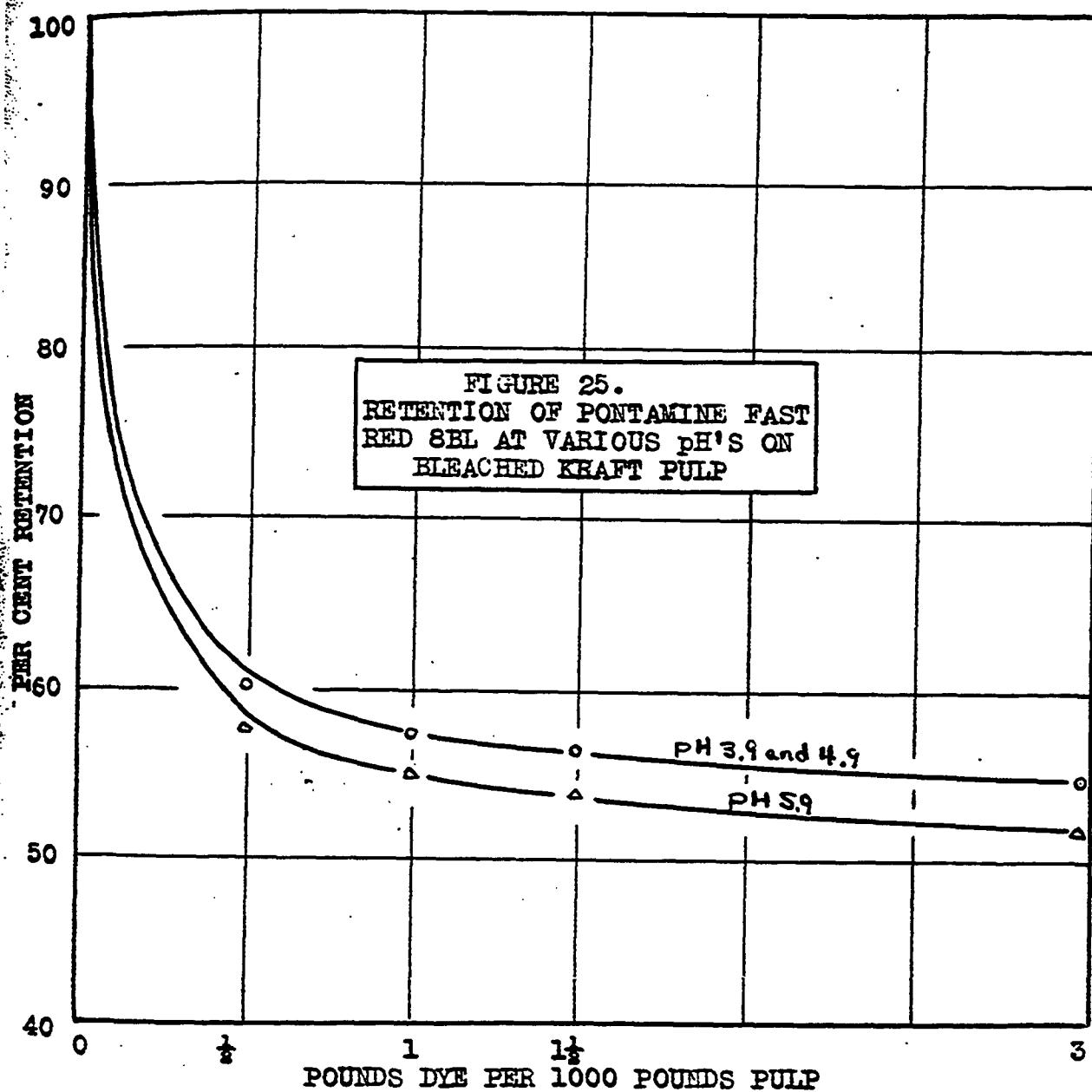












per cent for a pH of 5.9. It is apparent from these data that acid dyes require sufficient acid to set their color groups. From a pH of about 5.0 on down, however, more acid does not help.

Pontamine Fast Red 8BL on unbleached sulphite, figure 16, has approximately the same pH characteristics as Du Pont Brilliant Crocein FL on unbleached sulphite. No difference in retentions were shown for pH's of 3.9 and 4.9. At 5.9, however, the retention dropped a bit, but not nearly to the extent that it did with the acid dye. For one half pound dyeings, for example, the retentions were 58.2, 58.0, and 56.1 per cent for the pH's of 3.9, 4.9, and 5.9 respectively.

Du Pont Victoria Green SC, figure 17, showed a considerable drop in retention on bleached sulphite in relation to this same dye on unbleached sulphite. Retentions for pH's of 4.9 and 5.9 were for all practical purposes the same, but those for the pH of 3.9 were lower. For one half pound dyeings the retentions were 85.8, 87.9, and 87.8 per cent for pH's of 3.9, 4.9, and 5.9 respectively.

Du Pont Brilliant Crocein FL on bleached sulphite pulp, figure 18, showed the same general type of curves as for the the same dye on unbleached sulphite. The pH's of 3.9 and 4.9 showed the same retention while a pH of 5.9 showed a considerable



lowering in the retention. Retentions obtained on the unbleached stock were somewhat lower than those obtained on the bleached. For one half pound dyeings the retentions were 62.0 per cent for pH's of 3.9 and 4.9 and 54.0 per cent for a pH of 5.9.

Pontamine Fast Red 8BL on bleached sulphite pulp, figure 19, has a slightly higher retention for a pH of 3.9 than for a pH of 4.9. At a pH of 5.9 the retention is a bit lower. For example, for one half pound dyeings, the retentions are 64.4, 64.0, and 62.1 per cent for pH's of 3.9, 4.9, and 5.9 respectively. From these results it is apparent that the pH has very little effect on the retention of this dye.

Du Pont Victoria Green SC on unbleached kraft, figure 20, shows that practically the same retentions are obtained with pH's of 4.9 and 5.9. A lower pH, 3.9, produces lower retentions. For one half pound dyeings the retentions are 97.5, 98.3, and 98.2 per cent for pH's of 3.9, 4.9, and 5.9 respectively.

Du Pont Brilliant Crocein FL on unbleached kraft, figure 21, shows that for pH's of 3.9 and 4.9 the retentions are practically the same being 54.8 and 54.5 per cent respectively. At the pH

of 5.9 the retention is less, being 50.0 per cent.

Pontamine Fast Red 8BL on unbleached kraft, figure 22, shows practically no difference for pH's of 3.9 and 4.9 with only a small difference for a pH of 5.9. For example, with one half pound dyeings, the retentions are 50.1, 50.0 and 48.3 per cent for pH's of 3.9, 4.9, and 5.9 and in that order. The retention drop at 5.9 with this dye did not amount to any where near the drop experienced with the acid dye. Apparently, direct dyes are in general the least sensitive to pH.

Du Pont Victoria Green SC on bleached kraft, figure 23, showed a considerable drop in retention in relation to this dye on unbleached kraft. Retentions for pH's of 4.9 and 5.9 were for all practical purposes the same, with those for the pH of 3.9 being a bit lower. For one half pound dyeings the retentions were 88.2 per cent for the 3.9 pH and 88.5 per cent for the 4.9 and 5.9 pH's.

Du Pont Brilliant Crocein FL on bleached kraft, figure 24, showed the same type of curves on unbleached kraft. The dye at pH's of 3.9 and 4.9 was retained for all practical purposes in the same amounts for the same weight dyeings while the retention at a pH of 5.9 showed a considerable

lowering. For one half pound dyeings the retentions were 50.2, 49.9, and 44.6 per cent for pH's of 3.9, 4.9, and 5.9 respectively.

Pontamine Fast red 8BL on bleached kraft, figure 25, has the same retention for pH's of 3.9 and 4.9. At a pH of 5.9 the retention is a bit lower. For one half pound dyeings the retentions are 60.1, 60.0, and 57.8 per cent for pH's of 3.9, 4.9, and 5.9 respectively.

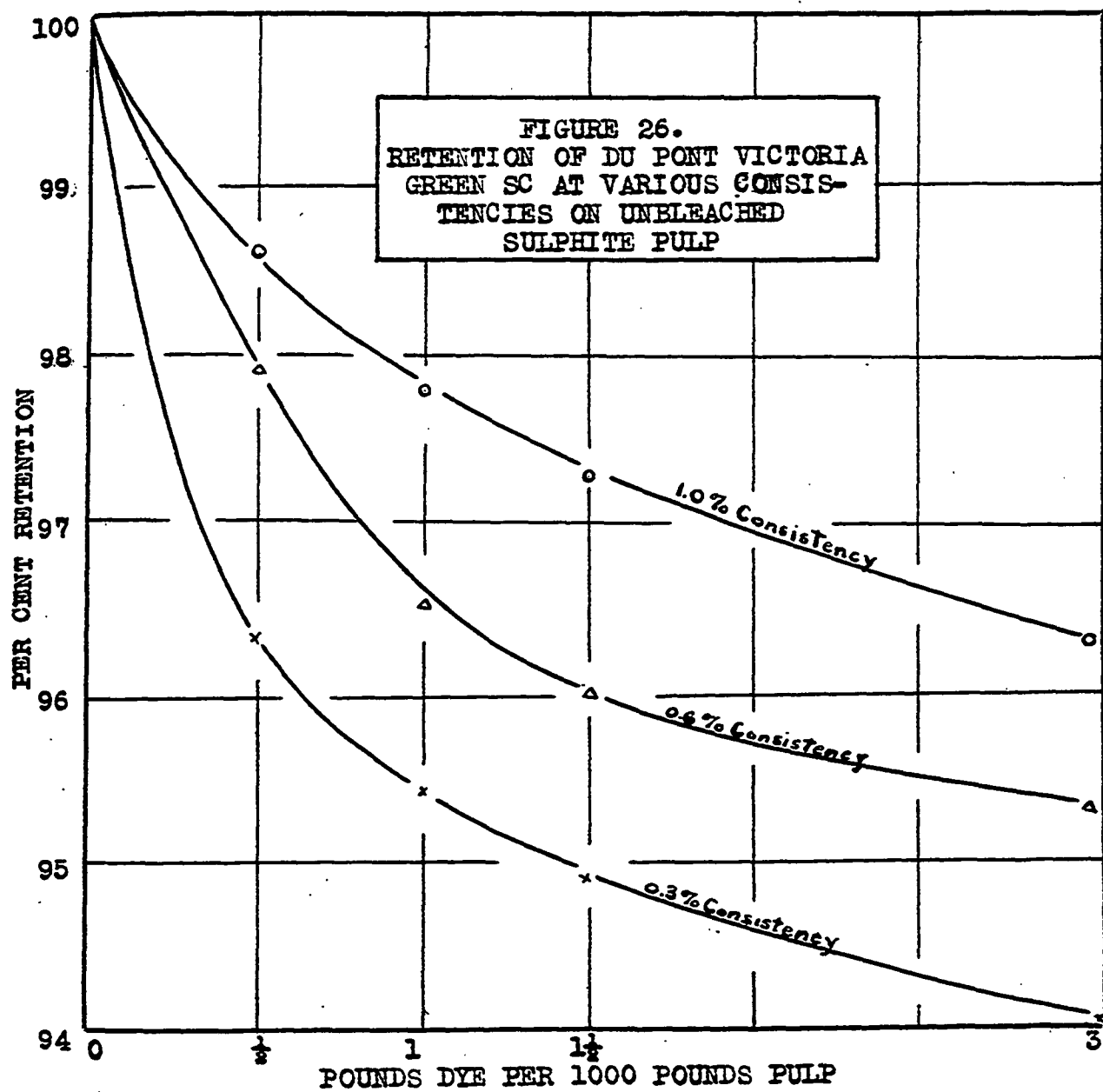
In summarizing the effect of pH on the retention of the three dyes when dyeing sulphite and kraft pulp it can be said that the acid dyes are by far the most affected by pH changes. Retention decreases rapidly above a pH of 5.0. Direct dyes are the least affected by pH changes while basic dyes, in general, have their optimum pH's near 5.0. In addition to these facts it has been apparent from the start that an unmistakable set of trends are present. For instance, with Pontamine Fast Red 8BL it has been noticed that the characteristics of its pH curves, when per cent retention is plotted against strength of dyeing, are always the same. For different pulps the retentions are different but the shape of the curves are the same. The retentions for the direct dye

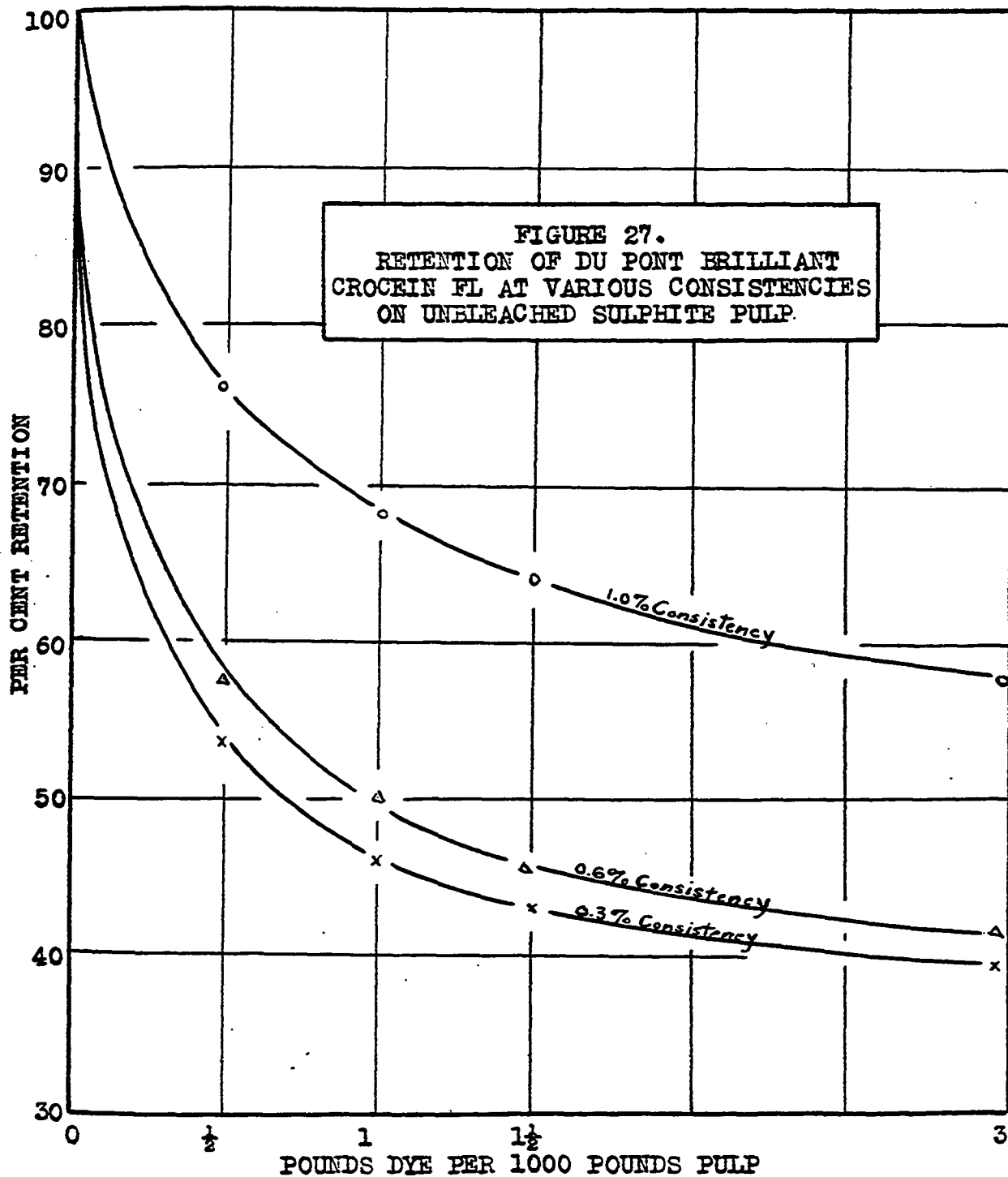
at pH's of 4.9 and 5.9 are always about the same while the retention at a pH of 5.9 is always a bit lower. Even though these generalizations have been made for Pontamine Fast Red 8BL, the retention of the dye is only affected slightly by pH in comparison with other dyes. The characteristics of the pH curves for Du Pont Brilliant Crocein FL are also similar regardless of the pulp being dyed. In every case the retentions for pH's of 3.9 and 4.9 are practically the same with the retention at a pH of 5.9 being lower. The characteristics of the pH curves for Du Pont Victoria Green SC are likewise similar, with the retentions at a pH of 4.9 and 5.9 in general being the same or with the pH of 4.9 being slightly higher. A dyeing made at a pH of 3.9 shows a lower retention. Only three dyes, one typical of each of the three main classes of dyestuffs, have been investigated and probably different dyes in the same class would act differently. However, in so far as this investigation is concerned, it can definitely be pointed out that for any given dye, the characteristics of its pH curves, when per cent retention is plotted against strength of dye, are always very nearly the same for all kinds of pulps.

5. Retention on Unbleached and Bleached Sulphite at Various Consistencies.

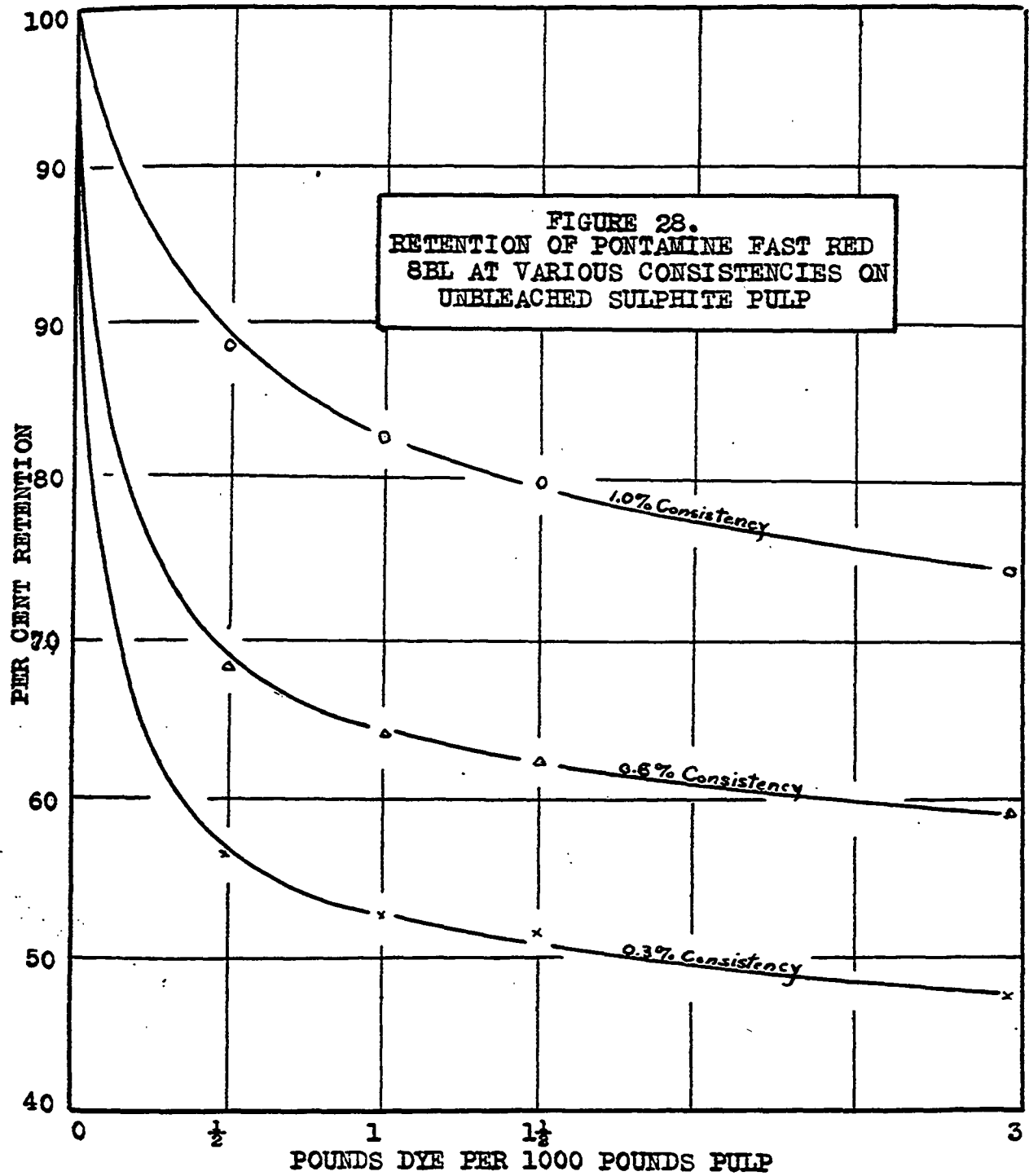
In the light of the facts brought out in section 4 above, it was decided that the use of unbleached and bleached sulphite alone would be sufficient to bring out the trends in dye retention due to any of the variables in the dyeing reaction. In order to study the effect of consistency on the dyeing reaction, unbleached and bleached sulphite were dyed at three consistencies, 0.3, 0.6, and 1.0 per cent with the three typical dyes in four different strengths ( $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings). The retentions are tabulated in table 17 and are shown graphically in figures 26 to 31 inclusive. As was expected, the stock dyed at one per cent consistency showed the highest retention because it had the greatest amount of dye present per unit volume of stock. Due to the fact that the dye was not being taken up by so much water, it naturally had a better chance of being absorbed by the fiber under the conditions of the experiment. The 0.6 per cent stock showed the next best retention for a given pulp and strength of dye, while that of the 0.3 per cent stock gave the poorest retention. In each case the one half pound dyeings gave higher retentions than those

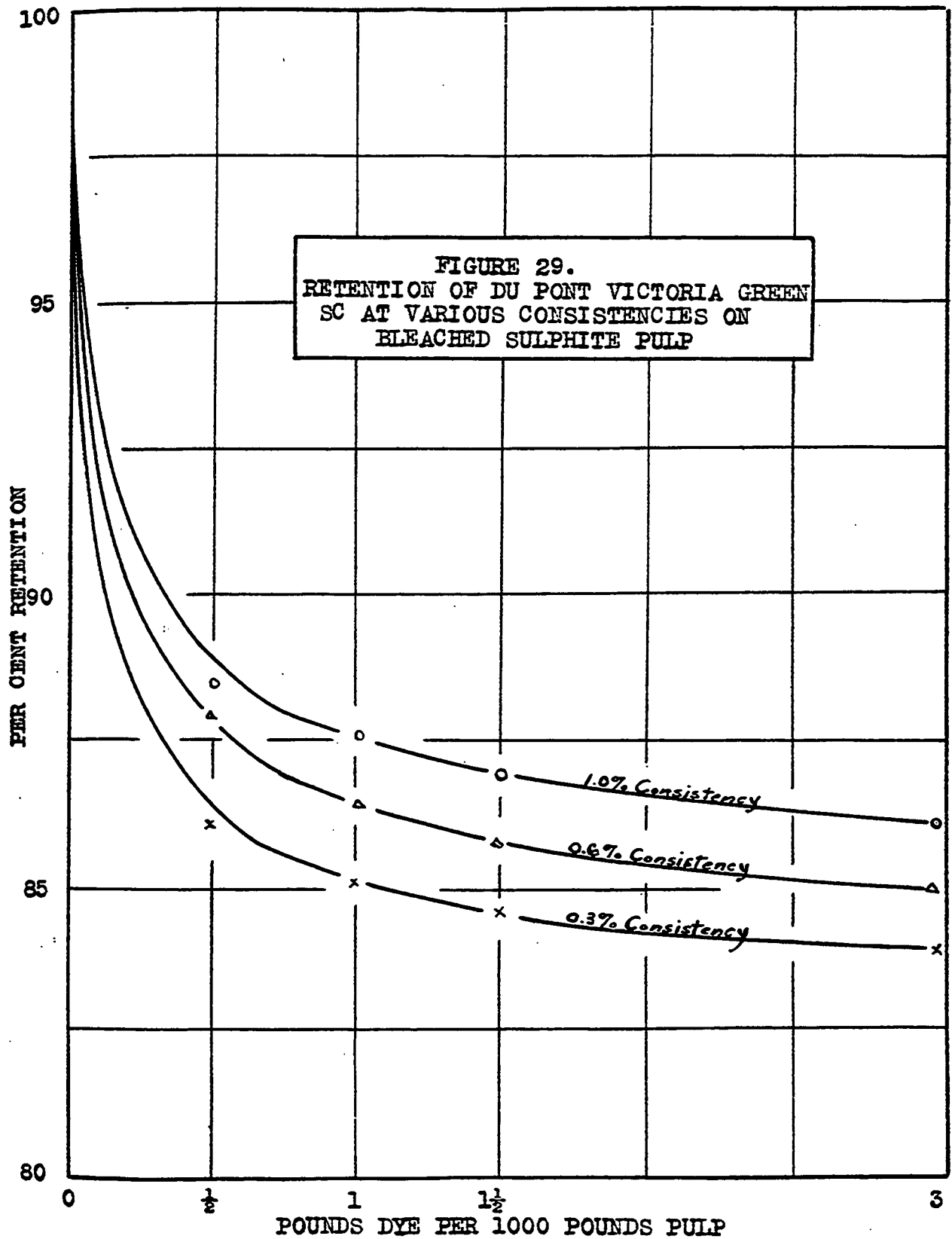
TABLE 17.				
RETENTION OF DYES ON UNBLEACHED AND BLEACHED				
SULPHITE PULP AT VARIOUS CONSISTENCIES				
Per Cent Consistency	Pounds Dye per 1000 Pounds Pulp	Du Pont Victoria Green SC Per Cent	Du Pont Brilliant Crocein FL Pef Cent	Pontamine Fast Red 8BL Per Cent
Unbleached Sulphite 0.3	$\frac{1}{2}$	96.3	53.3	46.4
	1	95.4	46.0	42.8
	$1\frac{1}{2}$	94.9	43.0	41.8
	3	94.0	39.6	37.8
Unbleached Sulphite 0.6	$\frac{1}{2}$	97.9	57.7	58.0
	1	96.5	50.0	54.0
	$1\frac{1}{2}$	96.0	45.7	52.2
	3	95.3	41.8	49.1
Unbleached Sulphite 1.0	$\frac{1}{2}$	98.6	76.3	78.7
	1	97.8	68.1	73.1
	$1\frac{1}{2}$	97.3	64.0	70.0
	3	96.3	58.0	64.7
Bleached Sulphite 0.3	$\frac{1}{2}$	86.1	58.8	52.4
	1	85.1	50.8	47.2
	$1\frac{1}{2}$	84.6	46.8	45.7
	3	83.9	42.0	41.7
Bleached Sulphite 0.6	$\frac{1}{2}$	87.9	62.0	64.0
	1	86.4	53.8	61.0
	$1\frac{1}{2}$	85.8	49.0	59.1
	3	85.0	43.3	56.2
Bleached Sulphite 1.0	$\frac{1}{2}$	88.4	81.3	91.0
	1	87.6	73.1	84.9
	$1\frac{1}{2}$	87.0	68.2	81.1
	3	86.1	61.3	75.8

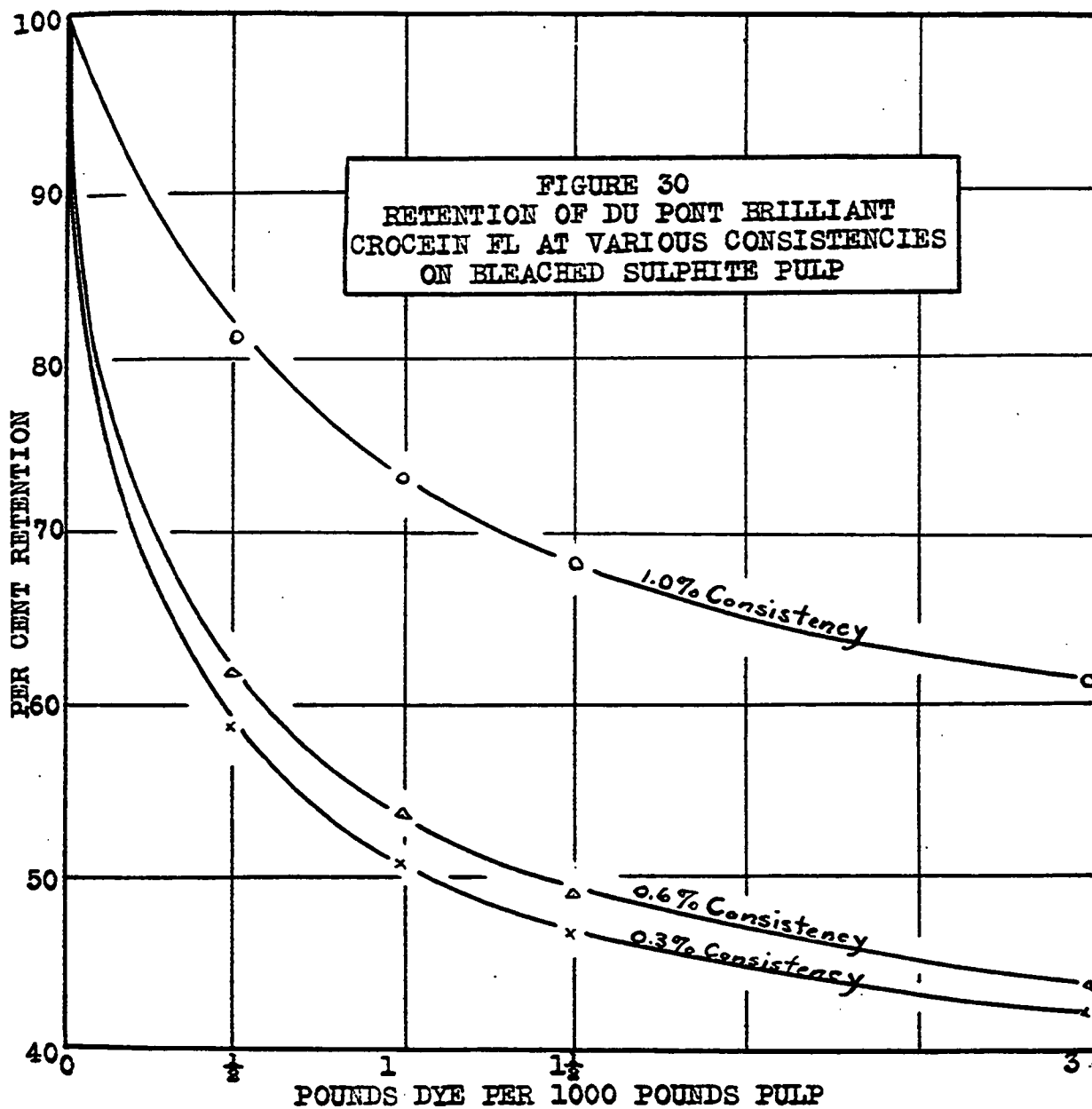


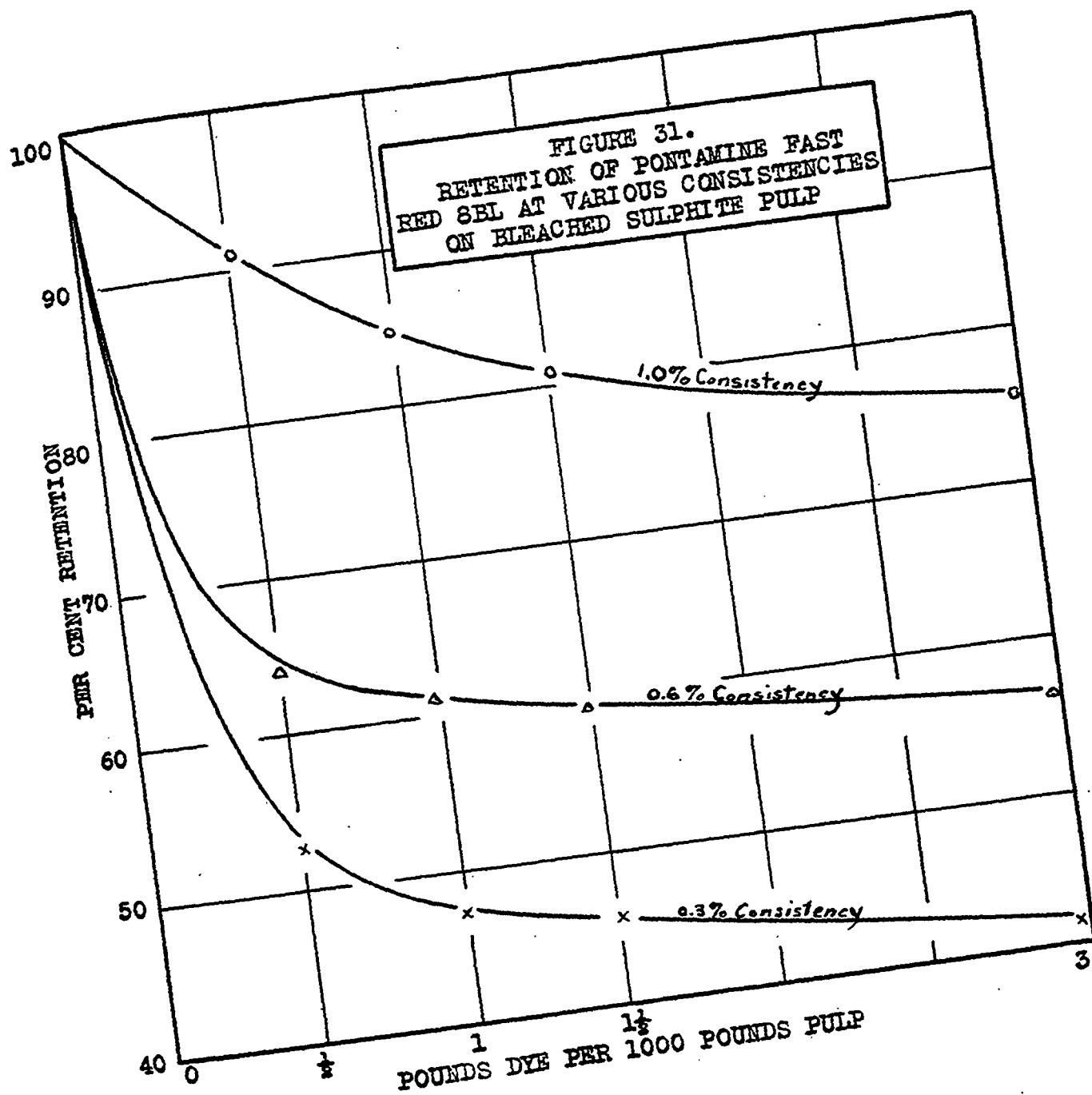












dyeings which contained greater amounts of dye.

Du Pont Victoria Green SC on unbleached sulphite, figure 26, showed retentions for one half pound dyeings, for example, of 96.3, 97.9, and 98.6 per cent for 0.3, 0.6, and 1.0 per cent consistencies respectively. It was noticed that the drop from the 0.6 per cent consistency standard to the 0.3 per cent consistency was quite appreciable and the rise to 1.0 per cent consistency was worth while.

Du Pont Brilliant Crocein FL on unbleached sulphite, figure 27, shows considerably larger differences in retention due to the consistency changes. For one half pound dyeings, the retentions were 53.3, 57.7, and 76.3 per cent for the 0.3, 0.6, and 1.0 per cent consistency stocks and in that order. It might be pointed out that as the consistency is raised from 0.3 per cent the changes are not exceptionally large until after 0.6 per cent is reached. After this point, however, large gains are noticed. From these results it can be seen that consistency is an important variable in regard to the retention of acid dyestuffs.

Pontamine Fast Red 8BL on unbleached sulphite, figure 28, shows that differences due to consistence changes are on the same order as those obtained with Du Pont Brilliant Crocein FL. For one half pound

dyeings, the retentions are 46.4, 58.0, and 78.7 per cent for pulp of 0.3, 0.6, and 1.0 per cent consistency. The raise from 0.3 to 1.0 per cent consistency is more uniform in this case than it is with the acid dye.

Du Pont Victoria Green SC on bleached sulphite, figure 29, is comparable with the results found for this dye on unbleached stock except that considerably less retention was obtained. For one half pound dyeings only 86.1, 87.9, and 88.4 per cent retentions were obtained for the 3 consistencies, 0.3, 0.6, and 1.0 per cent respectively in comparison with 96.3, 97.9, and 98.6 per cent obtained with the unbleached pulp.

Du Pont Brilliant Crocein FL on bleached sulphite, figure 30, is likewise comparable with the results obtained with the same dye on the unbleached pulp. The increase in retention does not increase rapidly from 0.3 to 0.6 per cent consistency, but large changes were noticed when the consistency advanced to 1.0 per cent. For one half pound dyeings the retentions were 58.8, 62.0, and 81.3 per cent for 0.3, 0.6, and 1.0 per cent consistency respectively. As will be noticed these results are slightly higher than the results obtained with the unbleached stock which is in

accord with the findings in section three above.

Pontamine Fast Red 8BL on bleached sulphite, figure 31, shows results which are comparable with those obtained with the same dye on unbleached sulphite except that they are higher. For one half pound dyeings a much higher rate of increase in retention is noticed for the 1.0 per cent over the 0.6 per cent consistency dyeing than in the three pound dyeings for the same consistencies. The reason for this is that the low concentration of dye when put into the higher pulp consistency is taken up rapidly, but when the higher concentration of dye is put into stock of higher consistency, the resulting dye concentration is lowered with a decrease in retention. Retentions obtained for one half pound dyeings were 52.4, 64.0, and 91.0 per cent for 0.3, 0.6, and 1.0 per cent consistencies respectively.

#### 6. Retention on Unbleached and Bleached Sulphite at Various Temperatures.

In studying the retention of dyes on pulps at various temperatures, unbleached and bleached sulphite were chosen to bring out the trends in the dye retention due to this variable. These pulps were dyed at three temperatures, 20, 40, and 60° C.,

with three typical dyes in four different strengths ( $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings). The retentions obtained are tabulated in table 18 and are shown graphically in figures 32 to 37 inclusive.

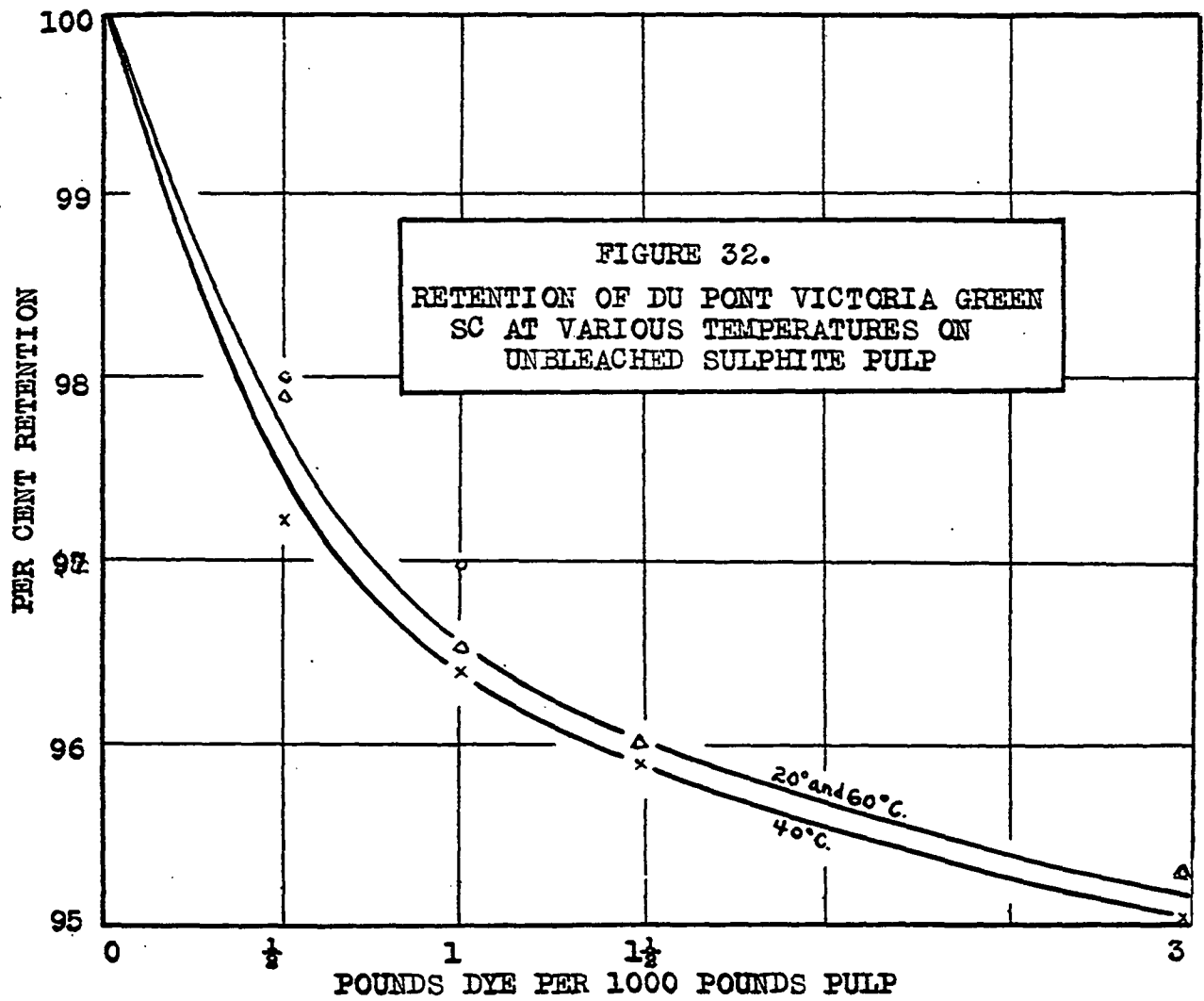
Du Pont Victoria Green SC on unbleached sulphite pulp, figure 32, shows that the retention of this basic dye at the temperatures studied does not vary much. It is not known why the retention shown by the 20 degree curve in figure 32 is between the 40 and 60 degree curves unless it is due to the limits of error in the procedure. For all practical purposes, the retentions of Du Pont Victoria Green SC on unbleached sulphite are constant in the range 20° to 60° C. For example, the retentions found for  $1\frac{1}{2}$  pound dyeings are 96.0 per cent for temperatures at 20° and 60° C. and 95.9 per cent for a temperature of 40° C.

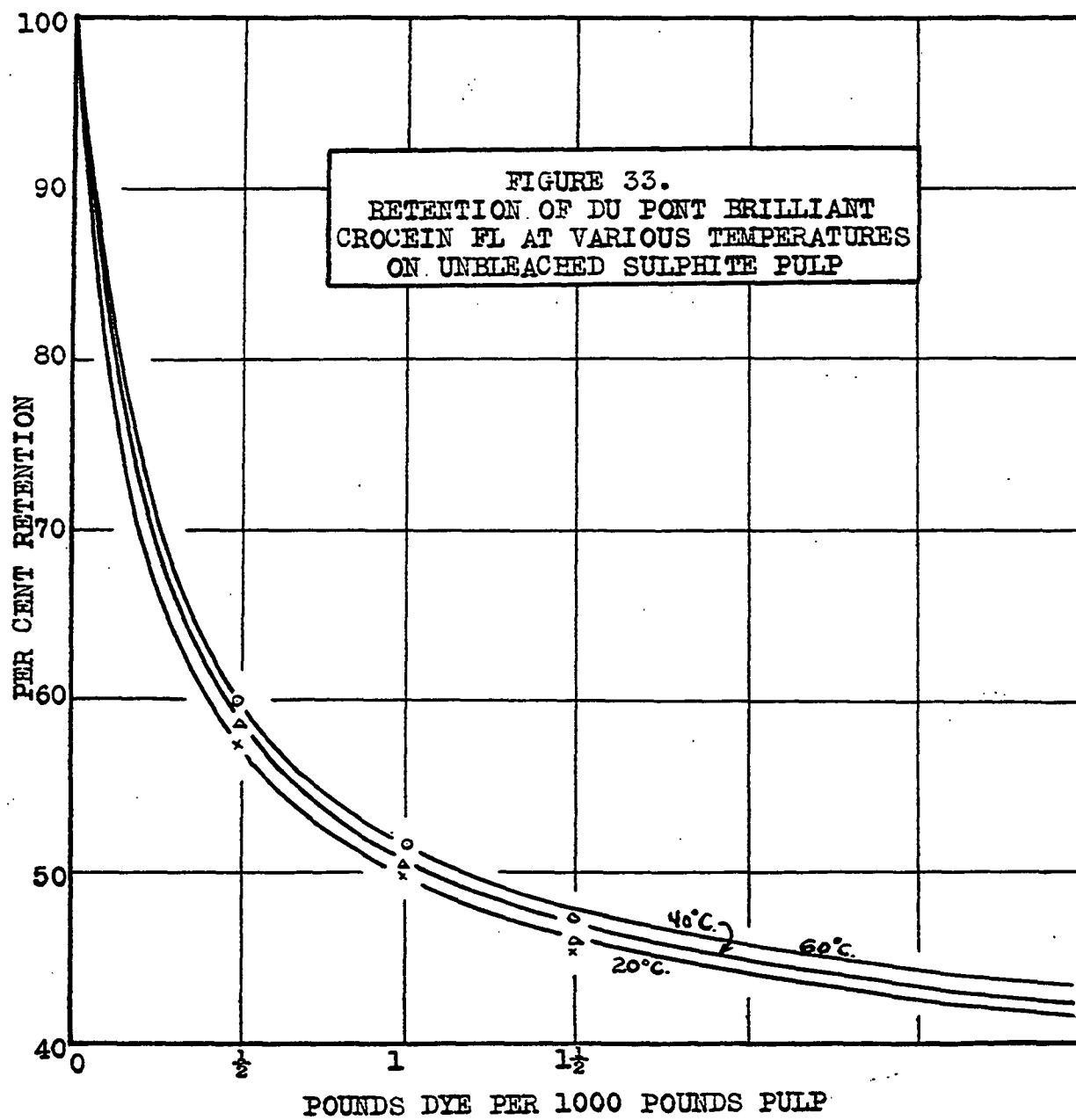
Du Pont Brilliant Crocein FL on unbleached sulphite, figure 33, shows that the higher the temperature of the dyeing reaction the higher the retention. For one half pound dyeings the retentions were 57.7, 58.6, and 60.0 per cent for 20°, 40°, and 60° C. temperatures respectively.

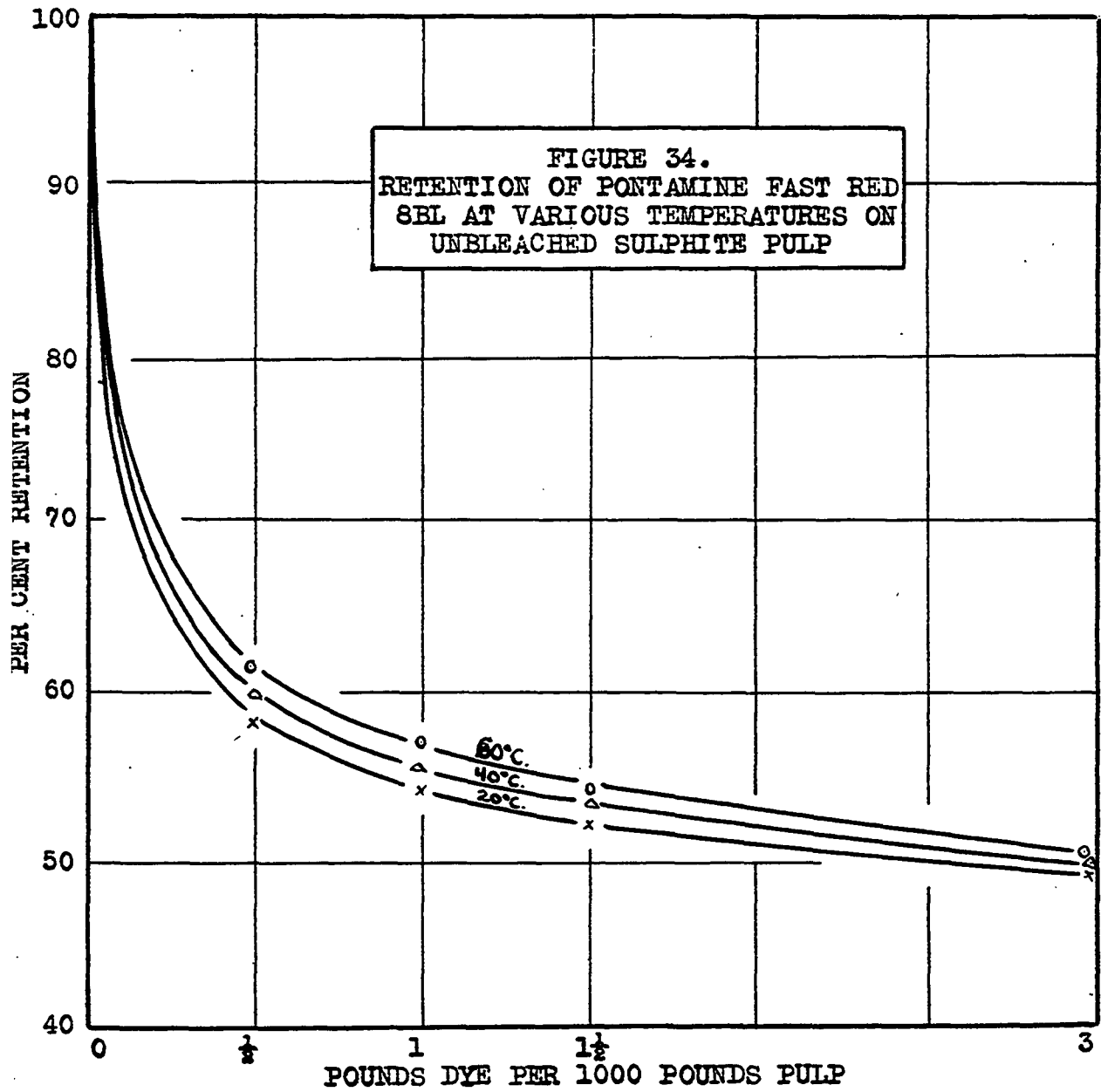
Pontamine Fast Red 8BL on unbleached sulphite, figure 34, also shows that increased temperature in

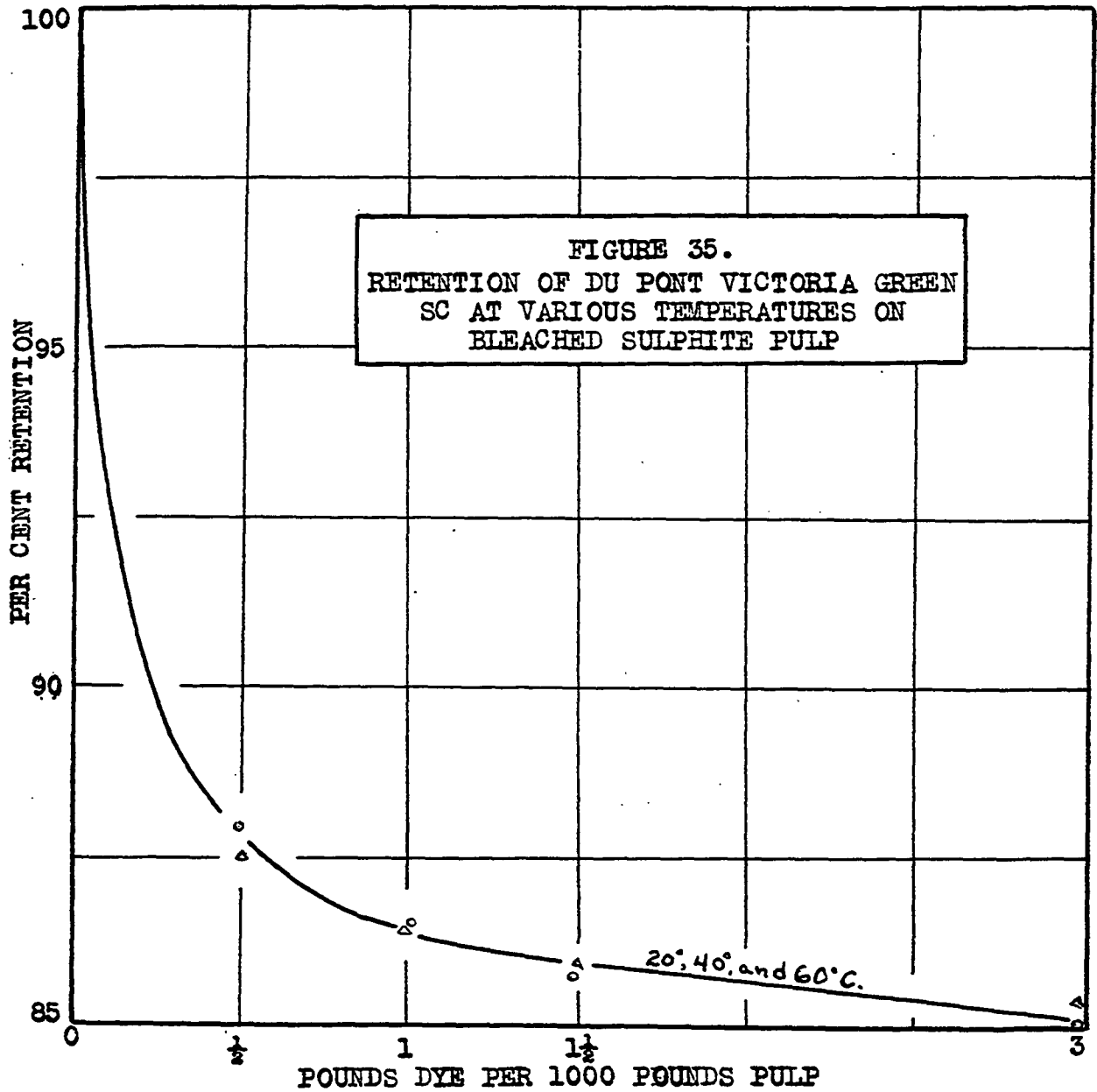


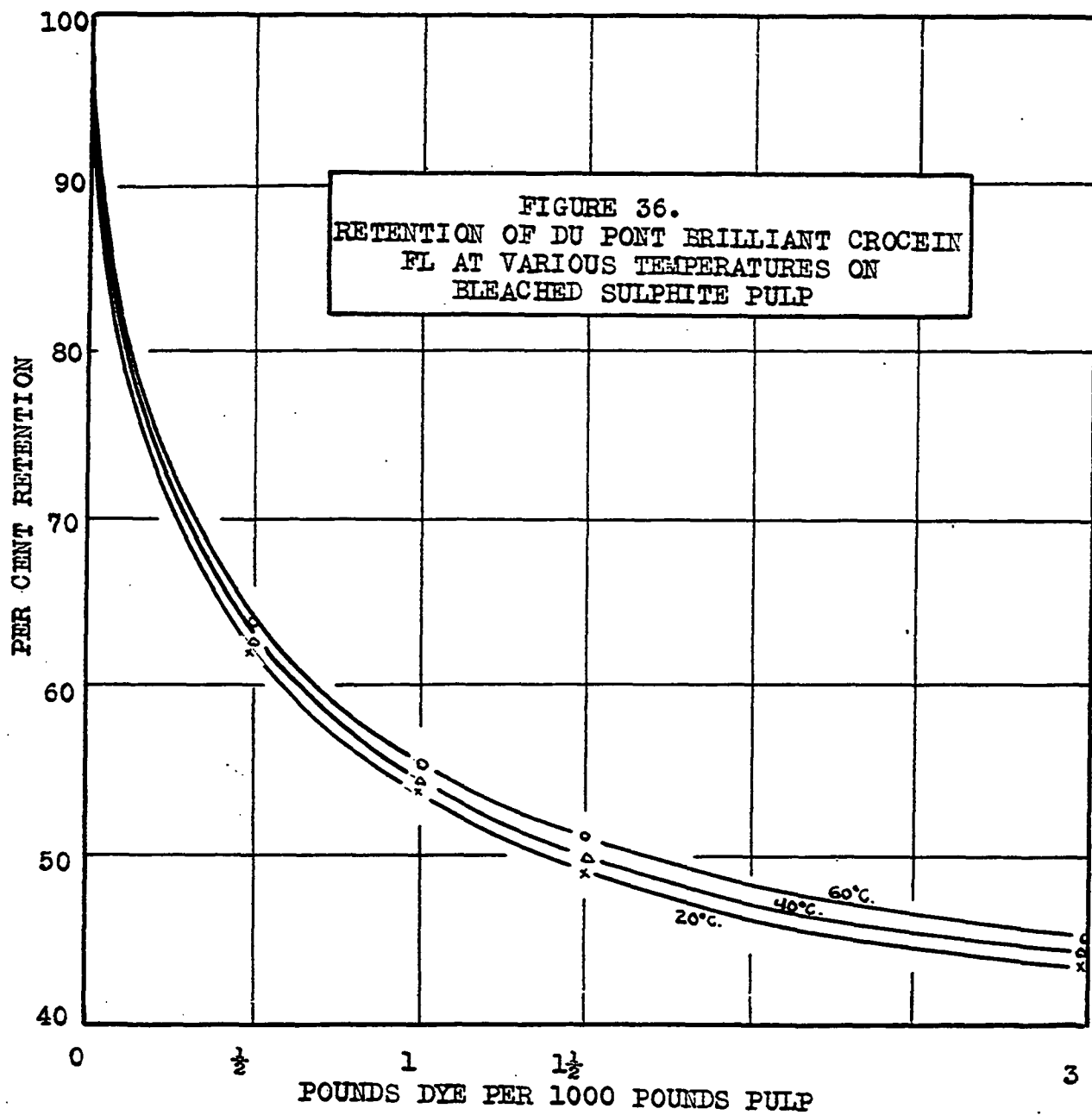
TABLE 18.				
RETENTION OF DYES ON UNBLEACHED AND BLEACHED				
SULPHITE PULP AT VARIOUS TEMPERATURES				
Temperature and Pulp	Pounds Dye per 1000 Pounds Pulp	Du Pont Victoria Green SC Per Cent	Du Pont Brilliant Crocein FL Per Cent	Pontamine Fast Red 8BL Per Cent
Unbleached Sulphite 20°	$\frac{1}{2}$	97.9	57.7	58.0
	1	96.5	50.0	54.0
	$1\frac{1}{2}$	96.0	45.7	52.2
	3	95.3	41.8	49.1
Unbleached Sulphite 40°	$\frac{1}{2}$	97.2	58.6	59.8
	1	96.4	50.5	55.5
	$1\frac{1}{2}$	95.9	46.0	53.4
	3	95.0	42.1	49.8
Unbleached Sulphite 60°	$\frac{1}{2}$	98.0	60.0	61.8
	1	97.0	51.9	57.1
	$1\frac{1}{2}$	96.0	47.5	54.3
	3	95.3	43.9	50.7
Bleached Sulphite 20°	$\frac{1}{2}$	87.9	62.0	64.0
	1	86.4	53.8	61.0
	$1\frac{1}{2}$	85.8	49.0	59.1
	3	85.0	43.3	56.2
Bleached Sulphite 40°	$\frac{1}{2}$	87.8	62.3	65.8
	1	86.5	54.1	61.8
	$1\frac{1}{2}$	85.7	49.5	60.1
	3	85.0	44.0	57.0
Bleached Sulphite 60°	$\frac{1}{2}$	87.5	63.8	68.4
	1	86.5	55.2	64.4
	$1\frac{1}{2}$	85.9	51.1	62.3
	3	85.4	45.0	59.5

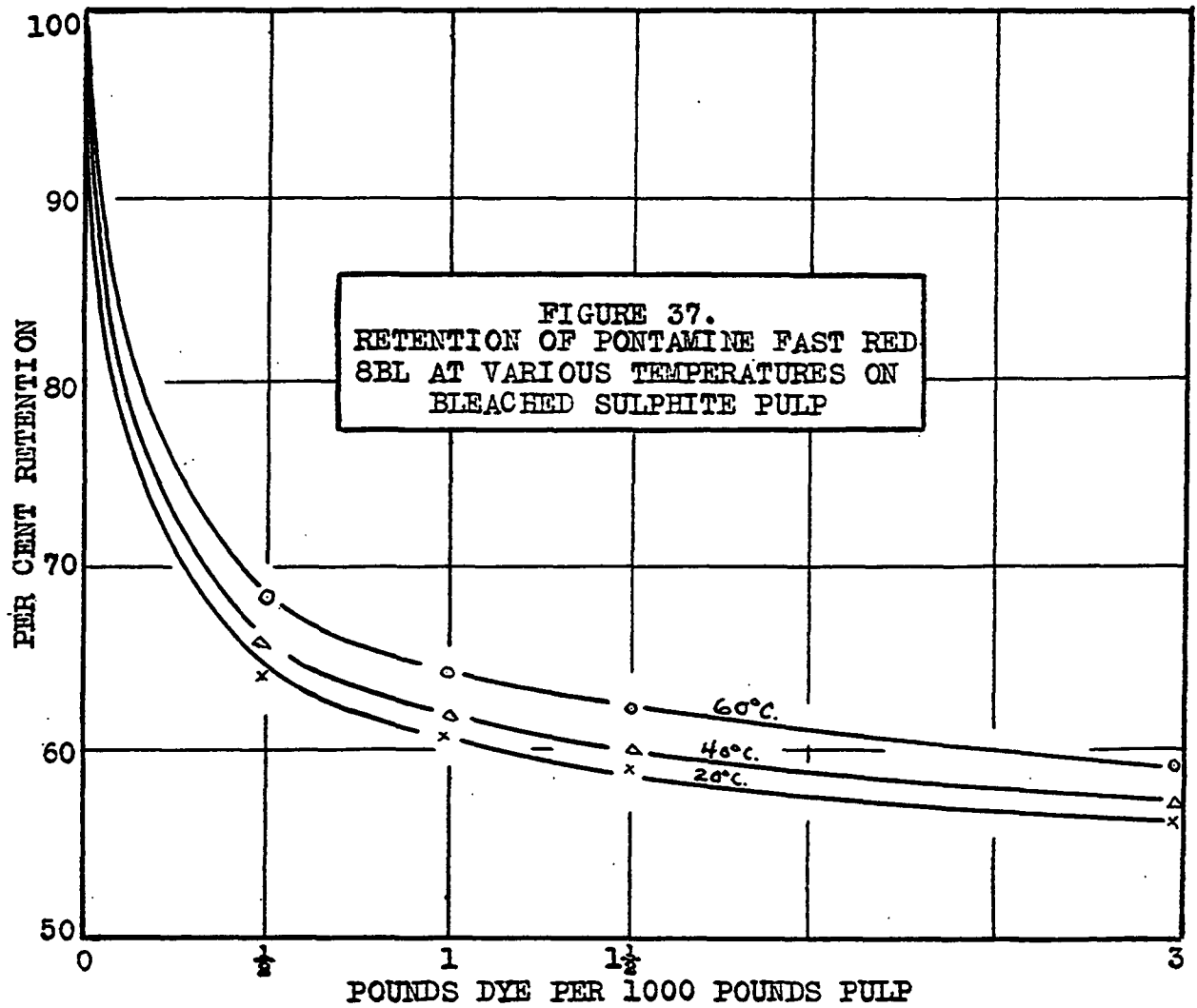












dyeing produces higher retentions of this dye. This effect was not as great as was expected changing only from 58.0 to 59.8 to 61.8 per cent for one half pound dyeings respectively for 20, 40, and 60° C. reaction temperatures.

Du Pont Victoria Green SC on bleached sulphite, figure 35, shows that, for all practical purposes, these changes in temperature had no affect on the amount of dye retained. For one half pound dyeings the retentions were 87.9, 87.8, and 87.5 per cent respectively for 20°, 40°, and 60° C. dyeing temperatures.

Du Pont Brilliant Crocein FL on bleached sulphite, figure 36, shows that increased temperature produces only slightly higher retention of this dye. For one half pound dyeings the retentions are 62.0, 62.3, and 63.8 per cent for temperatures of 20°, 40°, and 60° C. respectively.

Pontamine Fast Red 8BL on bleached sulphite, figure 37, shows that greater increases in retention than for any of the other dyes studied were obtained when the temperature was raised. For one half pound dyeings the retentions were 64.0, 65.8, and 68.4 per cent respectively for temperatures of 20°, 40°, and 60° C.



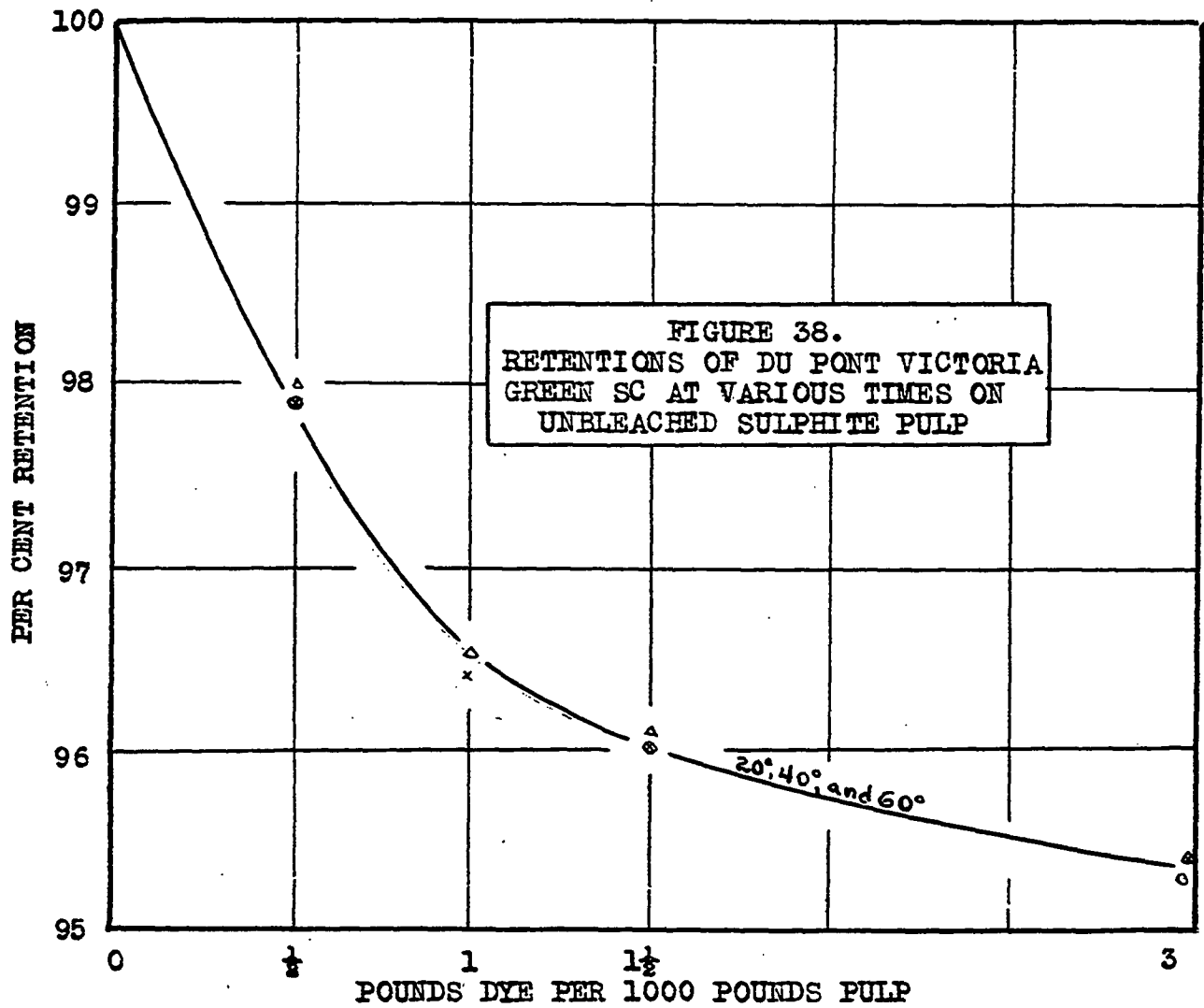
Results obtained by varying the temperature show that the retention of basic dyes are not affected from a practical standpoint by temperature; acid dyestuffs are retained very little better by increased temperatures of the dyeing reaction, while direct dyes are retained somewhat better at higher temperatures.

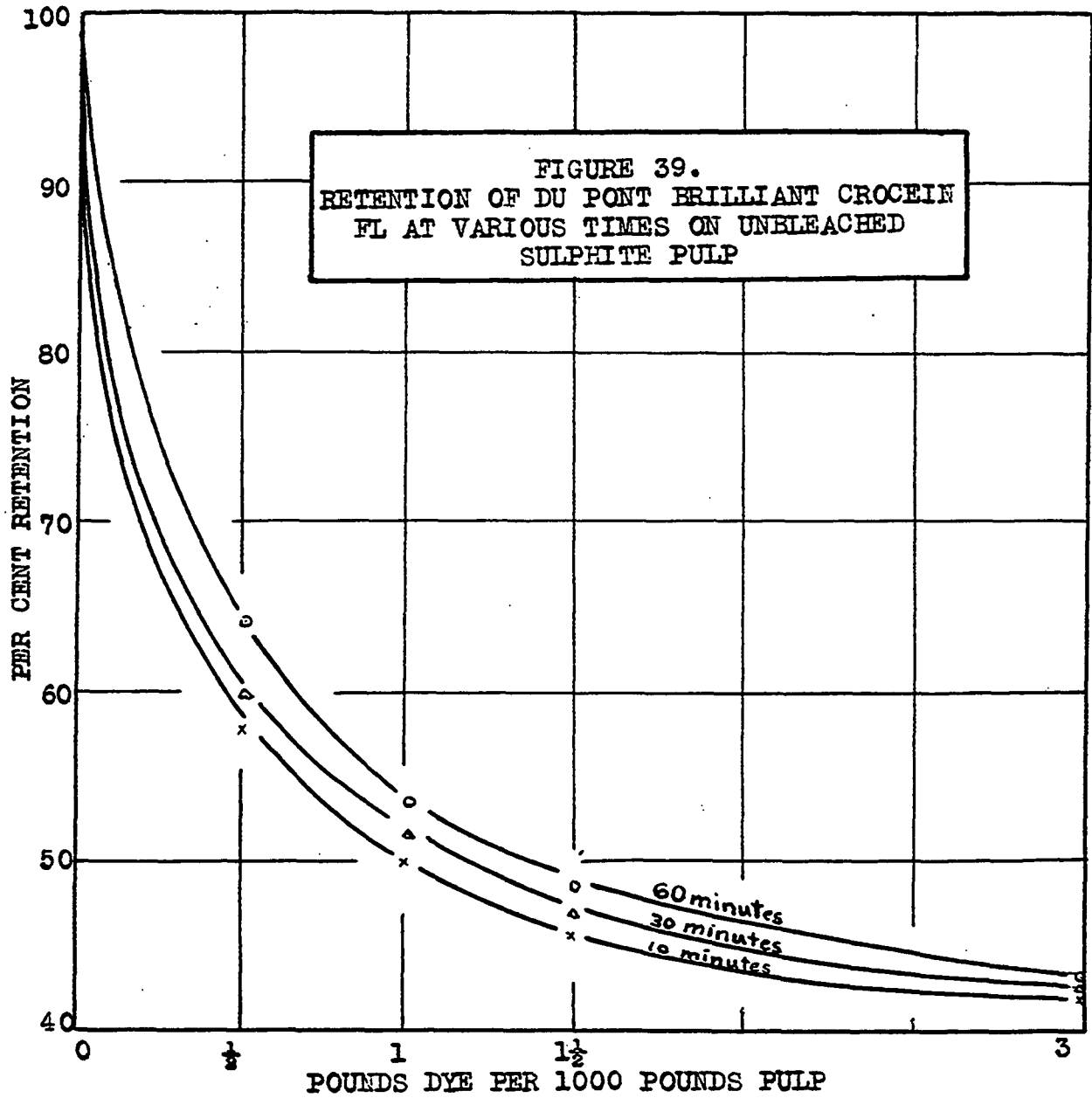
#### 7. Retention on Unbleached and Bleached Sulphite at Various Reaction Times.

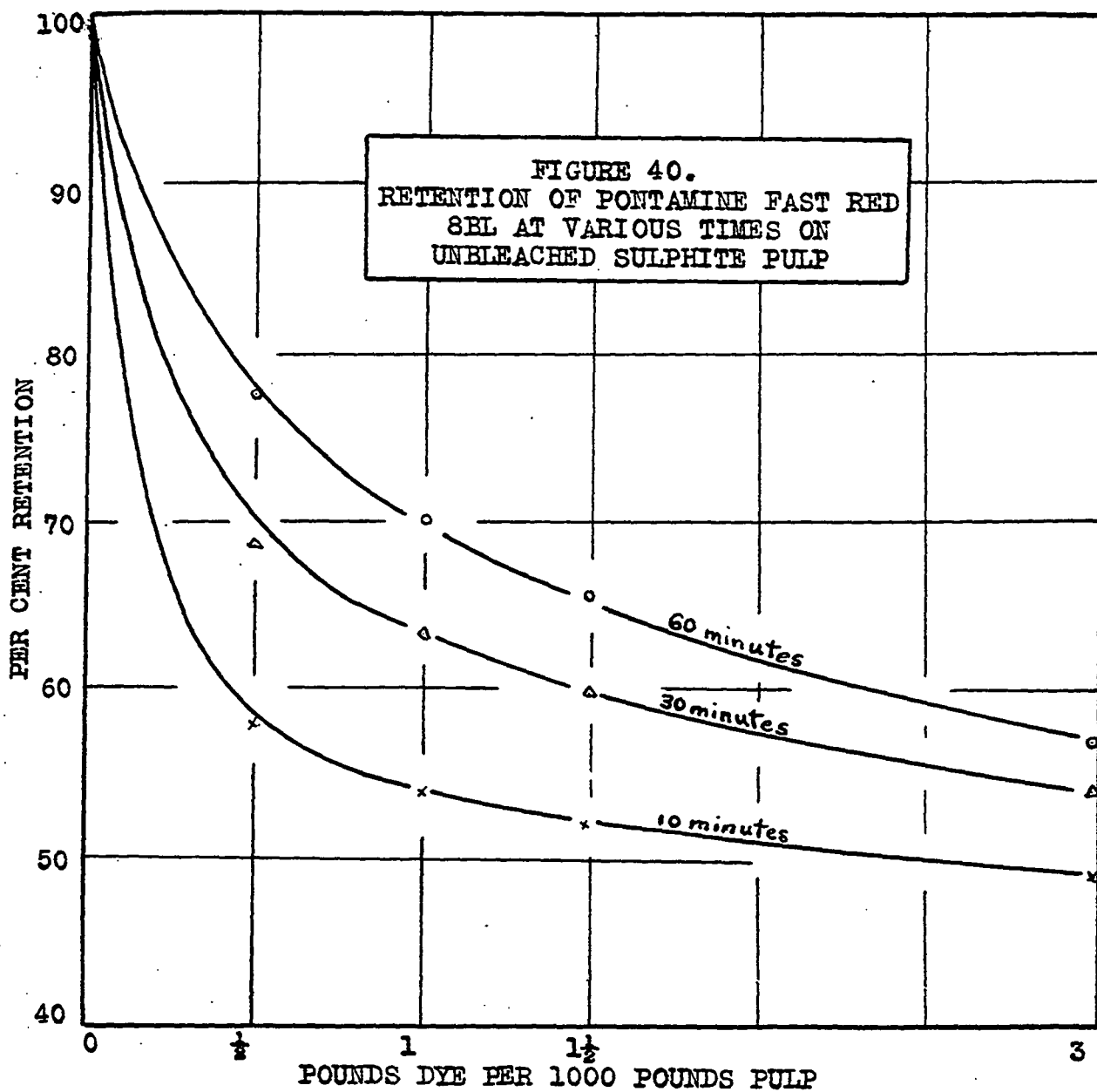
In studying the effect of time on the retention of dyes, unbleached and bleached sulphite pulp were dyed at different time periods, 10, 30, and 60 minutes, with the three typical dyes in four different concentrations ( $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings). The results of the retentions under the various conditions are shown in table 19 and are shown graphically in figures 38 to 43 inclusive.

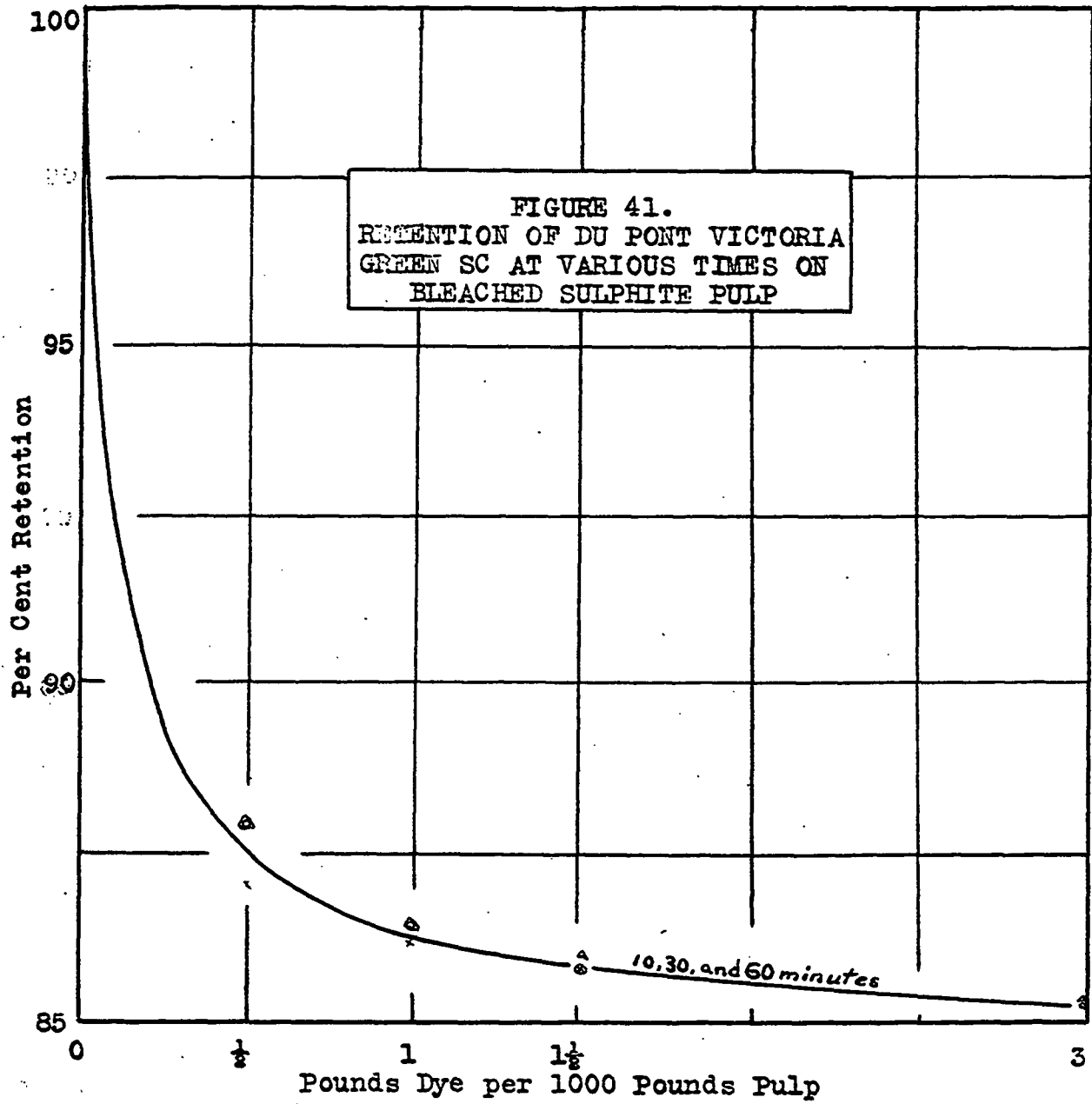
Du Pont Victoria Green SC on unbleached sulphite, figure 38, shows that time has very little effect on the retention of this dye. For one half pound dyeings the retentions were 97.9 per cent for the 10 and 30 minute reaction times and 98.0 per cent for the 60 minute reaction time. For all practical purposes the retention of this dye is the same for all time periods taken into consideration

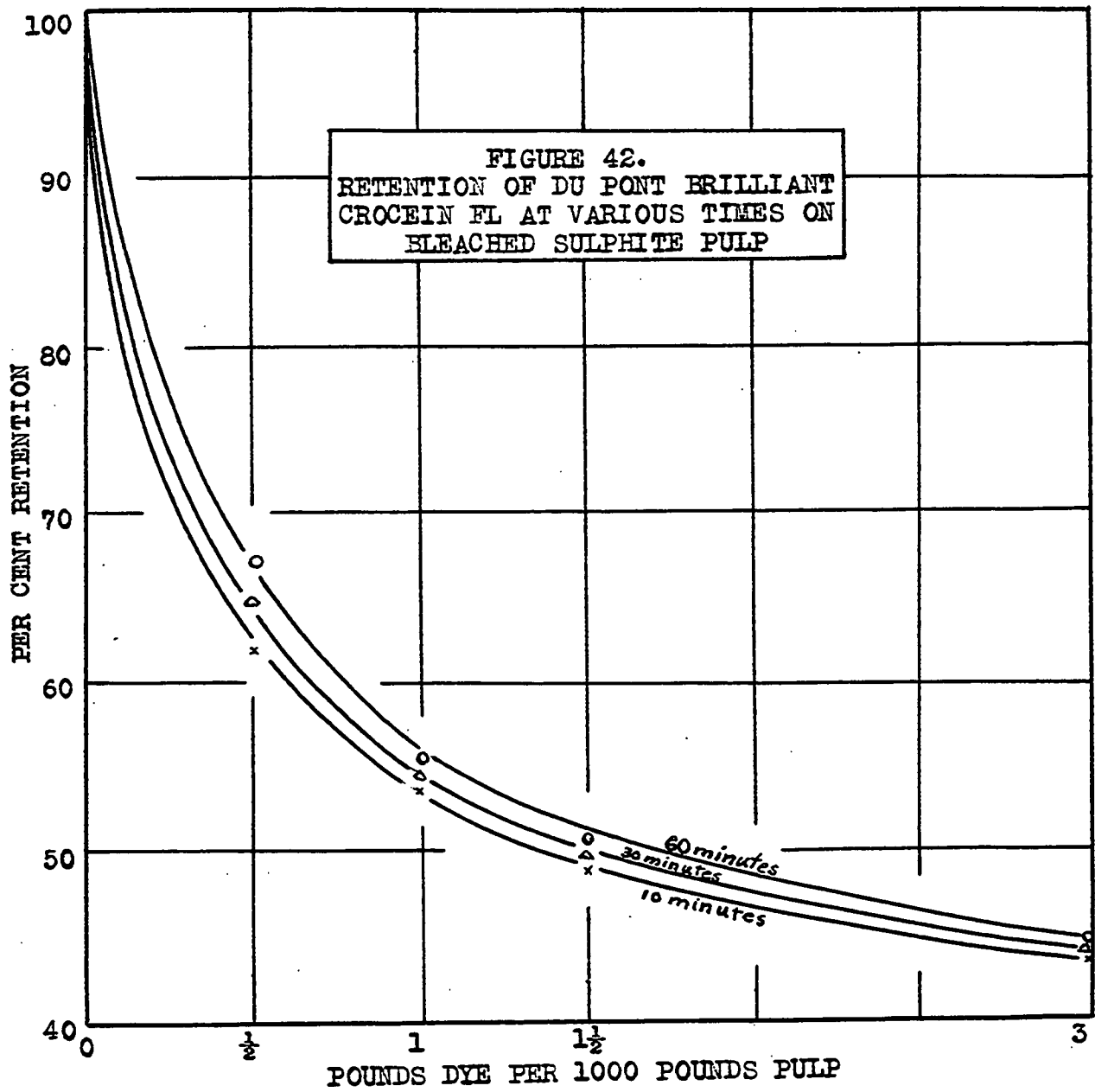
TABLE 19.				
RETENTION OF DYES ON UNBLEACHED AND BLEACHED				
SULPHITE PULP AT VARIOUS REACTION TIMES				
Time in Minutes and Pulp	Pounds Dye per 1000 Pounds Pulp	Du Pont Victoria Green SC Per Cent	Du Pont Brilliant Crocein Per Cent	Pontamine Fast Red 8BL Per Cent
Unbleached Sulphite 10	$\frac{1}{2}$	97.9	57.7	58.0
	1	96.5	50.0	54.0
	$1\frac{1}{2}$	96.0	45.7	52.2
	3	95.3	41.8	49.1
Unbleached Sulphite 30	$\frac{1}{2}$	97.9	59.7	68.7
	1	96.4	51.5	63.3
	$1\frac{1}{2}$	96.0	47.0	60.0
	3	95.4	42.3	53.9
Unbleached Sulphite 60	$\frac{1}{2}$	98.0	64.1	77.7
	1	96.5	53.5	70.1
	$1\frac{1}{2}$	96.1	48.3	65.5
	3	95.4	43.0	57.0
Bleached Sulphite 10	$\frac{1}{2}$	87.9	62.0	64.0
	1	86.4	53.8	61.0
	$1\frac{1}{2}$	85.8	49.0	59.1
	3	85.0	43.3	56.2
Bleached Sulphite 30	$\frac{1}{2}$	87.0	65.0	75.8
	1	86.2	54.5	68.5
	$1\frac{1}{2}$	85.8	49.8	64.7
	3	85.0	43.7	59.0
Bleached Sulphite 60	$\frac{1}{2}$	87.9	67.3	86.1
	1	86.4	55.2	77.4
	$1\frac{1}{2}$	86.0	50.7	71.3
	3	85.3	44.0	61.1

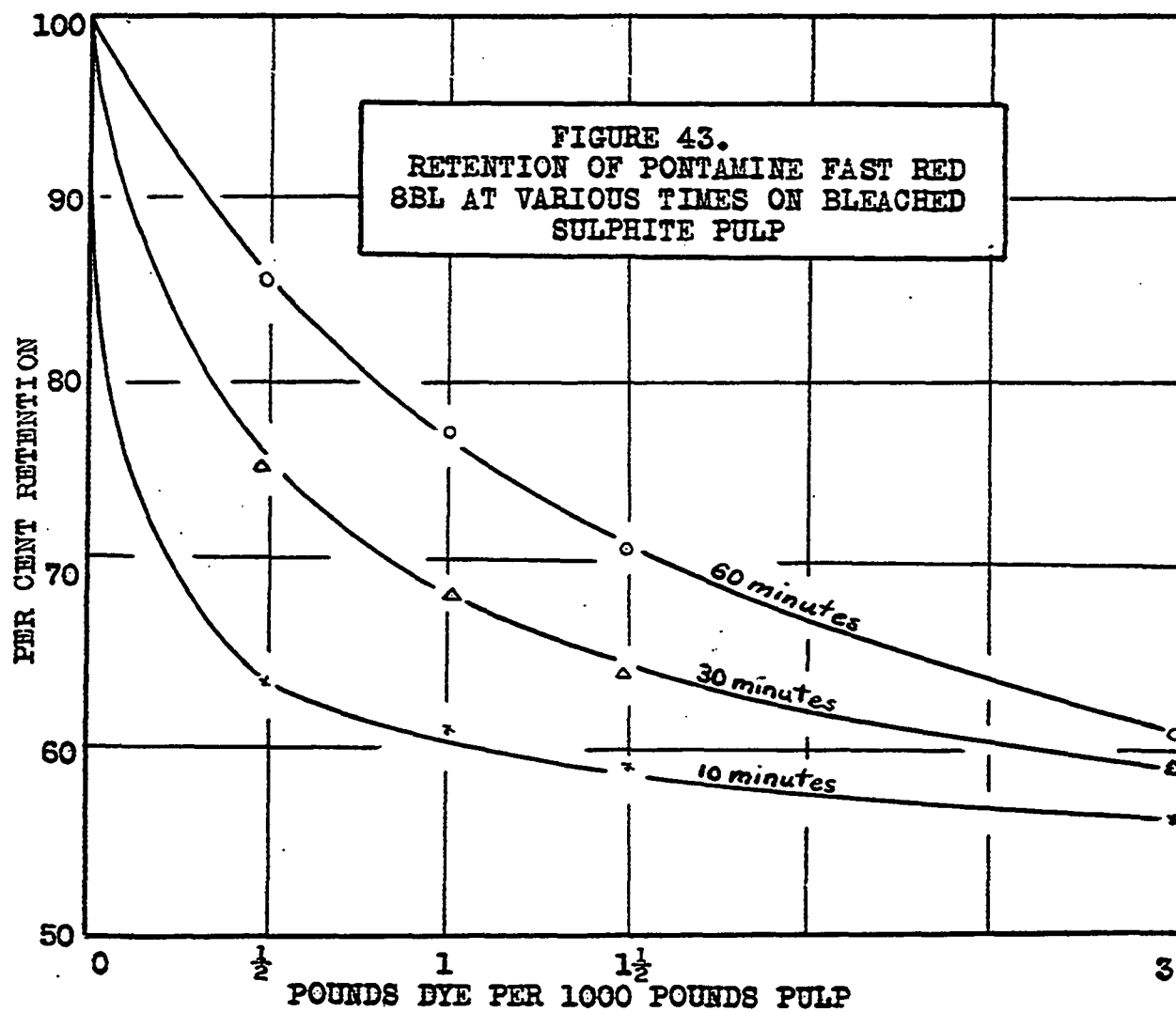














by this investigation.

Du Pont Brilliant Crocein FL on unbleached sulphite, figure 39, shows that, as the dyeing time increased, the retention is increased. For example, for one half pound dyeings, the retentions are 57.7, 59.7, and 64.1 per cent for the time periods of 10, 30, and 60 minutes respectively.

Pontamine Fast Red 8BL on unbleached sulphite, figure 40, shows that this dye is considerably affected by the time variable. The increases due to increased time of reaction are sizeable and can not go unnoticed. For one half pound dyeings the retention is increased from 58.0 to 68.7 to 77.7 per cent by changes in the time from 10 to 30 to 60 minutes respectively.

Du Pont Victoria Green SC on bleached sulphite, figure 41, shows the same type of results that were obtained with the unbleached pulp except that the retentions on bleached pulp are in general 10 per cent lower. The retention of this dye apparently is affected very little by time. For example, one half pound dyeings are retained 87.9 per cent for reaction times of 10 and 60 minutes, and 87.0 per cent for a reaction time of 30 minutes. From a practical standpoint the retention is the same in each case.

Du Pont Brilliant Crocein FL on bleached sulphite, figure 42, shows that with increased time of reaction the retention also increases just as was found for this dye with unbleached sulphite. For one half pound dyeings the retention is 62.0, 65.0, and 67.3 per cent respectively for 10, 30, and 60-minute reaction times.

Pontamine Fast Red 8BL on bleached sulphite, figure 43, shows that a sizeable change in retention can be obtained with this dye by allowing sufficient time to permit the dye to react with the fiber. For one half pound dyeings, for example, the retention was raised from 64.0 per cent for a 10-minute reaction time to 86.1 per cent for a 60-minute reaction time.

The results of this retention study, with time as a variable, show that Du Pont Victoria Green SC is not affected to any great extent by time of reaction. Du Pont Brilliant Crocein FL is somewhat affected, giving higher retentions for longer times, while Pontamine Fast Red 8BL produces sizeable increases in retention when the reaction time is lengthened.

8. Retention on Unbleached Sulphite at Various Freenesses.

To study the retention trends as the freeness

of the pulp was varied, unbleached sulphite was dyed at six different freenesses with the three typical dyes in just one strength, three pound dyeings. The results of the retentions under these conditions are tabulated in table 20 and are shown graphically in figures 44 to 46 inclusive.

Du Pont Victoria Green SC, figure 44, is retained only slightly better as the freeness of the pulp is decreased. For example, the retention increases from 95.3 to 98.3 per cent when the freeness of the pulp is changed from 860 to 250.

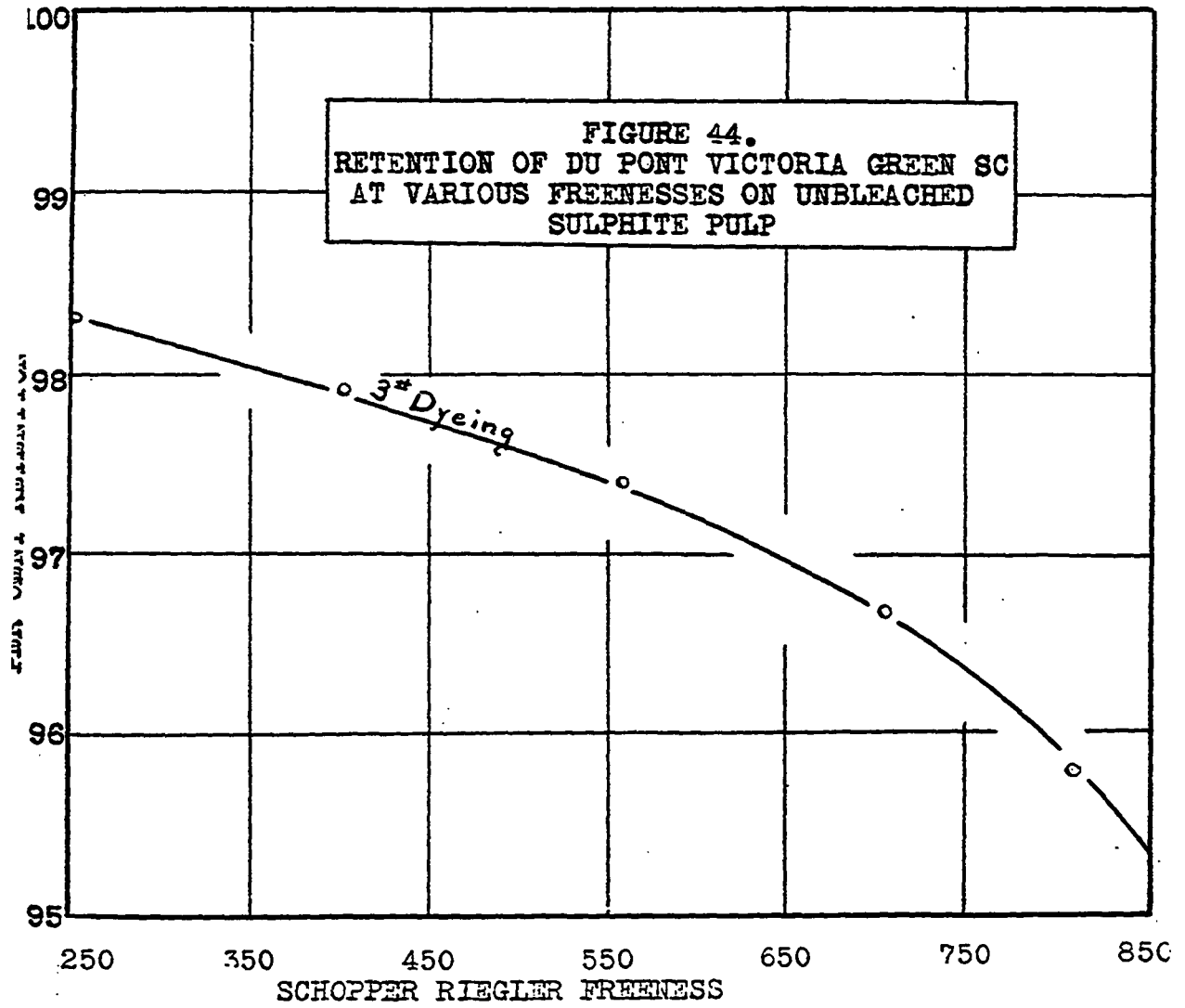
Du Pont Brilliant Crocein FL, figure 45, is retained better at lower freenesses than at high freenesses. The increase is considerable, changing steadily from 41.8 per cent retention at 860 freeness to 68.0 per cent at 250 freeness.

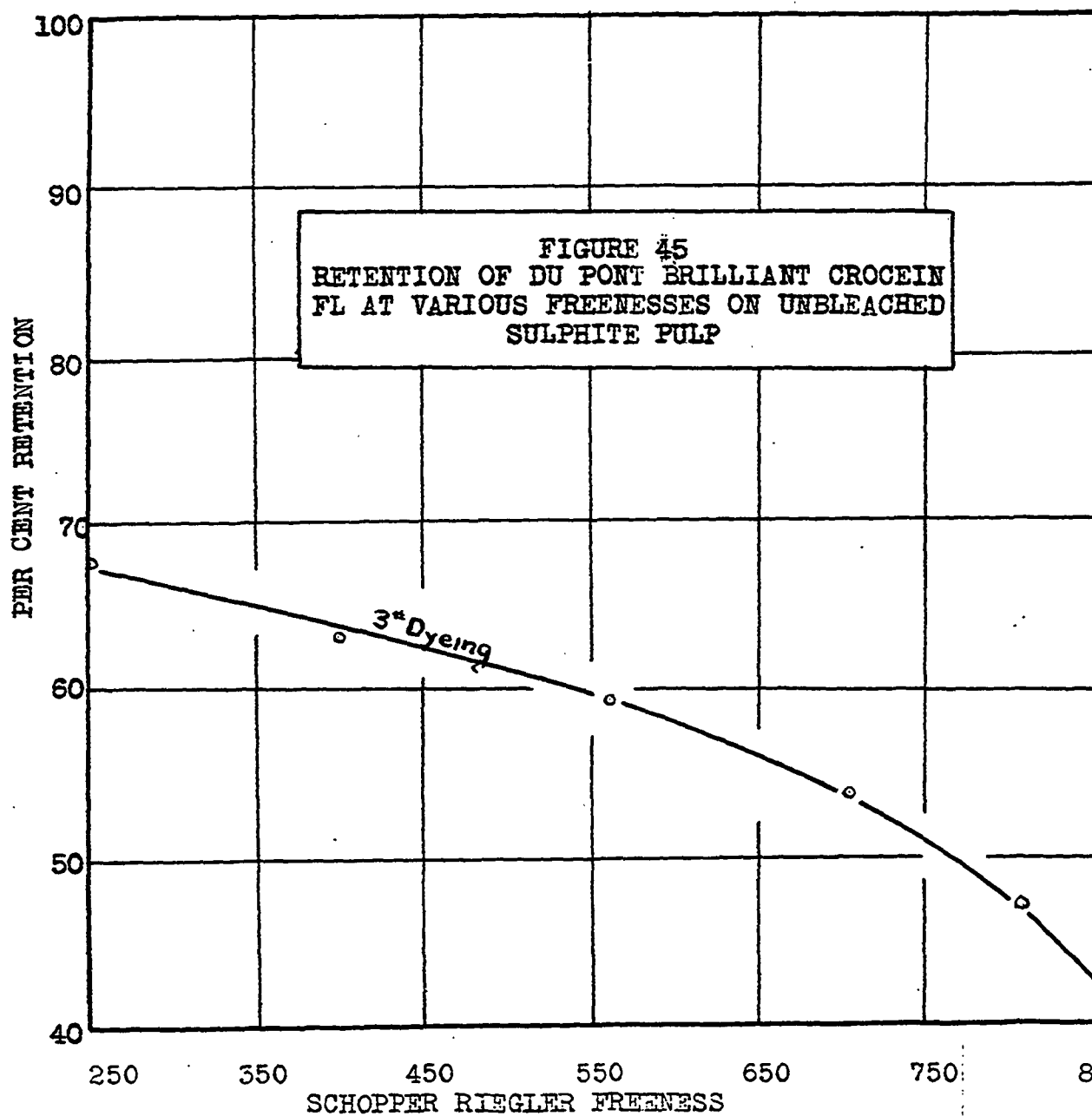
Pontamine Fast Red 8BL, figure 46, relative to the other dyes is retained to a much greater extent as the freeness is lowered. In this particular case the retention was only 49.1 per cent at a freeness of 860, but at a freeness of 250 the retention jumped to 84.5 per cent, all other conditions being equal.

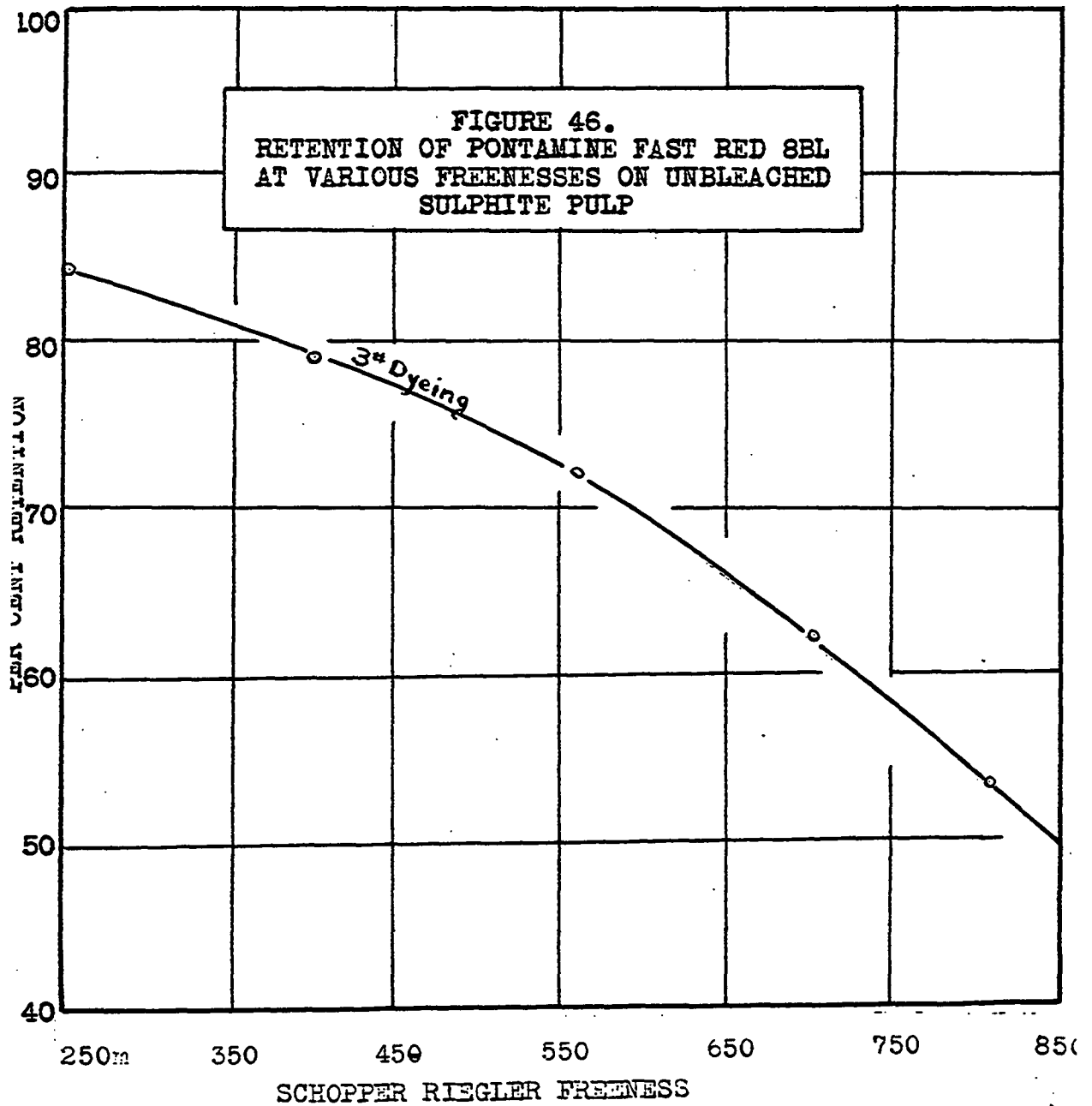
The results of this part of the investigation show that in every case the retention of a dye may

be increased by lowering the freeness of the pulp. Direct dyes such as Pontamine Fast Red 8BL are affected the most by freeness changes, acid dyes such as Du Pont Brilliant Crocein FL are next, while basic dyes such as Du Pont Victoria Green SC are affected the least.

TABLE 20.				
RETENTION OF DYES ON UNBLEACHED SULPHITE				
AT VARIOUS FREENESSES				
Time Minutes	Freeness	Du Pont Victoria Green SC Per Cent	Du Pont Brilliant Crocein FL Per Cent	Pontamine Fast Red 8BL Per Cent
0	860	95.3	41.8	49.1
10	810	95.8	47.5	53.4
20	705	96.7	54.0	62.5
30	560	97.4	59.5	72.0
40	400	97.9	63.5	79.3
50	250	98.3	68.0	84.5







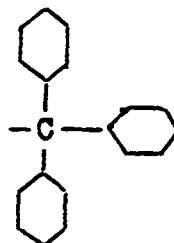
## 9. Retention of Various Dyes on Unbleached Sulphite.

In studying the retention of various dyes, three pound dyeings on unbleached sulphite were made with twenty different types of dyes obtained from the three main classes of dyestuffs. The results of the retentions are tabulated in table 21.

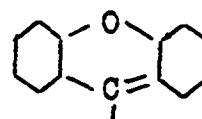
In picking the dyes, one and sometimes two were picked from each of the most important types of dyestuffs in each class as follows:

### Basic Dyes

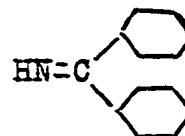
Du Pont Victoria Green SC  
Triphenyl Methane Grouping



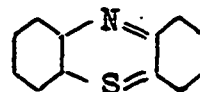
Du Pont Rhodamine B  
Xanthene Grouping



Du Pont Auramine Conc.  
Ketonamine Grouping



Du Pont Methylene Blue ZX  
Thiazine Grouping



Du Pont Basic Brown BR  
Disazo Grouping from Diamines

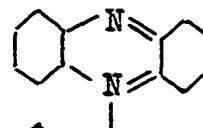




TABLE 21	
RETENTION OF VARIOUS DYES ON UNBLEACHED SULPHITE	
Dye	Per Cent Retention
Basic Dyes	
Du Pont Victoria Green SC	95.3
Du Pont Rhodamine B	85.4
Du Pont Auramine Conc.	94.8
Du Pont Methylene Blue ZX	96.3
Du Pont Basic Brown BR	53.8
Du Pont Safranine T Extra	94.8
Du Pont Methyl Violet NE	91.8
Acid Dyes	
Du Pont Brilliant Crocein FL	41.8
Du Pont Anthraquinone Blue B	5.6
Du Pont Nigrosine WSB Powder	15.9
Du Pont Quinoline Yellow Conc.	11.7
Du Pont Orange II Conc.	14.2
Pontacyl Violet S4B	36.1
Direct Dyes	
Pontamine Fast Red 8BL	49.1
Du Pont Purpurine 4B Conc.	76.1
Pontamine Fast Scarlet 4BS	55.7
Pontamine Black E	39.6
Du Pont Brilliant Paper Yellow Conc.	30.6
Pontamine Yellow SXP	25.9
Pontamine Fast Yellow>NNL	58.3

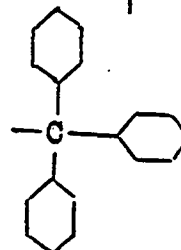
Du Pont Safranine T Extra

Azine Grouping



Du Pont Methyl Violet NE

Triphenyl Methane Grouping



### Acid Dyes

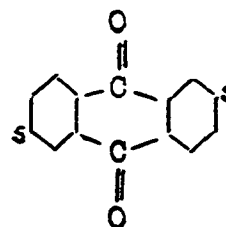
Du Pont Brilliant Crocein FL

Monoazo Grouping



Du Pont Anthraquinone Blue B

Anthraquinone Acid Grouping

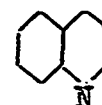


Du Pont Nigrosine WSB Powder

Very Complex Unknown Structure

Du Pont Quinoline Yellow Conc.

Quinoline Grouping



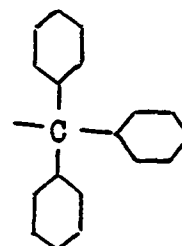
Du Pont Orange II Conc.

Monoazo Grouping



Pontacyl Violet 4SB

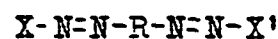
Triphenyl Methane Grouping



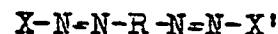
### Direct Dyes

Pontamine Fast Red 8BL

Disazo Grouping



Du Pont Purpurine 4B Conc.



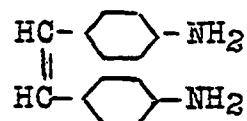
Disazo Grouping from Diamines

Pontamine Black E

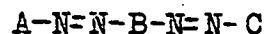


Trisazo Grouping

Du Pont Brilliant Paper  
Yellow Concentrated  
Mono Stilbene Grouping



Pontamine Fast Scarlet 4BS



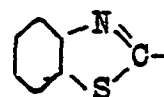
Disazo Grouping

Pontamine Yellow SXP

Di Stilbene Grouping

Pontamine Fast Yellow NNL

Thiazole Grouping



Complete structural formulas of these dyes are given in section C of the appendix. These dyes were picked so as to have a dye from each important grouping and at the same time have them overlap enough so as to determine whether or not dyestuffs having the same groupings would have approximately the same retentions.

The results show that dyes having the same groupings do not necessarily have the same retentions. For example, Du Pont Orange II Concentrated and

Du Pont Brilliant Crocein FL both have monoazo groupings but have widely differing retentions, 14.2 and 41.8 per cent respectively. All the basic dyes used in this investigation have retentions above 91.0 per cent except for Du Pont Rhodamine B which had a retention of 85.4 per cent and notably Du Pont Basic Brown which has the exceptionally low retention of 53.8 per cent. The acid dyestuffs have the lowest retentions of any class varying between 5.6 per cent for Du Pont Anthraquinone Blue B and 41.8 per cent for Du Pont Brilliant Crocein FL. The direct dyestuffs as a class are retained better than the acid dyes but not as well as the basic dyes. Du Pont Purpurine 4B Concentrated is exceptionally high with a retention of 76.1 per cent.

10. Retention on Unbleached Sulphite When Various Acids were Used With and Without Size.

In studying the retention of dyes when various acids were used with and without size, three pound dyeings were made with the dyes, Du Pont Victoria Green SC, Du Pont Brilliant Crocein FL, Du Pont Orange II, and Pontamine Fast Red 8BL, on unbleached sulphite. The acids investigated were sulphuric acid and alum (aluminum sulphate). In addition, the increase in retention obtained when size and

alum are both used was investigated. The results of the dye retentions obtained in this study are tabulated in table 22.

TABLE 22.				
RETENTION OF DYES ON UNBLEACHED SULPHITE				
WITH VARIOUS ACIDS				
Acid	Du Pont Victoria Green SC	Du Pont Brilliant Crocein FL	Du Pont Orange II	Pontamine Fast Red 8BL
Sulphuric Acid	95.3	41.8	14.3	49.1
Alum	95.2	41.9	14.7	48.2
Size and Alum	95.3	43.1	19.8	50.3

These results show quite conclusively that the retentions are the same, from a practical standpoint, when either alum or sulphuric acid are used. The use of size in connection with alum increases the retentions of the dyes except in the case of Du Pont Victoria Green SC, which is not affected. In the case of acid dyes, Du Pont Orange II shows a marked increase in its retention, changing from 14.7 per cent without size to 19.8 per cent with size. Du Pont Brilliant Crocein FL shows only a moderate increase, changing from 41.9 to 43.1 per cent retention. Pontamine Fast Red 8BL is also increased slightly by the addition of size to the

furnish. Size and alum were investigated in order to determine their effects on retention results. Sulphuric acid was used for the most part of this investigation because it produced results which were similar to those obtained with alum as has been shown above, and in addition, did not produce a flock in the white water. Alum did produce a flock in the white waters which necessitated an exceptionally large correction in the transmittance when examined spectrophotometrically and therefore was objectionable.

In summing up these results it can be said that the retentions obtained, when either sulphuric acid or alum is used, are the same from a practical standpoint. The use of alum and size together, however, produces an increase in the retention, the amount of which depends upon the dye under consideration

#### F. Applicability of the Kubelka and Monk Equations for Determining Dye Retentions.

In proving the applicability of the Kubelka and Monk equations, it appears that the best approach is to use the equations for determining dye retention. Such a method was developed and the results checked very well with those obtained by the retention method

using Beer's law. Since it was not necessary to check all the results obtained by the retention method using Beer's law to prove the applicability of the Kubelka and Monk equations, only  $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings of the three typical dyes on unbleached and bleached sulphite were considered in this part of the investigation. Hand sheets obtained when white waters were produced for these conditions were saved for this determination. The following section will consider in detail the equations, procedure, results, and errors of the method.

#### 1. Kubelka and Monk Equations.

Steele<sup>6</sup> has used the Kubelka and Monk equations to good advantage in studying the optical characteristics of paper. Due to the fact that the amount of dye in a sheet of paper affects the optical properties of the sheet, these equations have been studied in connection with this investigation. When light enters a diffusing medium, such as a sheet of paper, part of it is reflected, part of it is absorbed, and part of it is transmitted. In dealing with the color of paper, the light reflected is of primary importance. This is a function of the scattering and absorption char-

acteristics of the paper together with the thickness of the sheet and the reflecting characteristics of the background. Steele gives the derivation of the Kubelka and Monk equations concerning the reflection of light from diffusing media as follows:

"In a sheet of a translucent diffusing substance, such as paper, of a finite thickness,  $X$ , let us consider an element parallel to the surface of thickness  $dx$  and at  $x$  distance from the bottom. Let  $I_0$  be the intensity of illumination on the top,  $i_T$  the intensity of illumination on the element  $dx$  from the top due to transmission and  $i_R$  the intensity on the bottom of  $dx$  due to reflection from lower layers. Then in passing through  $dx$ ,  $i_T$  is decreased by  $i_T (S + K)dx$  and,  $i_R$  is decreased by  $i_R (S + K)dx$ , by definition, where  $S$  is the scattering coefficient and  $K$  is the absorption coefficient.

Similarly,

$i_T$  is increased by  $S i_R dx$ , and

$i_R$  is increased by  $S i_T dx$ . Then

$-di_T = -(S + K)i_T dx + S i_R dx$ , and

$di_R = -(S + K)i_R dx + S i_T dx$ .

Dividing by  $i_T$  and  $i_R$  respectively, and adding,



$$\frac{di_R}{i_R} - \frac{di_T}{i_T} = \frac{d \ln i_R}{i_T} = -2(S + K)dx + S \left( \frac{i_T}{i_R} + \frac{i_R}{i_T} \right) dx.$$

When  $I$  = intensity of reflected light, let

$$\frac{I}{i_0} = R. \text{ Then}$$

$$\frac{i_R}{i_T} = r, \text{ by analogy. Then, since}$$

$$d \ln r = \frac{dr}{r},$$

$$dr = (-2(S + K)r + S + Sr^2) dx;$$

$$\int_{R'}^R \frac{dr}{r^2 - \frac{2(S + K)r}{S} + 1} = S \int_0^K dx$$

which, when integrated between the limits shown gives  $R$ , the reflectance of the sheet, in terms of  $X$ , the thickness, and  $R'$  the reflectance of the background. Substituting

$$\frac{S + K}{S} = 1 + \frac{K}{S} = a,$$

and integrating between limits,

$$\ln \frac{(R - a - \sqrt{a^2 - 1})(R' - a \sqrt{a^2 - 1})}{(R' - a - \sqrt{a^2 - 1})(R - a \sqrt{a^2 - 1})} = 2SX \sqrt{a^2 - 1}.$$

When  $X = \infty$ ,  $R' = R = R_\infty$ , and

$$R_\infty = a - \sqrt{a^2 - 1}$$

giving, finally

$$R_{\infty} = 1 + \frac{K}{S} - \sqrt{\frac{K^2}{S^2} + 2\frac{K}{S}}, \text{ which is the general}$$

equation for reflectance at infinite thickness."

Rearranging and solving for  $\frac{K}{S}$ ,

$$\frac{K}{S} = \frac{(R_{\infty} - 1)^2}{2R_{\infty}} \text{ is obtained.}$$

This equation has been the basis of the method using the Kubelka and Monk equations for determining the retention of dyes.

## 2. Method of Determining Dye Retention by the Use of the Kubelka and Monk Equations.

The reflectance of a sheet of paper is dependent upon the optical properties of the materials present in it. Of all the materials that go into a sheet of paper, dye is one of the most important in this connection. The more dye in a sheet, the less the reflectance and vice-versa. Kubelka and Monk have derived a formula which shows the scattering and absorption coefficients of a sheet to be a function of its reflectance. With the help of this equation, another method of determining the dye retention has been developed. This method has already been given under "Procedures" but will be discussed at this point in order to furnish a

clear understanding of the way in which the calculations were made. Kubelka and Monk have used a quantity in their equation which is the absorption coefficient divided by the scattering coefficient for a given wavelength and is given the notation of  $K/S$ . This  $K/S$  value for a sheet can be calculated from its reflectivity, as derived above, and is dependent upon the material of which the sheet is made. With this in mind, it is assumed that the  $K/S$  value for the dye in a sheet may be determined by subtracting the  $K/S$  value of a natural sheet from the  $K/S$  value of a dyed sheet providing conditions, other than dye, have been kept equal. At this point a result from the method using Beer's law was used in order to obtain a constant. The product of the  $K/S$  value for the dye and a constant was set equal to the weight of dye known to be in that sheet as determined by the retention method using Beer's law. The constant is calculated from this equation and may be used for determining the weight of dye in any sheet using the same furnish and dye by simply multiplying it by the  $K/S$  value of the sheet in question. In tables 23, 24, and 25 are shown the calculations of the retentions for

Du Pont Victoria Green SC, Du Pont Brilliant Crocein FL, and Pontamine Fast Red 8BL respectively on unbleached and bleached sulphite pulp when  $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings were made. The constant in each case was determined from the results obtained for a one half pound dyeing by the retention method using Beer's law.

In table 26 is shown a summary of data determined by the retention method using the Kubelka and Monk equations along with the corresponding data determined by the retention method using Beer's law.

TABLE 23.

CALCULATION OF RETENTION OF DU PONT VICTORIA  
GREEN SC ON UNBLEACHED SULPHITE BY THE METHOD  
USING THE KUBELKA AND MONK EQUATIONS

Pounds Dye per 1000 Pounds Pulp	Reflecti- vity at 620 milli- microns	$\frac{K}{S}$ Value	$\frac{K}{S}$ Due to Dye	Gms. Dye Used	Calc. Gms. Dye in Sheet	Calc. Per Cent Reten- tion
0	71.0	0.059	0	0	0	100.0
$\frac{1}{2}$	22.3	1.352	1.293	0.0015	0.001469	97.9
1	14.1	2.616	2.557	0.0030	0.002900	96.7
$1\frac{1}{2}$	10.5	3.814	3.755	0.0045	0.004250	94.7
3	6.0	7.363	7.304	0.0090	0.008290	92.2

TABLE 23A.

CALCULATION OF RETENTION OF DU PONT VICTORIA  
GREEN SC ON BLEACHED SULPHITE BY THE METHOD  
USING THE KUBELKA AND MONK EQUATIONS

Pounds Dye per 1000 Pounds Pulp	Reflecti- vity at 620 milli- microns	$\frac{K}{S}$ Value	$\frac{K}{S}$ Due to Dye	Gms. Dye Used	Calc. Gms. Dye in Sheet	Calc. Per Cent Reten- tion
0	83.2	0.017	0	0	0	100.0
$\frac{1}{2}$	24.2	1.187	1.170	0.0015	0.001320	87.9
1	15.4	2.323	2.306	0.0030	0.002600	86.6
$1\frac{1}{2}$	11.7	3.332	3.315	0.0045	0.003740	83.1
3	6.6	6.609	6.592	0.0090	0.007430	82.6

TABLE 24.

CALCULATION OF RETENTION OF DU PONT BRILLIANT  
CROCEIN FL ON UNBLEACHED SULPHITE BY THE METHOD  
USING THE KUBELKA AND MONK EQUATIONS

Pounds Dye per 1000 Pounds Pulp	Reflecti- vity at 500 milli- microns	$\frac{K}{S}$ Value	$\frac{K}{S}$ Due to Dye	Gms. Dye Used	Calc. Gms. Dye in Sheet	Calc. Per Cent Retention
0	56.5	0.168	0	0	0	100.0
$\frac{1}{2}$	42.1	0.398	0.230	0.0015	0.000865	57.7
1	36.2	0.562	0.394	0.0030	0.001480	49.3
$1\frac{1}{2}$	32.3	0.710	0.542	0.0045	0.002040	45.5
3	24.9	1.133	0.965	0.0090	0.003630	40.4

TABLE 24A.

CALCULATION OF RETENTION OF DU PONT BRILLIANT  
CROCEIN FL ON BLEACHED SULPHITE BY THE METHOD  
USING THE KUBELKA AND MONK EQUATIONS

Pounds Dye per 1000 Pounds Pulp	Reflecti- vity at 500 milli- microns	$\frac{K}{S}$ Value	$\frac{K}{S}$ Due to Dye	Gms. Dye Used	Calc. Gms. Dye in Sheet	Calc. Per Cent Retention
0	71.0	0.059	0	0	0	100.0
$\frac{1}{2}$	46.1	0.315	0.256	0.0015	0.000931	62.0
1	38.0	0.506	0.447	0.0030	0.001627	54.2
$1\frac{1}{2}$	33.6	0.656	0.597	0.0045	0.002170	48.2
3	24.9	1.133	1.074	0.0090	0.003812	42.4

TABLE 25.

CALCULATION OF RETENTION OF PONTAMINE FAST  
RED 8BL ON UNBLEACHED SULPHITE BY THE METHOD  
USING THE KUBELKA AND MONK EQUATIONS

Pounds Dye per 1000 Pounds Pulp	Reflecti- vity at 520 milli- microns	$\frac{K}{S}$ Value	$\frac{K}{S}$ Due to Dye	Gms. Dye Used	Calc. Gms. Dye in Sheet	Calc. Per Cent Retention
0	57.5	0.157	0	0	0	100.0
$\frac{1}{2}$	35.0	0.603	0.446	0.0015	0.000870	58.0
1	26.8	1.000	0.843	0.0030	0.001645	54.8
$1\frac{1}{2}$	22.5	1.335	1.178	0.0045	0.002300	51.2
3	15.0	2.408	2.251	0.0090	0.004390	48.8

TABLE 25A.

CALCULATION OF RETENTION OF PONTAMINE FAST  
RED 8BL ON BLEACHED SULPHITE BY THE METHOD  
USING THE KUBELKA AND MONK EQUATIONS

Pounds Dye per 1000 Pounds Pulp	Reflecti- vity at 520 milli- microns	$\frac{K}{S}$ Value	$\frac{K}{S}$ Due to Dye	Gms. Dye Used	Calc. Gms. Dye in Sheet	Calc. Per Cent Retention
0	73.0	0.050	0	0	0	100.0
$\frac{1}{2}$	38.4	0.494	0.444	0.0015	0.000960	64.0
1	28.7	0.886	0.836	0.0030	0.001808	60.2
$1\frac{1}{2}$	23.4	1.253	1.203	0.0045	0.002600	57.8
3	15.5	2.303	2.253	0.0090	0.004880	54.2

TABLE 26.				
SUMMARY OF PER CENT RETENTION DATA DETERMINED BY THE METHOD USING BEER'S LAW AND THE CORRESPONDING DATA OBTAINED BY USING THE KUBELKA AND MONK EQUATIONS				
Pounds Dye per 1000 Pounds Pulp	$\frac{1}{2}$	1	$1\frac{1}{2}$	3
Victoria Green SC				
Unbleached Sulphite				
Method Using Beer's Law	97.9	96.5	96.0	95.3
Method Using K. & M. Eq.	97.9	96.7	94.7	92.2
Bleached Sulphite				
Method Using Beer's Law	87.9	86.4	85.8	85.0
Method Using K. & M. Eq.	87.9	86.6	83.1	82.6
Brilliant Crocein FL				
Unbleached Sulphite				
Method Using Beer's Law	57.7	50.0	45.7	41.8
Method Using K. & M. Eq.	57.7	49.3	45.5	40.4
Bleached Sulphite				
Method Using Beer's Law	62.0	53.8	49.0	43.3
Method Using K. & M. Eq.	62.0	54.2	48.2	42.4
Fast Red 8BL				
Unbleached Sulphite				
Method Using Beer's Law	58.0	54.0	52.2	49.1
Method Using K. & M. Eq.	58.0	54.8	51.2	48.8
Bleached Sulphite				
Method Using Beer's Law	64.0	61.0	59.1	56.2
Method Using K. & M. Eq.	64.0	60.2	57.8	54.2



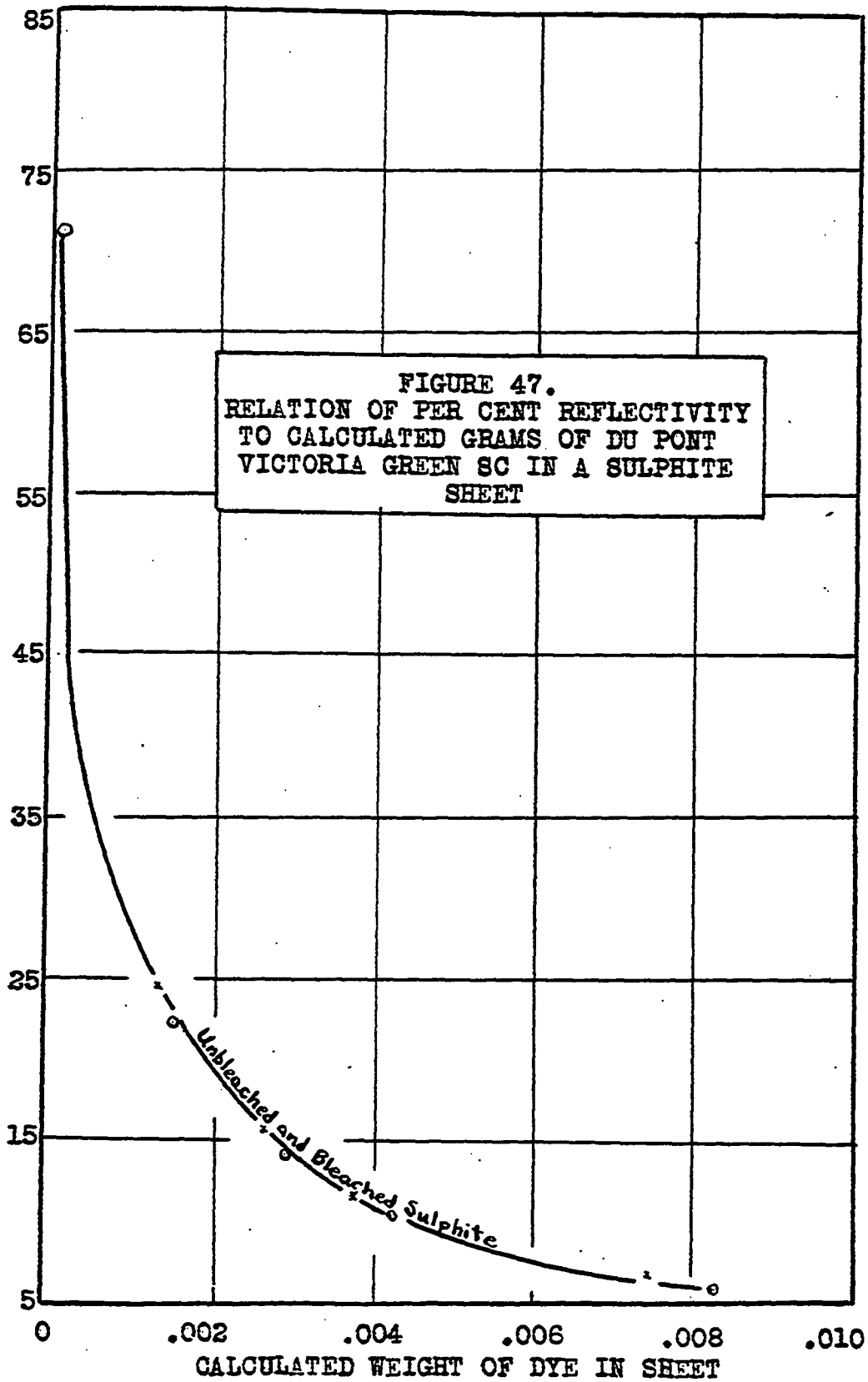
### 3. Limits of Error in the Retention Method Using the Kubelka and Monk Equations.

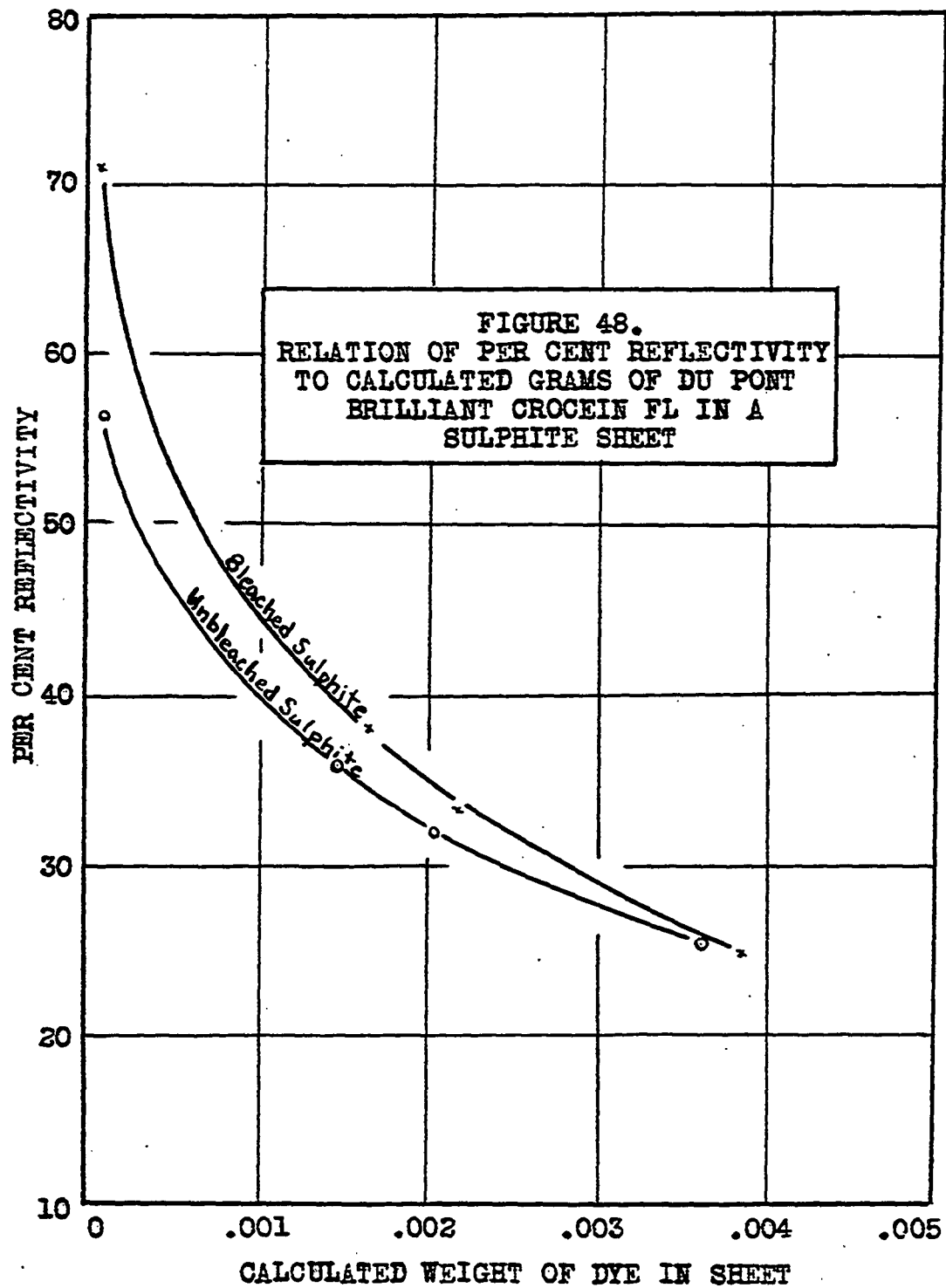
In examining the errors inherent in the method for determining dye retentions by using the Kubelka and Monk Equations, it was found that dyed paper could be produced experimentally under identical conditions which would not vary more than 0.1 per cent in reflectivity if the proper precautions were taken. The results in table 27 bear this point out. Reflectivities were taken at different points on each of five sheets dyed with one pound of Du Pont Brilliant Crocein FL and five sheets dyed with three pounds of the same dye.

TABLE 27										
REFLECTANCES OF VARIOUS SHEETS DYED UNDER IDENTICAL CONDITIONS WITH DU PONT BRILLIANT CROCEIN FL										
Light of Dyeing	1 Pound					3 Pound				
Sheet Number	1	2	3	4	5	1	2	3	4	5
Reflectance	36.3	36.3	36.1	36.2	36.3	24.9	24.9	25.0	24.7	24.9
	36.2	36.2	36.2	36.2	36.2	24.9	24.8	24.9	24.8	24.9
	36.2	36.3	36.2	36.2	36.2	24.8	24.8	25.0	24.8	24.9
	36.2	36.3	36.2	36.2	36.2	24.8	24.8	24.9	24.8	24.9
	36.2	36.2	36.2	36.2	36.2	24.8	24.9	24.9	24.8	24.9
Average	1 Pound Dyeings 36.204					3 Pound Dyeings 24.860				

It was also found that for any given point on a sheet, the color analyzer would reproduce itself exactly. With the above results in mind, the error in retention was determined when the reflectance was purposely changed by one tenth per cent for  $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings of Du Pont Victoria Green, Du Pont Brilliant Crocein FL, and Pontamine Fast Red 8BL on unbleached and bleached sulphite pulp. These three dyes were picked to be representative of the three common classes of dyestuffs and give a general idea of the errors to be expected with any dye. It was found in carrying out these calculations that the calculated amount of a given dye in a given sheet varied with the reflectance on a smooth curve. This is graphically shown in figures 47, 48, and 49. In the light of this fact, it is apparent that the per cent error at different reflectances will be different. The calculations of these errors and the results are shown in tables 28, 29, and 30 for Du Pont Victoria Green SC, Du Pont Brilliant Crocein FL, and Pontamine Fast Red 8BL respectively. The errors due to a one tenth per cent change in reflectance are, for the most part, less than six tenths per cent in retention with the notable exception of Du Pont Victoria

Green SC. Larger errors in the case of this dye are due to its high retention which necessitates extremely high K/S values and therefore pulls the curve out as shown in figure 50 until a slight change in reflectance indicates a large change in the weight of dye in the sheet.





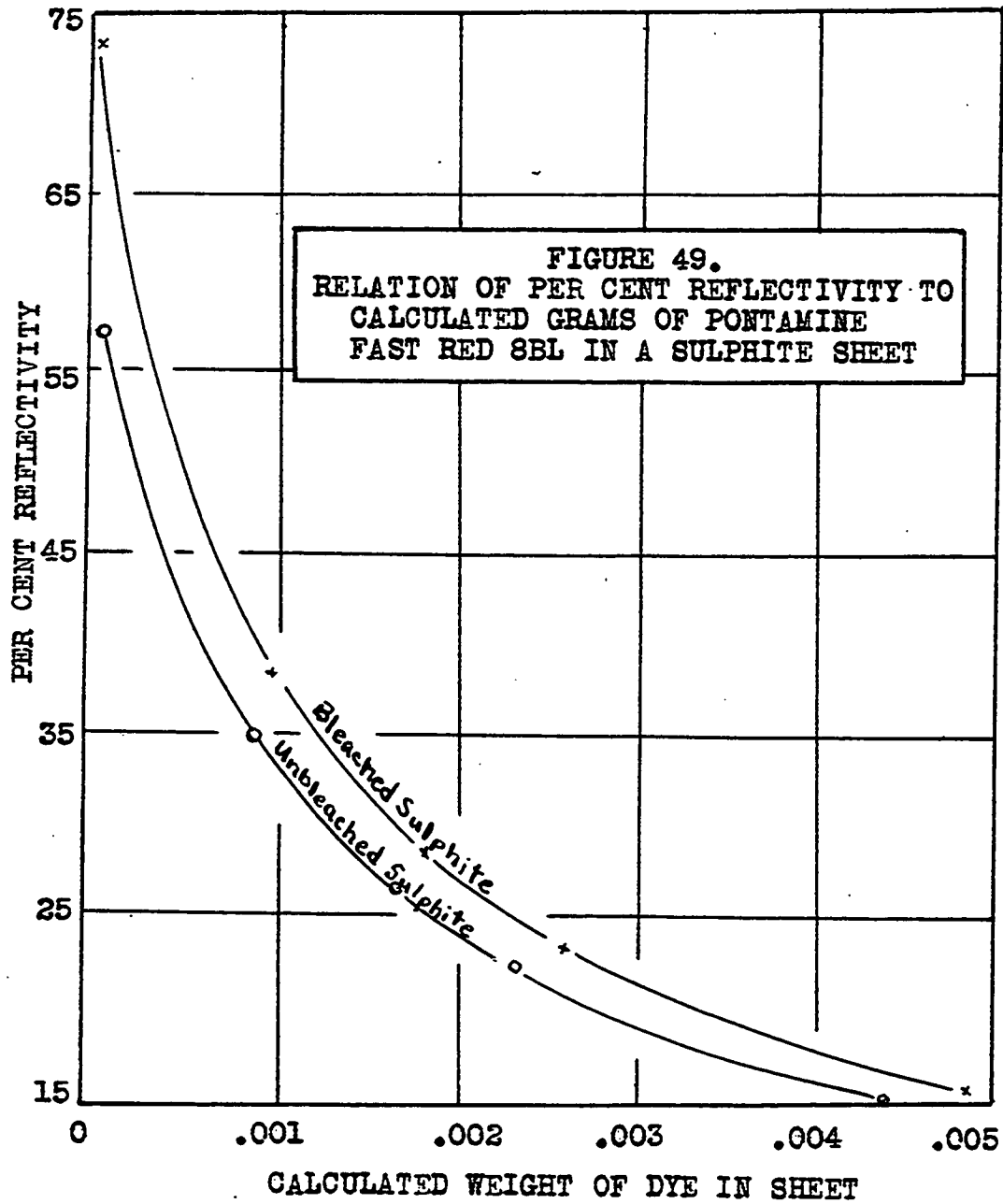


TABLE 28.						
CALCULATION OF ERROR OF RETENTION OF DU PONT VICTORIA GREEN SC ON UNBLEACHED SULPHITE AS REFLECTIVITY IS VARIED 0.1 PER CENT						
Pounds Dye per 1000 Pounds Pulp	Reflecti- vity	$\frac{K}{S}$	$\frac{K}{S}$ Due to Dye	Gms. Dye in Sheet	Per Cent Retention	Error
$\frac{1}{2}$	22.3	1.352	1.293	0.001469	97.9	0.7
	22.4	1.342	1.283	0.001457	97.2	
1	14.1	2.616	2.557	0.002900	96.7	1.0
	14.2	2.592	2.533	0.002870	95.7	
$1\frac{1}{2}$	10.5	3.814	3.755	0.004260	94.7	1.1
	10.6	3.770	3.711	0.004210	93.6	
3	6.0	7.363	7.304	0.008290	92.2	1.70
	6.1	7.227	7.168	0.008140	90.5	

TABLE 28A.						
CALCULATION OF ERROR OF RETENTION OF DU PONT VICTORIA GREEN SC ON BLEACHED SULPHITE AS REFLECTIVITY IS VARIES 0.1 PER CENT						
Pounds Dye per 1000 Pounds Pulp	Reflecti- vity	$\frac{K}{S}$	$\frac{K}{S}$ Due to Dye	Gms. Dye in Sheet	Per Cent Retention	Error
$\frac{1}{2}$	24.2	1.187	1.170	0.001320	87.9	0.5
	24.3	1.179	1.162	0.001311	87.4	
1	15.4	2.323	2.306	0.002600	86.6	0.6
	15.5	2.303	2.286	0.002578	86.0	
$1\frac{1}{2}$	11.7	3.332	3.315	0.003740	83.1	0.9
	11.6	3.368	3.351	0.003780	84.0	
3	6.6	6.609	6.592	0.007430	82.6	1.4
	6.5	6.725	6.708	0.00756	84.0	

TABLE 29.

CALCULATION OF ERROR OF RETENTION OF DU PONT BRILLIANT CROCEIN  
FL ON UNBLEACHED SULPHITE AS REFLECTIVITY IS VARIED 0.1 PER CENT

Pounds Dye per 1000 Pounds Pulp	Reflecti- vity	$\frac{K}{S}$	$\frac{K}{S}$ Due to Dye	Gms. Dye in Sheet	Per Cent Retention	Error
$\frac{1}{2}$	42.1	0.398	0.230	0.000865	57.7	0.6
	42.0	0.401	0.233	0.000876	58.3	
1	36.2	0.562	0.394	0.001480	49.3	0.3
	36.3	0.559	0.391	0.001469	49.0	
$1\frac{1}{2}$	32.3	0.710	0.542	0.002040	45.3	0.4
	32.4	0.705	0.537	0.002020	44.9	
3	24.9	1.133	0.965	0.003630	40.4	0.4
	25.0	1.125	0.957	0.003600	40.0	

TABLE 29A.

CALCULATION OF ERROR OF RETENTION OF DU PONT BRILLIANT CROCEIN  
FL ON BLEACHED SULPHITE AS REFLECTIVITY IS VARIED 0.1 PER CENT

Pounds Dye per 1000 Pounds Pulp	Reflecti- vity	$\frac{K}{S}$	$\frac{K}{S}$ Due to Dye	Gms. Dye in Sheet	Per Cent Retention	Error
$\frac{1}{2}$	46.1	0.315	0.256	0.000931	62.0	0.4
	46.0	0.317	0.258	0.000938	62.4	
1	38.0	0.506	0.447	0.001627	54.2	0.4
	38.1	0.503	0.444	0.001616	53.8	
$1\frac{1}{2}$	33.6	0.656	0.597	0.002170	48.2	0.4
	33.5	0.660	0.601	0.002190	48.6	
3	25.0	1.125	1.066	0.003877	43.1	0.7
	24.9	1.133	1.074	0.003812	42.4	



TABLE 30.						
CALCULATION OF ERROR OF RETENTION OF PONTAMINE FAST RED 8BL ON UNBLEACHED SULPHITE AS REFLECTIVITY IS VARIED 0.1 PER CENT						
Pounds Dye per 1000 Pounds Pulp	Reflectivity	$\frac{K}{S}$	$\frac{K}{S}$ Due to Dye	Gms. Dye in Sheet	Per Cent Retention	Error
$\frac{1}{2}$	35.0	0.603	0.446	0.000870	58.0	0.5
	35.1	0.600	0.443	0.000863	57.5	
1	26.8	1.000	0.843	0.001645	54.8	0.5
	26.9	0.993	0.836	0.001630	54.3	
$1\frac{1}{2}$	22.5	1.335	1.178	0.002300	51.2	0.5
	22.6	1.325	1.168	0.002280	50.7	
3	15.0	2.408	2.251	0.004390	48.8	0.5
	15.1	2.387	2.230	0.004350	48.3	

TABLE 30A.						
CALCULATION OF ERROR OF RETENTION OF PONTAMINE FAST RED 8BL ON BLEACHED SULPHITE AS REFLECTIVITY IS VARIED 0.1 PER CENT						
Pounds Dye per 1000 Pounds Pulp	Reflectivity	$\frac{K}{S}$	$\frac{K}{S}$ Due to Dye	Gms. Dye in Sheet	Per Cent Retention	Error
$\frac{1}{2}$	38.4	0.494	0.444	0.000960	64.0	0.5
	38.5	0.491	0.441	0.000954	63.5	
1	28.7	0.886	0.836	0.001808	60.2	0.5
	28.8	0.880	0.830	0.001693	59.7	
$1\frac{1}{2}$	23.4	1.253	1.203	0.002600	57.8	0.5
	23.5	1.245	1.195	0.002580	57.3	
3	15.5	2.303	2.253	0.004880	54.2	0.5
	15.6	2.283	2.233	0.004830	53.7	

#### G. Applicability of the Freundlich Adsorption Equation to Dye Retention Results.

The exact mechanism of the adsorption of dye on fiber is still a matter of dispute. Some of the available evidence shows that adsorption is due to chemical combination of the dye with the free valences of atoms on the surface of the fibers, while in other cases, the evidence indicates that adsorption is due to the retention of the dye by capillary action in the exceedingly fine pores of the fiber. No doubt, in some cases, the two phenomena coexist.

Whatever the mechanism of adsorption, the equilibrium between the dye and the adsorbing fiber has been expressed by the Freundlich equation<sup>8</sup>,

$$a = xc^{1/n}$$

where  $a$  is the amount of dye adsorbed per unit mass of adsorbing material, and  $c$  is the amount of dye in the white water which is in equilibrium with the reacting materials;  $x$  and  $n$  are constants. For convenience in this investigation  $a$  and  $c$  will be expressed in grams of dye. The value of  $n$  determines the firmness of the dye retention and  $x$  is proportional to the active surface of the fiber. While it has been claimed that this equation has theoretical basis, it is probably safer to regard it as empirical in

character. From the form of the equation, it is obvious that if one plot the logarithm of the weight of dye in a sheet against the logarithm of the weight of dye in the white water, the equation demands that the data follow a straight line, the slope of which is equal to the exponent,  $1/n$ . This offers a ready means of testing the applicability of the Freundlich equation to any given set of data.

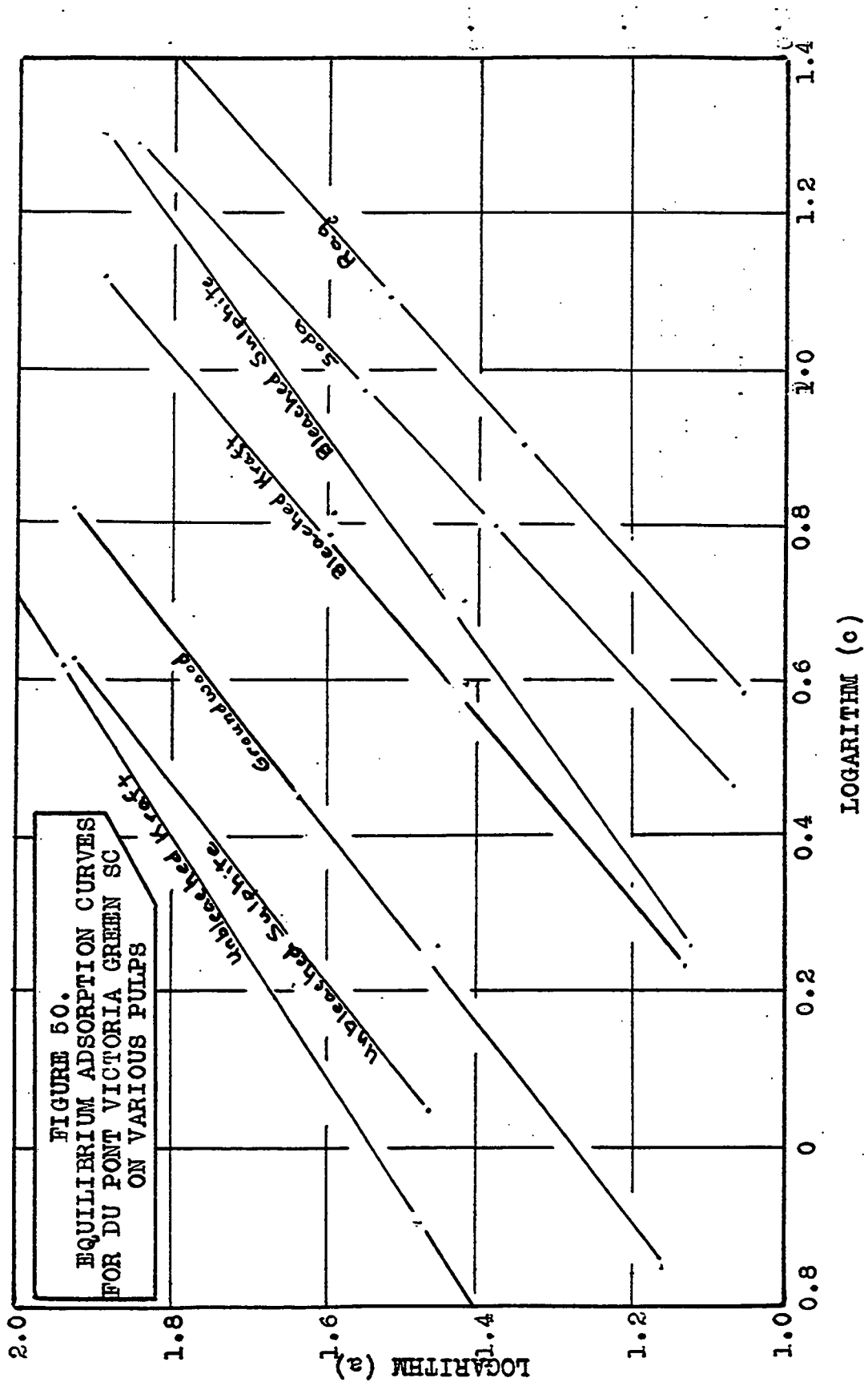
The quantitative data obtained in this investigation has given itself very favorably in application to this equation. Straight lines were obtained in every case with the exception of a few very slight deviations. As illustrations, the adsorption isotherms of seven pulps dyed with the three typical dyes in four different strengths ( $\frac{1}{2}$ , 1,  $1\frac{1}{2}$ , and 3 pound dyeings) have been calculated. The calculated data are tabulated in tables 31, 32, and 33, and are shown graphically in figures 50, 51, and 52.

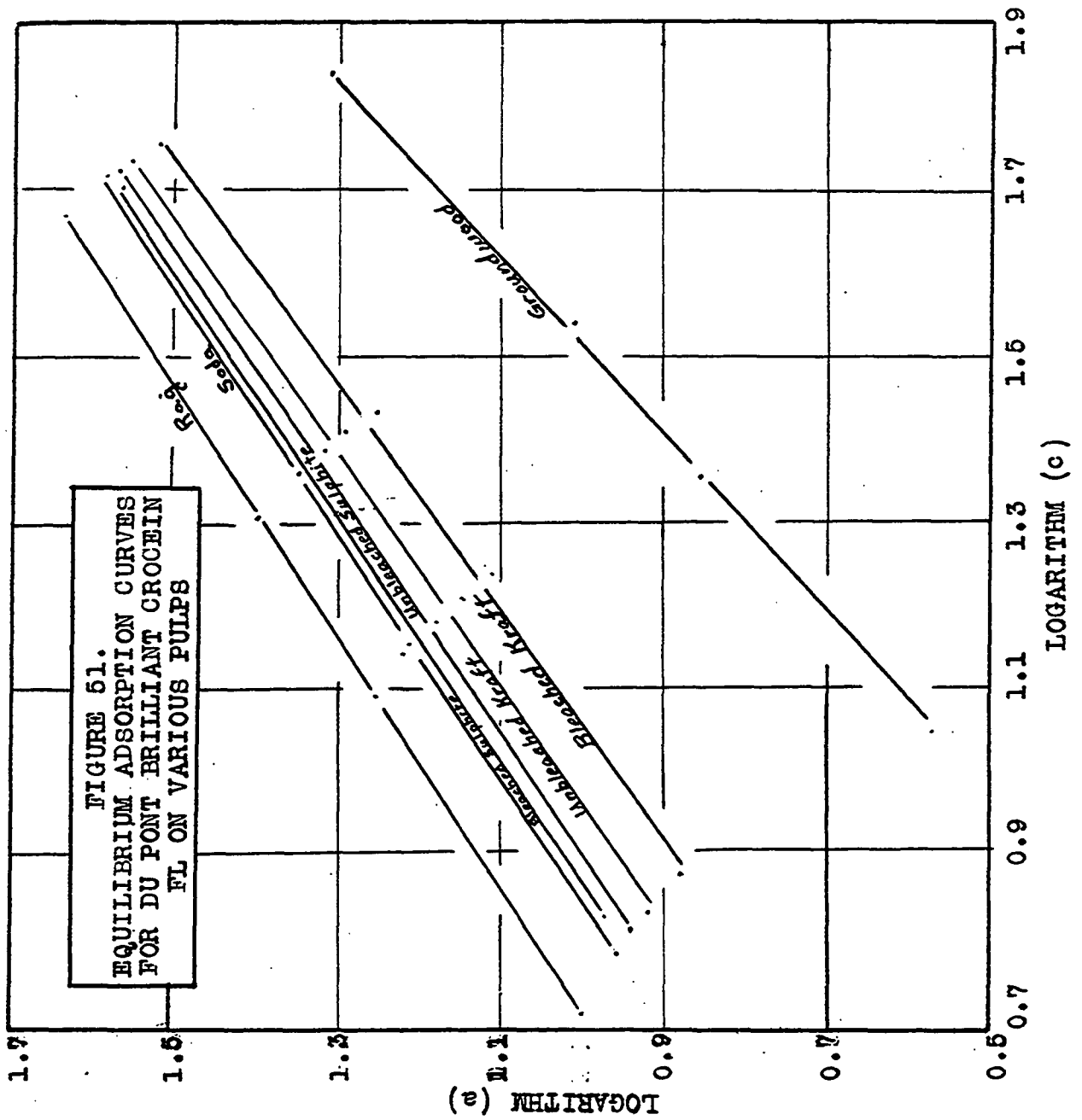
These results are typical of those which would be obtained under any of the conditions in this investigation and do show that the Freundlich adsorption equation represents the retention data.

TABLE 31.					
CALCULATION OF RETENTION DATA FOR DU PONT VICTORIA GREEN SC					
ON VARIOUS PULPS IN CONNECTION WITH THE FREUNDLICH ADSORPTION					
EQUATION					
Pulp	Pounds Dye per 1000 Pounds Pulp	Gms. Dye in Sheet(a)	Gms. Dye in White Water(b)	Logarithm 10000(a)	Logarithm 10000(c)
Soda	$\frac{1}{2}$	.00121	.00029	1.0828	0.4624
	1	.00237	.00063	1.3747	0.7993
	$1\frac{1}{2}$	.00355	.00095	1.5502	0.9777
	3	.00702	.00192	1.8463	1.2833
Rag	$\frac{1}{2}$	.00112	.00038	1.0492	0.5798
	1	.00220	.00080	1.3424	0.9031
	$1\frac{1}{2}$	.00327	.00123	1.5145	1.0899
	3	.00639	.00261	1.8055	1.4166
Ground- wood	$\frac{1}{2}$	.00143	.00007	1.1553	-1.8451
	1	.00282	.00018	1.4502	0.2553
	$1\frac{1}{2}$	.00422	.00028	1.6253	0.4472
	3	.00834	.00066	1.9212	0.8195
Unbleached Sulphite	$\frac{1}{2}$	.00147	.00003	1.1673	-1.4771
	1	.00289	.00011	1.4609	0.0414
	$1\frac{1}{2}$	.00432	.00018	1.6355	0.2553
	3	.00858	.00042	1.9335	0.6232
Bleached Sulphite	$\frac{1}{2}$	.00132	.00018	1.1206	0.2553
	1	.00259	.00041	1.4137	0.6096
	$1\frac{1}{2}$	.00386	.00064	1.5866	0.8062
	3	.00765	.00135	1.8337	1.01303
Unbleached Kraft	$\frac{1}{2}$	.00148	.00003	1.1688	-1.3979
	1	.00292	.00008	1.4654	-1.9031
	$1\frac{1}{2}$	.00434	.00016	1.6375	0.2041
	3	.00859	.00041	1.9340	0.6128
Bleached Kraft	$\frac{1}{2}$	.00133	.00017	1.1212	0.2355
	1	.00261	.00039	1.4166	0.5911
	$1\frac{1}{2}$	.00389	.00061	1.5899	0.7853
	3	.00771	.00129	1.8871	1.1106

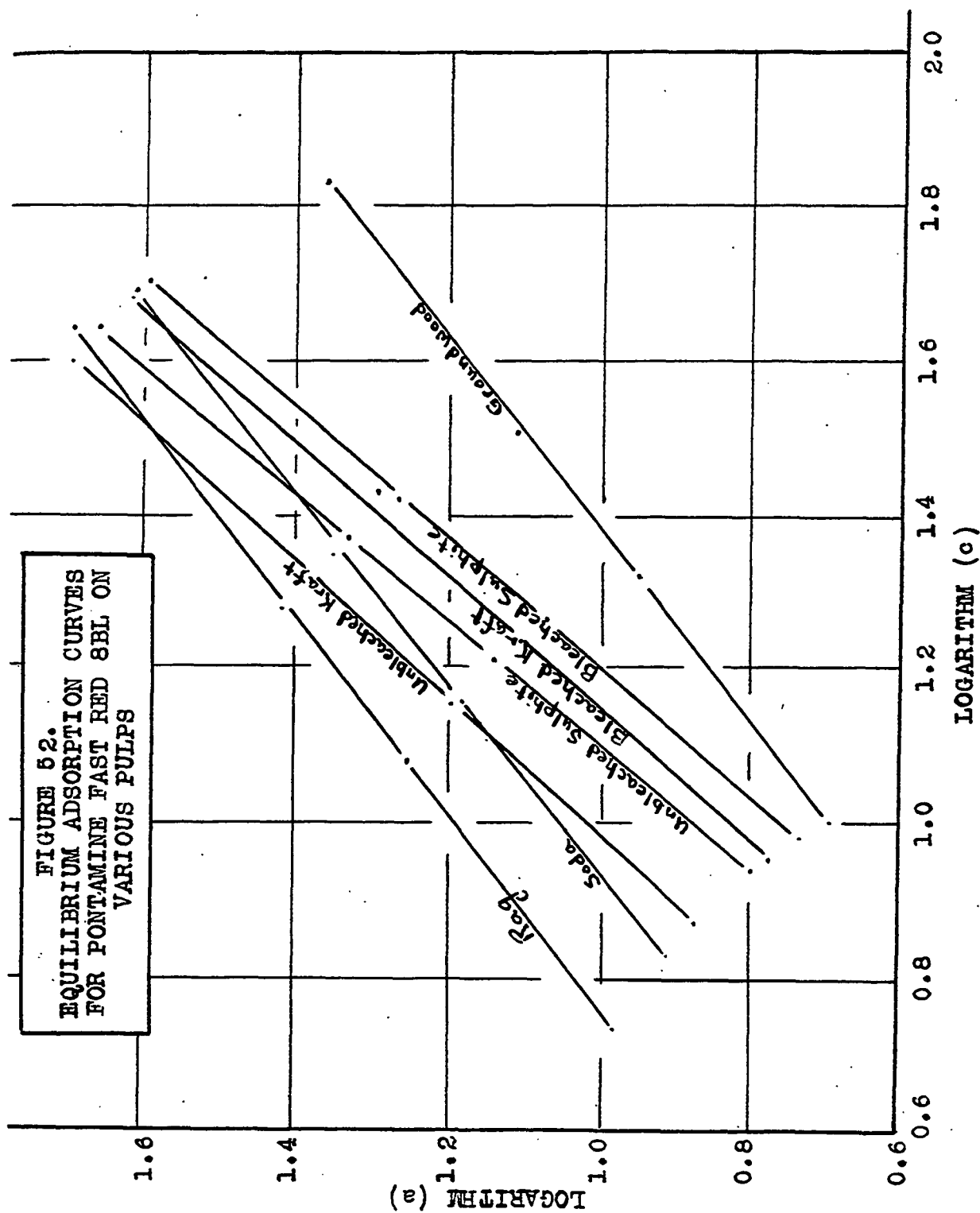
TABLE 32.					
CALCULATION OF RETENTION DATA FOR DU PONT BRILLIANT CROCEIN					
FL ON VARIOUS PULPS IN CONNECTION WITH THE FREUNDLICH					
ADSORPTION EQUATION					
Pulp	Pounds Dye per 1000 Pounds Pulp	Gms. Dye in Sheet(a)	Gms. Dye in White Water(c)	Logarithm 10000(a)	Logarithm 10000(c)
Soda	$\frac{1}{2}$	.00090	.00060	0.9542	0.7782
	1	.00162	.00138	1.2095	1.1399
	$1\frac{1}{2}$	.00223	.00227	1.3485	1.3560
	3	.00378	.00522	1.5775	1.7177
Rag	$\frac{1}{2}$	.00099	.00051	0.9956	0.7076
	1	.00179	.00122	1.2516	1.0855
	$1\frac{1}{2}$	.00248	.00202	1.3945	1.3054
	3	.00424	.00476	1.6274	1.6776
Ground- wood	$\frac{1}{2}$	.00038	.00112	0.5740	1.0512
	1	.00071	.00229	0.8537	1.3588
	$1\frac{1}{2}$	.00104	.00346	1.0151	1.5397
	3	.00206	.00694	1.3139	1.8414
Unbleached Sulphite	$\frac{1}{2}$	.00086	.00064	0.9370	0.8028
	1	.00150	.00150	1.1761	1.1761
	$1\frac{1}{2}$	.00206	.00244	1.3139	1.3874
	3	.00376	.00524	1.5752	1.7193
Bleached Sulphite	$\frac{1}{2}$	.00093	.00067	0.9685	0.8261
	1	.00161	.00139	1.2080	1.1415
	$1\frac{1}{2}$	.00221	.00229	1.3444	1.3598
	3	.00390	.00510	1.5611	1.7076
Unbleached Kraft	$\frac{1}{2}$	.00082	.00068	0.9128	0.8338
	1	.00139	.00161	1.1430	1.2068
	$1\frac{1}{2}$	.00192	.00258	1.2821	1.4123
	3	.00355	.00545	1.5502	1.7364
Bleached Kraft	$\frac{1}{2}$	.00075	.00075	0.8750	0.8750
	1	.00128	.00172	1.1060	1.2360
	$1\frac{1}{2}$	.00178	.00272	1.2504	1.4046
	3	.00329	.00571	1.5166	1.7570

TABLE 33					
CALCULATION OF RETENTION DATA FOR PONTAMINE FAST RED 8BL					
ON VARIOUS PULPS IN CONNECTION WITH THE FREUNDLICH					
ADSORPTION EQUATION					
Pulp	Pounds Dye per 1000 Pounds Pulp	Gms. Dye in Sheet(a)	Gms. Dye in White Water(c)	Logarithm 10000(a)	Logarithm 10000(c)
Soda	$\frac{1}{2}$	.00082	.00068	0.9165	0.8293
	1	.00156	.00144	1.1931	1.1584
	$1\frac{1}{2}$	.00226	.00224	1.3541	1.3504
	3	.00409	.00491	1.6117	1.6911
Rag	$\frac{1}{2}$	.00096	.00054	0.9814	0.7340
	1	.00180	.00120	1.2553	1.0792
	$1\frac{1}{2}$	.00261	.00189	1.4166	1.2765
	3	.00498	.00402	1.6972	1.6042
Ground- wood	$\frac{1}{2}$	.00049	.00101	0.6949	1.0022
	1	.00090	.00210	0.9542	1.3222
	$1\frac{1}{2}$	.00129	.00321	1.1106	1.5065
	3	.00239	.00670	1.3617	1.8261
Unbleached Sulphite	$\frac{1}{2}$	.00063	.00087	0.7993	0.9395
	1	.00138	.00162	1.1399	1.2095
	$1\frac{1}{2}$	.00215	.00235	1.3324	1.3711
	3	.00458	.00442	1.6609	1.6454
Bleached Sulphite	$\frac{1}{2}$	.00054	.00096	0.7324	0.9832
	1	.00117	.00183	1.0682	1.2625
	$1\frac{1}{2}$	.00184	.00266	1.2648	1.4249
	3	.00395	.00505	1.5966	1.7033
Unbleached Kraft	$\frac{1}{2}$	.00075	.00075	0.8751	0.8751
	1	.00157	.00143	1.1847	1.1553
	$1\frac{1}{2}$	.00241	.00209	1.3820	1.3201
	3	.00496	.00404	1.6955	1.6064
Bleached Kraft	$\frac{1}{2}$	.00060	.00090	0.7782	0.9542
	1	.00128	.00172	1.1072	1.2355
	$1\frac{1}{2}$	.00197	.00253	1.2945	1.4031
	3	.00413	.00487	1.6160	1.6875









## CHAPTER V

### CONCLUSIONS

In reviewing the retention data obtained above by analyzing white waters spectrophotometrically, a number of conclusions can be made.

At very low dye concentrations, relatively large percentages of dye are held on the fiber. As the dye concentration is increased, the amount held on the fiber at the end of a given time also increases, but at a much less rapid rate than would correspond to linear distribution.

Basic dyestuffs are attracted strongly by lignified fibers giving dye retentions above ninety per cent. These fibers, however, when bleached, attract this dyestuff to a lesser extent except when the character of the lignin has been changed by the addition of small amounts of bleach. It can be said without reservation that the fiber is the most important variable in so far as basic dyes are concerned. The temperature, time, and hydrogen ion concentration of the reaction mixture have very little effect on the retention of this class of dyestuff while increased consistencies and lower free nesses show considerable increases in retention.

With acid dyestuffs, the fiber variable is not so important to dye retention. Relatively small changes are obtained when the fiber is changed giving higher retentions for the purer pulps except in the case where the lignin present has been changed by the action of small amounts of bleach. The temperature of the reaction apparently has very little effect on the retention of acid dyes. Hydrogen ion concentration affects the retention probably more than any other variable, giving the higher results at a pH of less than five. With high consistencies and low freenesses, increases in retention are noticed, while increases in time have only small effects.

The retention of direct dyestuffs are affected some by the type of pulp being dyed. Bleached and refined pulps produce slightly higher retentions than the highly lignified pulps. If lignin is present, its character when slightly bleached decreases the retention to some extent when chlorinated. The hydrogen ion concentration and the temperature of the reaction mixture have very little effect on the retention. Increased time of reaction, increased consistency, and decreased freeness, however, produce considerable increases in the retention of this class of dyestuff.

In addition, a method for determining dye retention was developed which uses the Kubelka and Monk equation as its fundamental principle. The results obtained by this method were in good agreement with those obtained by the method using Beer's law. From this it can be concluded that the Kubelka and Monk equation, which shows scattering and absorption coefficients as a function of reflectivity, is sound from the standpoint of applicability.

With proof of the validity of this equation certain, it may possibly be used as a stepping stone in developing a method for predicting the shade which would be obtained for various dyes on various stocks. In addition, it is not at all impossible to conceive that this equation could be used in control work, especially in the case of coated papers where one hundred per cent retention is obtained.

In the course of this investigation the dye retention results were applied to the Freundlich adsorption equation. It was found that, when the logarithm of the weight of dye adsorbed is plotted against the logarithm of the weight of dye left in the white water, straight adsorption isotherms were obtained. No definite generalization concerning the

behavior of adsorption can be formulated, but, with such good agreement of these results with the equation in question, it is believed that the method as studied and described will be of value in future work dealing with a study of the mechanism of paper dyeing.

## CHAPTER VI

### SUMMARY

1. A spectrophotometric method for determining dye retention was developed which uses Beer's law as its fundamental principle.
2. Dye retentions for twenty dyes were determined on fourteen pulps under varying conditions of hydrogen ion concentration, consistency, temperature, time, and freeness.
3. A spectrophotometric method for determining dye retention was developed which uses the Kubelka and Monk equation as its fundamental principle. The results obtained by this method checked the results obtained by the method outlined above, thereby establishing the validity of the applicability of this equation.
4. The Freundlich adsorption equation was found to represent the retention data.

## CHAPTER VII

### BIBLIOGRAPHY

1. Clark, E.P., J. Assoc. Official Agr. Chem., 15: 136-40 (1932). C.A. 26: 2395 (1932).
2. Hardy, A.C. and Perrin, F.H., "The Principles of Optics," page 24-5, New York, McGraw-Hill Book Company, Inc., 1932.
3. Holzer, W.F., Tech. Assoc. Papers, 18, no. 1: 517-29 (1935).
4. Ritter, G.J., Seborg, R.M., and Mitchell, R.L., Ind. Eng. Chem. Anal. Ed., 4: 202-4 (1932).
5. Rys, S., Tech. Assoc. Papers, 11, no. 1: 181-7 (1928).
6. Steele, F.A., Tech. Assoc. Papers, 18, no. 1: 299-304 (1935).
7. T. A. P. P. I. Standard T 205m.
8. Walker, W.H., Lewis, W.K., and McAdams, W.H., "Principles of Chemical Engineering," page 645-7, New York, McGraw-Hill Book Company, Inc., 1927.
9. Wiles, R.H., Tech. Assoc. Papers, 17, no. 1: 146-8 (1934).
10. Willetts, W.R., Tech. Assoc. Papers, 14, no. 1: 121 (1931).

**APPENDIX A**  
**CALCULATIONS**



TABLE 1A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN  
ON SULPHITE PULP BLEACHED TO VARIOUS DEGREES

% BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
0	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.7	97.9	1.991	.000064	.000030	.0015	.001470	97.9
	1 1/2	82.7	96.8	1.986	.000234	.000110	.0030	.002890	96.5
	2	81.7	95.7	1.981	.000383	.000180	.0045	.004320	96.0
	3	79.0	92.5	1.965	.000995	.000420	.0090	.008580	95.3
3	0	82.4	98.3	1.993	0	0	0	0	100.0
	1	82.3	98.2	1.992	.000032	.000015	.0015	.001485	98.9
	1 1/2	81.7	97.5	1.989	.000117	.000055	.0030	.002945	98.2
	2	81.1	96.8	1.986	.000234	.000110	.0045	.004390	97.6
	3	78.8	94.0	1.973	.000638	.000300	.0090	.008700	96.7
6	0	85.0	98.3	1.993	0	0	0	0	100.0
	1	84.6	97.9	1.991	.000059	.000028	.0015	.001472	98.2
	1 1/2	83.9	97.0	1.987	.000191	.000090	.0030	.002910	97.0
	2	82.9	95.9	1.982	.000340	.000160	.0045	.004340	96.4
	3	80.3	92.9	1.968	.000830	.000390	.0090	.008610	95.7

(Continued)

TABLE 1A (Continued)

% BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS-MISSION	COR-RECTED TRANS-MISSION	LOG COR-RECTED TRANS-MISSION	CONCEN-TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN-TION
10	0	87.3	98.3	1.993	0	0	0	0	100.0
	1	86.7	97.7	1.990	.000040	.000040	.0015	.001460	97.3
	1 1/2	85.8	96.6	1.985	.000120	.000120	.0030	.002880	96.0
	2	84.5	95.1	1.978	.000220	.000220	.0045	.004280	95.1
	3	80.7	90.8	1.958	.001105	.000520	.0090	.008480	94.2
20	0	89.0	98.3	1.993	0	0	0	0	100.0
	1	87.7	96.8	1.986	.000223	.000105	.0015	.001395	93.0
	1 1/2	85.6	94.6	1.976	.000553	.000260	.0030	.002740	91.2
	2	83.2	92.0	1.964	.000936	.000440	.0045	.004060	90.2
	3	77.0	85.1	1.930	.002080	.000980	.0090	.008020	89.1
30	0	91.2	98.3	1.993	0	0	0	0	100.0
	1	88.8	95.7	1.981	.000383	.000180	.0015	.001320	87.9
	1 1/2	85.8	92.5	1.966	.000866	.000407	.0030	.002593	86.4
	2	82.8	89.3	1.951	.001360	.000640	.0045	.003860	85.8
	3	74.7	80.5	1.906	.002870	.001350	.0090	.007650	85.0

TABLE 2A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT OROCEIN FL  
ON SULPHITE PULP BLEACHED TO VARIOUS DEGREES

% BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
0	0 1 1 1 3	75.1 72.2 68.5 64.5 54.4	94.0 90.4 85.7 80.7 68.1	1.973 1.956 1.933 1.907 1.833	0 .001350 .003190 .005200 .011100	0 .000635 .001500 .002440 .005240	0 .0015 .0030 .0045 .0090	0 .000865 .001500 .002060 .003760	100.0 57.7 50.0 45.7 41.8
3	0 1 1 1 3	72.8 69.9 65.8 61.6 50.8	94.0 90.2 84.9 79.4 65.5	1.973 1.955 1.929 1.900 1.816	0 .001470 .003520 .005730 .012400	0 .000690 .001652 .002695 .005850	0 .0015 .0030 .0045 .0090	0 .000810 .001348 .001805 .003150	100.0 54.0 44.9 40.1 35.0
6	0 1 1 1 3	76.2 73.1 69.0 64.7 53.7	94.0 90.2 85.1 79.8 66.2	1.973 1.955 1.930 1.902 1.821	0 .001440 .003450 .005640 .012000	0 .000676 .001620 .002650 .005660	0 .0015 .0030 .0045 .0090	0 .000824 .001380 .001850 .003340	100.0 54.9 46.0 41.2 37.1

(Continued)

TABLE 2A (Continued)

% BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS-MISSION	COR-RECTED TRANS-MISSION	LOG COR-RECTED TRANS-MISSION	CONCENTRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETENTION
10	0	78.0	94.0	1.973	0	0	0	0	100.0
	1	75.0	90.4	1.956	.000654	.000654	.0015	.000846	56.4
	1 1/2	70.8	85.3	1.931	.001590	.001590	.0030	.001410	47.1
	2	66.5	80.2	1.904	.002570	.002570	.0045	.001930	42.8
	3	55.6	67.0	1.826	.005480	.005480	.0090	.003520	39.1
20	0	79.3	94.0	1.973	0	0	0	0	100.0
	1	71.4	90.6	1.957	.000600	.000600	.0015	.000900	60.0
	1 1/2	72.6	86.1	1.935	.001440	.001440	.0030	.001560	52.0
	2	68.5	81.3	1.910	.002360	.002360	.0045	.002140	47.6
	3	57.7	68.4	1.835	.005180	.005180	.0090	.003820	42.4
30	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.2	90.2	1.955	.000670	.000670	.0015	.000930	62.0
	1 1/2	75.0	86.5	1.937	.001385	.001385	.0030	.001615	53.8
	2	70.8	81.7	1.912	.002290	.002290	.0045	.002210	49.0
	3	59.4	68.5	1.836	.005100	.005100	.0090	.003900	43.3

TABLE 3A  
CALCULATION OF RETENTION OF PONTAMINE FAST RED 8BL  
ON SULPHITE PULP BLEACHED TO VARIOUS DEGREES

% BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS-MISSION	CORRECTED TRANS-MISSION	LOG CORRECTED TRANS-MISSION	CONCENTRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETENTION
0	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	72.9	89.9	1.954	.001340	.000630	.0015	.000870	58.0
	1 1/2	68.8	84.9	1.929	.002940	.001380	.0030	.001620	54.0
	3	64.8	80.0	1.903	.004570	.002150	.0045	.002350	52.2
3		53.8	66.4	1.822	.009750	.004580	.0090	.004420	49.1
	0	74.8	94.5	1.975	0	0	0	0	100.0
	1	70.4	88.9	1.949	.001640	.000773	.0015	.000727	48.4
	1 1/2	65.8	83.0	1.919	.003550	.001670	.0030	.001330	44.3
6		61.3	77.4	1.889	.005520	.002595	.0045	.001905	42.3
	3	49.2	62.1	1.793	.011650	.005480	.0090	.003520	39.1
	0	77.4	94.5	1.975	0	0	0	0	100.0
	1	73.2	89.3	1.951	.001535	.000722	.0015	.000778	51.8
	1 1/2	68.7	83.8	1.923	.003290	.001545	.0030	.001455	48.5
	3	64.3	78.5	1.895	.005130	.002410	.0045	.002090	46.5
		52.3	63.8	1.805	.010900	.005110	.0090	.003890	43.2

(Continued)

TABLE 3A (Continued)

% BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS-MISSION	CORRECTED TRANS-MISSION	LOG CORRECTED TRANSMISSION	CONCENTRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETENTION
10	0	79.9	94.5	1.975	0	0	0	0	100.0
	1	75.8	89.7	1.953	.001380	.000648	.0015	.000852	56.8
	1	71.7	84.7	1.928	.003020	.001420	.0030	.001580	52.6
	1	67.4	79.6	1.901	.004700	.002210	.0045	.002290	50.9
	3	56.4	66.7	1.824	.009670	.004640	.0090	.004360	48.4
20	0	81.2	94.5	1.975	0	0	0	0	100.0
	1	77.5	90.2	1.955	.001280	.000601	.0015	.000899	59.9
	1	73.5	85.5	1.932	.002770	.001300	.0030	.001700	56.7
	1	69.4	80.7	1.907	.004320	.002030	.0045	.002470	54.9
	3	58.3	67.8	1.831	.009240	.004340	.0090	.004660	51.8
30	0	83.1	94.5	1.975	0	0	0	0	100.0
	1	79.7	90.6	1.957	.001150	.000540	.0015	.000960	64.0
	1	75.9	86.3	1.936	.002490	.001170	.0030	.001830	61.0
	1	72.1	82.0	1.914	.003910	.001840	.0045	.002660	59.1
	3	61.3	69.7	1.843	.008400	.003950	.0090	.005050	56.2

TABLE 4A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SO  
ON KRAFT PULP BLEACHED TO VARIOUS DEGREES

% BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS-MISSION	COR-RECTED TRANS-MISSION	LOG COR-RECTED TRANS-MISSION	CONCENTRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETENTION
0	0	80.1	98.3	1.993	0	0	0	0	100.0
	1	79.7	97.9	1.991	.000053	.000025	.0015	.001475	98.3
	1 1/2	79.3	97.3	1.988	.000170	.000080	.0030	.002920	97.2
	2	78.1	95.9	1.982	.000340	.000160	.0045	.004340	96.6
	3	75.4	92.5	1.966	.000872	.000410	.0090	.008590	95.5
5	0	82.1	98.3	1.993	0	0	0	0	100.0
	1	81.6	97.7	1.990	.000085	.000040	.0015	.001460	97.3
	1 1/2	80.5	96.4	1.984	.000276	.000130	.0030	.002870	95.7
	2	79.4	95.1	1.978	.000490	.000230	.0045	.004270	94.8
	3	75.3	90.2	1.955	.001255	.000590	.0090	.008410	93.4
15	0	84.1	98.3	1.993	0	0	0	0	100.0
	1	83.0	97.0	1.987	.000185	.000087	.0015	.001413	94.2
	1 1/2	81.4	95.1	1.978	.000468	.000220	.0030	.002780	92.6
	2	79.7	93.1	1.969	.000788	.000370	.0045	.004130	91.6
	3	74.4	86.9	1.939	.001785	.000840	.0090	.008160	90.7

(Continued)

TABLE 4A (Continued)

% BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS-MISSION	COR-RECTED TRANS-MISSION	LOG COR-RECTED TRANS-MISSION	CONCEN-TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN-TION
30	0 1/4	86.3	98.3	1.993	0	0	0	0	100.0
	1 1/4	84.6	96.4	1.984	.000130	.0015	.001370	.001370	91.4
	1 1/2	82.5	94.0	1.973	.000310	.0030	.002690	.002690	89.8
	1 3/4	80.2	91.4	1.961	.000500	.0045	.004000	.004000	88.8
	3	73.2	83.4	1.921	.001120	.0090	.007880	.007880	87.6
40	0 1/4	88.1	98.3	1.993	0	0	0	0	100.0
	1 1/4	85.9	95.9	1.982	.000366	.0015	.001328	.001328	88.5
	1 1/2	83.2	92.9	1.968	.000830	.0030	.002610	.002610	87.1
	1 3/4	80.3	89.7	1.953	.001300	.0045	.003890	.003890	86.5
	3	72.8	81.3	1.910	.002740	.0090	.007710	.007710	85.7



TABLE 5A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT CROCEIN FL  
ON KRAFT PULP BLEACHED TO VARIOUS DEGREES

% BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS-MISSION	CORRECTED TRANS-MISSION	LOG CORRECTED TRANS-MISSION	CONCENTRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETENTION
0	0	70.2	94.0	1.973	0	0	0	0	100.0
	1	67.3	90.2	1.955	.001450	.000682	.0015	.000818	54.5
	1 1/2	63.5	85.1	1.930	.003430	.001610	.0030	.001390	46.3
	3	59.7	80.0	1.903	.005500	.002585	.0045	.001915	42.6
5	0	50.0	67.0	1.826	.011600	.005450	.0090	.003550	39.4
	1	73.4	94.0	1.973	0	0	0	0	100.0
	1 1/2	69.8	89.3	1.951	.001785	.000840	.0015	.000660	44.0
	3	65.5	83.8	1.923	.004020	.001890	.0030	.001110	37.0
15	0	61.2	78.3	1.894	.006240	.002930	.0045	.001570	34.2
	1	50.0	64.0	1.806	.013200	.006200	.0090	.002800	31.1
	1 1/2	75.1	94.0	1.973	0	0	0	0	100.0
	3	71.5	89.5	1.952	.001710	.000804	.0015	.000696	46.4
	0	67.2	84.1	1.925	.003840	.001805	.0030	.001195	39.8
	1	63.0	78.9	1.897	.006030	.002835	.0045	.001665	37.0
	1 1/2	51.9	65.0	1.813	.012640	.005940	.0090	.003060	34.0

(Continued)

TABLE 5A (Continued)

% BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS-MISSION	CORRECTED TRANS-MISSION	LOG CORRECTED TRANS-MISSION	CONCENTRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETENTION
30	0	77.5	94.0	1.973	0	0	0	0	100.0
	1	73.8	89.5	1.952	.001660	.000780	.0015	.000720	48.0
	1 1/2	69.7	84.5	1.927	.003770	.001775	.0030	.001225	40.8
	2	65.2	79.1	1.898	.005950	.002795	.0045	.001705	37.9
	3	54.0	65.5	1.816	.012450	.005860	.0090	.003140	34.9
40	0	80.1	94.0	1.973	0	0	0	0	100.0
	1	76.4	89.7	1.953	.001595	.000751	.0015	.000749	49.9
	1 1/2	72.0	84.5	1.927	.003660	.001722	.0030	.001278	42.6
	2	67.6	79.4	1.900	.005780	.002720	.0045	.001780	39.5
	3	56.2	65.9	1.819	.012150	.005713	.0090	.003287	36.5

TABLE 6A

CALCULATION OF RETENTION OF PONTAMINE FAST RED 8BL  
ON KRAFT PULP BLEACHED TO VARIOUS DEGREES

PERCENT BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS-MISSION	CORRECTED TRANS-MISSION	LOG CORRECTED TRANS-MISSION	CONCENTRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETENTION
0	0	71.9	94.5	1.975	0	0	0	0	100.0
	1	67.8	89.1	1.950	.001590	.000750	.0015	.000750	50.0
	1	63.7	83.8	1.923	.003340	.001570	.0030	.001430	47.6
	1	59.8	78.5	1.895	.005130	.002410	.0045	.002090	46.6
	3	49.2	64.6	1.810	.010540	.004960	.0090	.004040	45.0
5	0	75.0	94.5	1.975	0	0	0	0	100.0
	1	70.6	88.9	1.949	.001660	.000782	.0015	.000818	54.5
	1	67.1	84.5	1.927	.003070	.001444	.0030	.001556	51.9
	1	63.2	79.6	1.901	.004740	.002230	.0045	.002270	50.4
	3	52.1	65.6	1.817	.010100	.004640	.0090	.004360	48.4
15	0	76.9	94.5	1.975	0	0	0	0	100.0
	1	72.4	90.0	1.954	.001375	.000647	.0015	.000853	56.8
	1	68.2	84.9	1.929	.002930	.001375	.0030	.001625	54.2
	1	64.4	80.2	1.904	.004510	.002120	.0045	.002380	52.8
	3	53.9	67.1	1.827	.009450	.004440	.0090	.004560	50.7

TABLE 6A (Continued)

PERCENT BLEACH	POUNDS DYE PER 1000 POUNDS PULP	TRANS-MISSION	CORRECTED TRANS-MISSION	LOG CORRECTED TRANS-MISSION	CONCENTRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETENTION
30	0	79.0	94.5	1.975	0	0	0	0	100.0
	1	75.2	90.0	1.954	.000618	.0015	.0015	.000882	58.8
	1 1/2	71.3	85.3	1.931	.001310	.0030	.0030	.001690	56.3
	2	67.5	80.7	1.907	.002030	.0045	.0045	.002470	54.9
	3	57.0	68.2	1.834	.004230	.0090	.0090	.004770	53.0
40	0	81.9	94.5	1.975	0	0	0	0	100.0
	1	78.2	90.2	1.955	.000600	.0015	.0015	.000900	60.0
	1 1/2	74.3	85.7	1.933	.001280	.0030	.0030	.001720	57.4
	2	70.3	81.1	1.909	.001970	.0045	.0045	.002530	56.2
	3	59.3	68.4	1.835	.004130	.0090	.0090	.004870	54.1

TABLE 7A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SO  
ON BLEACHED SODA, RAG, AND GROUNDWOOD PULPS

PULP	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
Soda	0	84.7	98.3	1.993	0	0	0	0	100.0
	$\frac{1}{2}$	81.1	94.2	1.974	.000617	.000290	.0015	.001210	80.7
	1	77.0	89.5	1.952	.001340	.000630	.0030	.002370	79.0
	$1\frac{1}{2}$	73.6	85.5	1.932	.002020	.000950	.0045	.003550	78.9
Rag	3	63.8	74.1	1.870	.004080	.001920	.0090	.007020	78.0
	0	92.1	98.3	1.993	0	0	0	0	100.0
	$\frac{1}{2}$	87.0	92.9	1.968	.000810	.000380	.0015	.001120	74.7
	1	81.8	87.3	1.941	.001700	.000800	.0030	.002200	73.3
Ground- wood	$1\frac{1}{2}$	76.6	81.8	1.913	.002620	.001230	.0045	.003270	72.6
	3	62.7	67.0	1.826	.005550	.002610	.0090	.006390	71.0
	0	77.1	98.3	1.993	0	0	0	0	100.0
	$\frac{1}{2}$	76.3	97.3	1.988	.000149	.000070	.0015	.001430	95.2
	1	75.0	95.7	1.981	.000383	.000180	.0030	.002820	94.0
	$1\frac{1}{2}$	74.0	94.4	1.975	.000595	.000280	.0045	.004220	93.6
	3	69.9	89.1	1.950	.001400	.000660	.0090	.008340	92.8

TABLE 8A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT CROCEIN FL  
ON BLEACHED SODA, RAG, AND GROUNDWOOD PULPS

PULP	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
Soda	0	81.8	94.0	1.973	0	0	0	0	100.0
	1	78.9	90.6	1.957	.001275	.000600	.0015	.000900	60.0
	1 1/2	75.2	86.5	1.937	.002940	.001380	.0030	.001620	54.0
	2	71.1	81.7	1.912	.004830	.002270	.0045	.002230	49.5
Rag	3	59.3	68.1	1.833	.011100	.005220	.0090	.003780	42.0
	0	83.5	94.0	1.973	0	0	0	0	100.0
	1	80.8	91.0	1.959	.001084	.000510	.0015	.000990	66.0
	1 1/2	77.6	87.3	1.941	.002590	.001215	.0030	.001785	59.5
Ground- wood	2	73.7	83.0	1.919	.004300	.002020	.0045	.002480	55.0
	3	62.2	70.0	1.845	.010130	.004760	.0090	.004240	47.2
	0	74.1	94.0	1.973	0	0	0	0	100.0
	1	69.2	87.7	1.943	.002390	.001125	.0015	.000375	25.0
	1 1/2	64.4	81.7	1.912	.004870	.002286	.0030	.000714	23.8
	2	59.8	75.9	1.880	.007370	.003465	.0045	.001035	23.0
	3	48.2	61.1	1.786	.014760	.006940	.0090	.002060	22.9

TABLE 9A  
CALCULATION OF RETENTION OF PONTAMINE FAST RED 8BL  
ON BLEACHED SODA, RAG, AND GROUNDWOOD PULPS

PULP	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
Soda	0	82.3	94.5	1.975	0	0	0	0	100.0
	1	78.0	89.5	1.952	.001435	.000675	.0015	.000825	55.0
	1	73.6	84.5	1.927	.003060	.001440	.0030	.001560	52.0
	1	69.2	79.4	1.900	.004770	.002240	.0045	.002260	50.1
	3	56.6	64.9	1.812	.010430	.004910	.0090	.004090	45.5
Rag	0	85.2	94.5	1.975	0	0	0	0	100.0
	1	81.6	90.6	1.957	.001150	.000542	.0015	.000958	63.8
	1	77.5	86.1	1.935	.002550	.001200	.0030	.001800	60.0
	1	73.5	81.7	1.912	.004020	.001890	.0045	.002610	58.0
	3	71.4	69.3	1.841	.008550	.004020	.0090	.004980	55.3
Ground- wood	0	74.9	94.5	1.975	0	0	0	0	100.0
	1	69.4	87.5	1.942	.002140	.001005	.0015	.000495	33.0
	1	63.7	80.3	1.905	.004470	.002100	.0030	.000900	30.0
	1	58.5	73.8	1.868	.006830	.003210	.0045	.001290	28.6
	3	44.8	56.6	1.753	.014250	.006700	.0090	.002300	25.6

TABLE 10A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SO  
ON UNBLEACHED SULPHITE PULP AT VARIOUS pH'S

pH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.3	97.5	1.989	.000128	.000060	.0015	.001440	96.0
	1 1/2	81.9	95.9	1.982	.000362	.000170	.0030	.002830	94.5
	2	80.3	94.0	1.975	.000575	.000270	.0045	.004230	94.1
	3	77.4	90.6	1.957	.001190	.000560	.0090	.008440	93.8
4.9	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.5	97.9	1.991	.000068	.000032	.0015	.001468	97.9
	1 1/2	82.7	96.8	1.986	.000223	.000105	.0030	.002895	96.5
	2	81.7	95.7	1.981	.000383	.000180	.0045	.004320	96.0
	3	79.0	92.5	1.966	.000895	.000420	.0090	.008580	95.3
5.9	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.4	97.7	1.990	.000079	.000037	.0015	.001463	97.7
	1 1/2	82.6	96.8	1.986	.000240	.000113	.0030	.002887	96.2
	2	81.6	95.7	1.981	.000404	.000190	.0045	.004310	95.9
	3	78.4	91.8	1.963	.000980	.000460	.0090	.008540	94.9



TABLE 11A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN 80  
ON BLEACHED SULPHITE PULP AT VARIOUS PH'S

PH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCENTRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	91.2	98.3	1.993	0	0	0	0	100.0
	1	88.4	95.3	1.979	.000453	.000213	.0015	.001287	85.8
	1	85.0	91.6	1.962	.001010	.000475	.0030	.002525	84.8
	1	82.1	88.5	1.947	.001520	.000715	.0045	.003785	84.2
	3	73.0	78.7	1.896	.003210	.001510	.0090	.007490	83.2
4.9	0	91.2	98.3	1.993	0	0	0	0	100.0
	1	88.8	95.7	1.981	.000387	.000182	.0015	.001318	87.9
	1	85.8	92.5	1.966	.000872	.000410	.0030	.002590	86.4
	1	83.0	89.5	1.952	.001360	.000640	.0045	.003860	85.8
	3	74.7	80.5	1.906	.002870	.001350	.0090	.007650	85.0
5.9	0	91.2	98.3	1.993	0	0	0	0	100.0
	1	88.8	95.7	1.981	.000391	.000184	.0015	.001316	87.8
	1	85.8	92.5	1.966	.000880	.000413	.0030	.002587	86.3
	1	82.8	89.3	1.951	.001380	.000650	.0045	.003850	85.6
	3	74.8	80.7	1.907	.002850	.001340	.0090	.007660	85.1

TABLE 12A  
CALCULATION OF RETENTION OF BRILLIANT CROCEIN FL  
ON UNBLEACHED SULPHITE PULP AT VARIOUS PH'S

PH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	75.1	94.0	1.973	0	0	0	0	100.0
	1	72.2	90.4	1.956	.001350	.000635	.0015	.000865	57.7
	1	68.6	85.9	1.934	.003180	.001495	.0030	.001505	50.1
	1	64.5	80.7	1.907	.005220	.002450	.0045	.002050	45.6
	3	54.4	68.1	1.833	.011100	.005230	.0090	.003770	41.9
4.9	0	75.1	94.0	1.973	0	0	0	0	100.0
	1	72.2	90.4	1.956	.001350	.000635	.0015	.000865	57.7
	1	68.3	85.5	1.932	.003190	.001500	.0030	.001500	50.0
	1	64.5	80.7	1.907	.005200	.002445	.0045	.002055	45.7
	3	54.2	67.9	1.832	.011140	.005235	.0090	.003765	41.8
5.9	0	75.1	94.0	1.973	0	0	0	0	100.0
	1	71.7	89.7	1.953	.001590	.000748	.0015	.000752	50.1
	1	67.5	84.5	1.927	.003690	.001735	.0030	.001265	42.2
	1	63.2	79.1	1.898	.005930	.002785	.0045	.001715	38.1
	3	52.1	65.2	1.814	.012600	.005930	.0090	.003070	34.1

TABLE 13A  
CALCULATION OF RETENTION OF BRILLIANT CROCEIN FL  
ON BLEACHED SULPHITE PULP AT VARIOUS PH'S

PH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.7	90.8	1.958	.001200	.000570	.0015	.000930	62.0
	1	74.8	86.3	1.936	.002970	.001395	.0030	.001605	53.5
	1	71.0	82.0	1.914	.004900	.002300	.0045	.002200	48.9
	3	67.9	68.4	1.835	.010900	.005120	.0090	.003880	43.1
4.9	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.7	90.8	1.958	.001210	.000570	.0015	.000930	62.0
	1	74.7	86.3	1.936	.002950	.001385	.0030	.001615	53.8
	1	70.8	81.7	1.912	.004880	.002295	.0045	.002205	49.0
	3	59.3	68.5	1.836	.010850	.005100	.0090	.003900	43.3
5.9	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.1	90.2	1.955	.001470	.000690	.0015	.000810	54.0
	1	73.5	84.9	1.929	.003530	.001660	.0030	.001340	44.7
	1	69.0	79.6	1.901	.005700	.002680	.0045	.001820	40.5
	3	57.0	65.9	1.819	.012200	.005750	.0090	.003250	36.1

TABLE 14A  
CALCULATION OF RETENTION OF FAST RED 8BL ON  
UNBLEACHED SULPHITE PULP AT VARIOUS PH'S

pH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	72.9	90.0	1.954	.00134	.000627	.0015	.000873	58.2
	1 1/2	68.8	84.9	1.929	.00293	.001378	.0030	.001622	54.1
	2	64.8	80.0	1.903	.00458	.002150	.0045	.002350	52.2
	3	53.9	66.5	1.823	.00975	.004580	.0090	.004420	49.1
4.9	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	72.9	90.0	1.954	.00134	.000630	.0015	.000870	58.0
	1 1/2	68.8	84.9	1.929	.00294	.001380	.0030	.001620	54.0
	2	64.8	80.0	1.903	.00458	.002150	.0045	.002350	52.2
	3	53.9	66.5	1.823	.00975	.004580	.0090	.004420	49.1
5.9	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	72.7	89.7	1.953	.00140	.000659	.0015	.000841	56.1
	1 1/2	68.5	84.5	1.927	.00303	.001426	.0030	.001574	52.4
	2	64.5	79.6	1.901	.00474	.002230	.0045	.002270	50.4
	3	53.2	65.6	1.817	.01010	.004740	.0090	.004260	47.3

TABLE 15A  
CALCULATION OF RETENTION OF FAST RED 6BL ON  
BLEACHED SULPHITE PULP AT VARIOUS PH'S

PH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	83.1	94.5	1.975	0	0	0	0	100.0
	$\frac{1}{2}$	79.6	90.6	1.957	.000535	.000535	.0015	.000965	64.4
	1	76.0	86.5	1.937	.001155	.001155	.0030	.001845	61.5
	$1\frac{1}{2}$	72.1	82.0	1.914	.001830	.001830	.0045	.002670	59.4
	3	61.6	70.1	1.846	.003890	.003890	.0090	.005110	56.8
4.9	0	83.1	94.5	1.975	0	0	0	0	100.0
	$\frac{1}{2}$	79.6	90.6	1.957	.000540	.000540	.0015	.000960	64.0
	1	75.9	86.3	1.936	.001170	.001170	.0030	.001830	61.0
	$1\frac{1}{2}$	71.9	81.8	1.913	.001840	.001840	.0045	.002660	59.1
	3	61.4	69.8	1.844	.003950	.003950	.0090	.005050	56.2
5.9	0	83.1	94.5	1.975	0	0	0	0	100.0
	$\frac{1}{2}$	79.4	90.4	1.956	.000568	.000568	.0015	.000932	62.1
	1	75.5	85.9	1.934	.001231	.001231	.0030	.001769	58.9
	$1\frac{1}{2}$	71.6	81.5	1.911	.001920	.001920	.0045	.002580	57.3
	3	60.6	68.9	1.838	.004110	.004110	.0090	.004890	54.3

TABLE 16A

CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SO  
ON UNBLEACHED KRAFT PULP AT VARIOUS pH'S

pH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.4	97.7	1.990	.000083	.000039	.0015	.001461	97.5
	1 1/2	82.7	96.8	1.986	.000213	.000100	.0030	.002900	96.5
	2	81.7	95.7	1.981	.000383	.000180	.0045	.004320	96.0
	3	78.6	92.0	1.964	.000958	.000450	.0090	.008550	94.9
4.9	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.6	97.9	1.991	.000053	.000025	.0015	.001475	98.3
	1 1/2	83.1	97.3	1.988	.000170	.000080	.0030	.002920	97.2
	2	82.2	96.2	1.983	.000340	.000160	.0045	.004340	96.6
	3	79.0	92.5	1.966	.000873	.000410	.0090	.008590	95.5
5.9	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.6	97.9	1.991	.000059	.000028	.0015	.001472	98.2
	1 1/2	82.9	97.1	1.987	.000191	.000090	.0030	.002910	97.1
	2	81.9	95.9	1.982	.000362	.000170	.0045	.004330	96.4
	3	78.8	92.3	1.965	.000915	.000430	.0090	.008580	95.4

TABLE 17A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SO  
ON BLEACHED KRAFT PULP AT VARIOUS PH'S

PH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	91.2	98.3	1.993	0	0	0	0	100.0
	1	88.7	95.7	1.981	.000374	.000176	.0015	.001324	88.2
	1	86.1	92.9	1.968	.000819	.000385	.0030	.002615	87.1
	3	83.2	89.7	1.953	.001300	.000610	.0045	.003890	86.4
4.9	0	75.0	80.9	1.908	.002810	.001320	.0090	.007680	85.4
	0	91.2	98.3	1.993	0	0	0	0	100.0
	1	88.9	95.9	1.982	.000366	.000172	.0015	.001328	88.5
	1	86.1	92.9	1.968	.000819	.000385	.0030	.002615	87.1
5.9	0	83.5	90.0	1.954	.001290	.000605	.0045	.003895	86.5
	3	75.7	81.5	1.911	.002720	.001280	.0090	.007720	85.7
	0	91.2	98.3	1.993	0	0	0	0	100.0
	1	88.9	95.9	1.982	.000366	.000172	.0015	.001328	88.5
5.9	0	86.0	92.7	1.967	.000830	.000390	.0030	.002610	87.0
	1	83.5	90.0	1.954	.001290	.000605	.0045	.003895	86.5
	3	75.1	80.9	1.908	.002790	.001310	.0090	.007690	85.5

TABLE 18A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT OROCEIN FL  
ON UNBLEACHED KRAFT PULP AT VARIOUS PH'S

PH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	75.1	94.0	1.973	0	0	0	0	100.0
	$\frac{1}{2}$	72.0	90.2	1.935	.001440	.000679	.0015	.000821	54.8
	1	68.1	85.3	1.931	.003400	.001600	.0030	.001400	46.6
	$1\frac{1}{2}$	64.0	80.2	1.904	.005470	.002570	.0045	.001930	42.9
	3	53.8	67.3	1.828	.011500	.005400	.0090	.003600	40.0
4.9	0	75.1	94.0	1.973	0	0	0	0	100.0
	$\frac{1}{2}$	72.0	90.2	1.955	.001450	.000683	.0015	.000817	54.5
	1	68.0	85.1	1.930	.003420	.001610	.0030	.001390	46.3
	$1\frac{1}{2}$	63.8	80.0	1.903	.005500	.002585	.0045	.019150	42.6
	3	53.5	67.0	1.826	.011600	.005450	.0090	.003550	39.4
5.9	0	75.1	94.0	1.973	0	0	0	0	100.0
	$\frac{1}{2}$	71.7	89.7	1.953	.001590	.000750	.0015	.000750	50.0
	1	67.5	84.5	1.927	.003660	.001720	.0030	.001280	42.6
	$1\frac{1}{2}$	63.2	79.1	1.898	.005920	.002785	.0045	.017150	38.1
	3	52.4	65.6	1.817	.012400	.005840	.0090	.003160	35.1



TABLE 19A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT CROCEIN FL  
ON BLEACHED KRAFT PULP AT VARIOUS PH'S

PH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	77.7	89.7	1.953	.001585	.000747	.0015	.000753	50.2
	1 1/2	73.3	84.7	1.928	.003640	.001710	.0030	.001290	43.0
	2	68.8	79.4	1.900	.005750	.002700	.0045	.001800	40.0
	3	57.3	66.1	1.820	.012100	.005680	.0090	.003320	36.9
4.9	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	77.7	89.7	1.953	.001595	.000751	.0015	.000749	49.9
	1 1/2	73.2	84.5	1.927	.003660	.001720	.0030	.001280	42.6
	2	68.8	79.4	1.900	.005780	.002722	.0045	.001778	39.5
	3	57.1	65.9	1.819	.012140	.005710	.0090	.003290	36.5
5.9	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	77.3	89.3	1.951	.001770	.000832	.0015	.000668	44.6
	1 1/2	72.6	83.8	1.923	.003970	.001865	.0030	.001135	37.8
	2	67.8	78.3	1.894	.006270	.002950	.0045	.001550	34.4
	3	55.7	64.3	1.808	.013000	.006100	.0090	.002890	32.1

TABLE 20A

CALCULATION OF RETENTION OF PONTAMINE FAST RED 8BL  
ON UNBLEACHED KRAFT PULP AT VARIOUS PH'S

PH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	72.3	89.1	1.950	.001590	.000748	.0015	.000752	50.1
	1 1/2	67.8	83.6	1.922	.003340	.001570	.0030	.001430	47.6
	2	63.7	78.5	1.895	.005090	.002390	.0045	.002110	46.8
	3	52.5	64.7	1.811	.010500	.004950	.0090	.004050	45.0
4.9	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	72.3	89.1	1.950	.001595	.000750	.0015	.000750	50.0
	1 1/2	67.8	83.6	1.922	.003340	.001570	.0030	.001430	47.6
	2	63.5	78.3	1.894	.005120	.002405	.0045	.002095	46.6
	3	52.5	64.7	1.811	.010500	.004950	.0090	.004050	45.0
5.9	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	72.1	88.9	1.949	.00165	.000775	.0015	.000725	48.3
	1 1/2	68.0	83.8	1.923	.00330	.001550	.0030	.001450	45.9
	2	63.1	77.8	1.891	.00533	.002505	.0045	.001995	44.3
	3	51.8	63.8	1.805	.01090	.005120	.0090	.003880	43.1

TABLE 21A  
CALCULATION OF RETENTION OF PONTAMINE FAST RED 8BL  
ON BLEACHED KRAFT PULP AT VARIOUS pH'S

pH	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
3.9	0	83.1	94.5	1.975	0	0	0	0	100.0
	1	79.3	90.2	1.955	.000592	.000592	.0015	.000902	60.1
	1 1/2	75.3	85.7	1.933	.001260	.001265	.0030	.001735	57.8
	2	71.5	81.3	1.910	.004130	.001940	.0045	.002560	56.8
	3	60.7	69.0	1.839	.008700	.004080	.0090	.004920	54.7
4.9	0	83.1	94.5	1.975	0	0	0	0	100.0
	1	79.3	90.2	1.955	.001275	.000600	.0015	.000900	60.0
	1 1/2	75.3	85.5	1.932	.002720	.001280	.0030	.001720	57.4
	2	71.3	81.1	1.909	.004200	.001970	.0045	.002530	56.2
	3	60.3	68.7	1.837	.008800	.004130	.0090	.004870	54.1
5.9	0	83.1	94.5	1.975	0	0	0	0	100.0
	1	79.0	89.9	1.954	.001340	.000632	.0015	.000868	57.8
	1 1/2	74.8	85.1	1.930	.002870	.001350	.0030	.001650	55.0
	2	70.7	80.4	1.905	.004410	.002075	.0045	.002425	53.9
	3	59.4	67.6	1.831	.009210	.004330	.0090	.004670	51.9

TABLE 22A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SC ON  
UNBLEACHED SULPHITE PULP AT VARIOUS CONSISTENCIES

PERCENT CONSIST- ENCY	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
0.3	0 1/4	84.0	98.3	1.993	0	.000055	0	0	100.0
	1 1/4	83.3	97.5	1.989	.000117	.000055	.0015	.001445	96.3
	1 1/2	82.3	96.4	1.984	.000298	.000140	.0030	.002860	95.4
	1 3/4	81.3	95.1	1.978	.000490	.000230	.0045	.004270	94.9
	3	77.7	90.8	1.958	.001150	.000540	.0090	.008460	94.0
0.6	0 1/4	84.0	98.3	1.993	0	.000032	0	0	100.0
	1 1/4	83.7	98.0	1.991	.000068	.000032	.0015	.001468	97.9
	1 1/2	82.7	96.8	1.986	.000234	.000110	.0030	.002890	96.5
	1 3/4	81.7	95.7	1.981	.000383	.000180	.0045	.004320	96.0
	3	79.0	92.5	1.966	.000895	.000420	.0090	.008580	95.3
1.0	0 1/4	84.0	98.3	1.993	0	.000021	0	0	100.0
	1 1/4	83.9	98.2	1.992	.000045	.000021	.0015	.001479	98.6
	1 1/2	83.3	97.5	1.989	.000138	.000065	.0030	.002935	97.8
	1 3/4	82.6	96.6	1.985	.000255	.000120	.0045	.004380	97.3
	3	80.2	93.8	1.972	.000703	.000330	.0090	.008670	96.3

TABLE 23A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SO ON  
BLEACHED SULPHITE PULP AT VARIOUS CONSISTENCIES

PERCENT CONSIST- ENCY	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
0.3	0	91.2	98.3	1.993	0	.000208	0	0	100.0
	1	88.4	95.3	1.979	.000442	.000208	.0015	.001292	86.1
	1 1/2	85.3	92.0	1.964	.000958	.000450	.0030	.002550	85.1
	2	82.3	88.7	1.948	.001470	.000690	.0045	.003810	84.6
	3	73.8	79.6	1.901	.003060	.001440	.0090	.007560	83.9
0.6	0	91.2	98.3	1.993	0	.000183	0	0	100.0
	1	88.4	95.7	1.981	.000390	.000410	.0015	.001317	87.9
	1 1/2	85.8	92.5	1.966	.000873	.000410	.0030	.002590	86.4
	2	82.8	89.3	1.951	.001360	.000640	.0045	.003860	85.8
	3	74.6	80.5	1.906	.002870	.001350	.0090	.007650	85.0
1.0	0	91.2	98.3	1.993	0	.000174	0	0	100.0
	1	88.7	95.7	1.981	.000370	.000370	.0015	.001326	88.4
	1 1/2	86.4	93.1	1.969	.000787	.000370	.0030	.002630	87.6
	2	83.7	90.2	1.955	.001250	.000590	.0045	.003910	87.0
	3	75.8	81.7	1.912	.002660	.001250	.0090	.007750	86.1

TABLE 24A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT CROCEIN FL  
ON UNBLEACHED SULPHITE PULP AT VARIOUS CONSISTENCIES

PERCENT CONSIST- ENCY	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
0.3	0	75.1	94.0	1.973	0	0	0	0	100.0
	1	72.1	90.2	1.955	.001485	.000698	.0015	.000802	53.3
	1 1/2	68.0	85.1	1.930	.003430	.001610	.0030	.001390	46.0
	2	64.1	80.2	1.904	.005460	.002565	.0045	.001935	43.0
	3	53.7	67.1	1.827	.011550	.005440	.0090	.003560	39.6
0.6	0	75.1	94.0	1.973	0	0	0	0	100.0
	1	72.2	90.4	1.956	.001350	.000634	.0015	.000866	57.7
	1 1/2	68.5	85.7	1.933	.003190	.001500	.0030	.001500	50.0
	2	64.5	80.7	1.907	.005200	.002440	.0045	.002060	45.7
	3	54.3	67.9	1.832	.011150	.005240	.0090	.003760	41.8
1.0	0	75.1	94.0	1.973	0	0	0	0	100.0
	1	74.0	92.5	1.966	.000788	.000370	.0015	.001130	76.3
	1 1/2	70.9	88.7	1.948	.002040	.000960	.0030	.002040	68.1
	2	68.0	85.1	1.930	.003440	.001620	.0045	.002880	64.0
	3	59.6	74.5	1.872	.008050	.003780	.0090	.005220	58.0

TABLE 25A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT CROCEIN FL  
ON BLEACHED SULPHITE PULP AT VARIOUS CONSISTENCIES

PERCENT CONSIST- ENCY	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
0.3	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.5	90.6	1.957	.001320	.000622	.0015	.000878	58.8
	1 1/2	74.2	85.7	1.933	.003140	.001475	.0030	.001525	50.8
	2	70.1	80.9	1.908	.005130	.002410	.0045	.002090	46.8
	3	67.7	68.1	1.833	.011100	.005220	.0090	.003780	42.0
0.6	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.7	90.8	1.958	.001210	.000569	.0015	.000931	62.0
	1 1/2	74.7	88.7	1.948	.002950	.001385	.0030	.001615	53.8
	2	70.7	81.5	1.911	.004900	.002300	.0045	.002200	49.0
	3	59.5	68.6	1.836	.010850	.005100	.0090	.003900	43.3
1.0	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	80.1	92.5	1.966	.000575	.000270	.0015	.001230	81.3
	1 1/2	77.4	89.3	1.951	.001720	.000810	.0030	.002190	73.1
	2	74.6	86.1	1.935	.003040	.001430	.0045	.003070	68.2
	3	65.8	75.9	1.880	.007400	.003480	.0090	.005520	61.3

TABLE 26A  
CALCULATION OF RETENTION OF PONTAMINE FAST RED 8BL ON  
UNBLEACHED SULPHITE PULP AT VARIOUS CONSISTENCIES

PERCENT CONSIST- ENCY	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
0.3	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	71.9	88.7	1.948	.001710	.000804	.0015	.000696	46.4
	1 1/2	67.0	82.6	1.917	.003650	.001715	.0030	.001285	42.8
	2	62.7	77.3	1.888	.005570	.002620	.0045	.001880	41.8
	3	50.1	61.7	1.790	.011900	.005600	.0090	.003400	37.8
0.6	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	72.7	89.7	1.953	.001340	.000630	.0015	.000870	58.0
	1 1/2	68.7	84.7	1.928	.002940	.001380	.0030	.001620	54.0
	2	64.9	80.0	1.903	.004580	.002150	.0045	.002350	52.2
	3	53.8	66.4	1.822	.009750	.004580	.0090	.004420	49.1
1.0	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	74.5	91.8	1.963	.000680	.000320	.0015	.001180	78.7
	1 1/2	71.8	88.5	1.947	.001720	.000810	.0030	.002190	73.1
	2	68.8	84.9	1.929	.002870	.001350	.0045	.003150	70.0
	3	60.1	74.1	1.870	.006760	.003180	.0090	.005820	64.7



TABLE 27A  
CALCULATION OF RETENTION OF PONTAMINE FAST RED SBL ON  
BLEACHED SULPHITE PULP AT VARIOUS CONSISTENCIES

PERCENT CONSIST- ENCY	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
0.3	0	83.1	94.5	1.975	0	0	0	0	100.0
	1	78.4	89.1	1.950	.001520	.000713	.0015	.000787	52.4
	1 1/2	73.3	83.4	1.921	.003380	.001585	.0030	.001415	47.2
	3	68.8	78.2	1.893	.005210	.002450	.0045	.002050	45.7
0.6	0	55.4	63.0	1.799	.011130	.005240	.0090	.003760	41.7
	1	83.1	94.5	1.975	0	0	0	0	100.0
	1 1/2	80.0	91.0	1.956	.001147	.000540	.0015	.000960	64.0
	3	75.7	86.1	1.935	.002490	.001170	.0030	.001830	61.0
1.0	0	72.0	81.8	1.913	.003910	.001840	.0045	.002660	59.1
	1	61.4	69.8	1.844	.008370	.003940	.0090	.005060	56.2
	1 1/2	83.1	94.5	1.975	0	0	0	0	100.0
	3	82.1	93.3	1.970	.000291	.000137	.0015	.001363	91.0
	0	80.0	91.0	1.959	.000968	.000455	.0030	.002545	84.9
	1 1/2	77.7	88.3	1.946	.001820	.000855	.0045	.003645	81.1
	3	70.2	79.8	1.902	.004620	.002170	.0090	.006830	75.8

TABLE 28A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SO  
ON UNBLEACHED SULPHITE PULP AT VARIOUS TEMPERATURES

TEMPER- ATURE	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
20	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.7	98.0	1.991	.000068	.000032	.0015	.001468	97.9
	1 1/2	82.7	96.8	1.986	.000234	.000110	.0030	.002890	96.5
	2	81.7	95.7	1.981	.000383	.000180	.0045	.004320	96.0
	3	79.0	92.5	1.966	.000895	.000420	.0090	.008580	95.3
40	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.5	97.7	1.990	.000087	.000041	.0015	.001459	97.2
	1 1/2	82.7	96.8	1.986	.000234	.000110	.0030	.002890	96.4
	2	81.6	95.5	1.980	.000404	.000190	.0045	.004310	95.9
	3	78.8	92.3	1.965	.000957	.000450	.0090	.008550	95.0
60	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.7	98.0	1.991	.000064	.000030	.0015	.001470	98.0
	1 1/2	82.8	97.0	1.987	.000191	.000090	.0030	.002910	97.0
	2	81.8	95.7	1.981	.000383	.000180	.0045	.004320	96.0
	3	78.8	92.3	1.965	.000910	.000430	.0090	.008570	95.3

TABLE 29A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN 80  
ON BLEACHED SULPHITE PULP AT VARIOUS TEMPERATURES

TEMPER- ATURE	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
20	0	91.2	98.3	1.993	0	0	0	0	100.0
	1	88.8	95.7	1.981	.000390	.000183	.0015	.001317	87.9
	1 1/2	85.8	92.5	1.966	.000873	.000410	.0030	.002590	86.4
	2	82.8	89.3	1.951	.001360	.000640	.0045	.003860	85.8
	3	74.6	80.5	1.906	.002870	.001350	.0090	.007650	85.0
40	0	91.2	98.3	1.993	0	0	0	0	100.0
	1	88.7	95.7	1.981	.000387	.000182	.0015	.001318	87.8
	1 1/2	86.4	93.1	1.967	.000861	.000405	.0030	.002595	86.5
	2	82.8	89.3	1.951	.001380	.000650	.0045	.003850	85.7
	3	74.7	80.5	1.906	.002880	.001350	.0090	.007650	85.0
60	0	91.2	98.3	1.993	0	0	0	0	100.0
	1	88.6	95.5	1.980	.000396	.000186	.0015	.001314	87.5
	1 1/2	86.0	92.7	1.967	.000861	.000405	.0030	.002595	86.5
	2	83.0	89.5	1.952	.001360	.000640	.0045	.003860	85.9
	3	74.5	80.4	1.905	.002900	.001360	.0090	.007640	85.4

TABLE 30A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT CROCEIN FL  
ON UNBLEACHED SULPHITE PULP AT VARIOUS TEMPERATURES

TEMPER- ATURE	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
20	0	75.1	94.0	1.973	0	0	0	0	100.0
	1	72.2	90.4	1.956	.001350	.000635	.0015	.000865	57.7
	1 1/2	68.5	85.7	1.933	.003190	.001500	.0030	.001500	50.0
	2	64.5	80.7	1.907	.005250	.002470	.0045	.002030	45.7
	3	54.3	67.9	1.832	.001115	.005240	.0090	.003760	41.8
40	0	75.1	94.0	1.973	0	0	0	0	100.0
	1	72.2	90.4	1.956	.001320	.000620	.0015	.000880	58.6
	1 1/2	68.7	85.9	1.934	.003160	.001485	.0030	.001515	50.5
	2	64.7	80.9	1.908	.005170	.002430	.0045	.002070	46.0
	3	54.2	67.9	1.832	.011100	.005210	.0090	.003790	42.1
60	0	75.1	94.0	1.973	0	0	0	0	100.0
	1	72.4	90.6	1.957	.001270	.000600	.0015	.000900	60.0
	1 1/2	68.8	86.1	1.935	.003060	.001440	.0030	.001560	51.9
	2	64.9	81.3	1.910	.005020	.002360	.0045	.002140	47.5
	3	54.8	68.7	1.837	.010700	.005050	.0090	.003950	43.9

TABLE 31A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT CROCEIN FL  
ON BLEACHED SULPHITE PULP AT VARIOUS TEMPERATURES

TEMPER- ATURE	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
20	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.7	90.8	1.958	.001210	.000569	.0015	.000931	62.0
	1 1/2	74.7	96.3	1.936	.002950	.001385	.0030	.001615	53.8
	2	70.7	81.5	1.911	.004900	.002300	.0045	.002200	49.0
	3	59.5	68.6	1.836	.010850	.005100	.0090	.003900	43.3
40	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.7	90.8	1.958	.001200	.000565	.0015	.000935	62.3
	1 1/2	74.9	86.5	1.937	.002940	.001380	.0030	.001620	54.1
	2	70.7	81.7	1.912	.004820	.002270	.0045	.002230	49.5
	3	59.5	68.7	1.837	.010700	.005030	.0090	.003970	44.0
60	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.8	91.0	1.959	.001150	.000543	.0015	.000957	63.8
	1 1/2	75.1	86.7	1.938	.002860	.001342	.0030	.001658	55.2
	2	71.0	82.0	1.914	.004680	.002200	.0045	.002300	51.1
	3	59.9	69.2	1.840	.010500	.004950	.0090	.004050	45.0

TABLE 32A  
CALCULATION OF RETENTION OF PONTAMINE FAST RED 8BL ON  
UNBLEACHED SULPHITE PULP AT VARIOUS TEMPERATURES

TEMPER- ATURE	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
20	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	72.7	89.7	1.953	.001340	.000630	.0015	.000870	58.0
	1	68.7	84.7	1.928	.002940	.001380	.0030	.001620	54.0
	1	64.9	80.0	1.903	.004580	.002150	.0045	.002350	52.2
	3	53.8	66.4	1.822	.009750	.004580	.0090	.004420	49.1
40	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	73.2	90.2	1.955	.001280	.000604	.0015	.000896	59.8
	1	69.2	85.3	1.931	.002840	.001335	.0030	.001665	55.5
	1	65.1	80.3	1.905	.004470	.002100	.0045	.002400	53.4
	3	54.1	66.7	1.824	.009610	.004520	.0090	.004480	49.8
60	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	73.2	90.4	1.956	.001220	.000574	.0015	.000926	61.8
	1	69.3	85.5	1.932	.002750	.001290	.0030	.001710	57.1
	1	65.4	80.7	1.907	.004360	.002050	.0045	.002450	54.3
	3	54.4	67.1	1.827	.009450	.004440	.0090	.004560	50.7

TABLE 33A  
CALCULATION OF RETENTION OF FONTAMINE FAST RED SBL ON  
BLEACHED SULPHITE PULP AT VARIOUS TEMPERATURES

TEMPER- ATURE	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
20	0	83.1	94.5	1.975	0	0	0	0	100.0
	$\frac{1}{2}$	80.0	91.0	1.956	.001147	.000540	.0015	.000960	64.0
	1	75.7	86.1	1.935	.002490	.001170	.0030	.001830	61.5
	$1\frac{1}{2}$	72.0	81.8	1.913	.003910	.001840	.0045	.002660	59.1
	3	61.4	69.8	1.844	.008370	.003940	.0090	.005060	56.2
40	0	83.1	94.5	1.975	0	0	0	0	100.0
	$\frac{1}{2}$	79.7	90.6	1.957	.001150	.000514	.0015	.000986	65.8
	1	76.1	86.5	1.937	.002440	.001145	.0030	.001855	61.8
	$1\frac{1}{2}$	72.3	82.2	1.915	.003830	.001860	.0045	.002700	60.1
	3	61.7	70.1	1.846	.008220	.003860	.0090	.005140	57.0
60	0	83.1	94.5	1.975	0	0	0	0	100.0
	$\frac{1}{2}$	80.0	91.0	1.959	.001010	.000475	.0015	.001025	68.4
	1	76.4	86.9	1.939	.002280	.001070	.0030	.001930	64.4
	$1\frac{1}{2}$	72.8	82.8	1.918	.003620	.001700	.0045	.002800	62.3
	3	62.8	71.4	1.854	.007740	.003640	.0090	.005360	59.5

TABLE 34A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SO  
ON UNBLEACHED SULPHITE PULP AT VARIOUS TIMES

TIME IN MINUTES	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
10	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.7	98.0	1.991	.000068	.000032	.0015	.001468	97.9
	1 1/2	82.7	96.8	1.986	.000234	.000110	.0030	.002890	96.5
	2	81.7	95.7	1.981	.000383	.000180	.0045	.004320	96.0
	3	79.0	92.5	1.966	.000895	.000420	.0090	.008580	95.3
30	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.7	98.0	1.991	.000066	.000031	.0015	.001469	97.9
	1 1/2	82.7	96.8	1.986	.000234	.000110	.0030	.002890	96.4
	2	81.7	95.7	1.981	.000383	.000180	.0045	.004320	96.0
	3	79.0	92.5	1.966	.000895	.000420	.0090	.008580	95.4
60	0	84.0	98.3	1.993	0	0	0	0	100.0
	1	83.7	98.0	1.991	.000064	.000030	.0015	.001470	98.0
	1 1/2	82.7	96.8	1.986	.000224	.000105	.0030	.002895	96.5
	2	81.9	95.9	1.982	.000362	.000170	.0045	.004330	96.1
	3	77.1	90.2	1.955	.000895	.000420	.0090	.008580	95.4



TABLE 35A  
CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SC  
ON BLEACHED SULPHITE PULP AT VARIOUS TIMES

TIME IN MINUTES	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
10	0	91.2	98.3	1.993	0	0	0	0	100.0
	$\frac{1}{4}$	88.8	95.7	1.981	.000390	.000183	.0015	.001317	87.9
	1	85.8	92.5	1.966	.000873	.000410	.0030	.002590	86.4
	$1\frac{1}{4}$	82.8	89.3	1.951	.001360	.000640	.0045	.003860	85.8
	3	74.6	80.5	1.906	.002870	.001350	.0090	.007650	85.0
30	0	91.2	98.3	1.993	0	0	0	0	100.0
	$\frac{1}{4}$	88.5	95.5	1.980	.000415	.000195	.0015	.001305	87.0
	1	85.5	92.3	1.965	.000925	.000435	.0030	.002585	86.2
	$1\frac{1}{4}$	82.9	89.5	1.952	.001360	.000640	.0045	.003860	85.8
	3	74.6	80.5	1.906	.002870	.001350	.0090	.007650	85.0
60	0	91.2	98.3	1.993	0	0	0	0	100.0
	$\frac{1}{4}$	91.0	98.2	1.992	.000385	.000181	.0015	.001319	87.9
	1	85.8	92.5	1.966	.000870	.000410	.0030	.002590	86.4
	$1\frac{1}{4}$	83.0	89.5	1.952	.001340	.000630	.0045	.003870	86.0
	3	75.0	80.9	1.908	.002810	.001320	.0090	.007680	85.3

TABLE 36A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT CROCEIN FL  
ON UNBLEACHED SULPHITE PULP AT VARIOUS TIMES

TIME IN MINUTES	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
10	0	75.1	94.0	1.973	0	0	0	0	100.0
	$\frac{1}{2}$	72.2	90.4	1.956	.001350	.000635	.0015	.000865	57.7
	1	68.5	85.7	1.933	.003190	.001500	.0030	.001500	50.0
	$1\frac{1}{2}$	64.5	80.7	1.907	.005250	.002470	.0045	.002030	45.7
	3	54.3	67.9	1.832	.011150	.005240	.0090	.003760	41.8
30	0	75.1	94.0	1.973	0	0	0	0	100.0
	$\frac{1}{2}$	72.4	90.6	1.957	.001280	.000605	.0015	.000895	59.7
	1	68.7	85.9	1.934	.003100	.001455	.0030	.001545	51.5
	$1\frac{1}{2}$	64.8	81.1	1.909	.005080	.002390	.0045	.002110	47.0
	3	54.5	68.2	1.834	.011000	.005190	.0090	.003810	42.3
60	0	75.1	94.0	1.973	0	0	0	0	100.0
	$\frac{1}{2}$	72.7	91.0	1.959	.001150	.000539	.0015	.000961	64.1
	1	69.0	86.3	1.936	.002970	.001395	.0030	.001605	53.5
	$1\frac{1}{2}$	65.0	81.3	1.910	.004950	.002330	.0045	.002170	48.3
	3	54.7	68.4	1.835	.010900	.005130	.0090	.003870	43.0

TABLE 37A  
CALCULATION OF RETENTION OF DU PONT BRILLIANT CROCEIN FL  
ON BLEACHED SULPHITE PULP AT VARIOUS TIMES

TIME IN MINUTES	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
10	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.7	90.8	1.958	.001210	.000569	.0015	.000931	62.0
	1 1/2	74.7	86.3	1.936	.002950	.001385	.0030	.001615	53.8
	2	70.7	81.5	1.911	.004900	.002300	.0045	.002200	49.0
	3	59.5	68.6	1.836	.010850	.005100	.0090	.003900	43.3
30	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	78.8	91.0	1.959	.001118	.000525	.0015	.000975	65.0
	1 1/2	74.9	86.5	1.937	.002900	.001365	.0030	.001635	54.5
	2	70.7	81.7	1.912	.004810	.002260	.0045	.002240	49.8
	3	59.5	68.7	1.837	.010750	.005070	.0090	.003930	43.7
60	0	81.4	94.0	1.973	0	0	0	0	100.0
	1	79.3	91.6	1.962	.000900	.000424	.0015	.001076	67.3
	1 1/2	75.0	86.7	1.938	.002860	.001342	.0030	.001658	55.2
	2	71.0	82.0	1.914	.004720	.002220	.0045	.002280	50.7
	3	59.7	68.9	1.838	.010700	.005040	.0090	.003960	44.0

TABLE 38A  
CALCULATION OF RETENTION OF PONTAMINE FAST RED 8BL  
ON UNBLEACHED SULPHITE PULP AT VARIOUS TIMES

TIME IN MINUTES	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
10	0	76.7	94.5	1.975	0	0	0	0	100.0
	1	72.7	89.7	1.953	.001340	.000630	.0015	.000870	58.0
	1 1/2	68.7	84.7	1.928	.002940	.001380	.0030	.001620	54.0
	3	64.9	80.0	1.903	.004580	.002150	.0045	.002350	52.2
30	0	53.8	66.4	1.822	.009750	.004580	.0090	.004420	49.1
	1	76.7	94.5	1.975	0	0	0	0	100.0
	1 1/2	73.7	91.0	1.959	.001000	.000470	.0015	.001030	68.7
	3	70.2	86.7	1.938	.002340	.001100	.0030	.001900	63.3
60	0	66.6	82.2	1.915	.003830	.001800	.0045	.002700	60.0
	1	55.7	68.7	1.837	.008800	.004140	.0090	.004860	53.9
	1 1/2	76.7	94.5	1.975	0	0	0	0	100.0
	3	74.6	92.0	1.964	.000713	.000335	.0015	.001165	77.7
	0	71.4	88.1	1.945	.001910	.000900	.0030	.002100	70.1
	1	68.0	83.8	1.923	.003300	.001550	.0045	.002950	65.5
	1 1/2	56.8	70.1	1.846	.008230	.003870	.0090	.005130	57.0

TABLE 39A  
CALCULATION OF RETENTION OF PONTAMINE FAST RED 8BL  
ON BLEACHED SULPHITE PULP AT VARIOUS TIMES

TIME IN MINUTES	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
10	0	83.1	94.5	1.975	0	0	0	0	100.0
	1	80.0	91.0	1.956	.001147	.000540	.0015	.000960	64.0
	1 1/2	75.7	86.1	1.935	.002420	.001170	.0030	.001830	61.0
	1 1/4	72.0	81.8	1.913	.003910	.001840	.0045	.002660	59.1
	3	61.4	69.8	1.844	.008370	.003940	.0090	.005060	56.2
30	0	83.1	94.5	1.975	0	0	0	0	100.0
	1	80.8	91.8	1.963	.000770	.000362	.0015	.001138	75.8
	1 1/2	77.1	87.7	1.943	.002010	.000945	.0030	.002055	68.5
	1 1/4	73.5	83.6	1.922	.003380	.001590	.0045	.002910	64.7
	3	62.5	71.1	1.852	.007830	.003680	.0090	.005320	59.0
60	0	83.1	94.5	1.975	0	0	0	0	100.0
	1	81.7	92.9	1.968	.000447	.000210	.0015	.001290	86.1
	1 1/2	78.7	89.5	1.952	.001450	.000680	.0030	.002320	77.4
	1 1/4	75.2	85.5	1.932	.002740	.001290	.0045	.003210	71.3
	3	63.4	72.1	1.858	.007450	.003500	.0090	.005500	61.1

TABLE 40A

CALCULATION OF RETENTION OF DU PONT VICTORIA GREEN SO, DU PONT  
BRILLIANT CROCEIN FL, AND PONTAMINE FAST RED SBL ON  
UNBLEACHED SULPHITE PULP AT VARIOUS FREENESSES

DYE	BEATING TIME IN MINUTES	FREENESS S.R.	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
Victoria Green	0	860	79.0	92.5	1.965	.000895	.000420	.009	.008580	95.3
	10	810	79.4	92.9	1.968	.000808	.000380	.009	.008620	95.8
	20	705	80.5	94.2	1.974	.000638	.000300	.009	.008700	96.7
	30	560	81.3	95.1	1.978	.000490	.000230	.009	.008770	97.4
	40	400	81.7	95.7	1.981	.000404	.000190	.009	.008810	97.9
	50	250	82.2	96.2	1.983	.000319	.000150	.009	.008850	98.3
Brilliant Crocein FL	0	860	54.2	67.9	1.832	.011140	.005235	.009	.003765	41.8
	10	810	56.0	70.1	1.846	.010070	.004730	.009	.004270	47.5
	20	705	58.2	72.8	1.862	.008800	.004140	.009	.004860	54.0
	30	560	59.7	74.8	1.874	.007770	.003650	.009	.005350	59.5
	40	400	61.2	76.6	1.884	.006980	.003280	.009	.005720	63.5
	50	250	62.8	78.7	1.896	.006120	.002880	.009	.006120	68.0
Fast Red SBL	0	860	53.9	66.5	1.823	.009750	.004580	.009	.004420	49.1
	10	810	55.5	68.4	1.835	.008940	.004200	.009	.004800	53.4
	20	705	59.1	72.9	1.863	.007180	.003380	.009	.005620	62.5
	30	560	63.2	77.8	1.891	.005360	.002520	.009	.006480	72.0
	40	400	66.4	81.8	1.913	.003960	.001860	.009	.007140	79.3
	50	250	68.7	84.7	1.928	.002980	.001400	.009	.007600	84.5

TABLE 41A

## CALCULATION OF RETENTION OF VARIOUS DYES ON UNBLEACHED SULPHITE PULP

DYE STUFF	WAVE LENGTH EXAM. mu	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
Du Pont Victoria Green 80	620	79.0	92.5	1.965	.000895	.000420	.009	.008580	95.7
Du Pont Rhodamine B	522	70.4	86.9	1.939	.002790	.001310	.009	.007690	85.4
Du Pont Auramine Conc.	434	62.7	83.8	1.923	.001000	.000470	.009	.008530	94.8
Du Pont Methylene Blue ZX	660	77.4	89.5	1.952	.000702	.000330	.009	.008670	96.3
Du Pont Basic Brown BR	460	66.1	68.2	1.834	.008850	.004160	.009	.004840	53.8
Du Pont Safranin T Extra	520	70.6	87.1	1.940	.001000	.000470	.009	.008530	94.8
Du Pont Methyl Violet NE	580	66.3	78.5	1.895	.001572	.000740	.009	.008260	91.8
Du Pont Brilliant Crocein FL	500	54.4	68.1	1.823	.011100	.005240	.009	.003760	41.8
Du Pont Anthraquinone Blue B	600	59.7	70.0	1.845	.018050	.008496	.009	.000504	5.6
Du Pont Nigrosine WSB Powder	570	62.0	73.8	1.868	.016050	.007570	.009	.001430	15.9
Du Pont Quinoline Yellow Conc.	440	44.0	57.8	1.762	.016900	.007950	.009	.001050	11.7
Du Pont Orange II Conc.	490	31.1	39.1	1.592	.016400	.007720	.009	.001280	14.2
Pontoyl Violet 54B	550	42.8	51.6	1.713	.012230	.005750	.009	.003250	36.1
Pontamine Fast Red 8BL	520	53.8	66.4	1.822	.009750	.004580	.009	.004420	49.1
Du Pont Purpurine 4B Conc.	500	60.9	76.2	1.882	.004600	.002160	.009	.006840	76.1
Pontamine Fast Scarlet 4BS	500	58.2	72.8	1.862	.008500	.003990	.009	.005010	55.7
Pontamine Black E	500	63.9	80.0	1.903	.011600	.005440	.009	.003560	39.6
Du Pont Brill. Paper Yellow Conc.	420	37.1	49.9	1.698	.015420	.007250	.009	.002750	30.6
Pontamine Yellow SXP	460	58.7	75.9	1.880	.014180	.006670	.009	.002330	25.9
Pontamine Yellow>NNL	460	66.4	85.9	1.934	.007980	.003750	.009	.005250	58.3

TABLE 42A  
CALCULATION OF RETENTION OF VARIOUS DYES ON UNBLEACHED SULPHITE  
PULP USING SULPHURIC ACID, ALUM, AND SIZE AND ALUM

DYE	ACID	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
VG	H <sub>2</sub> SO <sub>4</sub> Alum Alum and Size	0	84.0	98.3	1.993	0	0	0	0	100.0
		3	79.0	92.5	1.966	.000895	.000420	.009	.008580	95.3
		0	84.0	98.3	1.993	0	0	0	0	100.0
		3	78.5	92.0	1.964	.000936	.000440	.009	.008560	95.2
		0	81.9	98.3	1.993	0	0	0	0	100.0
		3	77.0	92.5	1.966	.000895	.000420	.009	.008580	95.3
FL	H <sub>2</sub> SO <sub>4</sub> Alum Alum and Size	0	75.1	94.0	1.973	0	0	0	0	100.0
		3	54.3	67.9	1.832	.001115	.005240	.009	.003760	41.8
		0	75.1	94.0	1.973	0	0	0	0	100.0
		3	53.0	66.4	1.822	.011080	.005220	.009	.003780	41.9
		0	72.3	94.0	1.973	0	0	0	0	100.0
		3	53.8	70.0	1.845	.010880	.005120	.009	.003880	43.1

(Continued)



TABLE 42A (Continued)

DYE	ACID	POUNDS DYE PER 1000 POUNDS PULP	TRANS- MISSION	COR- RECTED TRANS- MISSION	LOG COR- RECTED TRANS- MISSION	CONCEN- TRATION OF WHITE WATER	GRAMS DYE IN WHITE WATER	GRAMS DYE USED	GRAMS DYE IN SHEET	PERCENT RETEN- TION
II	H <sub>2</sub> SO <sub>4</sub>	0	74.4	94.0	1.973	0	0	0	0	100.0
	Alum	3	30.8	38.9	1.590	.016420	.007720	.009	.001280	14.2
		0	74.4	94.0	1.973	0	0	0	0	100.0
	Alum and Size	3	31.1	39.3	1.594	.016300	.007670	.009	.001330	14.7
		0	71.5	94.0	1.973	0	0	0	0	100.0
8BL		3	31.5	41.3	1.616	.015350	.007220	.009	.001780	19.8
	H <sub>2</sub> SO <sub>4</sub>	0	76.7	94.5	1.975	0	0	0	0	100.0
	Alum	3	53.8	66.4	1.822	.009750	.004580	.009	.004420	49.1
		0	76.7	94.5	1.975	0	0	0	0	100.0
	Alum and Size	3	53.6	66.1	1.820	.009910	.004660	.009	.004340	48.2
		0	74.3	94.5	1.975	0	0	0	0	100.0
		3	52.7	67.1	1.827	.009500	.004470	.009	.004530	50.3

**APPENDIX B.**  
**DYE-STANDARD GRAPHS**

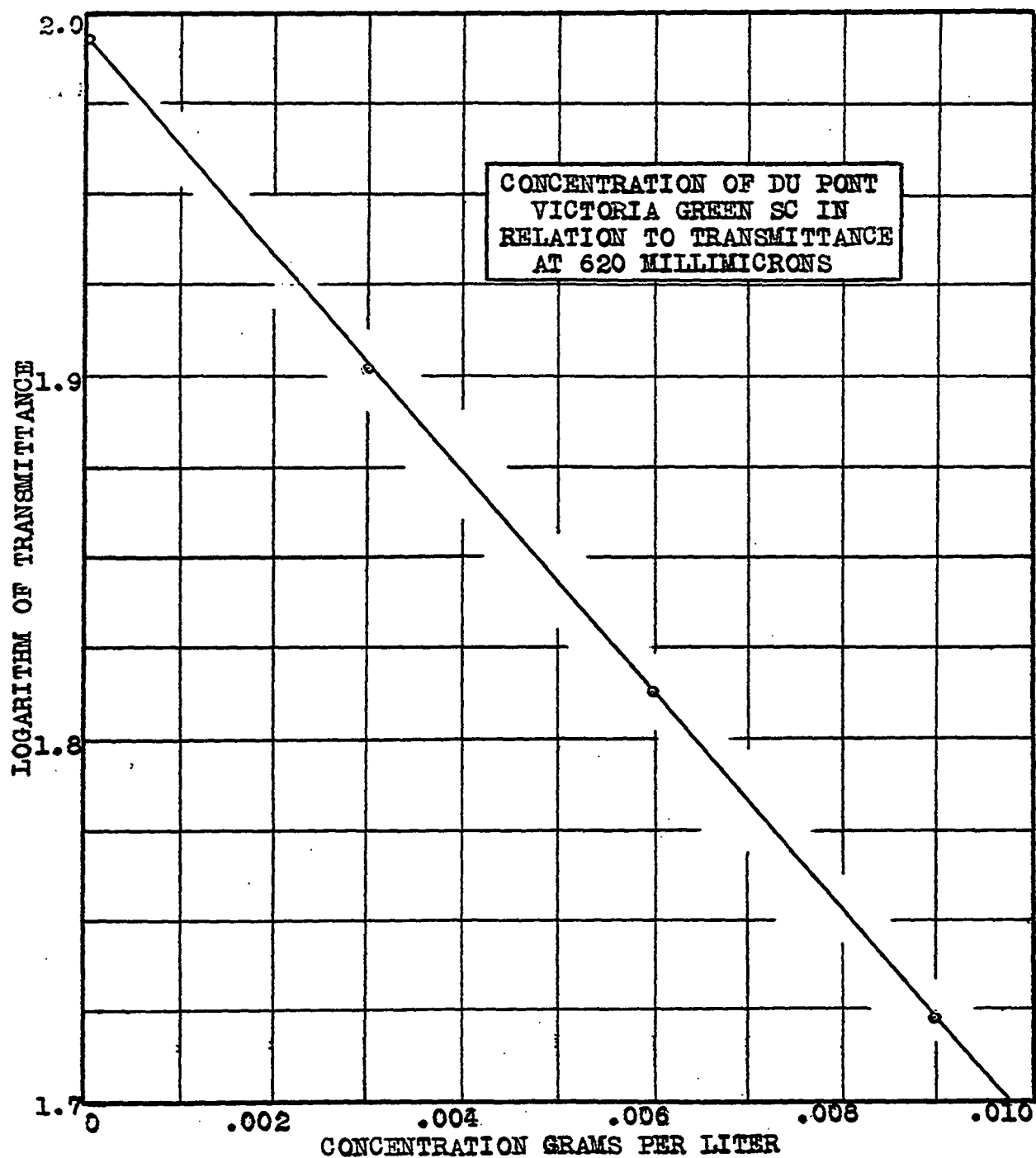
TRANSMITTANCE DATA OF THE STANDARD SOLUTIONS AT VARIOUS CONCENTRATIONS					
VICTORIA GREEN SC AT 620 MILLIMICRONS			RHODAMINE B- AT 522 MILLIMICRONS		
Conc. Grams per L.	Trans- mittance	Log Trans- mittance	Conc. Grams per L.	Trans- mittance	Log Trans- mittance
0	98.3	1.993	0	95.9	1.982
.003	79.8	1.902	.003	86.3	1.936
.006	65.0	1.813	.006	77.6	1.890
.009	53.0	1.724	.009	70.0	1.845
AURAMINE CONC. AT 434 MILLIMICRONS			METHYLENE BLUE ZX AT 660 MILLIMICRONS		
Conc. Grams per L.	Trans- mittance	Log Trans- mittance	Conc. Grams per L.	Trans- mittance	Log Trans- mittance
0	93.4	1.970	0	99.0	1.997
.003	67.6	1.831	.003	64.3	1.808
.006	49.2	1.692	.006	41.8	1.621
.009	36.0	1.556	.009	27.5	1.435
PONTAMINE FAST SCARLET 4BS AT 500 MILLIMICRONS					
Conc. Grams per L.	Trans- mittance	Log Trans- mittance			
0	94.0	1.973			
.006	78.7	1.896			
.012	66.1	1.820			
.018	55.3	1.743			

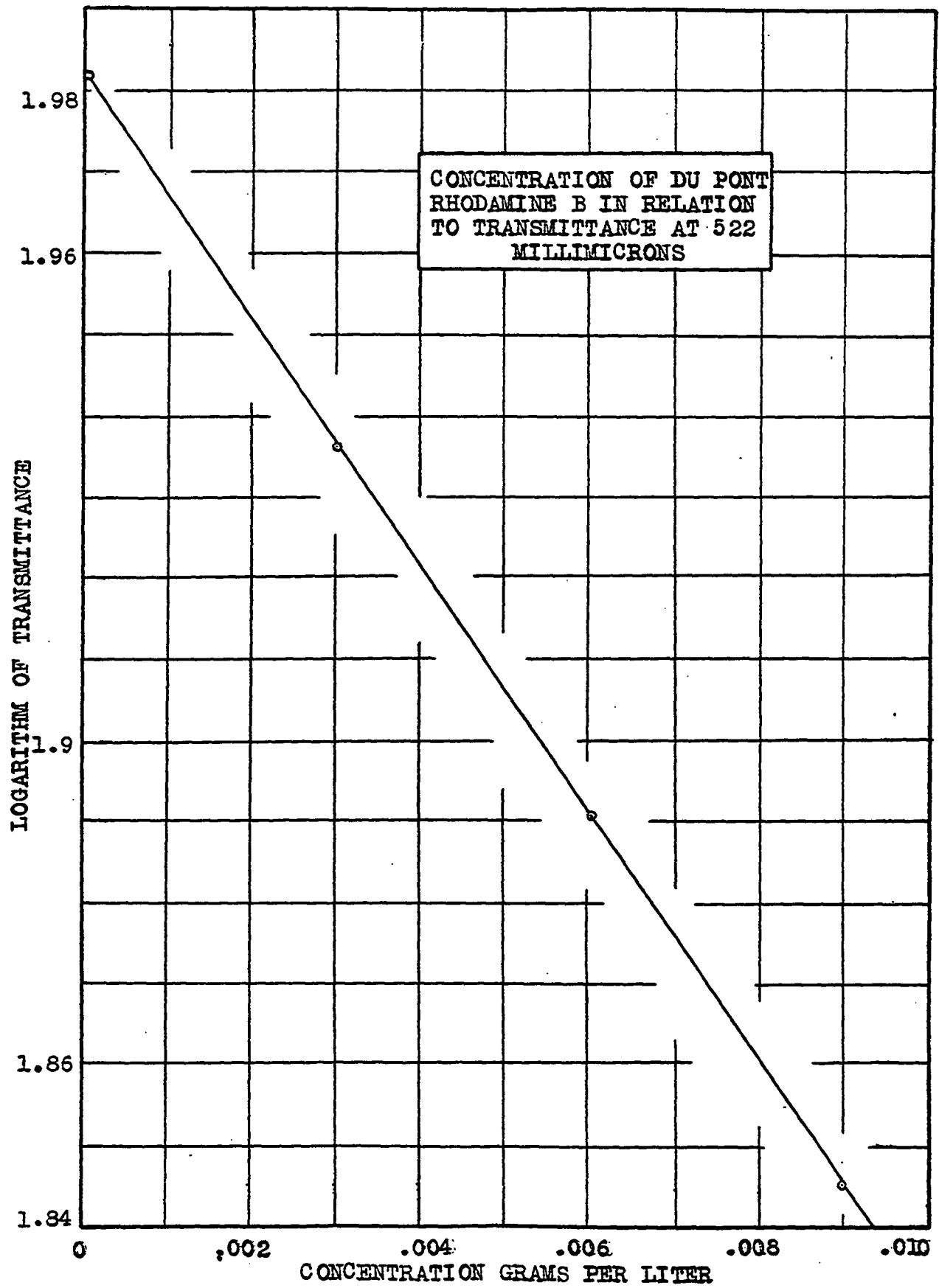
TRANSMITTANCE DATA OF THE STANDARD SOLUTIONS AT VARIOUS CONCENTRATIONS					
BASIC BROWN BR AT 460 MILLIMICRONS			SAFRANINE T EX. AT 520 MILLIMICRONS		
Conc. Grams per L.	Trans- mittance	Log Trans- mittance	Conc. Grams per L.	Trans- mittance	Log Trans- mittance
0	93.0	1.969	0	94.5	1.975
.003	83.8	1.923	.003	74.2	1.870
.006	75.4	1.877	.006	58.2	1.765
.009	67.9	1.832	.009	45.7	1.660
METHYL VIOLET NE AT 580 MILLIMICRONS			BRILLIANT CROCEIN FL AT 500 MILLIMICRONS		
Conc. Grams per L.	Trans- mittance	Log Trans- mittance	Conc. Grams per L.	Trans- mittance	Log Trans- mittance
0	97.0	1.987	0	94.0	1.973
.003	64.6	1.810	.006	78.9	1.897
.006	43.3	1.637	.012	66.2	1.821
.009	29.2	1.465	.018	55.7	1.746

BRILLIANT PAPER YELLOW CONC. AT 420 MILLIMICRONS		
Conc. Grams per L.	Trans- mittance	Log Trans- mittance
0	94.1	1.974
.006	73.5	1.866
.012	57.3	1.758
.018	44.9	1.652

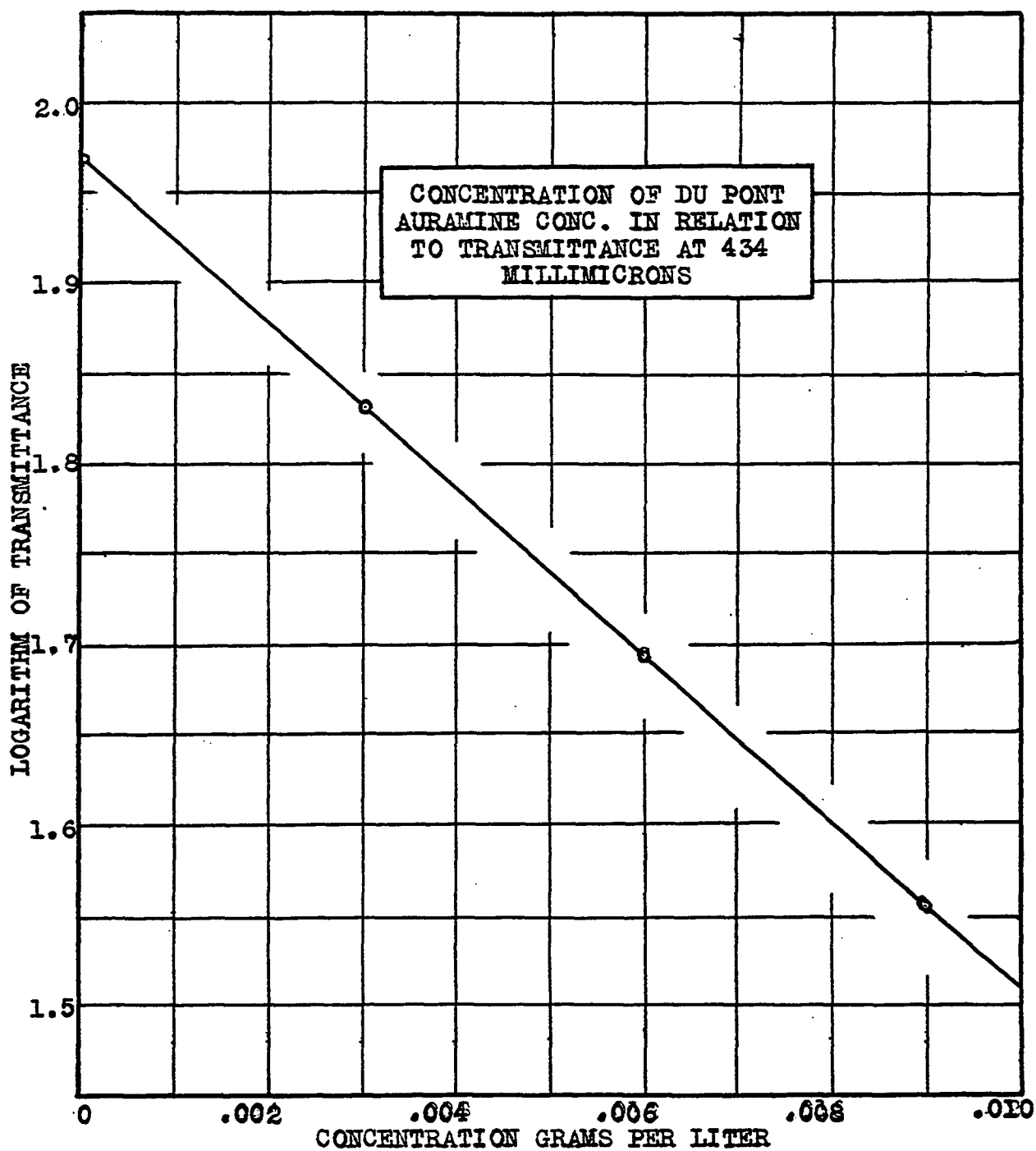
TRANSMITTANCE DATA OF THE STANDARD SOLUTIONS AT VARIOUS CONCENTRATIONS					
ANTHRAQUINONE BLUE B AT 600 MILLIMICRONS			NIGROSINE WSB POWDER AT 570 MILLIMICRONS		
Conc. Grams per L.	Trans- mittance	Log Trans- mittance	Conc. Grams per L.	Trans- mittance	Log Trans- mittance
0	97.4	1.989	0	96.5	1.985
.006	87.3	1.941	.006	87.3	1.941
.012	78.2	1.893	.012	78.9	1.897
.018	70.0	1.845	.018	71.5	1.854
ORANGE II CONC. AT 490 MILLIMICRONS			QUINOLINE YELLOW CONC. AT 440 MILLIMICRONS		
Conc. Grams per L.	Trans- mittance	Log Trans- mittance	Conc. Grams per L.	Trans- mittance	Log Trans- mittance
0	94.0	1.973	0	93.2	1.969
.006	68.2	1.834	.006	78.5	1.895
.012	49.3	1.693	.012	66.4	1.822
.018	35.7	1.553	.018	56.1	1.749
PONTAMINE FAST YELLOW>NNL AT 460 MILLIMICRONS					
Conc. Grams per L.	Trans- mittance	Log Trans- mittance			
0	93.0	1.969			
.006	87.7	1.943			
.012	82.6	1.917			
.018	77.8	1.891			

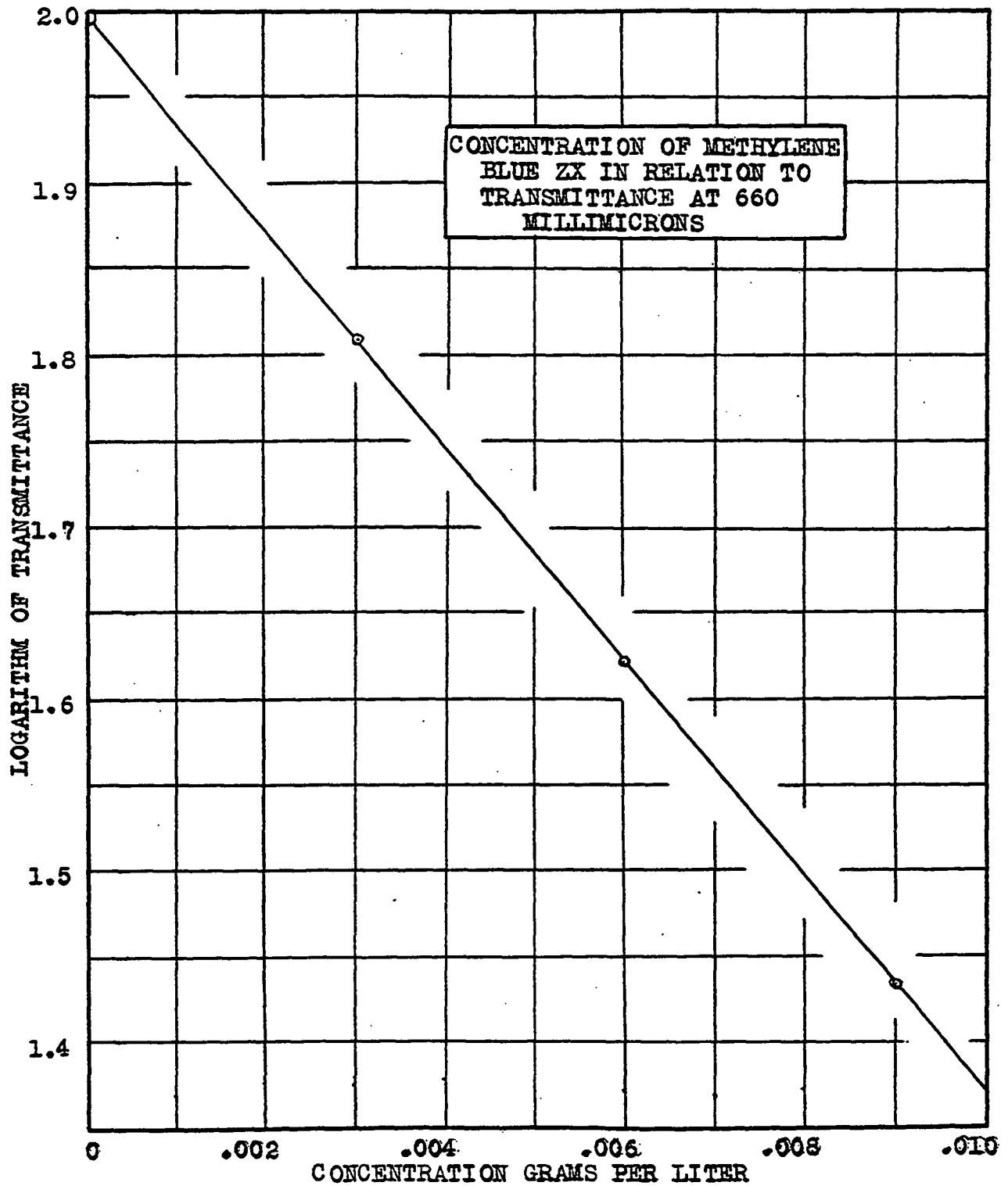
TRANSMITTANCE DATA OF THE STANDARD SOLUTIONS AT VARIOUS CONCENTRATIONS					
PONTAGYL VIOLET S4B AT 550 MILLIMICRONS			PURPURINE 4B CONC. AT 500 MILLIMICRONS		
Conc. Grams per L.	Trans- mittance	Log Trans- mittance	Conc. Grams per L.	Trans- mittance	Log Trans- mittance
0	95.9	1.982	0	94.0	1.973
.006	70.8	1.850	.006	71.6	1.855
.012	52.4	1.719	.012	54.7	1.738
.018	38.5	1.586	.018	41.8	1.621
PONTAMINE BLACK E AT 500 MILLIMICRONS			PONTAMINE YELLOW SXP AT 460 MILLIMICRONS		
Conc. Grams per L.	Trans- mittance	Log Trans- mittance	Conc. Grams per L.	Trans- mittance	Log Trans- mittance
0	94.0	1.973	0	93.0	1.969
.006	86.1	1.935	.006	85.3	1.931
.012	79.1	1.898	.012	78.3	1.894
.018	72.5	1.860	.018	71.8	1.856
PONTAMINE FAST RED 8BL AT 520 MILLIMICRONS					
Conc. Grams per L.	Trans- mittance	Log Trans- mittance			
0	94.5	1.975			
.006	76.0	1.881			
.012	61.4	1.788			
.018	49.5	1.695			

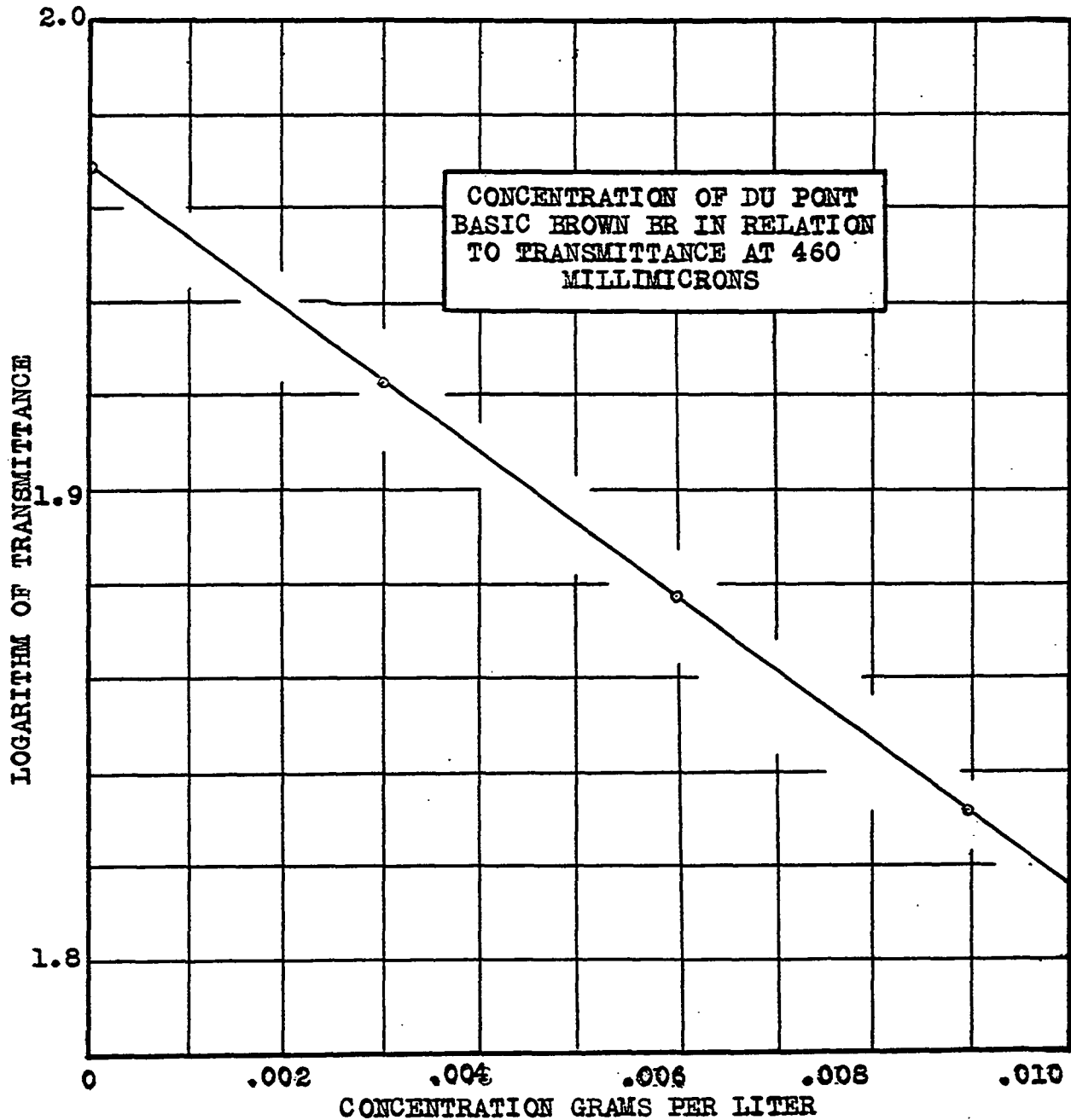


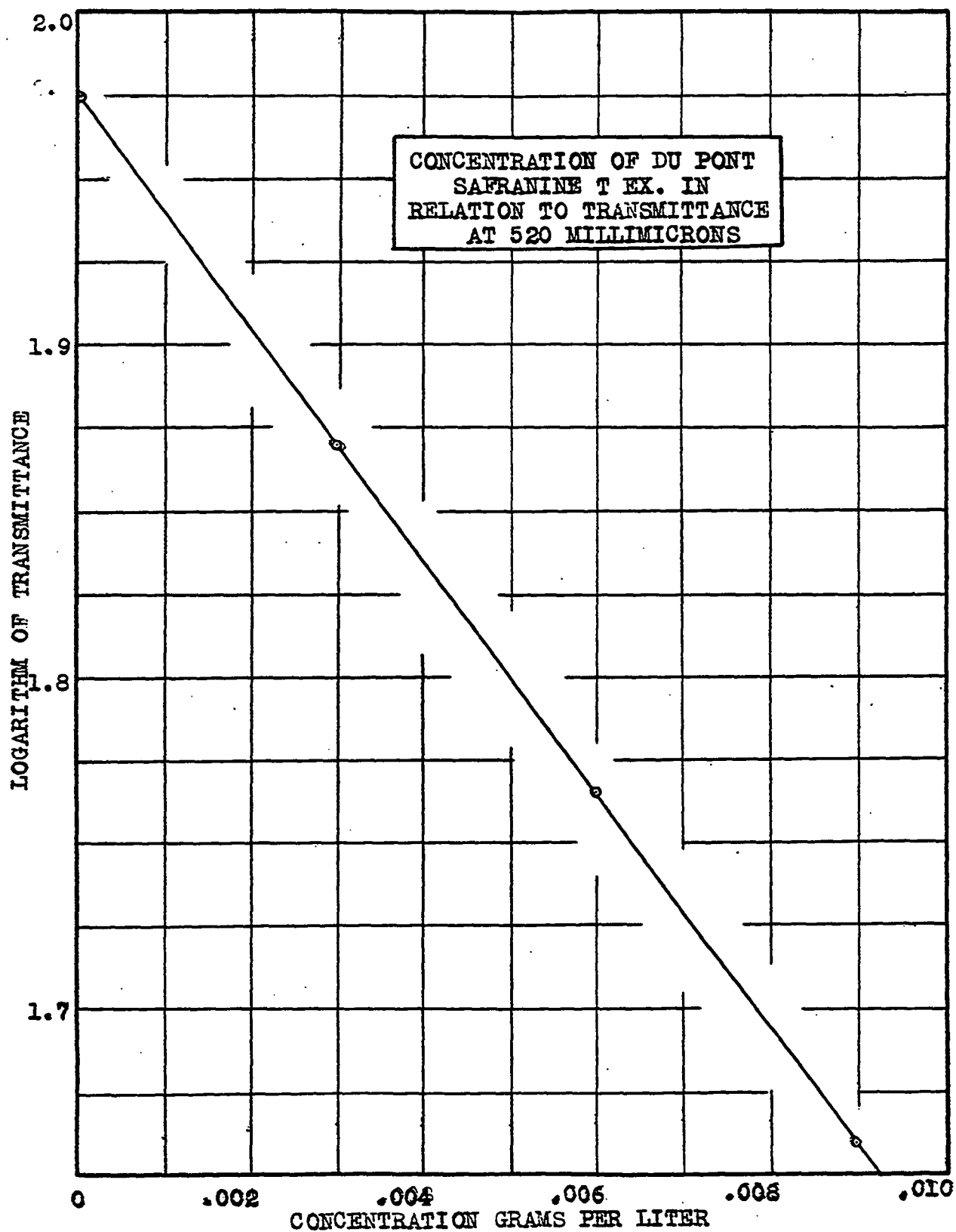


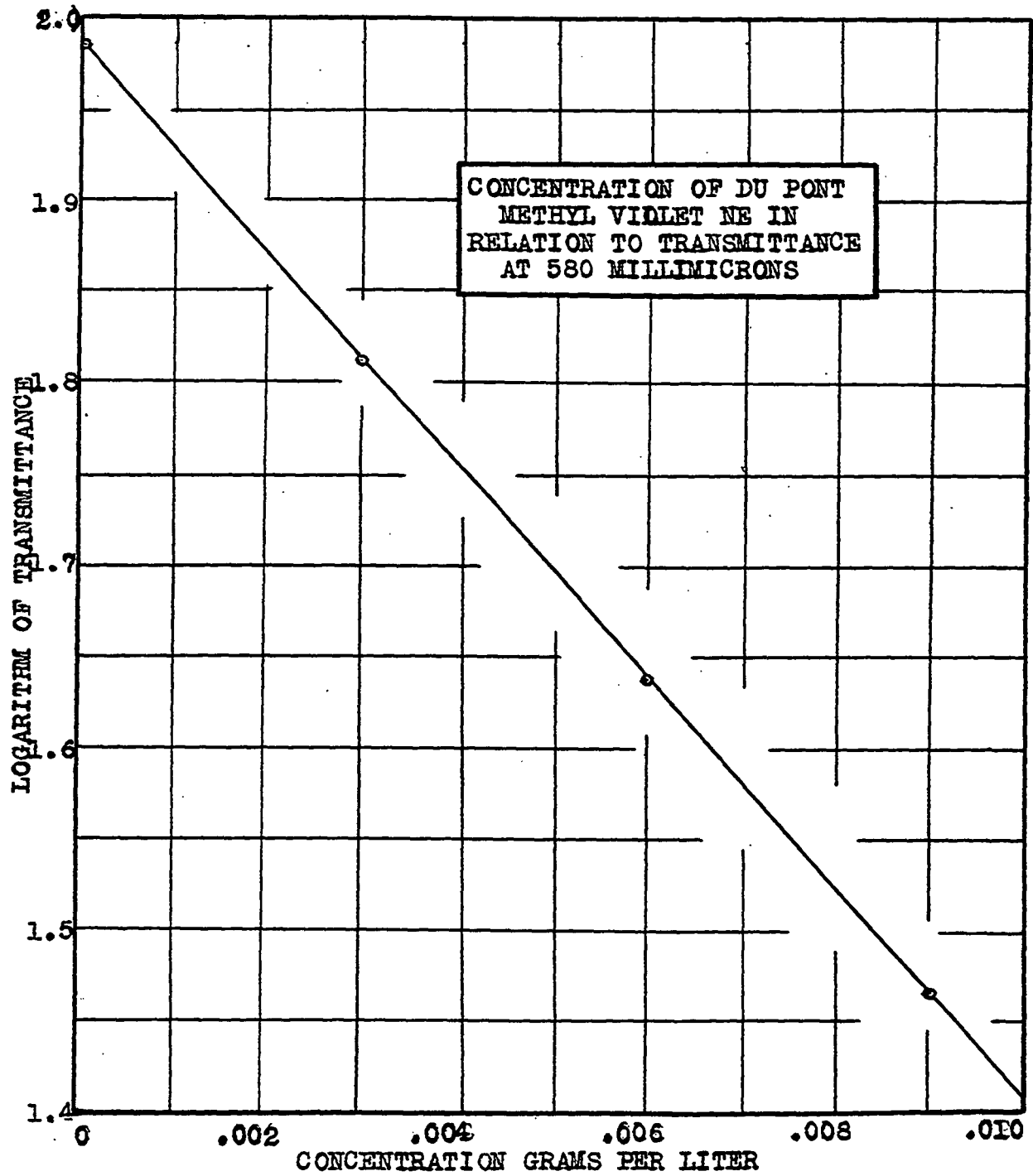


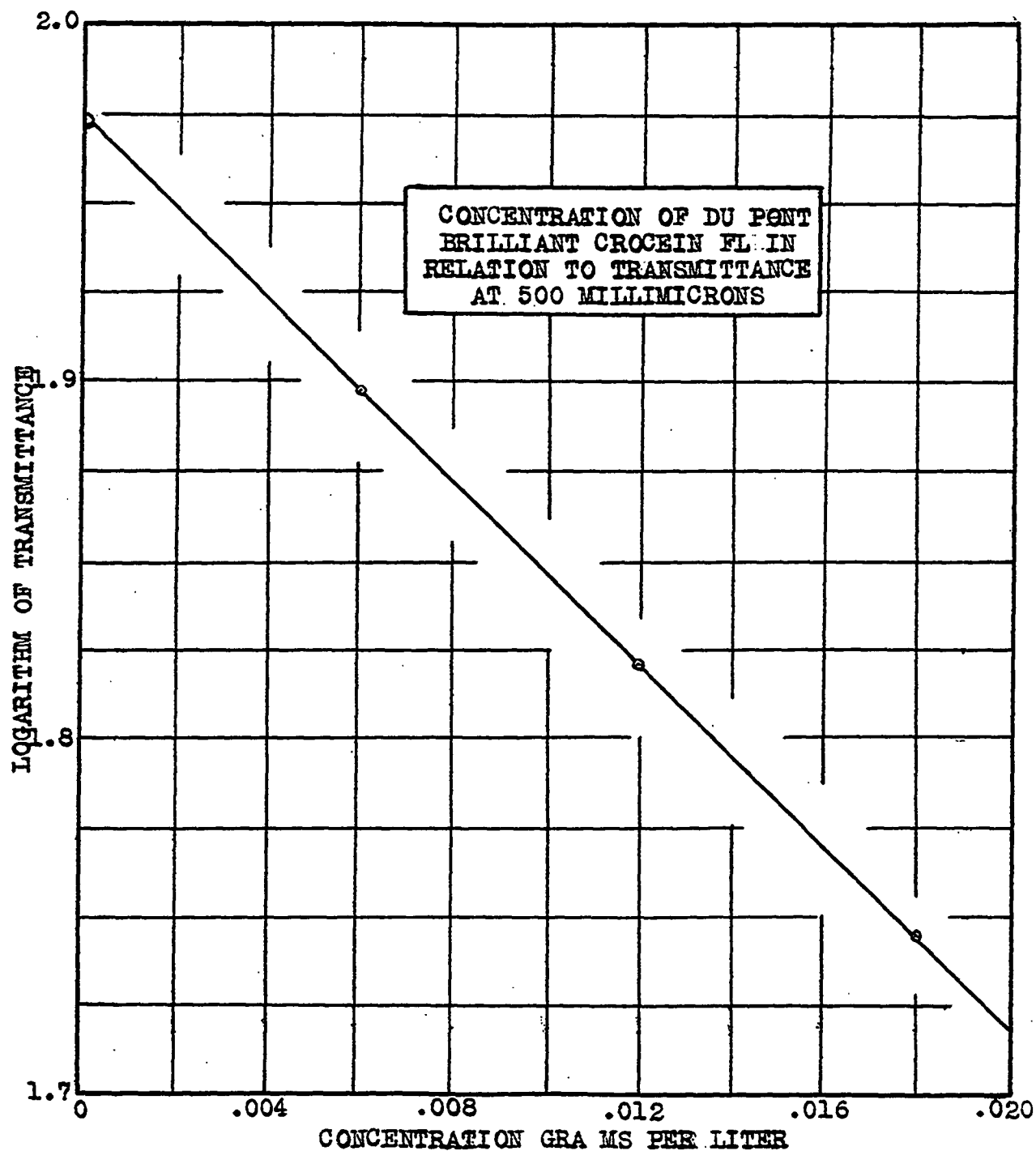


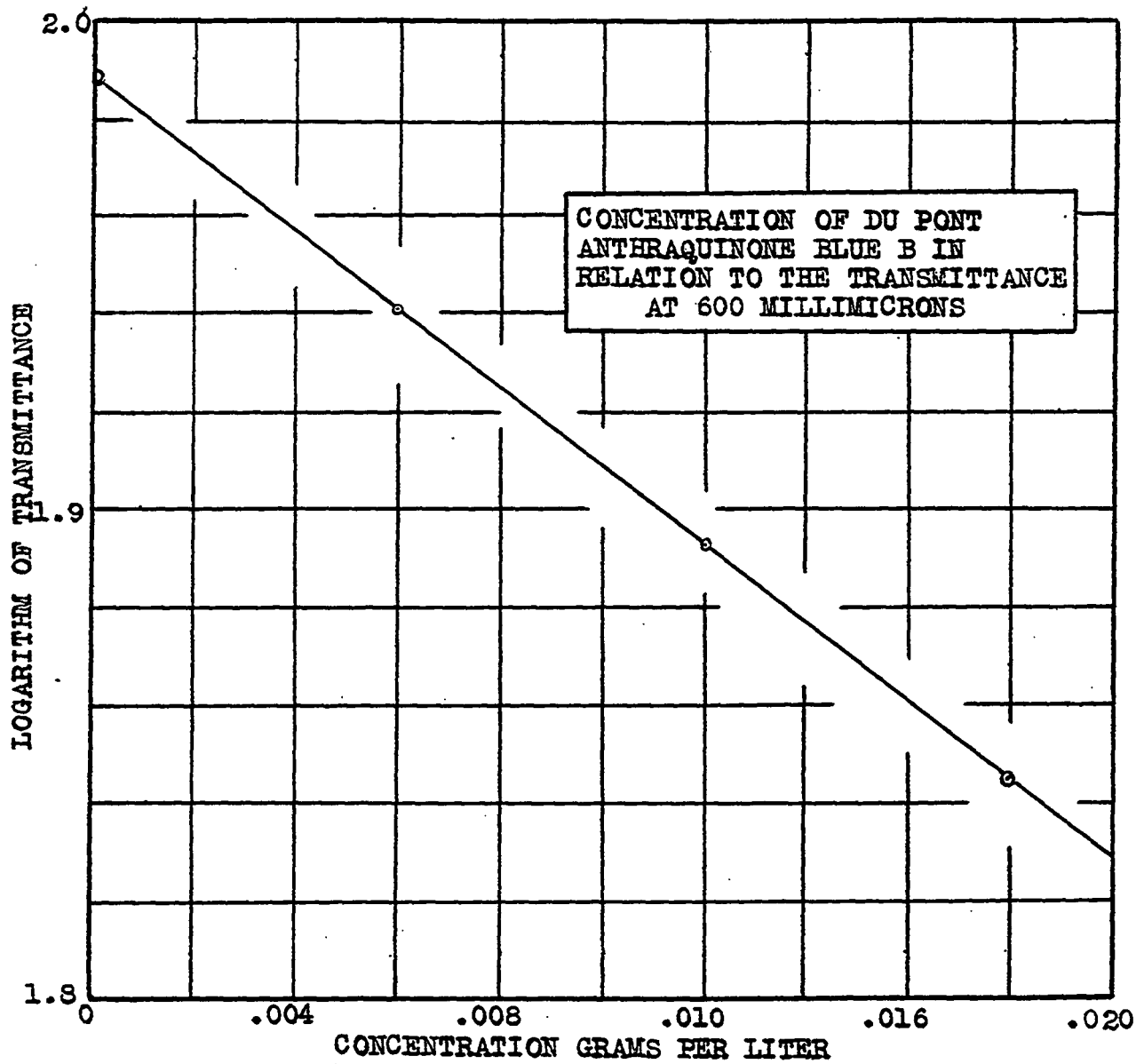


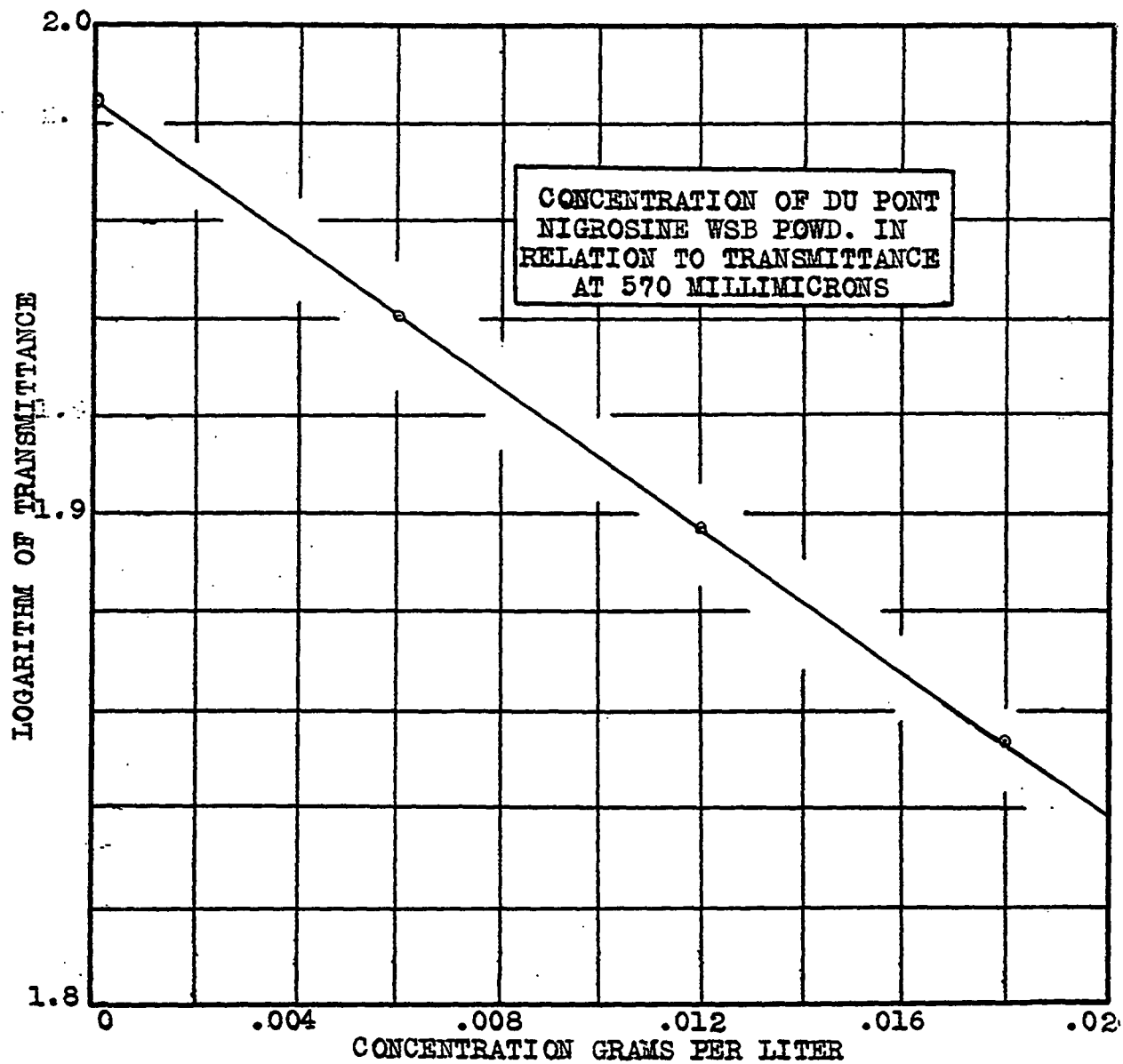




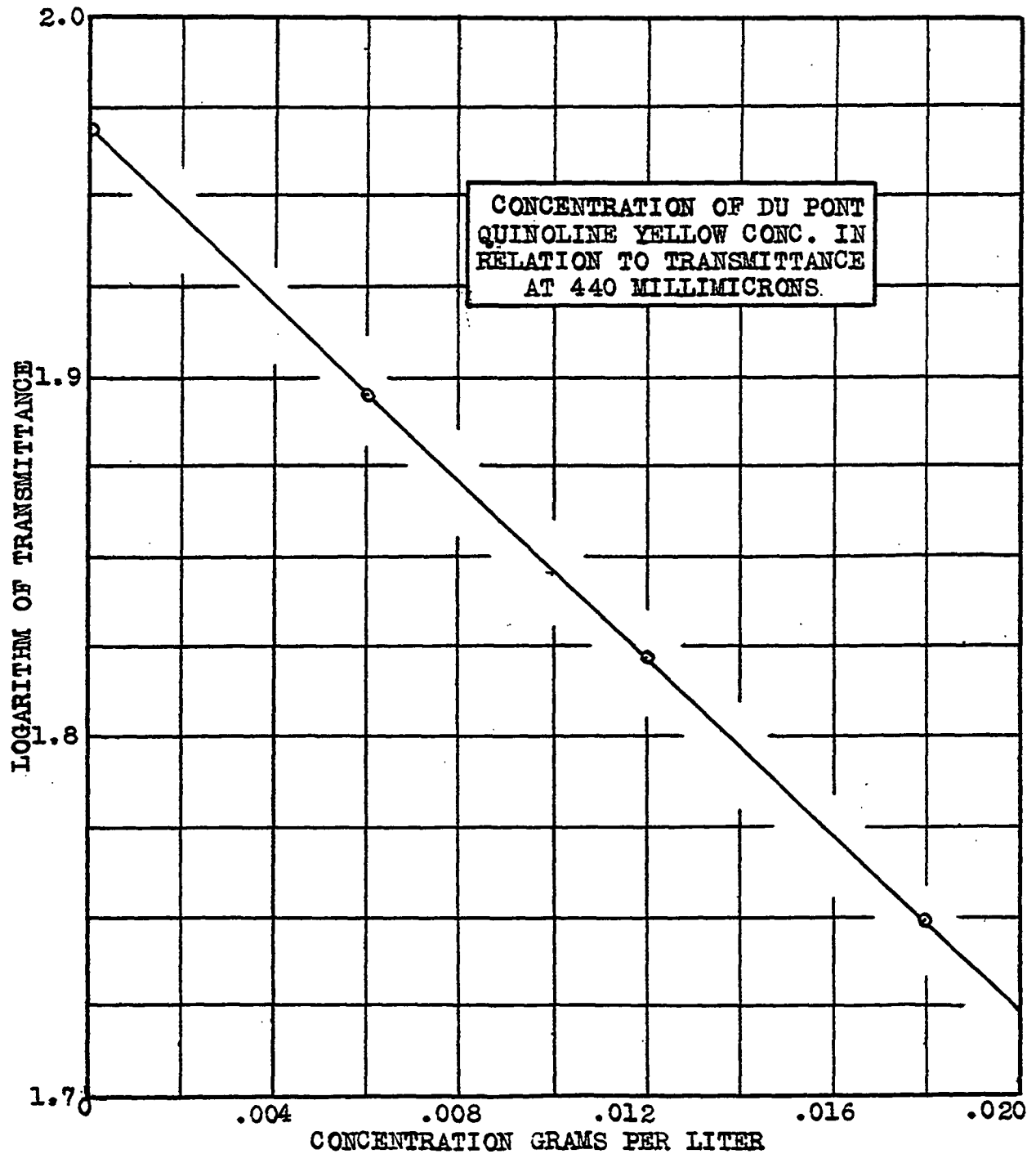


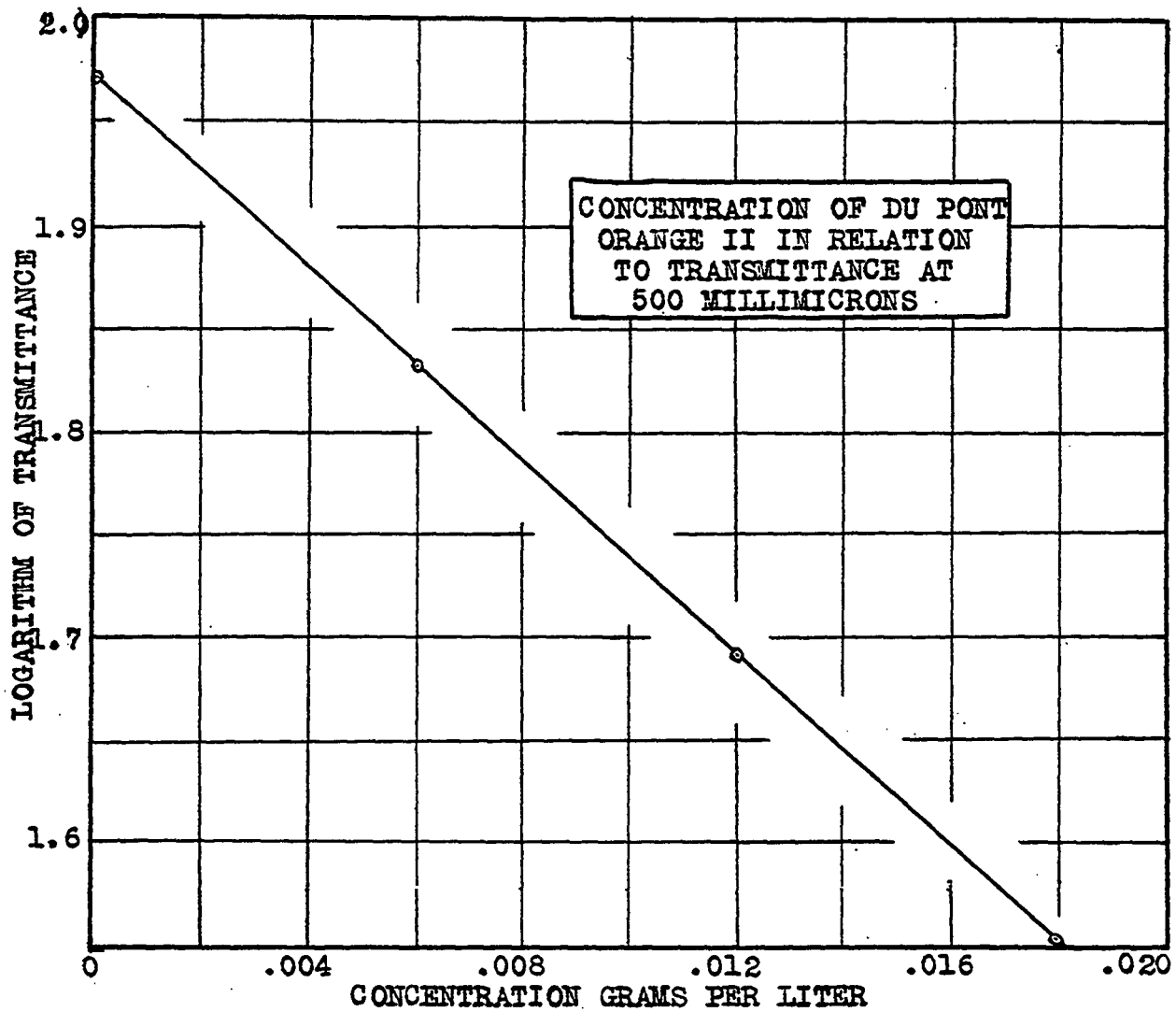


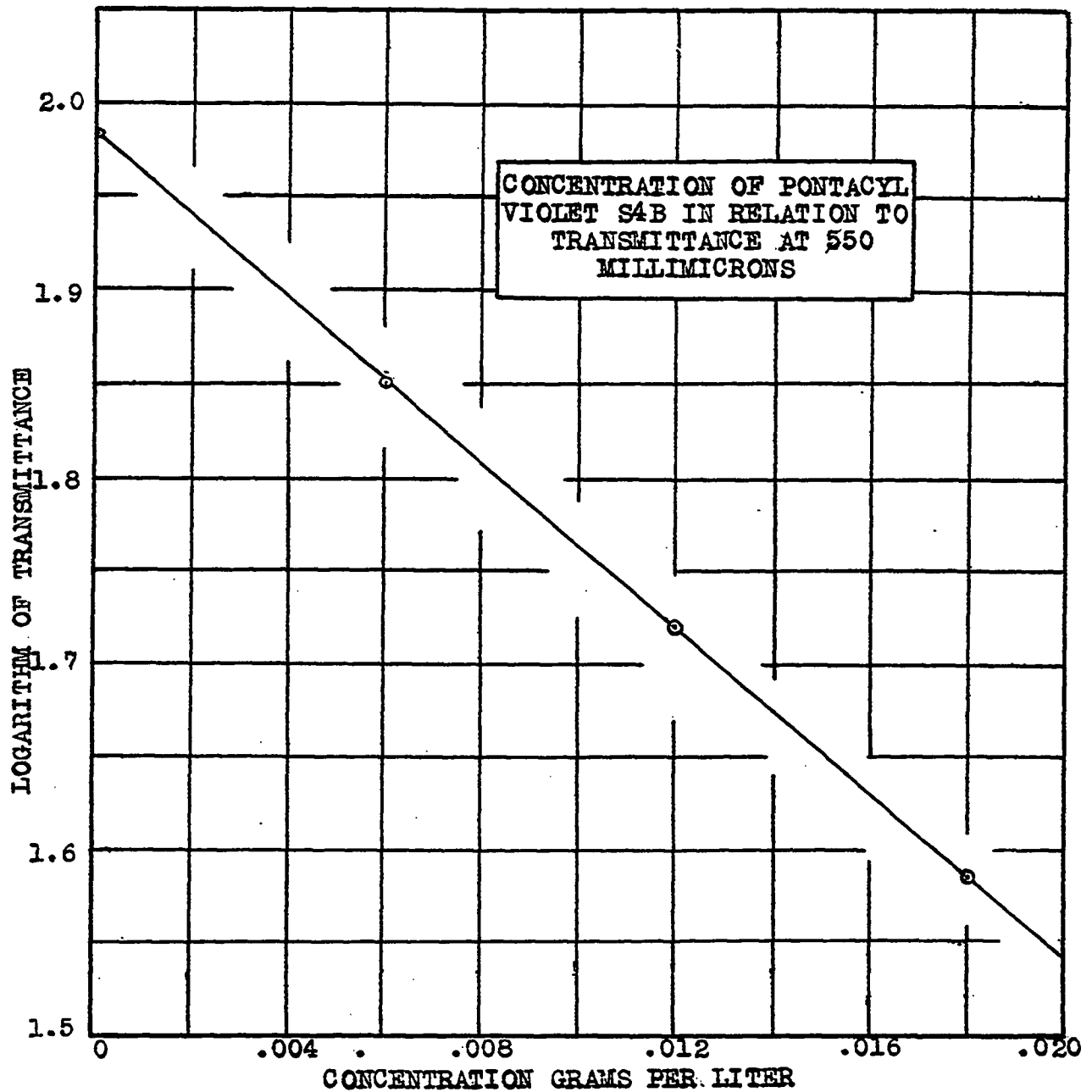


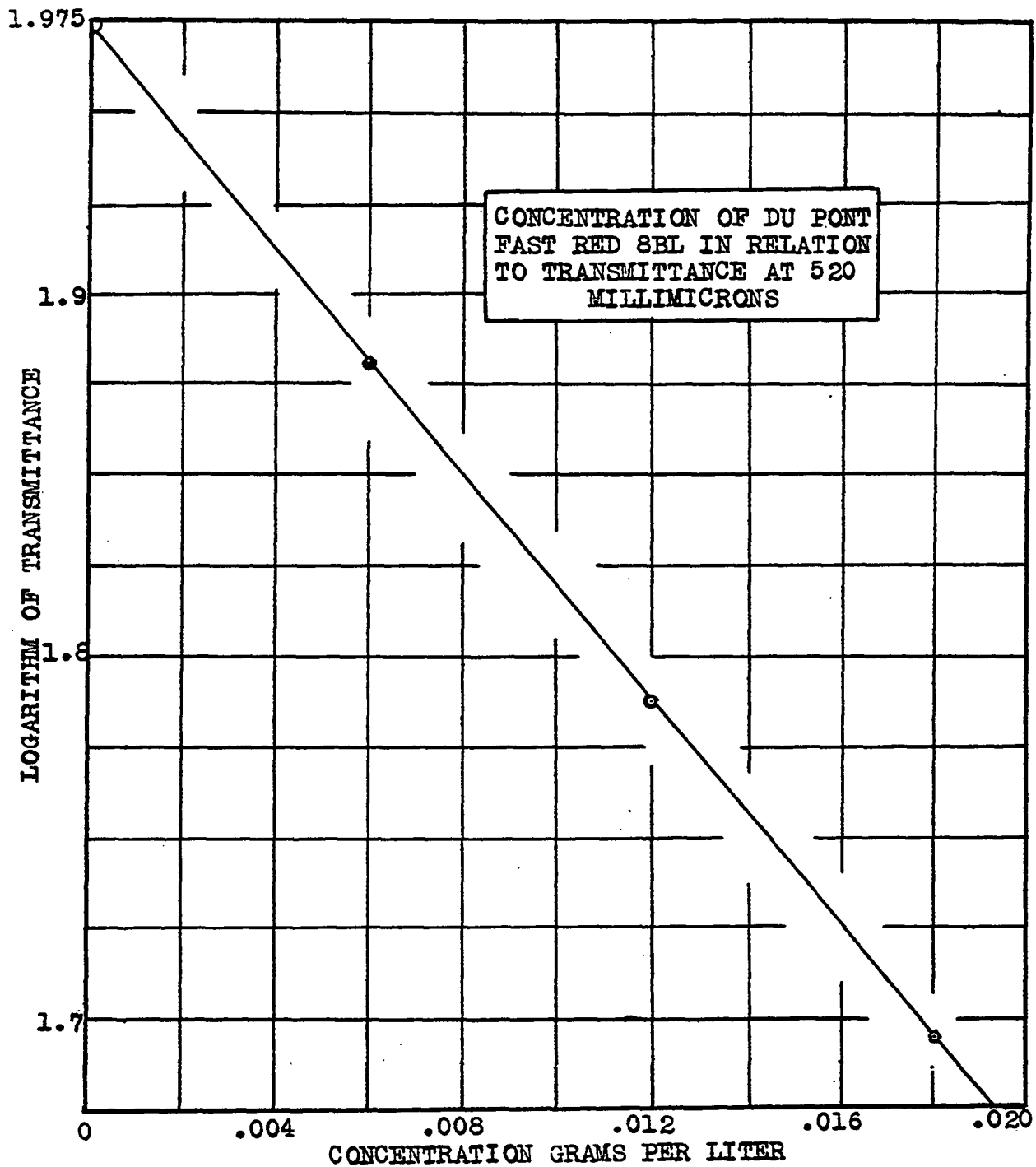


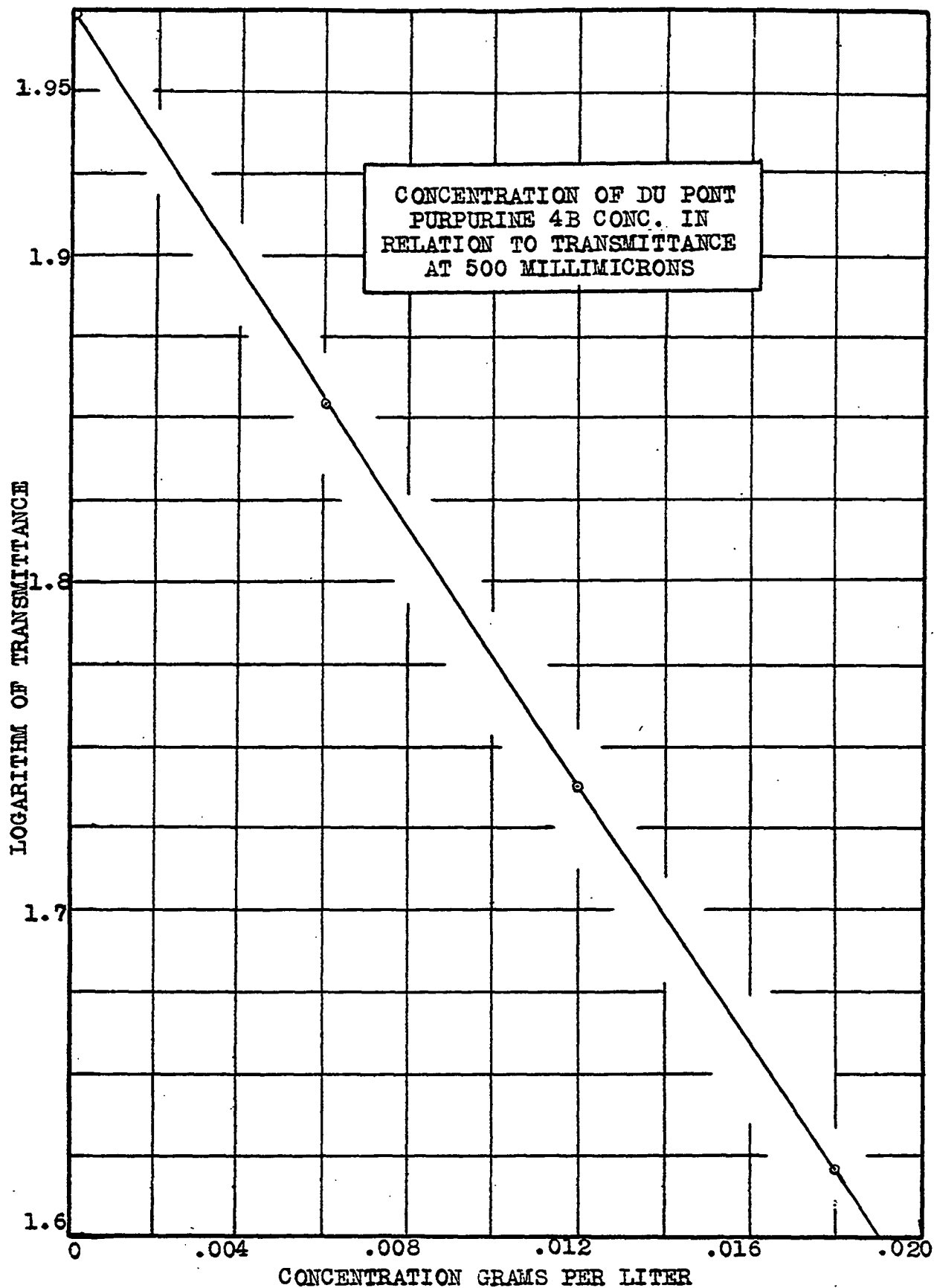


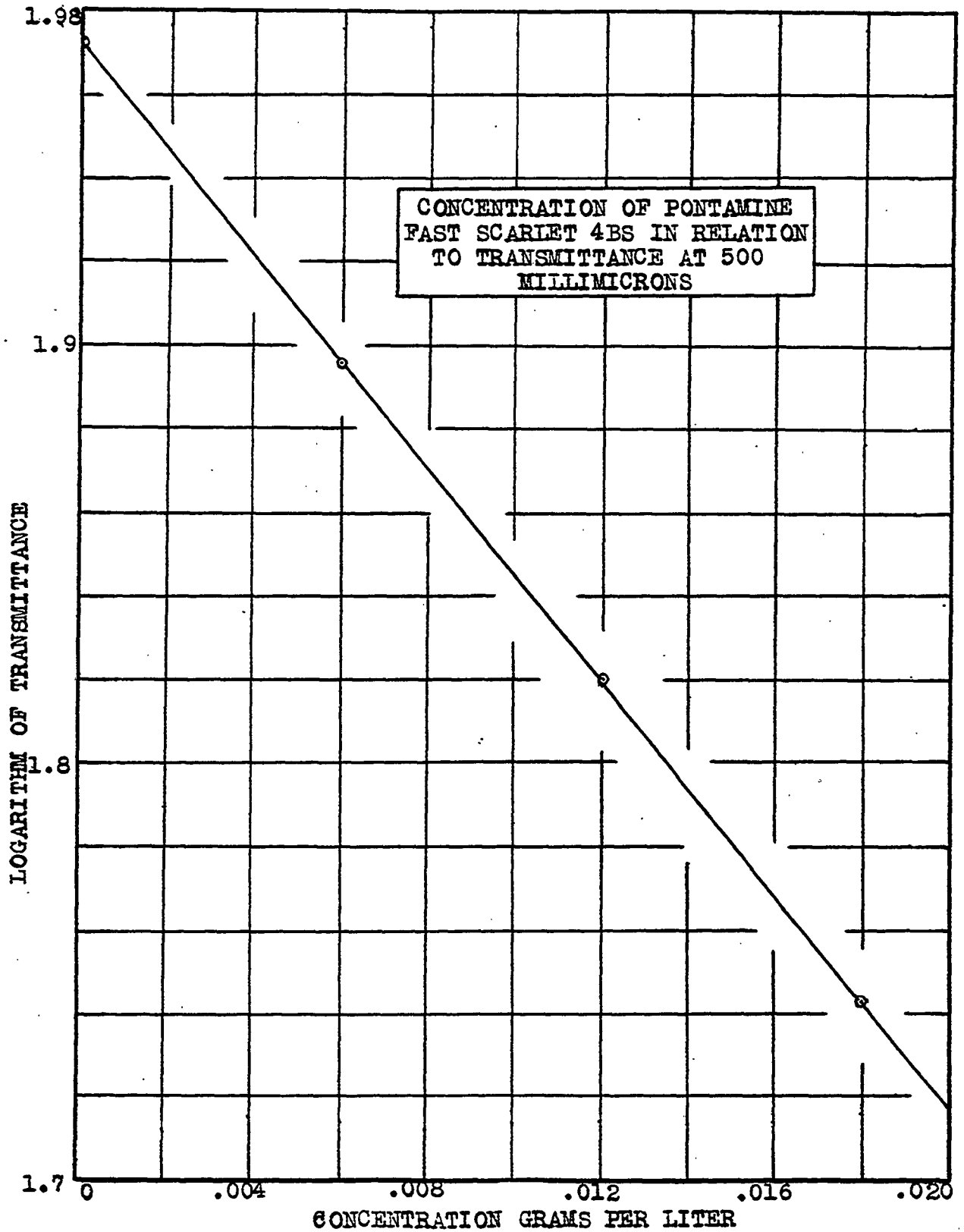


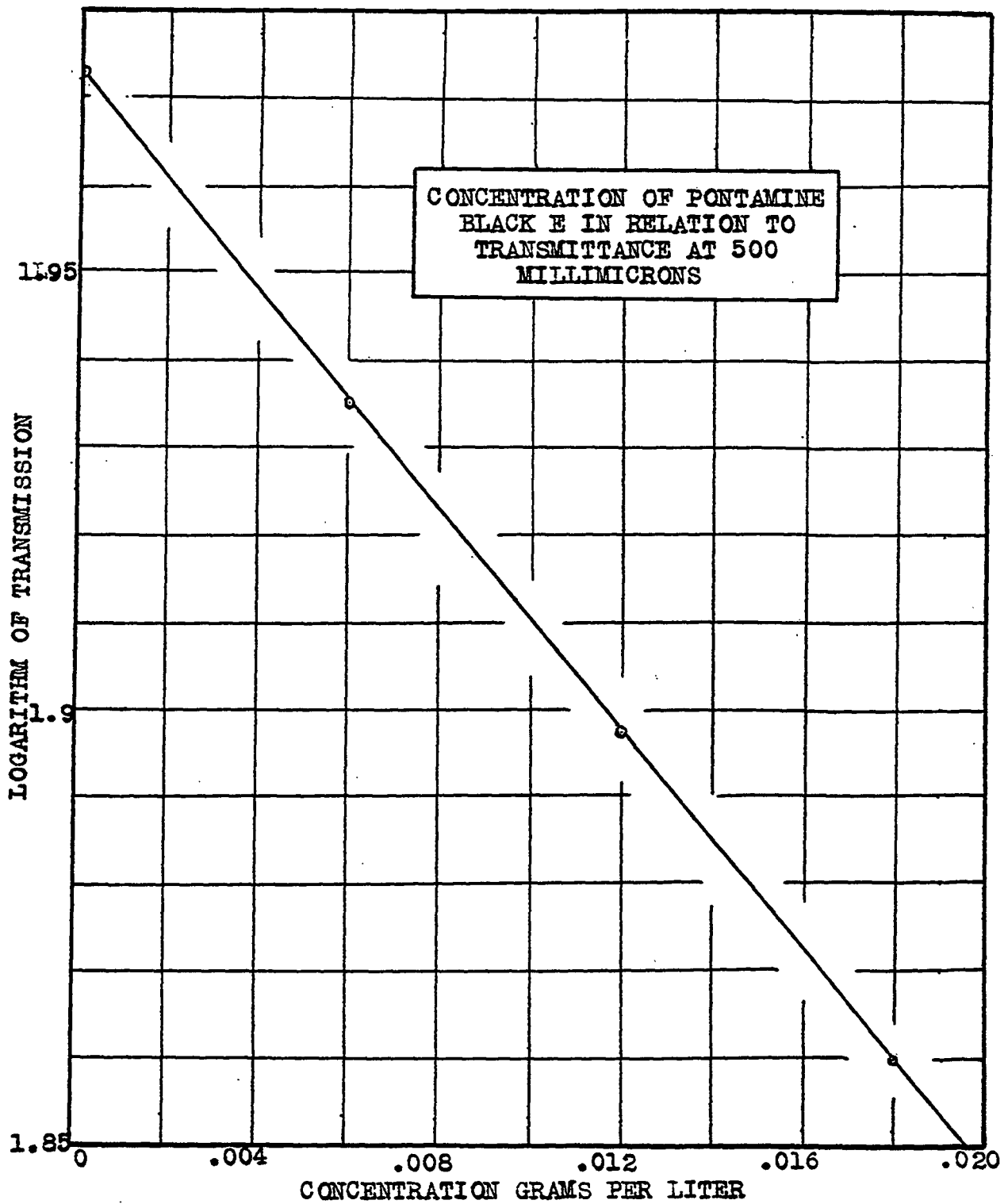


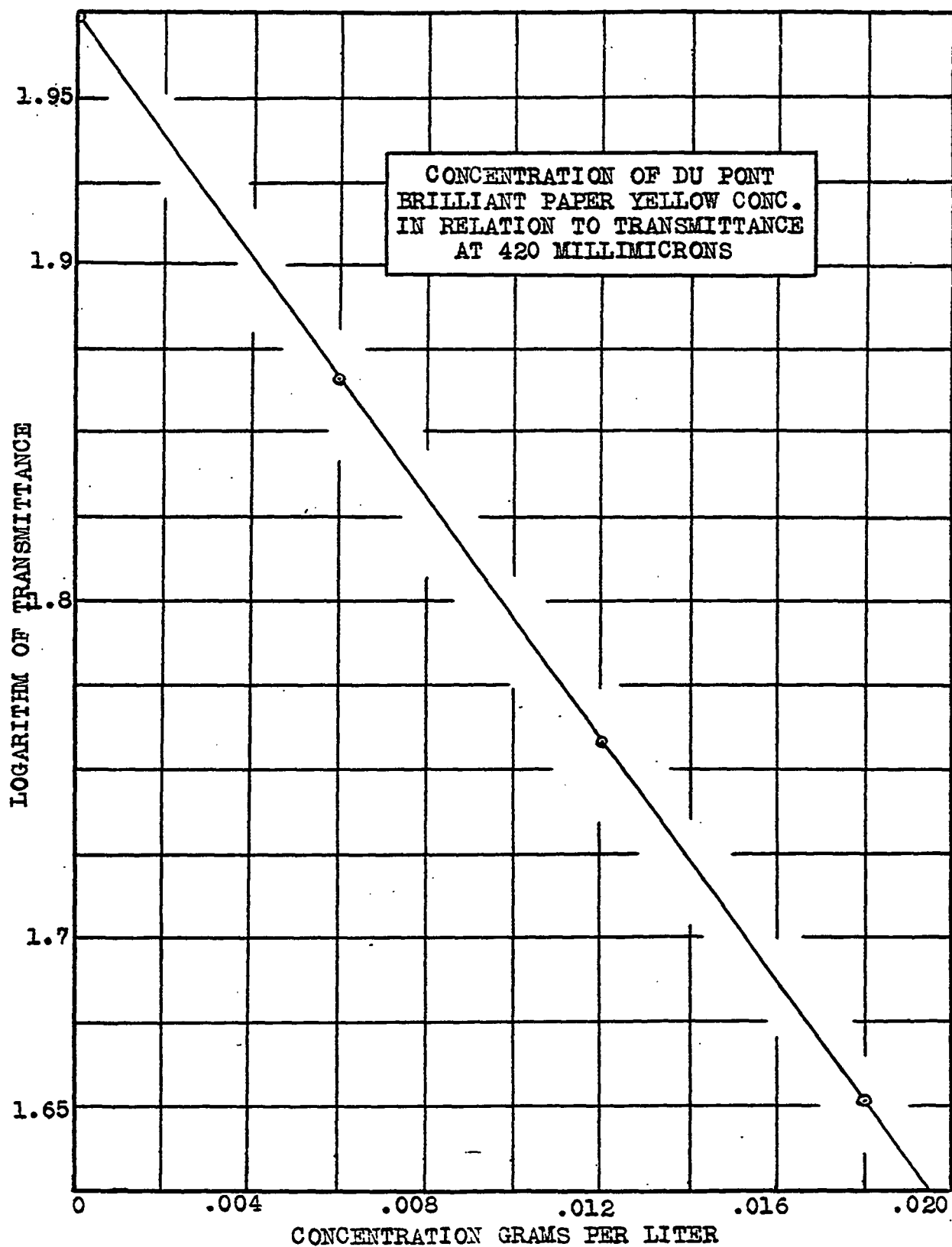




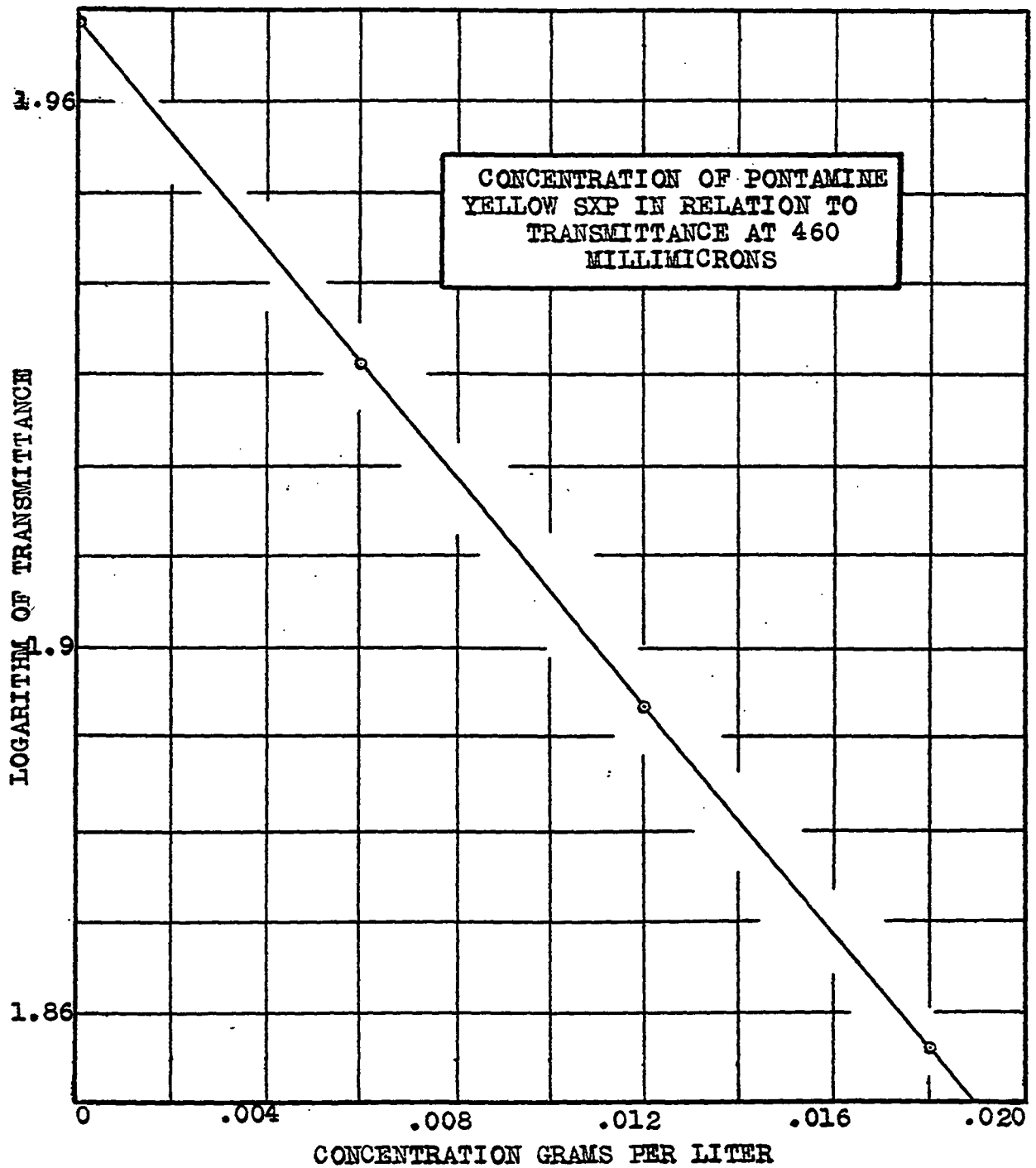


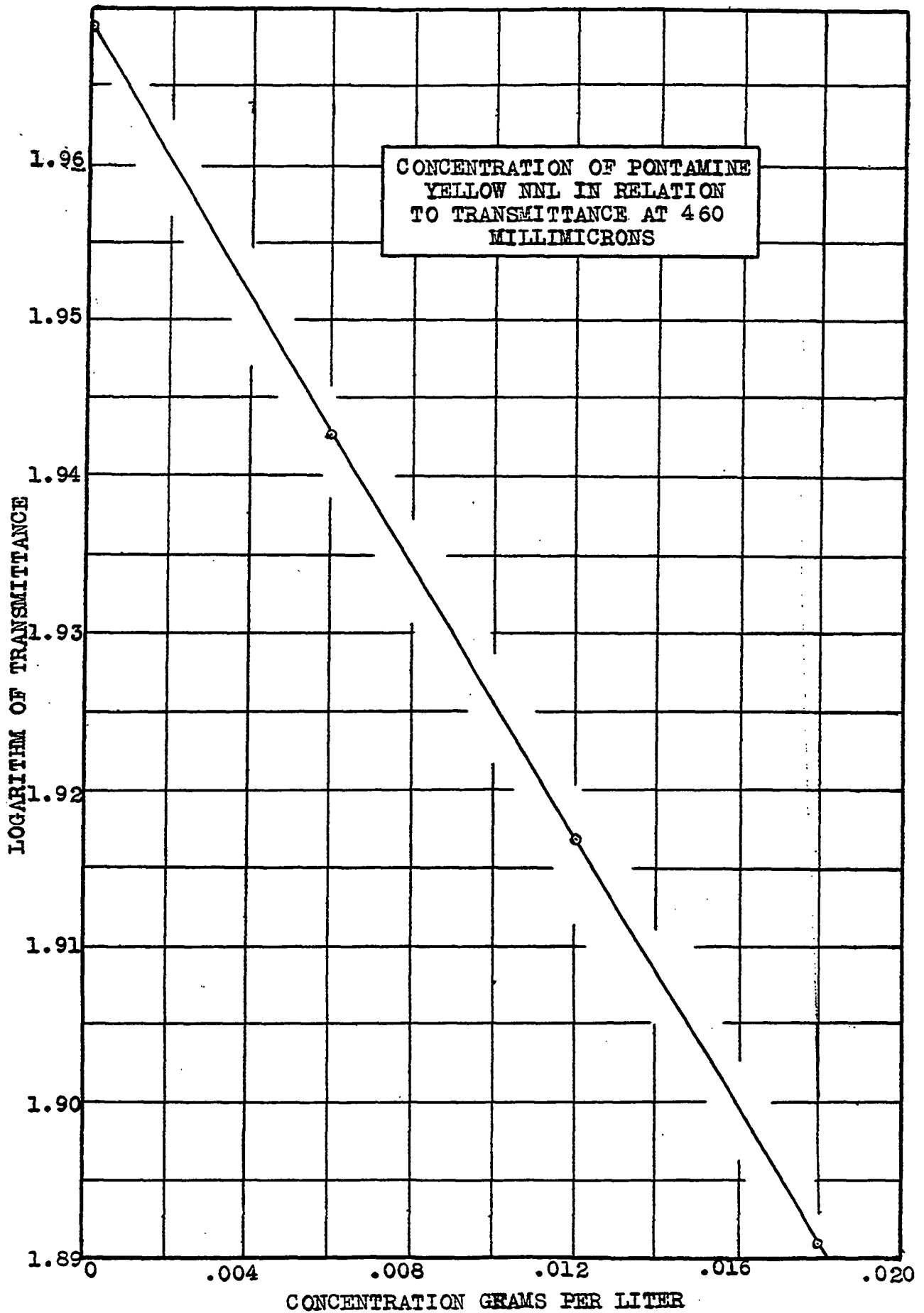








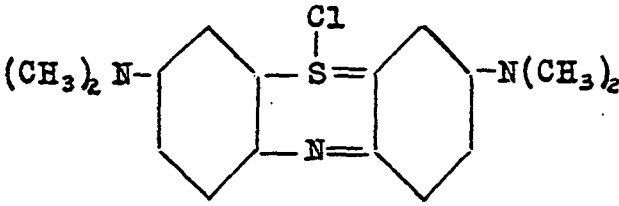
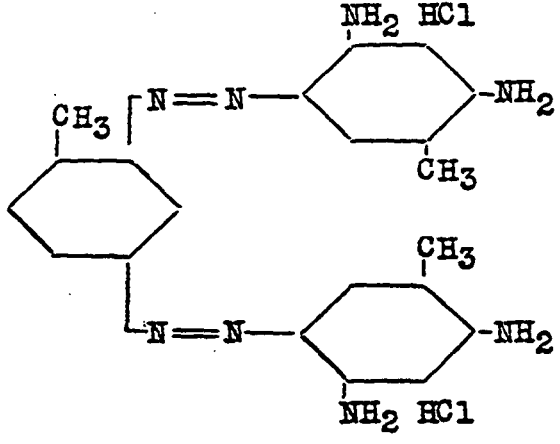
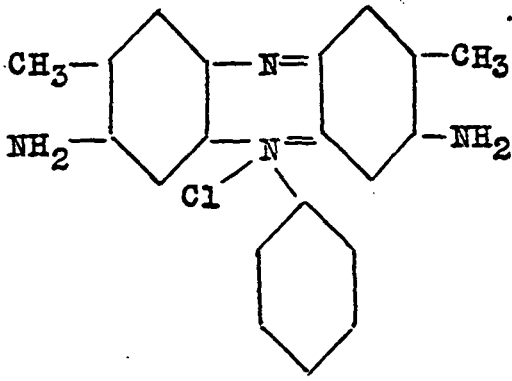
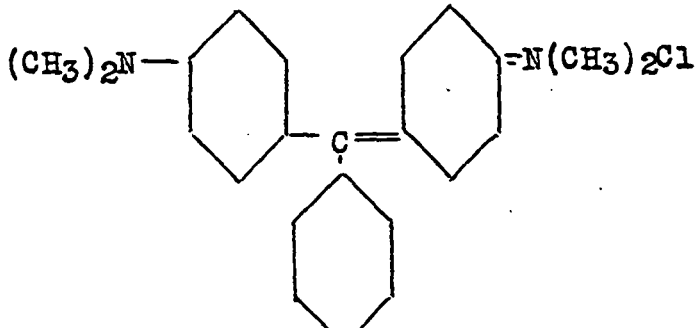


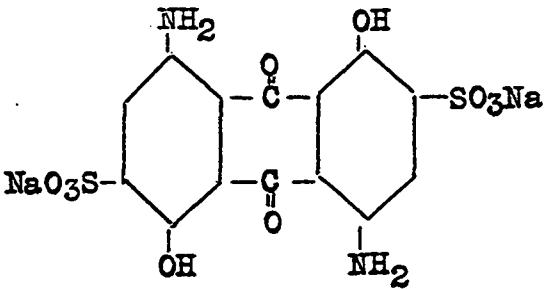
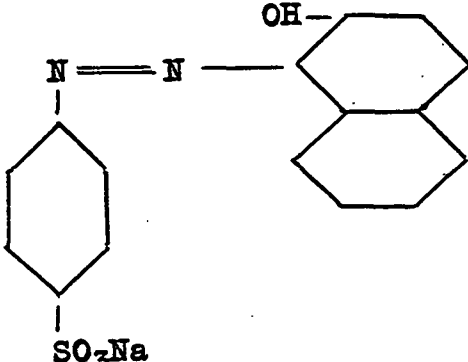
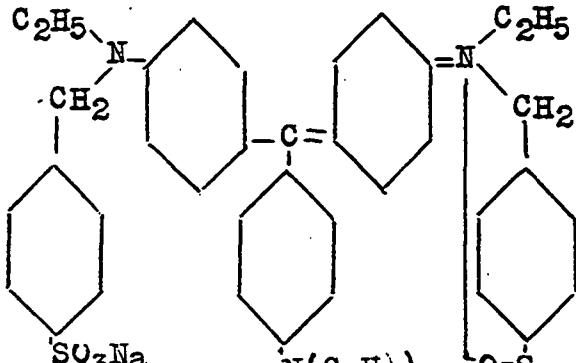


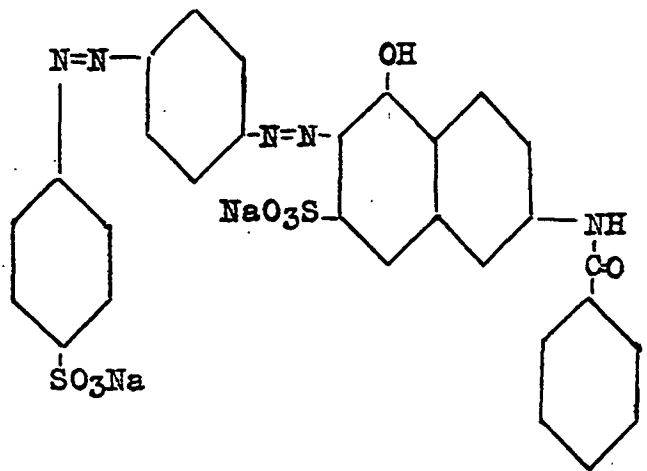
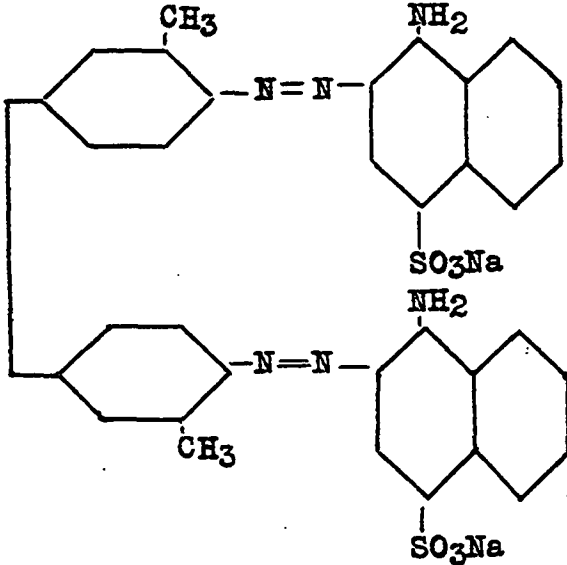
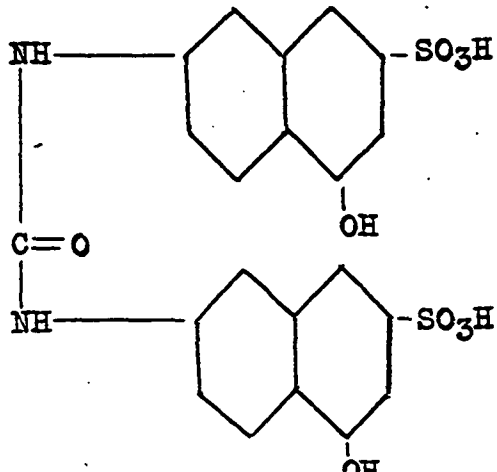
APPENDIX C  
DYE FORMULAS

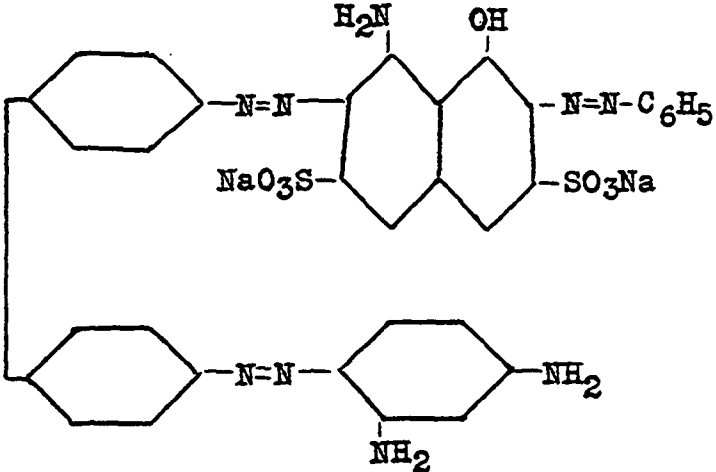
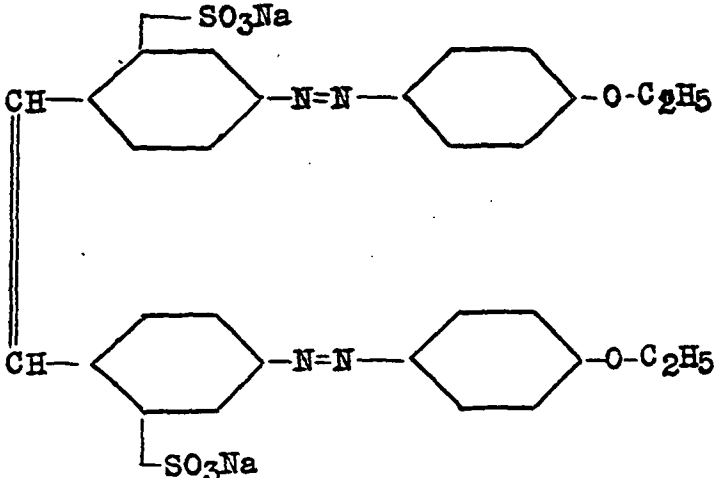
STRUCTURAL FORMULAS OF DYES AS SHOWN IN THE GUSTAV  
COLOR TABLE-7TH ED.

SCHULTZ COLOR NUMBER	DYE AND FORMULA	COLOR INDEX NUMBER
754	<p>DU PONT VICTORIA GREEN SC</p> <chem>CN(C)C1=CC=CC=C1C(=C2C=CC=CC=C2N(C)C)C3CCCCC3</chem>	657
864	<p>DU PONT RHODAMINE B</p> <chem>CCN(CC)C1=CC=CC=C1C(=C2C=CC(=C2C(=O)O)N(CC)CC)OC3CCCCC3N(CC)CC</chem>	749
752	<p>DU PONT AURAMINE CONC.</p> <chem>CN(C)C1=CC=CC=C1C(=C2C=CC=CC=C2N(C)C)C3CCCCC3</chem>	655

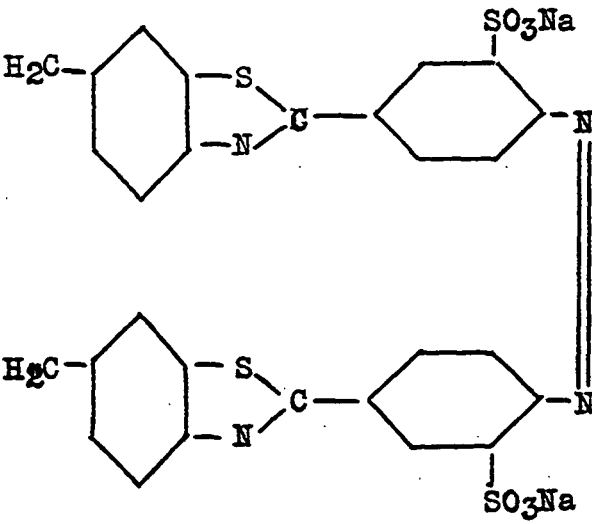
SCHULTZ COLOR NUMBER	DYE AND FORMULA	COLOR INDEX NUMBER
1038	DU PONT METHYLENE BLUE ZX 	922
318	DU PONT BASIC BROWN BR 	332
967	DU PONT SAFRANINE T EXTRA 	841
783	DU PONT METHYL VIOLET 6B 	680

SCHULTZ COLOR NUMBER	DYE AND FORMULA	COLOR INDEX NUMBER
1187	DU PONT ANTHRAQUINONE BLUE B 	1054
986	DU PONT NIGROSINE WSB POWDER STRUCTURE VERY COMPLEX	865
918	DU PONT QUINOLINE YELLOW CONC. $C_{18}H_9NO_8S_2Na$ - SODIUM SALT OF THE DISULFONIC ACID NO STRUCTURE GIVEN	801
189	DU PONT ORANGE II CONCENTRATED 	151
806	PONTACYL VIOLET S4B 	698

SCHULTZ COLOR NUMBER	DYE AND FORMULA	COLOR INDEX NUMBER
566	<p>PONTAMINE FAST RED 8BL</p> 	278
448	<p>DU PONT PURPURINE 4B CONCENTRATED</p> 	448
305	<p>PONTA MINE FAST SCARLET 4BS</p> 	326

SCHULTZ COLOR NUMBER	DYE AND FORMULA	COLOR INDEX NUMBER
671	<p data-bbox="631 386 963 415">PONTAMINE BLACK E</p> 	581
-	<p data-bbox="573 1020 1117 1050">DU PONT BRILLIANT CROCEIN FL</p> <p data-bbox="459 1087 1149 1182">COLOR NOT GIVEN IN SCHULTZ NOR LIST- ED IN THE COLOR INDEX</p>	-
724	<p data-bbox="459 1346 1149 1375">DU PONT BRILLIANT PAPER YELLOW CONC.</p> 	364



S CHULTZ COLOR NUMBER	DYE AND FORMULA	COLOR INDEX NUMBER
-	<p>PONTAMINE YELLOW SXP</p> <p>COLOR NOT GIVEN IN SCHULTZ NOR LISTED IN THE COLOR INDEX</p>	-
935	<p>PONTAMINE FAST YELLOW&gt;NNL</p> 	814

APPENDIX D.  
STANDARD WHITE WATER DATA

TRANSMITTANCE DATA OF DISTILLED WATER AND NATURAL WHITE WATERS AT VARIOUS WAVELENGTHS					
WAVE LENGTHS MILLIMICRONS	DISTILLED WATER	UNBLEACHED SULPHITE	BLEACHED SULPHITE 3% BLEACH	BLEACHED SULPHITE 6% BLEACH	BLEACHED SULPHITE 10% BLEACH
400	95.9	69.3	67.0	70.0	71.9
410	95.0	69.5	67.3	70.3	72.0
420	94.1	70.0	67.7	70.9	72.5
430	93.8	70.2	68.0	71.3	72.9
440	93.2	70.9	68.5	71.8	73.3
450	93.0	71.2	68.9	72.1	74.0
460	93.0	71.8	69.6	72.6	74.8
470	93.1	72.4	70.2	73.3	75.5
480	93.3	73.5	71.0	74.4	76.2
490	93.6	74.4	72.0	75.3	77.0
500	94.0	75.1	72.8	76.2	78.0
510	94.2	75.8	73.8	77.0	78.9
520	94.5	76.7	74.8	77.4	79.9
530	95.1	77.8	75.6	78.6	80.9
540	95.5	78.8	76.7	79.8	82.0
550	95.9	79.7	77.9	80.7	82.9
560	96.1	80.5	78.8	81.9	83.8
570	96.5	81.1	79.5	82.6	84.3
580	97.0	81.9	80.1	83.0	85.0
590	97.2	82.1	80.5	83.5	85.5
600	97.4	83.0	81.1	83.9	86.2
610	97.8	83.2	81.7	84.2	86.8
620	98.3	84.0	82.4	85.0	87.3
630	98.5	85.1	82.9	85.6	87.8
640	98.7	85.3	83.2	86.1	88.4
650	99.0	85.4	83.4	86.9	89.0
660	99.0	85.6	83.6	87.0	89.2
670	99.0	85.6	83.7	87.1	89.3
680	99.0	85.8	83.8	87.2	89.4
690	99.0	85.9	83.9	87.3	89.5
700	99.0	86.0	83.9	87.3	89.5

TRANSMITTANCE DATA OF DISTILLED WATER AND NATURAL WHITE WATERS AT VARIOUS WAVELENGTHS						
WAVELENGTH MILLIMICRONS	BLEACHED SULPHITE 20% BLEACH	BLEACHED SULPHITE 30% BLEACH	RAG	SODA	GROUND- WOOD	UNBLEACHED SULPHITE WITH SIZE AND ALUM
400	73.4	75.0	76.0	76.0	69.0	66.5
410	73.9	75.5	76.4	76.5	69.5	66.8
420	74.4	76.0	77.0	77.2	70.2	67.2
430	74.9	76.6	77.7	78.0	71.1	67.5
440	75.3	77.0	78.3	78.8	71.8	68.0
450	76.0	77.5	79.4	79.4	72.1	68.3
460	76.3	78.1	80.4	79.9	72.5	69.0
470	76.8	78.7	81.0	80.4	73.1	69.7
480	77.5	79.8	82.2	81.0	73.4	70.5
490	78.5	80.5	82.8	81.4	73.8	71.5
500	79.3	81.4	83.5	81.8	74.1	72.3
510	80.1	82.2	84.3	82.0	74.5	73.3
520	81.2	83.1	85.2	82.3	74.9	74.3
530	82.8	84.3	86.0	82.8	75.2	75.0
540	83.6	85.0	86.7	83.1	75.6	76.2
550	84.4	85.8	87.5	83.5	76.0	77.4
560	85.1	86.7	88.5	83.9	76.3	78.3
570	86.0	87.5	89.1	84.0	76.7	79.0
580	86.8	88.4	89.8	84.2	76.9	79.5
590	87.2	89.0	90.1	84.3	77.0	80.0
600	87.8	89.5	90.8	84.4	77.0	80.6
610	88.4	90.4	91.5	84.5	77.1	81.2
620	89.0	91.2	92.1	84.7	77.1	81.9
630	89.6	91.8	92.8	84.9	77.2	82.4
640	90.1	92.2	93.2	85.0	77.4	82.7
650	90.8	92.3	93.9	85.1	77.7	82.8
660	91.2	92.4	94.2	85.2	77.9	83.1
670	91.3	92.5	94.6	85.3	77.9	83.2
680	91.4	92.6	94.7	85.4	78.0	83.3
690	91.5	92.6	95.0	85.5	78.1	83.4
700	91.5	92.6	95.0	85.5	78.1	83.4

TRANSMITTANCE DATA OF DISTILLED WATER AND NATURAL WHITE WATERS AT VARIOUS WAVELENGTHS					
WAVELENGTH MILLIMICRONS	UNBLEACHED KRAFT	KRAFT 5% BLEACH	KRAFT 15% BLEACH	KRAFT 30% BLEACH	KRAFT 40% BLEACH
400	63.0	65.8	69.1	71.3	74.8
410	64.0	66.5	69.6	72.0	75.5
420	64.8	67.5	70.3	73.0	76.0
430	65.5	68.3	71.0	73.6	76.4
440	66.4	69.1	71.8	74.0	76.8
450	67.1	69.8	72.4	74.5	77.0
460	67.6	70.6	73.0	75.0	77.3
470	68.3	71.3	73.5	75.6	78.0
480	69.1	72.2	74.1	76.2	78.5
490	69.5	72.9	74.6	76.8	79.4
500	70.2	73.4	75.1	77.5	80.1
510	71.1	74.1	76.0	78.1	81.0
520	71.9	75.0	76.9	79.0	81.9
530	72.9	75.9	77.9	80.0	82.6
540	74.0	77.0	78.9	80.9	83.1
550	75.1	77.9	79.6	81.5	84.0
560	76.2	78.8	80.2	82.2	84.6
570	77.0	79.3	81.0	82.9	85.3
580	78.0	80.0	81.4	83.5	86.0
590	78.8	80.8	82.2	84.0	87.0
600	79.3	81.4	83.0	84.8	87.3
610	79.8	82.0	83.8	85.9	87.9
620	80.1	82.1	84.1	86.3	88.1
630	80.2	82.3	84.3	86.4	88.5
640	80.4	82.4	84.5	86.5	88.6
650	80.5	82.5	84.6	86.6	88.6
660	80.6	82.6	84.7	86.7	88.7
670	80.7	82.7	84.9	86.8	88.8
680	80.8	82.9	85.0	86.9	88.9
690	80.9	82.9	85.1	87.0	89.0
700	80.9	83.0	85.1	87.0	89.0